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C. Franco, A. Wustrow, B. Xia, A. M. Baccarella, F. Burgos, J. Nicasio, E. Dooryhee, J. R. Neilson, and J. W. Simonson Phys. Rev. Materials **4**, 045404 — Published 13 April 2020 DOI: 10.1103/PhysRevMaterials.4.045404

Optimized in situ crystal growth and disordered quasi-one dimensional magnetism in $Li_2Mn_2(MoO_4)_3$

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(Dated: March 23, 2020)

The quasi-one dimensional structure of Li₂Mn₂(MoO₄)₃ consists of three mutually distinct chains of Li_{1-x}Mn_x-centered polyhedra in which Mn ostensibly adopts a J = 5/2 Mn²⁺ configuration. In situ x-ray scattering experiments carried out as crystallites emerge from a molten oxide solution facilitate the synthesis of large single crystals. Ex situ x-ray diffraction finds no evidence of long range Li/Mn occupancy ordering, suggesting that the structure is effectively composed of finite chains of Mn moments of statistically varying lengths. UV/Vis diffuse reflectance spectroscopy measurements establish a wide 3.43(12) eV direct charge gap consistent with the local polyhedral coordination of the nominally Mn²⁺ species. The temperature T dependence of the DC magnetic susceptibility χ reveals a fluctuating moment of only 2.74 ± 0.01 μ_B/Mn – dramatically reduced from the 5.9 μ_B/Mn expected for Mn²⁺. Meanwhile, the Weiss temperature $\Theta_W = -89 \pm 1$ K reveals antiferromagnetic fluctuations that are stymied from reaching an ordered state apparently by the chemical disorder intrinsic to the polyhedral chains. Measurements of magnetization vs. field H at $T \leq 10$ K are far from saturation even at H = 5 T and are strongly non-Brillouin like, instead scaling as $H/T^{0.24(3)}$ and suggesting the presence of quantum fluctuations associated with an eventual quasi-one dimensional, disordered magnetic phase.

I. INTRODUCTION

In low-dimensional or frustrated magnetic systems, quantum zero point fluctuations can be instrumental in determining the magnetic ground state¹. In such materials, the combined effects of frustration and low dimensionality often reduce the energy scale of long range magnetic order and give rise to novel or unconventional low-temperature magnetic states, including spin liquids, spin glasses, or quantum orbital spin chains².

 $Li_{4-x}M_x(MoO_4)_3$ -type compounds (M=Sc, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Al, or Ga) are examples of such compounds in which one-dimensional crystallographic motifs lead to high conductivities, fast ion transport channels, and ultimately potential technological applications $^{3-6}$. These practically important properties stem in part from the low dimensional, lyonsite-like structures often adopted by these phases and characterized by one-dimensional interstitial channels passing between vertex-sharing polyhedra and promoting the conduction of small cations like Li⁺ or Na⁺. In lyonsite-structured $Li_{4-x}M_x(MoO_4)_3$ -type systems, these interstitial tunnels additionally serve to segregate infinite, parallel chains of Li/M face-sharing octahedra, Li/M edge-sharing octahedra, and Li/M edge-sharing trigonal prisms, providing three distinct quasi-one dimensional structural motifs, all arrayed along the same crystallographic direction⁷.

Substitutional disorder between the Li and M sites is intrinsic across a wide swath of these systems and typically serves to stabilize the lyonsite structure while preserving electroneutrality⁸. Accordingly, when M is a valence-flexible 3d transition metal, its oxidation state and hence magnetic character can be controlled by the Li:M ratio. For instance, the temperature dependence of the magnetic susceptibility χ of Li₃Cr(MoO₄)₃ confirms the expected Cr³⁺ valence with an effective fluctuating moment of 3.91(1) μ_B/Cr and eventual weak ferromagnetism emerging below T = 10 K, perhaps due to short range order⁹. Magnetic properties of the presumably mixed-valent $Li_{1.8}Cr_{1.2}(MoO_4)_3$, however, have not been reported. Isostructural $Li_3Fe(MoO_4)_3$ reveals a similar Fe³⁺ oxidation state and corresponding weak ferromagnetism below ~ 12 K^{10,11}. On the other hand, the replacement of Li with a larger alkali metal like K or Rb obstructs the formation of mixed-occupancy sites, and their absence instead distorts the overall structure. Even so, exotic magnetic states can be achieved. In $Rb_2Cu_2Mo_3O_{12}$, for instance, the requirement for overall electroneutrality demands a J = 1/2 Cu²⁺ state, and two quantum critical regimes can be tuned with H at the upper- and lower-field boundaries of an ordered phase, with quasi-one dimensional quantum fluctuations dominating above the high-H quantum critical point QCP¹². Similarly, the J = 1 system Ni²⁺ ANi₂Mo₃O₁₂ (A = Rb or K) lacks substitutional disorder, is likewise distorted from the lyonsite structure, and engenders a correlated spin-1 tetramer ground state¹³.

