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Tunable magnetic order in low-symmetry SeO₃ ligand linked $TM_3(SeO_3)_3H_2O$ (TM = Mn, Co and Ni) compounds

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The ability to design crystal lattices with specific sub-lattice geometries has long been sought after especially in the context of harnessing magnetic frustration to elicit emergent physics. One approach which has seen some success recently, is to design magnetic sub-lattices chemically through the use of non-magnetic linker ligands as scaffolding for the magnetic ions. Here we report the magnetic properties of one such family of materials, the transition metal (TM) selenite hydrates with chemical formula $TM_3(SeO_3)_3H_2O$. These materials link highly distorted TMO_6 octahedra via non-magnetic $[SeO_3]^{2+}$ linkers. Studying members with TM = Mn, Co and Ni we use magnetic susceptibility and neutron powder diffraction to identify antiferromagnetic order in all three compounds with moderate frustration indexes. A comparison of the magnetic structures suggests that while the overall structure of the SeO_3 scaffolding remains unchanged the TM effects changes in the magnetic properties through tuning the internal bonding parameters and the single ion physics. Changing from Mn to Co is found to be particularly consequential manifesting changes in both the ordered moment direction and in the direction of the ordering vector. Field dependent measurements of the susceptibility and heat capacity reveal metamagnetic transitions in both Mn₃(SeO₃)₃H₂O and Co₃(SeO₃)₃H₂O indicating nearby magnetic ground states accessible under relatively small applied fields. Density functional theory calculations broadly confirm these results, showing both a sensitivity of the magnetic structure to the TM and its local environment. Although no spin liquid behavior is achieved, these results suggest the fruitfulness of such synthesis philosophies and encourage future work to engender higher frustration in these materials via doping, field, pressure or larger linker ligands.

I. INTRODUCTION

Recently the search for orders beyond the Landau paradigm has garnered much interest as such orders are predicted to offer both novel physics and enormous potential for quantum device design [1]. While several methods have emerged to find such order, one particularly fruitful pathway (at least for some classes of topological materials) has been to realize frustrated magnetism through competing magnetic interactions [2–4]. Materials can host competing interactions either through specific geometric arrangements of the magnetic sublattice, magnetic Hamiltonians with multiple interaction terms with different preferred couplings or length scales or combinations of both [4, 5]. For instance, when spin carriers are located on lattices based on corner or edge shared triangles such as Kagomé and triangular lattices, spin frustration will be present and three-dimensional longrange magnetic order tends to be suppressed, especially for a spin- $\frac{1}{2}$ value which can enhance quantum fluctuations [6, 7].

One controllable route to generate such geometric magnetic frustration, is through using non-magnetic bridging ligands such as $[VO_4]^{3-}$, $[PO_4]^{3-}$, $[GeO_4]^{4-}$ and [SeO₃]²⁻ which can link magnetic ions, acting as a structural scaffolding, and give rise to a variety of possible magnetic sub-lattices [8-16]. The hope of using such a route to generate potentially frustrated magnetic materials is to gain the control to design magnetic sublattices from the outset rather than search for frustrated physics in existing compounds. For instance, recently a series of novel magnetic compounds has been grown using vanadium oxides as the non-magnetic linkers. The variability of the VO_x scaffolding allowed for a zoo of TM sub-lattice structures ranging from alternating stacked triangular and honeycomb Mn layers in K₂Mn₃(VO₄)₂CO₃ to a onedimensional Mn chain in SrMn(VO₄)(OH) [8, 9]. Such an ability to generate different TM magnetic sub-lattices is hugely advantageous when trying to engender frustrated magnetic properties and demonstrates the need for more work exploring the structural space accessible via these techniques.

One linker ligand which has not enjoyed much study is SeO_3 . SeO_3 can be used to synthesize a series of

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novel transition metal (TM) selenite hydrate compounds with stoichiometry $TM_3(SeO_3)_3H_2O$ and oxidation state TM^{2+} [13–16]. These compounds form low symmetry structures with lower dimensional TM layers networked into three dimensional (3D) TM lattices. Previously, they were shown to exhibit a useful receptivity to different TM site ions, thus potentially allowing for some tunability to any magnetic properties. Furthermore, the SeO_3 linkers create a complex network of TM-O-TMbonds with numerous distinct TM sites per unit cell. Such a structure may allow multiple competing pathways for both direct and superexchange interactions. This is especially intriguing, for the TM oxidation state of ²⁺ which for the case of TM = Co would provide a route to build a magnetic sub-lattice potentially in a low-spin configuration with the much sought after $\frac{1}{2}$ state. In this paper we present the results of comprehen-

sive magnetic, transport and diffraction measurements on $Mn_3(SeO_3)_3H_2O$, $Co_3(SeO_3)_3H_2O$ and $Ni_3(SeO_3)_3H_2O$ evaluating them for signs of frustrated magnetism and studying the sensitivity of their magnetic properties to the specific TM ion. Using magnetic susceptibility and neutron diffraction we identify low temperature magnetic order in all three compounds, with evidence of moderate frustration. Characterization of this order reveals a general sensitivity of the stabilized magnetic structure to the TM site which tunes not only the local ion physics but also the internal bonding parameters of the overall structure. Specifically, we find the magnetic structure changes significantly between the Mn and Co/Ni members with a change in the magnetic ordering vector and a change from a layered structure to a fully 3D magnetic sublattice. Additionally, field dependent work suggests low field metamagnetic transitions in both Mn₃(SeO₃)₃H₂O and $Co_3(SeO_3)_3H_2O$, encouraging future work. While our results show long range order in all compounds, several behaviors of Co₃(SeO₃)₃H₂O suggest it has competing magnetic ground states and may be easily tuned between them.

