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Facile integration of giant exchange bias in $\text{Fe}_{0.5}\text{Ge}_{0.5}\text{Te}_{0.2}\text{MoO}_4$ heterostructures by atomic layer deposition

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2 Fe₅GeTe₂/oxide heterostructures by atomic layer
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22 **Keywords:** exchange bias, van der Waals magnets, Fe₅GeTe₂, atomic layer deposition, surface
23 oxidation

24 **ABSTRACT**

25 Exchange bias arises from the interfacial exchange coupling in ferromagnet-antiferromagnet
26 bilayers and manifests as a horizontal shift of the magnetic hysteresis loop, constituting a critical
27 component underpinning a broad range of magnetoresistive logic and memory devices. The facile
28 implementation of exchange bias in van der Waals (vdW) magnets would be a key step towards
29 practical devices for emerging vdW spintronics. Here, we report an easy approach to establishing
30 strong exchange bias in the vdW magnet Fe_5GeTe_2 by a single-step process – atomic layer
31 deposition (ALD) of oxides. We successfully created exchange bias of 300~1500 Oe in
32 $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$, $\text{Fe}_5\text{GeTe}_2/\text{ZnO}$, and $\text{Fe}_5\text{GeTe}_2/\text{V}_2\text{O}_5$ heterostructures, at 130K. Control
33 experiments showed that increasing the oxidant pulse duration in each ALD cycle or utilizing the
34 stronger oxidant O_3 can enhance the exchange bias strength, revealing the key role of the ALD
35 oxidants. Our systematic work elucidates the essential role of ALD-enabled oxidization of
36 Fe_5GeTe_2 in the formation of exchange bias, and establishes ALD of oxides as a facile,
37 controllable, and generally effective approach to creating giant exchange bias in vdW magnets,
38 representing an integral advance towards practical vdW spintronic devices.

39 I. INTRODUCTION

40 Exchange bias, a magnetic phenomenon emerging at the interface between a ferromagnet (FM)
41 and an antiferromagnet (AFM), originates from the pinning of spins in the FM layer by the adjacent
42 AFM layer [1,2], effectively stabilizing the FM against the environmental stray magnetic fields.
43 This phenomenon has been implemented in a wide range of technological applications including
44 magnetic memories [3], magnetic read heads [4], and sensors [5], and plays the foundational role
45 for advanced spintronics [6–12]. The recently emerged van der Waals (vdW) magnets [13–20]
46 constitute unique platforms that promise the next generation of atomically thin magnetoelectric
47 and magneto-optic devices; however, the large interlayer spacing poses a fundamental challenge
48 regarding how to establish strong exchange bias in vdW systems.

49 Thus far, the exchange bias has been studied in multiple vdW FM/AFM heterostructures
50 including $\text{Fe}_3\text{GeTe}_2/\text{CrI}_3$ [21], $\text{Fe}_3\text{GeTe}_2/\text{FePS}_3$ [22], and $\text{Fe}_3\text{GeTe}_2/\text{MnPX}_3$ ($X = \text{S}$ and
51 Se) [23,24], which all rely on the sophisticated stacking of dissimilar vdW flakes, and have
52 relatively small magnitudes of exchange bias (e.g., 200 Oe at 10 K in $\text{Fe}_3\text{GeTe}_2/\text{CrI}_3$). A facile
53 approach that can generate strong exchange bias, while being compatible with the Si-based
54 fabrication technology, is highly desirable. Recently, the exchange bias was observed in vdW FM
55 Fe_3GeTe_2 after being annealed in the air [25], suggesting the possibility of using surface oxidation
56 to convert FM Fe_3GeTe_2 layers into AFM oxidized Fe_3GeTe_2 . Compared with thermal annealing
57 in the air with complex gases, atomic layer deposition (ALD) is a cleaner and more rigorous
58 technique for surface reactions due to its use of pure gas precursors (e.g., H_2O or O_3) and the
59 precise time control of each oxidant gas pulse. Moreover, ALD represents an industry-favored
60 technique for growing ultrathin conformal dielectrics on electronic materials [26]. If the exchange
61 bias can be created in vdW magnets by the ALD process, it would reveal not only useful knowledge

62 for applying vdW magnets to spintronic and electronic devices but also practical guidance on
63 device fabrication. In addition, given the exchange bias in air-annealed Fe_3GeTe_2 still drops below
64 250 Oe at 100 K [25], new vdW magnets beyond Fe_3GeTe_2 should be explored to achieve strong
65 exchange bias.