Naïvely, we could expect compositions with the largest moments and the greatest transition metal occupancies to yield the most robust magnetic states, as might be realized with $J = 5/2 \text{ Mn}^{2+}$. Indeed, a detailed electron spin resonance ESR study on polycrystalline samples of Li₂Mn₂(MoO₄)₃ found the expected Mn²⁺ state along with the concomitant absence of any ESR signal corresponding to Mo ions, indicating the expected nonmagnetic Mo⁶⁺ configuration¹⁴. Most interestingly, χ measurements show a modest downturn below T = 1.4 K in this compound, which the study authors interpret as the onset of antiferromagnetism. It remains to be seen which of the three Li/Mn-centered polyhedral sites participate in this transition or if the ordered state indeed corresponds to long range antiferromagnetic order. Furthermore, it was very recently reported that Na₂Mn₂(MoO₄)₃ forms with a unique triclinic structure, and measurements of χ likewise suggest the onset of antiferromagnetic ordered state at Néel temperature $T_N = 6.5(5)$ K¹⁵.

We report here the results of *in situ* x-ray scattering experiments that have accelerated the synthesis of large, high quality single crystals of $Li_2Mn_2(MoO_4)_3$ suitable for spectroscopic and magnetic measurements. Subsequent ex situ x-ray diffraction measurements confirm the expected composition $Li_{2.004(6)}Mn_{1.996(6)}(MoO_4)_3$ to high precision, indicating that integer stoichiometry is indeed favored by $3d^5$ Mn²⁺, but bond valence sums uncover significant discrepancies from the ideal divalent state. UV/Vis spectroscopy reveals a wide gap consistent with the expected Mn atomic configuration, including multiplet features typically observed in octahedral MnO₆ systems. Finally, measurements of $\chi(T)$ divulge a paramagnetic fluctuating moment far smaller than the free ion value, corroborating that a full picture of the Mn valence is more complicated, while M(H,T) at T < 10 K is far from free ion behavior, revealing the importance of collective phenomena along the quasi-one dimensional Li/Mn polyhedral chains at the lowest temperatures.

II. METHODS

We grew single crystals of $Li_2Mn_2(MoO_4)_3$ as large as 5 mm \times 1 mm \times 1 mm from a molten eutectic solution of commercial MnO, Li₂MoO₄, and MoO₃ powders along the experimentally-determined Li₂O-MoO₃ pseudo-binary line¹⁶. In situ x-ray scattering experiments informing the crystal growth process were performed at the National Synchrotron Light Source II XPD beamline (28-ID-2) with an incident wavelength $\lambda = 0.18202$ Å. The *in situ* measurements were carried out on premixed powders sealed in air in 3 mm OD, 2 mm ID quartz capillaries, with the sample position constrained by quartz rods. Samples were heated with a hot air blower from 300 to 973 K at 0.2 K/s until melting was observed and subsequently cooled from 973 K to 773 K at 0.033 K/s to probe the crystal growth process. Once ideal growth temperatures had been identified, further *ex situ* crystal synthesis routines were undertaken in 3N Ag tubing to avoid quartz devitrification observed when samples were exposed to elevated temperatures for dozens of hours. As a precaution, x-ray fluorescence measurements carried out with a Niton FXL x-ray florescence spectrometer were unable to detect any Ag impurity in the resulting crystals with an estimated detection floor on the order of 10 ppm. The crystals were transparent

yellow and rod-like in habit and could be mechanically extracted from the oxide regulus.

We determined the crystal structure of $Li_2Mn_2(MoO_4)_3$ with an Oxford Gemini single crystal diffractometer with Mo-K α radiation. Some 59,815 reflections were collected with 99.4% completeness at T = 297 K, of which 2968 independent reflections were used to solve the structure via a charge-flipping algorithm^{17–19}. We collected UV-Vis diffuse reflectance spectra with photon energies $h\nu = 1.38$ to 4.96 eV in a Cintra 40 double-beam spectrometer equipped with an integrating sphere coated with $BaSO_4$. The resolution in λ was 1.0 nm. We performed measurements of DC magnetization M on a 2.64 mg collection of single crystals enclosed within a gold sachet via a Quantum Design Magnetic Property Measurement System from temperature T = 1.8 to 300 K. Magnetic measurements were carried out in both zero field cooled ZFC and field cooled FC conditions.