II. EXPERIMENTAL METHODS

A. Synthesis

A low-temperature hydrothermal technique was employed to synthesize the single crystals of $TM_3(SeO_3)_3H_2O$ (TM=Mn,Co,Ni). All the reactions were performed in 23 mL poly(tetrafluroethylene)-lined pressure vessels (Parr instruments) at 220 °C for 2 weeks. To synthesize single crystals of each phase, a mixture of K_2CO_3 (Aldrich, 99%), $TMCl_3$ and SeO_2 (AlfaAesar, 99.4%), were prepared in a molar ratio of 1 : 2 : 3. Here, K_2CO_3 acts as a mineralizer which improves the quality of the single crystals (we note that any further increase of the K_2CO_3 amount in the reaction had a very little effect on size of the obtained single crystals.) For each reaction, a total of

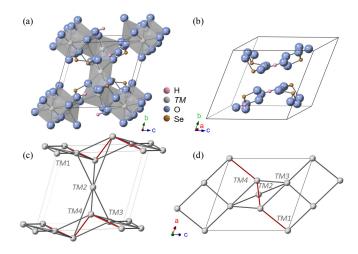


FIG. 1. (a) Crystal structure of the $TM_3(\mathrm{SeO_3})_3\mathrm{H_2O}$ compounds viewed along the a lattice direction. (b) Unit cell with TM site removed showing the $\mathrm{SeO_3}$ linkers and $\mathrm{H_2O}$ positions. View of TM sublattice (c) along the a and (d) b lattice directions. In panels (c) and (d) the red highlighted distances indicate the TM1-TM4 quasi-one-dimensional sublattice. The TM1-TM4-TM3 planes mentioned in the text are visible in panel (c) roughly coinciding with the crystallographic (010) plane. Images were generated with CrystalMaker.

0.5 g of reactants were mixed with 6 mL of de-ionized (DI) water. The reaction mixture was loaded to the poly(tetrafluroethylene)-lined pressure vessel and sealed After the reaction period, columnar shaped single crystals (average size 0.5 x 0.2 x 0.1 mm³) were isolated using suction filtration method by washing with DI water and acetone. Each reaction produced a very homogeneous mixture of single crystals of the target phases. Phase purity and elemental composition were confirmed using powder X-ray diffraction and a Hitachi S-3400 scanning electron microscope, respectively. Due to the small size of the obtained crystals, single crystal neutron diffraction was eschewed for neutron powder diffraction. In order to obtain enough sample mass for the neutron powder diffraction experiments, these reactions were repeated multiple times to obtain 2 g of single crystals from each phase which were then gently ground to obtain powders.

B. Neutron Diffraction Experiments

Neutron powder diffraction (NPD) measurements were performed using the time-of-flight high-resolution POW-GEN diffractometer of Oak Ridge National Laboratory's Spallation Neutron Source [17]. Using the high-intensity beam guide a wavelength band centered at 2.665 Å was used to maximize flux while allowing access to the large d-spacings needed to search for magnetic Bragg reflections.

Analysis of the neutron powder diffraction data was performed using the Rietveld method as implemented in the FullProf and GSASII software suites[18, 19]. A convolution of a Pseudo-Voight and back-to-back exponentials was used in the refinements to model the instrument resolution. In addition to profile fitting, the atomic positions and isotropic atomic displacement parameters of all sites were refined. For magnetic structure determination the Simulated Annealing and Representational Analysis (SARAh) software was used as well as the Bilbao Crystallographic Server [20–23]. Visualization of the nuclear crystal structure was performed using CrystalMaker[®] (CrystalMaker, Software Ltd www.crystalmaker.com) while the magnetic structure was visualized using VESTA [24].

C. First Principles Calculations

First principles calculations were performed using the all-electron density functional theory (DFT) code WIEN2K [25]. The generalized gradient approximation of Perdew et al was used. [26]. To perform such calculations using an all-electron code is arduous given the very small Hydrogen muffin-tin sphere radius used here of just 0.54 Bohr (in addition to the large and low-symmetry crystal structure) and accordingly a smaller RK_{max} value of just 3.12 was used. Here RK_{max} is the product of the smallest sphere radius and the largest planewave expansion vector. However, despite the small RK_{max} value these calculations are considered accurate as the sphere radii for O, Se and Co (and Ni) are much larger at 1.21, 1.57 and 1.92 Bohr respectively, so that the effective RK_{max} for these atoms is much larger and in the range of 7.0 to over 11, in keeping with standard calculation practice.