66 Recently, an itinerant vdW ferromagnet Fe_5GeTe_2 has garnered significant attention due to its
67 high magnetic ordering temperature compared to Fe_3GeTe_2 [20,27], meanwhile its exchange bias
68 remains to be explored. Here, via growing non-magnetic oxides by ALD, we demonstrate giant
69 exchange bias in the vdW ferromagnet Fe_5GeTe_2 . Through reflectance magnetic circular dichroism
70 (RMCD) studies, we found that the exchange bias in the $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ structure reaches 460 Oe
71 at 130 K after field cooling, and the effect persists at 160 K. The strong exchange bias was
72 reproducible by ALD of ZnO and V_2O_5 on Fe_5GeTe_2 , suggesting that our ALD integration of
73 exchange bias is generically effective. The exchange bias is potentially formed at the interface
74 between oxidized Fe_5GeTe_2 (AFM) and unoxidized Fe_5GeTe_2 (FM) layers. Remarkably, by
75 increasing the reaction time of oxidant H_2O with Fe_5GeTe_2 or utilizing O_3 as a stronger oxidant
76 during ALD, we enhanced the exchange bias effectively (up to ~ 1500 Oe). These observations
77 strongly suggest the critical role of initial oxidization in forming and enhancing exchange bias.
78 Our results demonstrated a facile, controllable, and generally effective method to create giant
79 exchange bias for vdW spintronics devices.

80 **II. EXPERIMENTAL DETAILS**

81 Bulk Fe_5GeTe_2 single crystals were synthesized by a chemical vapor transport method; the
82 crystals of average composition near $\text{Fe}_{4.7(2)}\text{GeTe}_2$ were quenched by controlling the growth
83 temperature and have a bulk Curie temperature (T_C) near 270 - 310 K [28]. Fe_5GeTe_2 flakes were

84 mechanically exfoliated by scotch tape and transferred onto 260-nm-thick-SiO₂/Si chips. Atomic
85 force microscopy confirmed the thickness of Fe₅GeTe₂ flakes to be ~30 to 80 nm. Temperature-
86 dependent RMCD on an exfoliated pristine flake indicates a T_C of ~243 K (Fig. S1 in the
87 Supplemental Material [29]). Consistent with the literature [20], T_C of exfoliated flakes can be
88 lowered in a range of 30-40 K compared to T_C of bulk Fe₅GeTe₂ (270 – 310 K). To avoid unwanted
89 oxidation in the air, we stored the samples in a glovebox filled with N₂ gas, with oxygen and
90 moisture levels < 0.1 ppm.

91 The Fe₅GeTe₂/oxide heterostructures were prepared by ALD in BENEQ TFS 500 reactor with
92 a base pressure of 2 mbar. For ALD of Al₂O₃, Al(CH₃)₃ and H₂O were used as the aluminum
93 precursor and the oxidizing agent, respectively. The pulse of Al(CH₃)₃ or H₂O was controlled by
94 regular ALD valves, which introduce N₂ to the precursor supply vessel, allowing the vapor
95 pressure of Al(CH₃)₃ or H₂O to be established in the N₂ headspace, followed by the injection of
96 the headspace gas to the downstream ALD valves. The 10-nm Al₂O₃ was grown at 150 °C with 87
97 cycles, each consisting of 0.2 s Al(CH₃)₃ pulse, 0.5 s N₂ purge, 0.2 s H₂O pulse, and 0.5 s N₂ purge.
98 For ALD of 10-nm-thick ZnO, all settings were kept the same as the Al₂O₃ growth including the
99 H₂O pulse length of 200 ms, except that the ALD precursor became Zn(C₂O₅)₂. Last, for ALD of
100 V₂O₅, VO(OC₃H₇)₃ was used as the vanadium precursor, whereas Ozone was the oxidant instead
101 of water. A MKS O3MEGA ozone delivery subsystem was employed to provide a stable 17.3 wt%
102 O₃ from a pure O₂ source. V₂O₅ of 10 nm was grown at 170 °C with 222 cycles, each having 0.5
103 s VO(OC₃H₇)₃ pulse, 1 s N₂ purge, 2 s O₃ pulse, and 5 s N₂ purge as reported [30].

104 Furthermore, to study the impact of ALD oxidants on the exchange bias effect, we prepared
105 Fe₅GeTe₂/Al₂O₃ samples with different pulse lengths of H₂O per ALD cycle (i.e., 200 ms, 800 ms,
106 and 1600 ms) while keeping the rest of the parameters the same. To avoid the incomplete removal