III. RESULTS AND DISCUSSION

A. In situ crystal growth optimization

Optimizing de novo growth processes to obtain large, high quality single crystals from solution is a laborintensive process that may require months of work in the laboratory. For the present system, we identified an ideal composition $(MnO)_{10}(Li_2MoO_4)_{49.5}(MoO_3)_{51.5}$ from oxide phase diagram literature^{16,20} and carried out preliminary *ex situ* experiments to verify crystal formation, which resulted in sub-mm crystals readily identified by their yellow color. Then, as shown in figure 1, we carried out an eight-hour series of *in situ* x-ray scattering measurements in which we observed the melting of the oxide solution and the resultant precipitation and growth of the desired crystalline phase, permitting us to tune the growth temperature range via a single experiment.

Fig. 1(a) shows the scattered intensities of this series of in situ measurements. At T = 373 K – the onset of the experiment – we observe diffracted peaks corresponding to three phases: rhombohedral $R\overline{3}h$ -type Li₂MoO₄. which is best identified by peaks at Q = 1.37, 1.52, and 1.76 Å⁻¹; *Pnma*-type MoO₃, best characterized by peaks at Q = 1.63 and 1.95 Å⁻¹; and $Fm\overline{3}m$ -type MnO, with peaks seen at Q = 2.50, 2.93, and 4.03 Å^{-1} . We see the emergence of new intensity, particularly at Q = 0.78and 1.01 $Å^{-1}$ in the 823 K dataset, which we ascribe to the formation of monoclinic $P12_11$ -type Li₂Mo₄O₁₃. The peaks corresponding to all three of initial phases, as well as those of $Li_2Mo_4O_{13}$ simultaneously disappear between T = 823 and 873 K, signifying melting of the oxide solution. Coincident with this phase transition we see the emergence of weak, coherent scattering at Q = 2.37, 3.38, and 4.12 $Å^{-1}$, which are consistent with strong $\{133\}, \{226\}, \text{ and } \{333\}$ reflections expected from our structure solution of $Li_2Mn_2(MoO_4)_3$ that we discuss be-



FIG. 1. (Color online) (a) A color plot of the observed intensities from *in situ* x-ray scattering as a function of the magnitude of the scattering wavevector Q. Experiments were carried out as T was increased from 373 to 973 K with measurements in 50 K increments followed by cooling to 773 K with measurements in 10 K increments, as plotted from top to bottom in the figure. Indigo represents low intensity, and the color is incremented in rainbow order from red, orange, yellow, green, blue, as intensity increases. (b) A schematic of the phases present in (a) as function of temperature as determined by indexing the diffracted peaks.

low. These Li₂Mn₂(MoO₄)₃-derived peaks persist as the flux is heated even to 973 K and then cooled to 773 K as the crystals continue grow as solubility decreases in concert with temperature. Finally, we begin to observe the reemergence of weak oxide peaks indicating recrystallization of the molten oxide solution at 803 K and below, suggesting that the ideal crystal growth range is $873 \rightarrow 803$ K and allowing us to grow large single crystals for subsequent measurements.

B. Disordered polyhedral chains and charge fluctuations

Figure 2 shows the crystal structure of $\text{Li}_2\text{Mn}_2(\text{MoO}_4)_3$, which consists of three unique, quasi-one dimensional chains of mixed occupancy Li/Mn positions that are oriented along the crystallographic *a* direction and mutually connected by vertex-sharing MoO₄ tetrahedra. Our charge-flipping algorithm rapidly and reliably converges to the unit cell shown in Fig. 2(a) with orthorhombic lattice

parameters a = 5.18710(1) Å, b = 10.5856(3) Å, c = 17.8521(4) Å, V = 980.23(4) Å³, and space group Pnma, in excellent agreement with previous reports of polycrystalline $Li_{2-2x}Mn_{2+x}(MoO_4)_3$ -type compounds and single crystal $Li_{1.60}Mn_{2.20}(MoO_4)_3^{21}$. Both the quasi-one dimensional nature of the structure and the mixed occupancy of the Li/Mn sites are more clearly evident in Fig. 2(b), which shows a central $Li_{0.669(3)}Mn_{0.331(3)}O_6$ octahedral chains surrounded by a ring-like structure consisting of Li_{0.393(2)}Mn_{0.607(2)}O₆ octahedral chains and chains of $Li_{0.548(4)}Mn_{0.452(4)}O_6$ triangular prisms. The central chains consist of facesharing octahedra, as show in Fig. 2(c), while the ring member chains (Fig. 2(d)) and prisms (Fig. 2(e)) are edge-sharing. Refinement yields an overall composition $Li_{2.004(6)}Mn_{1.996(6)}(MoO_4)_3$, which is indistinguishable from an integer occupancy to 0.3% precision. In the solution-growth environment, the integer-stoichiometry $Li_2Mn_2(MoO_4)_3$ composition appears to be favored. Our largest crystals were obtained from solutions with excess Li in a 10:1 Li:Mn ratio, and even when we adjusted this ratio, the composition of the resulting crystals remained unchanged. We conclude that during growth the crystals extract the necessary Li from the solution to maintain the favored Mn^{2+} oxidation state.