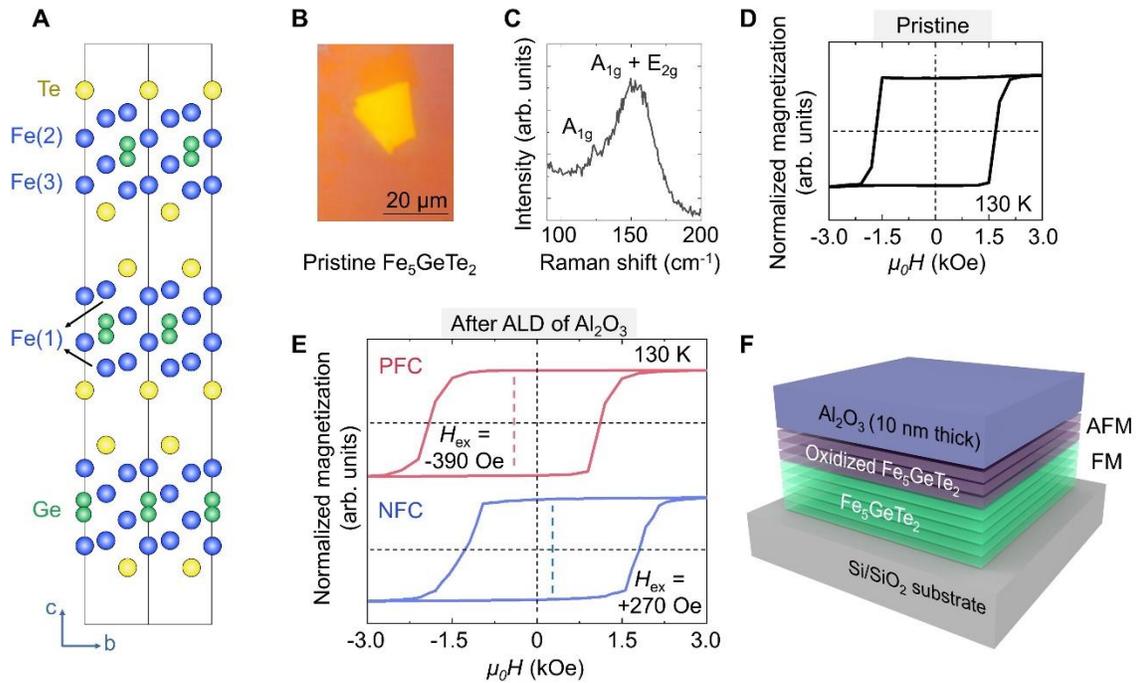
107 of H₂O, we set the N₂ purge to be 4 s instead of 0.5 s after the H₂O pulse for each ALD cycle. And
108 we also compared the exchange bias in two Fe₅GeTe₂/Al₂O₃ heterostructures where Al₂O₃ was
109 grown by different oxidants (i.e., H₂O vs. O₃). For a fair comparison, the 10-nm Al₂O₃ was grown
110 at 150 °C with 87 cycles, each consisting of 0.2 s Al(CH₃)₃ pulse, 0.5 s N₂ purge, 0.2 s H₂O or O₃
111 pulse, and 0.5 s N₂ purge.

112 The sample's out-of-plane magnetization was probed by RMCD under the out-of-plane
113 magnetic field up to 300 mT in a Montana cryostat ($< 10^{-4}$ Torr). A HeNe laser (633 nm, optical
114 power of 7 μ W) was focused onto the samples via a 50 \times objective of numerical aperture 0.5 to
115 achieve a sub-micrometer spot size. A photoelastic modulator at 50 kHz was used to modulate the
116 helicity of the optical excitation between left and right, and a photodiode detected the reflected
117 light from the sample. The RMCD was determined by the ratio between an AC signal at 50 kHz
118 and a low-frequency AC signal at 237 Hz of the reflected light intensity, which was measured by
119 two different lock-in amplifiers.

120 **III. RESULTS AND DISCUSSION**

121 Fig. 1(a) shows a schematic of the Fe₅GeTe₂ crystal structure with three-layer periodicity along
122 the c axis (out-of-plane direction), where nonequivalent Fe sites are labeled as Fe(1), Fe(2), and
123 Fe(3). Two locations of Fe(1) represent the split sites, and their occupation (either up or down)
124 leads to the adjustment of Ge atom (down or up, respectively). Thus, the crystal structure of
125 Fe₅GeTe₂ was found to be more complex than Fe₃GeTe₂ due to those split sites and vacancy
126 disorder [20]. In this work, Fe₅GeTe₂ flakes (~30 - 80 nm thick) with lateral areas above $8 \times 8 \mu\text{m}^2$
127 were mechanically exfoliated from bulk crystals onto the Si substrates with 260-nm thick SiO₂,
128 and subsequently examined by optical microscopy (Fig. 1(b)). Raman spectroscopy of Fe₅GeTe₂

129 (Fig. 1(c)) shows the A_{1g} peak (123 cm^{-1}) and the combined $A_{1g} + E_{2g}$ peak (153 cm^{-1}), similar to
 130 the Fe_3GeTe_2 feature [31]. For a pristine Fe_5GeTe_2 flake, its out-of-plane magnetization was
 131 probed by RMCD at 130 K after a zero-field cooling (ZFC). As expected, the magnetic hysteresis
 132 loop remains symmetric to the zero magnetic field (Fig. 1(d)), indicating the absence of exchange
 133 bias in pristine Fe_5GeTe_2 .