We considered the possibility that crystallographic disorder associated with mixed Li/Mn occupancy within the three chain structures shown in the figure could be relieved by lowering the space group symmetry. We would expect such lowering to be easily resolvable given the substantial difference in x-ray scattering cross-sections of Li and Mn. Manifestations of lowered symmetry might be a supercell or an incommensurate structural modulation, most likely along the chain direction, *i.e.* the crystallographic *a* axis. The $h0\ell$ reciprocal space map shown in fig. 1(f), however, shows no additional reflections as would be associated with a doubling or tripling of the cell along this direction, nor do we observe such reflections as would accompany an incommensurate structural modulation.

Along these lines, in figs. 1(g) and 1(h), we show cuts of fig. (f) to clarify the absence of a larger cell or modulated structure. Even with the intensity of the scattered x-rays plotted on a logarithmic scale, we see no evidence of weak peaks that cannot be indexed by the lattice parameters of $Li_2Mn_2(MoO_4)_3$. We do note that the observed odd reflections at (100), (300), (003), (005), and (007) should be systematically absent from the space group Pnma. In this case, however, the observed intensity of these peaks is consistent with $\lambda/2$ diffraction of the orders more intense (200), (600), (006), (0010), and (0014) reflections, respectively²², and we accordingly modeled the the effects of $\lambda/2$ contamination in our structural refinement. In any case, a simple doubling of the unit cell in all directions could not alone bring to order the non-half Li/Mn occupancies of the three chains. We likewise considered the possibility of reduced symmetry within a single unit cell by permitting refinement in



FIG. 2. (Color online) (a) The unit cell of $\text{Li}_{2.004(6)} \text{Mn}_{1.996(6)} (\text{MoO}_4)_3$ with Li/Mn-centered octahedra and right triangular prisms colored in red, Mo-centered tetrahedra colored in gray, and O drawn in black, as indicated. Refinement was carried out on structure factor F^2 and refinement parameters are residual R = 0.0336, weighted residual wR = 0.0822, and goodness of fit S = 2.100. (b) The primary tubular motif of the structure, consisting of three separated chain-like structures, colors as in (a). (c) The chain consisting of $\text{Li}_{0.669(3)}\text{Mn}_{0.331(3)}$ face-sharing octahedra, colors as in (a). (d) The chain consisting of the $\text{Li}_{0.393(2)}\text{Mn}_{0.607(2)}$ edge-sharing octahedra, colors as in (a). (e) The chain consisting of the $\text{Li}_{0.548(4)}\text{Mn}_{0.452(4)}\text{O}_6$ right triangular prisms, colors as in (a). (f) An unwarped image of the $h0\ell$ reciprocal space map with all reflections indexed by lattice parameters showing absence of a supercell or modulated structure to accommodate disorder. (g) A cut of (f) along (h00) plotted on a logarithmic scale to clarify the absence of a supercell or modulated structure along the direction of the chains. (h) A cut of (f) along the (00\ell) direction similarly clarifying the absence of a supercell or modulated structure outside the direction of the chains. (i) A typical single crystal on 1 mm grid paper.

any of the seven maximally non-isomorphic subgroups of *Pnma*, but in no case did introducing the associated extra degrees of freedom appreciably improve the quality of the refinement. From these results, we conclude that the occupancy of all three Li/Mn polyhedral chains remains statistical, and the structure contains finite chains of the magnetic species in the form of Mn, Mn-Mn, Mn-Mn, etc. with decreasing likelihood.

It was previously reported that polycrystalline samples could be sintered with various Li/Mn compositions and that sub-mm single crystals with composition $\text{Li}_{1.60}\text{Mn}_{2.20}(\text{MoO}_4)_3$ could be grown from such presintered polycrystalline precursors²¹. We note that the solution growth procedure we describe here has only been successful in producing crystals with a global Li:Mn ratio indistinguishable from 2:2, despite an approximately 10:1 molar ratio of Li:Mn in the molten growth solution. Elementary valence counting suggests the 2:2 composition is associated with Mn²⁺ and may therefore be the composi-

tion made energetically favorable by Hund's stabilization of the Mn $3d^5$ configuration.