134

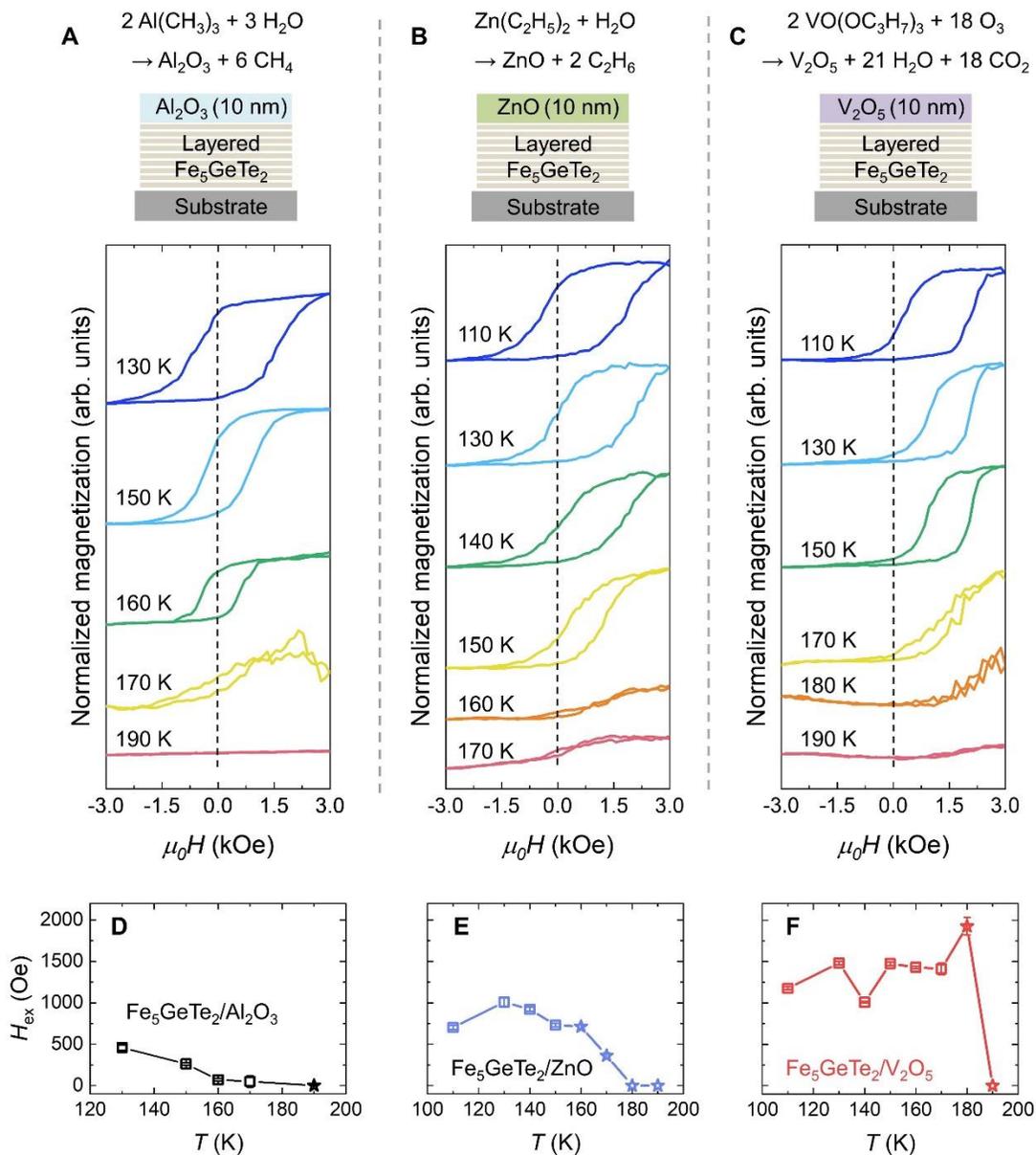
135 **FIG. 1.** Characterizations of Fe_5GeTe_2 and $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$. (a) Side view of Fe_5GeTe_2 crystal structure
 136 where Fe(1) and Ge are split sites. The optical image (b) and Raman spectrum (c) of a pristine Fe_5GeTe_2
 137 flake. (d) RMCD measurement of the out-of-plane magnetization of the pristine Fe_5GeTe_2 at 130 K after
 138 ZFC. The vertical and horizontal black dashed lines serve as the eye guide for zero magnetic field ($x = 0$)
 139 and zero magnetization ($y = 0$), respectively. No exchange bias was found in pristine Fe_5GeTe_2 . (e) RMCD
 140 measurements of a representative $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructure at 130 K after PFC and NFC,
 141 respectively. The red (blue) dashed line highlights the shifted center line of the magnetic hysteresis loop
 142 after PFC (NFC). The magnitude of the cooling field is 3000 Oe in this study. After ALD of 10-nm-thick
 143 Al_2O_3 on Fe_5GeTe_2 , negative (positive) exchange bias emerges after PFC (NFC) in the heterostructure. (f)
 144 Schematics of the $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructure consisting of oxidized Fe_5GeTe_2 . The Fe_5GeTe_2 layers

145 at the $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ interface are likely oxidized upon ALD and serve as the antiferromagnetic pinning
146 layers to induce the exchange bias effect.