Bond valence sums BVS are in agreement to first order with the $3d^5$ picture, but we nevertheless observe substantial discrepancies from the expected in-Specifically, BVS are 2.427(6) for teger values. Mn in the $Li_{0.669(3)}Mn_{0.331(3)}$ face-sharing octahedral chains, 2.246(4) for Mn in the Li_{0.393(2)}Mn_{0.607(2)} edgesharing octahedral chains, and 2.008(4) for Mn in the $Li_{0.548(4)}Mn_{0.452(4)}O_6$ prisms. Thus, even if all Mn BVS are in reasonably good agreement with the expectation of Mn²⁺, both octahedra are substantially overbonded. A potential explanation for the discrepancy is that the observed substantial Li occupancies of these two positions lead to shorter bond distances on the local level. Tables of ionic radii typically give Li^{1+} as 90 pm and high-spin Mn^{2+} as 97 pm if both species are octahedrally coordinated, so overbonding at the 5-10% level is to be expected. The $Li_{0.669(3)}Mn_{0.331(3)}$ face-

sharing octahedra, however, are more than 20% overbonded, to an extent that size effects alone are unable to explain. We conclude that Mn in this system is not precisely divalent, and accordingly that electroneutrality must require an equivalent valence change elsewhere. In a similar situation, in situ x-ray scattering experiments of structurally-related $LiCr(MoO_4)_2$ revealed Li intercalation and deintercalation to be accompanied by $Cr^{3+/2+}$ and $Mo^{6+/5+}$ redox so that overall charges remain balanced⁵. As therefore might be expected, our calculated BVS for the two tetrahedrally coordinated Mo sites in $Li_2Mn_2(MoO_4)_3$ are 5.815(14) and 5.763(16), despite full Mo occupancy on both sites. This modest underbonding of Mo^{6+} in concert with the substantial overbonding of Mn^{2+} suggests that the valence picture of $(\text{Li}^{1+})_{2.004(6)}(\text{Mn}^{2+})_{1.996(6)}(\text{Mo}^{6+})_3(\text{O}^{2-})_{12}$ is not as simple as previously thought and that transition metal valence fluctuations may play a role in the stabilization of the lyonsite crystal structure in this case.

C. Mn^{2+} -like excitations and the charge gap

As we show in figure 3, $Li_2Mn_2(MoO_4)_3$ is an insulator with a wide, direct charge gap, and its UV/Vis spectrum is broadly consistent with the expected localized exitons of a primarily Mn^{2+} state. Fig. 3(a) presents the diffuse reflectance R spectrum after background subtraction from a collection of infinitely $(\sim 1 \text{ mm})$ thick crystals with light directed normal to the (001) crystallographic surface. R approaches 1 at incident light energies $h\nu$ below 2 eV – indicating that the crystals are largely transparent to red and near infrared photons – and falls off broadly as $h\nu$ is increased, most likely as charge carriers are excited across the gap. An inflection in R at 2.0(1) eV is consistent with the transition in binary MnO of octahedrallycoordinated Mn^{2+} from the ground state to the first excited quartet state, ${}^{6}A_{1g} \rightarrow {}^{4}T_{1g}{}^{23,24}$, suggesting that the octahedrally-coordinated Mn in Li₂Mn₂(MoO₄)₃ are likewise predominantly divalent. This transition is spinforbidden by Pauli exclusion and is therefore accomplished by spin-spin or spin-orbit coupling. The resulting broad shoulder in R around 2.2(2) eV is likely the source of the observed vellow color of the crystals and may be associated with the charge transfer states we infer from BVSs.

Like the presumably ${}^{6}A_{1g} \rightarrow {}^{4}T_{1g}$ derived inflection at $h\nu = 2.0(1)$, the observed minimum in R near $h\nu = 3$ eV is again consistent with known d - d optical transitions in binary MnO, reinforcing the Mn²⁺ octahedral picture for Li₂Mn₂(MoO₄)₃. By comparison to MnO, we infer that the minimum in R corresponds to transitions from the ${}^{6}A_{1g}$ ground state to the ${}^{4}A_{1g}$ and ${}^{6}E_{g}$ excited states, which remain degenerate in the octahedral crystal field²⁵, and are expected at $h\nu = 2.95 \text{ eV}^{23}$. Above 3 eV, R increases again over a wide energy scale of 2 eV and may just reach a maximum near the upper $h\nu$ limit of our measurement, a feature likely associated with the



FIG. 3. (Color online) (a) The UV-Vis diffuse reflectance R spectrum of Li₂Mn₂(MoO₄)₃ (red) plotted as a function of incident photon energy $h\nu$ from 1.38 to 4.96 eV, corresponding to wavelength $\lambda = 900$ to 250 nm. The green arrow denotes the position of a likely transition of octahedral Mn²⁺ to its first excited state. (b) The square of the Kubelka-Munk function $f(R)^2$ plotted versus $h\nu$ (red), where the solid blue line is a best fit of the Tauc relation to the linear region (see text).

 $3d^5 \rightarrow 3d^4 + 4s^1$ transition observed in direct reflectance measurements of MnO²⁶. These spectroscopic results are consistent with the excitations expected of localized, octahedrally-coordinated Mn²⁺, though we stress that they do not require a uniform divalent configuration. Even prototypically divalent MnO has substantial d^6 and even d^7 ligand hole character^{23,27}, and both experimental and theoretical investigations of more complicated materials reveal even greater charge fluctuations in otherwise ostensibly Mn²⁺ systems ^{28–31}.

The square of the Kubelka-Munk function $f(R)^2$ plotted in fig. 3(b) permits us to estimate the magnitude of optical charge gap of Li₂Mn₂(MoO₄)₃ as $E_g = 3.43(12)$ eV. We found lines of best fit tangent to $f(R)^2$ for $h\nu > 4.05$ eV via linear regression to the Tauc relation $\alpha h\nu = C(h\nu - E_g)^{\beta}$, for absorption coefficient α and a sample geometry-dependent fitting parameter C. The nature of the allowed transitions across the gap determines the value of the exponent β in this expression, with $\beta = 1/2$ corresponding to direct transitions and $\beta = 2$ indirect. For Li₂Mn₂(MoO₄)₃, the regression is substantially improved when $\beta = 1/2$, suggesting that the optical gap is direct.

D. Disorder and correlated magnetic fluctuations

Magnetic measurements presented in figure 4 further dispel the simple picture of independent local moments in fixed-valent $(\text{Li}^{1+})_2(\text{Mn}^{2+})_2(\text{Mo}^{6+})_3(\text{O}^{2-})_{12}$. As expected of a Curie-Weiss paramagnet, the dc magnetic susceptibility $\chi = M/H$ shown in fig. 4(a) falls off rapidly as T is increased from our 1.8 K base temperature. A non-negligible temperature-independent contribution $\chi_0 = 1.4 \times 10^{-3}$ emu/mol Mn remains present across the entire measurement, presumably the sum of Van Vleck and core diamagnetic terms. Once χ_0 is sub-



FIG. 4. (Color online) (a) The temperature T dependence of the dc magnetic susceptibility $\chi = M/H$ with applied field H = 1000 Oe of a collection of Li₂Mn₂(MoO₄)₃ single crystals (red circles). Measurements carried out in FC and ZFC configurations are overlaid and indistinguishable. (b) The Tdependence of $1/(\chi - \chi_0)$ (red circles), where χ_0 is the sum of T-independent contributions to χ . The blue solid line is a fit to the Curie-Weiss law for T > 100 K corresponding to a fluctuating moment of 2.74(1) μ_B /Mn and Weiss temperature $\theta_W = -89 \pm 1$ K, as indicated. (c) Magnetization $M - M_0$ plotted as a function of H/T to illustrate deviation from Brillouin function-like behavior. M_0 is the temperature-independent contribution to M as in (b). Colors are T = 1.8 K (red), 4 K (orange), 6 K (green), and 8 K (blue), as indicated. The upper and lower solid violet lines are the q = 2, J = 5/2 Brillouin function with magnitude arbitrarily adjusted to the T = 8 K and T = 1.8 K data, respectively. (d) $M - M_0$ plotted as a function of H/T^{γ} with the critical exponent $\gamma = 0.24(3)$, colors as in (c).

tracted, the remaining susceptibility corresponds to the Curie Weiss law $\chi - \chi_0 = C/(T + \theta_W)$, as shown in fig. 4(b), where the magnitude of the effective fluctuating moment is derived from the Curie constant C, and θ_W is the Weiss temperature. The fit of this relation is in excellent agreement with the data for T > 100 K, yielding a fluctuating moment of only 2.74(1) μ_B/Mn – less than half the 5.9 μ_B expected for the Mn²⁺ free ion. This moment is notably less than even the 4.9 μ_B expected for the Mn³⁺ free ion. Observation of both static and fluctuating moments substantially less than free ion values is not atypical in superficially Mn²⁺-based insulators, for example LaMnPO²⁸, CaMn₂Sb₂^{32,33}, LaMnAsO³⁴, and BaMn₂As₂^{35,36}. Factors responsible for reduction of magnetic moments in these systems include valence fluctuations^{28,29}, effective dimensionality³⁷, and orbital hybridization, the first two of which our crystallographic and spectroscopic measurements suggest to be important in $\text{Li}_2\text{Mn}_2(\text{MoO}_4)_3$.