147 Next, a 10-nm thick Al_2O_3 was grown by ALD on top of Fe_5GeTe_2 , involving surface chemical
148 reactions between the metal precursor $\text{Al}(\text{CH}_3)_3$ and the oxidant H_2O [32,33]. The magnetic
149 hysteresis loops of a $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructure were then measured by RMCD at 130 K after
150 the positive- and negative-field cooling (PFC and NFC), respectively. In stark contrast to the
151 pristine Fe_5GeTe_2 that has a symmetric magnetic hysteresis loop (Fig. 1(d)), significant loop shifts
152 of the $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructure after PFC and NFC clearly confirm the emergence of
153 exchange bias (Fig. 1(e)), with the exchange bias field (H_{ex}) being negative (positive) after PFC
154 (NFC) as expected [2]. We observed the exchange bias at 130 K in multiple $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$
155 heterostructures after PFC and NFC (Fig. S2 in the Supplemental Material [29]), suggesting that
156 the phenomena are well reproducible. As further evidence, exchange bias also appears in
157 $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructures after the first ZFC, with random signs and amplitudes (Fig. S3
158 in the Supplemental Material [29]). Because Al_2O_3 itself is non-magnetic, the observed exchange
159 bias indicates the AFM nature of the oxidized Fe_5GeTe_2 layers formed upon ALD of Al_2O_3 , as
160 illustrated by Fig. 1(f). This agrees with the recent report of exchange bias in the vdW magnet
161 Fe_3GeTe_2 after being annealed in the air at 100 °C for 30 min [25].

162 We continued to systematically examine the temperature dependence of the magnetic hysteresis
163 loops and the exchange bias in a representative $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructure. As shown in Fig.
164 2(a), raising the temperature decreases the coercivity (H_c) by introducing stronger thermal
165 fluctuations, and the hysteresis loop finally disappears at 190 K. The decreasing trend is also
166 observed for H_{ex} as the temperature increases (Fig. 2(d)). Notably, H_{ex} reaches 460 Oe at 130 K
167 after NFC and persists to 160 K with a magnitude of 50 Oe, indicating a blocking temperature (T_B)

168 around 160 - 170 K at which the exchange bias disappears. In short, these results suggest that ALD
 169 of Al₂O₃ on top of Fe₅GeTe₂ leads to a strong exchange bias with the relatively high T_B.



170
 171 **FIG. 2.** Temperature dependence of the exchange bias in three types of Fe₅GeTe₂/oxide heterostructures.
 172 (a - c) Temperature dependence of magnetic hysteresis loops measured by RMCD in three types of
 173 Fe₅GeTe₂/oxides heterostructures: Fe₅GeTe₂/Al₂O₃ (a), Fe₅GeTe₂/ZnO (b), and Fe₅GeTe₂/V₂O₅ (c).
 174 Positive shifts of hysteresis loops emerge after NFC for all Fe₅GeTe₂/oxides systems. The black dashed
 175 lines serve as the eye guide for zero magnetic field ($x = 0$). The corresponding ALD reactions for each type
 176 of Fe₅GeTe₂/oxide are illustrated in the top schematics. (d - f) The exchange bias H_{ex} as a function of