In addition to these departures on the local scale from Mn^{2+} physics, $\chi(T)$ suggests that the substitutional disorder observed in our crystallographic characterization plays a role in frustrating the onset of the eventual magnetic phase. Specifically, $\theta_W = -89 \pm 1$ K is nearly two orders of magnitude above the reported ordering temperature 1.4 K¹⁴, indicating a significant suppression of magnetic order that may originate from the inherent chemical disorder of the three Li/Mn polyhedrally-coordinated sites. The prevailing magnetic fluctuations are antiferromagnetic in nature, as indicated by $\theta_W < 0$, in agreement with the previous report. The magnitude of θ_W that we determined from $\chi(T)$ somewhat higher than was previously obtained from ESR measurements carried out on polycrystalline powders ($\theta_W = -56$ to -65 K)¹⁴. This observation is in line with our expectation of higher θ_W in single crystal samples, which by definition lack the grain boundaries and associated impurities that typically reduce θ_W ³⁸. We conclude that the absence of magnetic order between our base temperature T = 1.8 K and $-\theta_W = 89 \pm 1$ K is unrelated grain boundaries, contamination, or issues of crystal quality and instead stems from intrinsic crystallographic disorder that persists to the unit cell level. In the complete absence of such chemical disorder and frustration, one would expect $Li_2Mn_2(MoO_4)_3$ to transition to a long-range antiferromagnetic phase with T_N on the order of 100 K.

On the other hand, measurements of M below T =10 K demonstrate that the dominant magnetic excitations at the lowest temperatures are definitively not those of local paramagnetic moments, regardless of Mn atomic configuration. We plot in fig. 4(c) M after subtracting the same T-independent term $M_0 = \chi_0 H$ as in fig. 4(b). The remaining quantity $M(H,T) - M_0$ is invariably far from saturation, reaching only 0.23 μ_B /Mn even at T = 1.8 K and H = 5 T. The Brillouin function, which describes the magnetization of local paramagnetic moments as a function of H and T, would be 99% saturated at these values T = 1.8 K and H = 5 T, clearly in stark disagreement with our observations. Additionally, as the figure shows, the scaling of $M - M_0$ with the quantity H/T expected of independently fluctuating local moments is nearly completely absent. We must conclude that the Mn moments in $Li_2Mn_2(MoO_4)_3$ are correlated and that the observed fluctuations are instead the result of collective phenomena, presumably stemming from the onset of the magnetic transition reported previously at 1.4 K¹⁴. We note that our measurements of $\chi(T \ge 1.8 \text{ K})$ show no evidence of reaching this transition, which could occur at a temperature higher than 1.4 K in single crystal samples. Furthermore, the absence of a Brillouinlike contribution to M indicates that none of the three crystallographically-independent, quasi-linear sublattices of Mn (fig. 2(c-e)) hosts individually fluctuating moments, suggesting that all three may participate in this eventual transition.

Our subsequent scaling analysis shown in fig. 4(d) suggests a strong one-dimensional character to these correlated fluctuations above the eventual ordering temperature. As the figure demonstrates, $M - M_0$ scales as H/T^{γ} with $\gamma = 0.24(3)$ instead of the H/T dependence expected of independently fluctuating local moments. We note that there is no qualitative change in this result if we take $M_0 = 0$ as a sanity check. We recently observed similar phenomena, albeit with a larger value of $\gamma = 0.68(4)$, in Bi₂CrAl₃O₉³⁹. In that case, the sole magnetic ion Cr^{3+} is octahedrally-coordinated by O, and disordered, quasi-one dimensional chains of magnetic Crand non-magnetic Al-centered, edge-sharing octahedra are arrayed along a single crytallographic direction, much like the chains of Li- and Mn-centered polyhedra we find in $Li_2Mn_2(MoO_4)_3$. The apparent absence of any Cr/Al occupancy ordering along the octahedra of the former system leads to finite chains of Cr-centered octahedra with statistically varying lengths, which form the fundamental magnetic unit and manifest in M(H,T) as $\gamma < 1$.