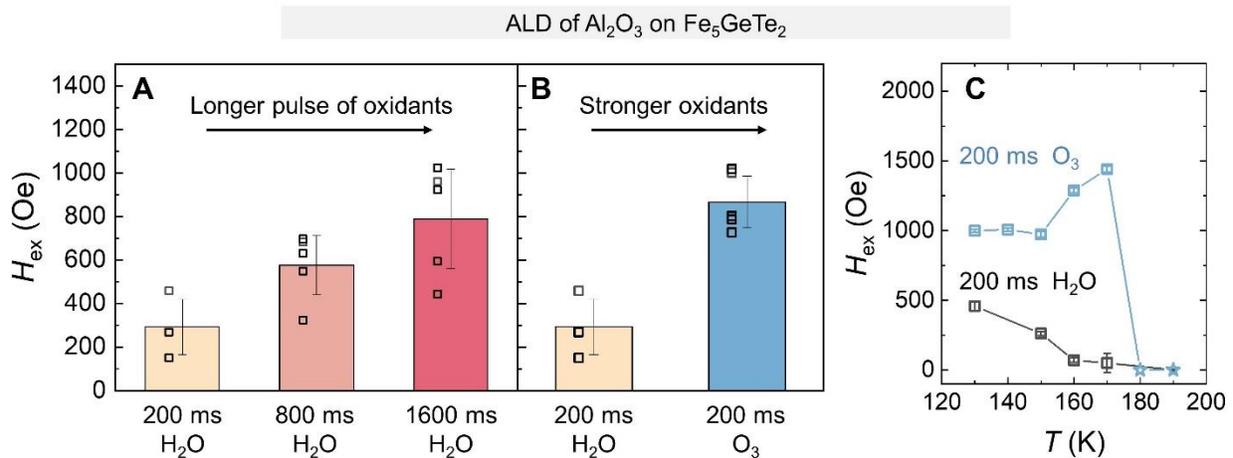
177 temperature for the three types of Fe₅GeTe₂/oxides. To indicate the possible uncertainty in extracted values,
178 H_{ex} data points are marked by stars when the coercivity drops to zero. Error bars represent the standard
179 deviation above the mean of the extracted data.

180 To prove that our method to generate exchange bias is not oxide-specific, we extended to ALD
181 of ZnO [34] and V₂O₅ [30], as illustrated by the top schematics in Fig. 2. Indeed, magnetic
182 hysteresis loops with large positive shifts were observed after NFC for both Fe₅GeTe₂/ZnO and
183 Fe₅GeTe₂/V₂O₅ heterostructures. For Fe₅GeTe₂/ZnO, the hysteresis loop shrinks with the
184 increasing temperature and disappears at ~160 K (Fig. 2(b)). H_{ex} reaches 1007 Oe at 130 K,
185 decreases gradually as the temperature increases, and remains at 710 Oe when H_c drops to zero at
186 160 K (Fig. 2(e)). For Fe₅GeTe₂/V₂O₅, H_{ex} persists at the level of ~1500 Oe to at least 170 K (Figs.
187 2(c) and 2(f)), showing an even stronger exchange bias effect compared to Fe₅GeTe₂/Al₂O₃ and
188 Fe₅GeTe₂/ZnO in this work. Remarkably, such a large H_{ex} in Fe₅GeTe₂/V₂O₅ is 600~1000 Oe
189 greater than the reported H_{ex} in other vdW magnets [21–25,35] and is achievable at high
190 temperatures (e.g., 150 to 170 K). The general existence of exchange bias in Fe₅GeTe₂/oxides
191 fabricated by ALD further suggests the scenario that the observed exchange bias is due to the
192 oxidation of top Fe₅GeTe₂ layers during the ALD process.

193 In addition, there is no noticeable dependence of the exchange bias strength on the Fe₅GeTe₂
194 layer thickness in Fe₅GeTe₂/oxides (Fig. S4 in the Supplemental Material [29]). This finding is in
195 contrast to the observations in conventional FM/AFM bilayer systems where H_{ex} magnitude
196 depends inversely on the FM thickness [2], but agrees with the recent observations on the exchange
197 bias in vdW magnet Fe₃GeTe₂ [21,25], showing a unique magnetic behavior due to the weak
198 interlayer exchange coupling inherent to layered vdW magnets. Therefore, the thickness variance
199 among Fe₅GeTe₂ samples should not interfere with the major finding that the exchange bias in
200 Fe₅GeTe₂/oxides is mainly due to the ALD induced oxidation in Fe₅GeTe₂. Overall, these

201 observations trigger an interesting question regarding whether we can control the exchange bias
 202 strength in Fe_5GeTe_2 by tuning the ALD conditions.

203 It is worth highlighting that the ALD of V_2O_5 adopted O_3 (Figs. 2(c) and 2(f)), a stronger oxidant
 204 than H_2O , and the resultant exchange bias in $\text{Fe}_5\text{GeTe}_2/\text{V}_2\text{O}_5$ is clearly stronger than that in
 205 $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ (Figs. 2(a) and 2(d)) and $\text{Fe}_5\text{GeTe}_2/\text{ZnO}$ (Figs. 2(b) and 2(e)). This again indicates
 206 the critical role of ALD-enabled oxidization of Fe_5GeTe_2 in the final formation of exchange bias.
 207 We conducted two sets of control experiments to unravel the underlying mechanism further. First,
 208 using $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ as the model system, we varied the ALD pulse duration of oxidant H_2O and
 209 studied its impact on the exchange bias strength (see Experimental Details). We increased the H_2O
 210 pulse duration per cycle from 200 to 1600 ms while keeping the rest of the parameters the same
 211 (e.g., temperature, pulse duration of the precursor $\text{Al}(\text{CH}_3)_3$, and cycle numbers). In Fig. 3(a), we
 212 summarized all the H_{ex} measured at 130 K after NFC on randomly selected multiple samples for
 213 each H_2O pulse duration. By increasing the H_2O pulse duration from 200 to 1600 ms, H_{ex} increases
 214 by about 170% from 290 ± 130 Oe to 790 ± 230 Oe (Fig. 3(a)), demonstrating an effective tuning
 215 of the resultant exchange bias strength by adjusting the ALD oxidant's pulse duration.



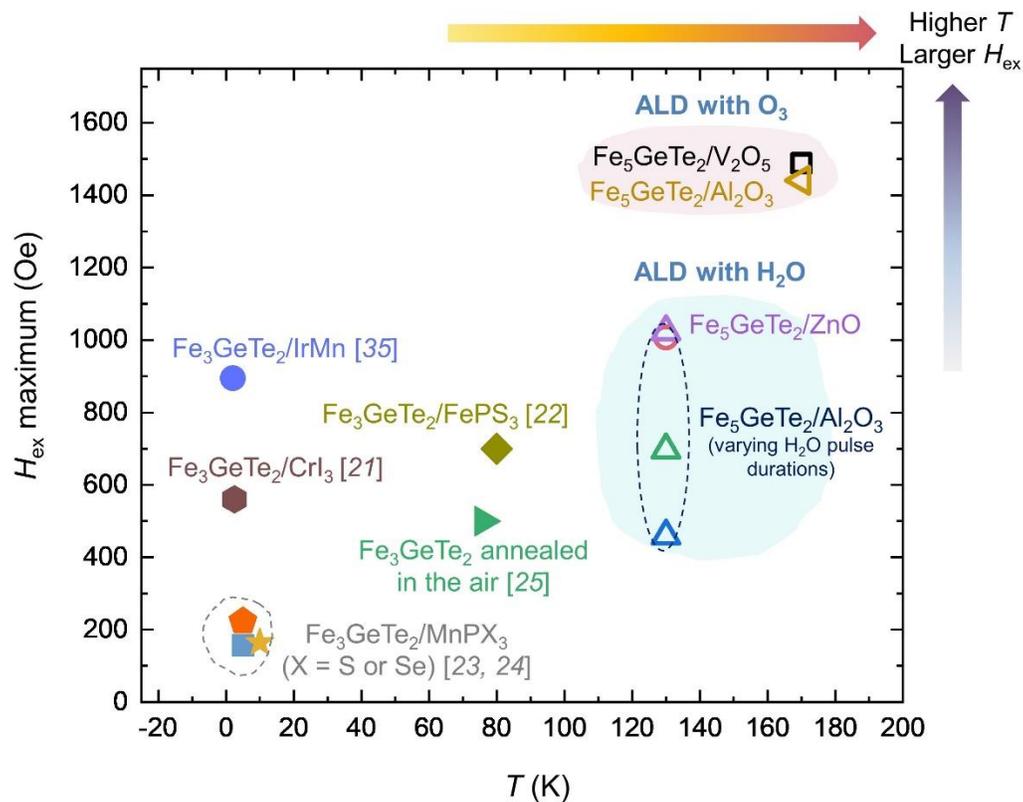
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217 **FIG. 3.** Tuning the exchange bias strength in $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructures by adjusting the oxidants
 218 during ALD. (a - b) Summary of H_{ex} in $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructures where oxides were prepared using

219 the H₂O pulse durations of 200, 800, and 1600 ms per ALD cycle (a) or using different ALD oxidants (i.e.,
220 H₂O vs. O₃) (b). For each set of Fe₅GeTe₂/Al₂O₃ heterostructures, three to five randomly selected sample
221 flakes were measured at 130 K after NFC to show the variance. The average H_{ex} for each set is shown by
222 the histogram, with the original data points shown on the left and the error bar representing the standard
223 deviation above the mean on top of each histogram. The resultant exchange bias is effectively enhanced by
224 increasing the pulse duration of the ALD oxidant (i.e., longer H₂O pulse) or utilizing O₃ as a stronger
225 oxidant during ALD of Al₂O₃. (c) The exchange bias H_{ex} after NFC as a function of temperature for
226 Fe₅GeTe₂/Al₂O₃ prepared by H₂O and O₃. To indicate the possible uncertainty in extracted values, H_{ex} data
227 points are marked by stars when the coercivity drops to zero. Error bars represent one standard deviation
228 above the mean of the extracted data.