As a point of comparison, $\gamma < 1$ is found throughout the disordered phase of so-called random transverse-field Ising spin chains (RTISC). RITSCs are exactly soluble in both ordered and disordered phases across a wide range of parameters^{40,41}, which makes for a convenient theoretical approximation to experimentally-accessible systems like $Bi_2CrAl_3O_9$ and $Li_2Mn_2(MoO_4)_3$. Specifically, χ diverges in disordered RTISCs at low temperatures as $\chi(T) \sim 1/T^{\gamma}$ ($\gamma < 1$), suggesting that M adopts a universal function of H/T^{γ} , just as we observe in $Li_2Mn_2(MoO_4)_3$ in fig. 4(d). These results to not imply $Li_2Mn_2(MoO_4)_3$ is an RTISC by any means but merely that compositional disorder and low-dimensionality, as we find from our crystallographic measurements are potential roots of the H/T^{γ} dependence. In true RTISC systems, γ varies continuously with a parameter δ that quantifies the system's distance from criticality with $\gamma = 0$ at the QCP. If the same arguments can be applied to $Bi_2CrAl_3O_9$ and $Li_2Mn_2(MoO_4)_3$, the smaller $\gamma = 0.24(3)$ we find for the latter would imply that this system is more weakly disordered, which is perhaps consistent with the Li/Mn occupancies of the individual chains being further from $Cr_{0.50}Al_{0.50}$ as observed in the former. Regardless of whether such a comparison can be made, it appears that $M(H/T^{\gamma})$ in Li₂Mn₂(MoO₄)₃ arises from a superposition of the thermal and quantum fluctuations of an ordered state, frustrated by chemical disorder, that coalesces only below T < 1.8 K.

IV. CONCLUSIONS

The growth of large crystals of $\text{Li}_2\text{Mn}_2(\text{MoO}_4)_3$ was facilitated and accelerated by *in situ* x-ray scattering measurements that identified the onset of nucleation, clarified the ideal growth temperature range, and demonstrated a path to avoid the nucleation of unwanted secondary phases, such as could potentially derail magnetic characterization. The crystals resulting from the optimized synthesis routine were large enough for spectroscopic characterization and for magnetic measurements even in a paramagnetic system with a small moment far from saturation at large fields.

Accordingly, we report that $Li_2Mn_2(MoO_4)_3$ is an insulator with a $3.43(12)~{\rm eV}$ direct charge gap. Elementary charge counting suggests an Mn^{2+} valence state, which is supported by previous ESR measurements. Moreover, this state is consistent with the excitations we observe in UV/Vis spectroscopy, paralleling those of the prototypical divalent system MnO. The full picture is more complicated, however, as is the case for other ostensibly divalent Mn-based insulators, including MnO itself. Ex situ single crystal x-ray diffraction measurements reveal substantial overbonding of two of the Mn positions, accompanied by underbonding of Mo. suggesting that the system hosts considerable charge fluctuations. These fluctuations are borne out in magnetic measurements that reveal a high temperature fluctuating moment less than half the value expected of the Mn^{2+} free ion. Across the various lyonsite-structured and related materials, nature compromises when necessary to preserve the structuretype by introducing substitutional disorder between the alkali metal and transition metal sites. When this compromise is circumvented with large radii alkali metals, the structure is forced to distort. We posit that charge fluctuations may be another, perhaps concurrent, avenue towards stabilizing the apparently fragile lyonsite structure.

The magnetic properties of Li₂Mn₂(MoO₄)₃ likewise reveal the importance of chemical disorder in frustrating the magnetic ordering temperature far below θ_W . Instead of ideal, long range antiferromagnetism, we detect finite, correlated chains of magnetic ions collectively fluctuating above an eventual ordered state that only occurs below T < 1.8 K. Li₂Mn₂(MoO₄)₃ appears to be a member of a broader class of finite magnetic chain systems potentially identifiable by H/T^{γ} scaling with $\gamma < 1$ at sufficiently low T, which may extend across a wide variety of compositions and structure-types. It remains to be seen if a QCP associated with $\gamma \rightarrow 0$ and controlled by disorder can be realized experimentally in this class of materials.

ACKNOWLEDGMENTS

This work was supported as part of GENESIS: A Next Generation Synthesis Center, an Energy Frontier Research Center funded by the U.S. Department of Energy, Office of Science, Basic Energy Sciences under Award Number DE-SC0019212. This research used the X-ray Powder Diffraction beamline of the National Synchrotron Light Source II, a U.S. Department of Energy (DOE) Office of Science User Facility operated for the DOE Office of Science by Brookhaven National Laboratory under Contract No. DE-SC0012704. The Stony Brook University single crystal diffractometer was obtained through the support of the National Science Foundation grant CHE-0840483. The authors are grateful for insightful conversations with P. G. Khalifah and Y. Janssen.

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CF and JWS conceived of the work. The crystal growth process was designed by CF, and samples were prepared by CF, FB, AMB, and JN. *In situ* x-ray scattering measurements were carried out by ED and analyzed by FB, ED, and JWS. The crystal structure was solved by BX. UV/Vis measurements were performed by CF, AMB, and FB. AW and JRN carried out the magnetic measurements, which were analyzed by CF, AW, JRN, and JWS. All authors contributed to writing the manuscript.

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