229 As a second set of control experiments, we studied the exchange bias strengths in
230 Fe₅GeTe₂/Al₂O₃ samples prepared by two different ALD oxidants (i.e., H₂O versus O₃) while
231 maintaining all other ALD parameters the same. Under the same oxidant pulse duration of 200 ms,
232 the statistic study summarized in Fig. 3(b) shows that H_{ex} increases from an average of 290 ± 130
233 Oe to 870 ± 120 Oe by switching from H₂O to O₃, with a percentage increase of 200%. Moreover,
234 we measured the temperature dependence of H_{ex} in Fe₅GeTe₂/Al₂O₃ prepared by O₃ (hysteresis
235 loops in Fig. S5 in the Supplemental Material [29]) and compared the H_{ex} results with
236 Fe₅GeTe₂/Al₂O₃ prepared by H₂O in Fig 3(c). The maximum H_{ex} reaches 1450 Oe at 170 K after
237 NFC for Fe₅GeTe₂/Al₂O₃ prepared by O₃, which is comparable with that achieved at 170 K in O₃-
238 prepared Fe₅GeTe₂/V₂O₅ (see Fig. 2(f)) but about 3 times larger than the maximum exchange bias
239 obtained at 130 K in Fe₅GeTe₂/Al₂O₃ prepared by H₂O. This result suggests that ALD oxidants
240 (i.e., O₃ and H₂O) play a major role in determining the exchange bias strength. It has been well-
241 established that O₃, as a stronger oxidant, can react more with the basal plane of many vdW
242 materials than H₂O during the nucleation stage of ALD [26,36–38]. Thus, similar effects by O₃ are
243 expected for ALD of oxides on vdW magnets in this work, which could lead to a stronger oxidation
244 to enhance the exchange bias effect.

245 The giant exchange bias in $\text{Fe}_5\text{GeTe}_2/\text{oxides}$ discovered in this work has been unprecedented for
 246 vdW magnets. In Fig. 4, we summarized the maximum values of H_{ex} measured in vdW
 247 magnets/heterostructures and their measurement temperatures. For many all-vdW
 248 heterostructures, H_{ex} exceeding 200 Oe can only be achieved at low temperatures (e.g., 10
 249 K) [21,23,24,35]. In contrast, H_{ex} with the maximum magnitude in the range from 400 to 1000 Oe
 250 at 130 K can be easily induced in Fe_5GeTe_2 by ALD of Al_2O_3 or ZnO using the oxidant H_2O with
 251 varying H_2O pulse durations. By means of a stronger oxidant O_3 , the maximum values of H_{ex} reach
 252 ~ 1500 Oe at 170 K in $\text{Fe}_5\text{GeTe}_2/\text{V}_2\text{O}_5$ and $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ heterostructures, which are about 1000
 253 Oe larger than the reported H_{ex} in most Fe_3GeTe_2 -based heterostructures as summarized in Fig. 4.



254
 255 **FIG. 4.** The maximum exchange bias versus the measurement temperature for different vdW magnets and
 256 heterostructures (data adapted from refs^{21-25,35}). H_{ex} data of this work when H_c becomes zero is not included.
 257 The exchange bias in our $\text{Fe}_5\text{GeTe}_2/\text{oxides}$ is strong and retained at relatively high temperatures.

258 **IV. CONCLUSIONS**

259 In summary, we utilized ALD of oxides to achieve a facile integration of giant exchange bias in
260 vdW magnet Fe_5GeTe_2 . Through ALD of Al_2O_3 , we induced a sizable exchange bias of 460 Oe at
261 130 K. The exchange bias effect is reproducible by ALD of two other oxides (ZnO and V_2O_5),
262 confirming the general effectiveness of our approach. Through control experiments, we
263 demonstrated that the exchange bias strength can be enhanced by increasing the oxidant pulse
264 duration in each ALD cycle or utilizing the stronger oxidant O_3 . In particular, the maximum
265 exchange bias reaches ~ 1500 Oe at 170 K for $\text{Fe}_5\text{GeTe}_2/\text{V}_2\text{O}_5$ and $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ that were both
266 prepared by O_3 , about three times larger than the maximum exchange bias achieved at 130 K for
267 $\text{Fe}_5\text{GeTe}_2/\text{Al}_2\text{O}_3$ by H_2O , highlighting the critical role of the ALD oxidants in the formation of the
268 exchange bias. Our results demonstrate a method that is simple, generally effective, and integrable
269 to the existing Si-based fabrication technology, for creating giant exchange bias persisting up to
270 relatively high temperatures in Fe_5GeTe_2 . This ALD-based method allows a one-step integration
271 of giant exchange bias during dielectric growth for simple device assembly, beneficial for
272 ultracompact vdW spintronics devices. We envision that the future optimization of this approach
273 could benefit from the advancement in unravelling the detailed relationship between the chemical,
274 atomic, and magnetic properties of the oxide phase at the interface. Our discovery of the one-step
275 ALD integration of giant exchange bias in Fe_5GeTe_2 represents an important step towards practical
276 vdW spintronic devices.

277

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