III. RESULTS AND DISCUSSION

A. Structure

The TM selenite hydrate structure reported in previous structural work is shown in Fig. 1[13–16]. As discussed, the TM occupies four distinct symmetry sites, one each on the special 1a:(0,0,0) and $1h:(\frac{1}{2},\frac{1}{2},\frac{1}{2})$ Wyckoff positions and two in the general 2i:(x,y,z) position. The first two positions describe a body centered like cell, with the second two defining a tilted distorted square planar-like arrangement around the 1h site. Each of the TM sites is coordinated by six O atoms (with ten distinct O sites in the unit cell) making highly distorted TMO₆ octahedra. In the planar arrangement of TM, along one diagonal the octrahedra are corner sharing while along the other they are edge-sharing, this central structure is then corner-sharing with the 1a site octrahedra. The Se sites act as 'linkers' between the octahedra (as shown in Fig. 1(b)) with the Se lone pair of electrons from the SeO₃ trigonal pyramids occupying the apparent 'channels' running along the a lattice direction as seen in

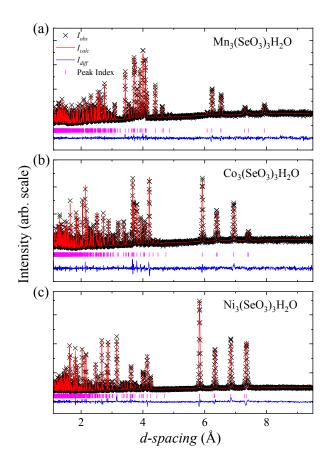


FIG. 2. Best model fits of neutron powder diffraction patterns collected on POWGEN at 100, 20 and 100 K for (a) $Mn_3(SeO_3)_3H_2O$, (b) $Co_3(SeO_3)_3H_2O$ and (c) $Ni_3(SeO_3)_3H_2O$. The data and peak indexes are shown as black and magneta '×' and '|' markers respectively while the diffraction patterns generated from the models and the difference curves ($I_{Calculated} - I_{Observed}$) are shown as red and blue curves respectively.

Fig. 1(a). An insightful in-depth description of the nuclear structure is given in Ref 14

Alternatively, (as suggested in Ref. 14), the TM metal sub-lattice can be considered as built of smaller quasione and quasi-two dimensional structures which connect to make the full three-dimensional sub-lattice. Such a description will be helpful in later discussions involving magnetic structures in different lattice settings. Starting with the TM1 (1a) and TM4 (2i) sites, a quasione-dimensional zigzag chain structure is seen which runs along the $(10\overline{1})$ direction (Fig. 1(d)). These chains are then connected into planes coinciding with the (010) crystallographic plane via the TM3 (2i) site. As shown in Fig. 1(d) the TM3 site links TM1 to TM4 and TM4 to TM1 on neighboring chains. To finish the full threedimensional sub-lattice these planes are then connected via the TM2 (1h) site as which sits halfway along the b axis between neighboring TM1 - TM4 - TM3 planes as seen in Fig. 1(c).

In order to check the reported nuclear structure and

TABLE I. Crystallographic parameters of $Mn_3(SeO_3)_3H_2O$, $Co_3(SeO_3)_3H_2O$ and $Ni_3(SeO_3)_3H_2O$ at 100, 20 and 100 K, respectively. Parameters determined from Rietveld refinements performed using NPD collected on the POWGEN diffractometer with the frame centered at $\lambda=2.665$ Å. The atomic displacement parameters are reported in units of Å².

	F	<u> </u>	
TM	Mn	Со	Ni
Space Group	$P\overline{1}$	$P\overline{1}$	$P\overline{1}$
T	$100 \mathrm{K}$	$20 \mathrm{~K}$	100 K
R_{wp}	2.26~%	2.73 %	2.95~%
a (Å)	8.3257(3)	8.0919(5)	7.9823(3)
b (Å)	8.2654(3)	8.2163(4)	8.1154(3)
c (Å)	8.9891(4)	8.5401(5)	8.4077(3)
$\alpha \text{ (deg)}$	68.714(1)	69.192(1)	69.610(1)
β (deg)	65.395(1)	62.820(1)	62.577(1)
$\gamma \text{ (deg)}$	67.815(1)	67.217(1)	67.730(1)
$V(Å^3)$	504.99(1)	454.54(1)	437.24(1)
TM1(1a)	. ,	. ,	, ,
x	0	0	0
y	0	0	0
z	0	0	0
U	0.004(1)	0.002(2)	0.001(1)
TM2 (1h)			
x	0.5	0.5	0.5
y	0.5	0.5	0.5
z	0.5	0.5	0.5
U	0.004(1)	0.002(2)	0.001(1)
TM3 (2i)			
x	0.658(1)	0.654(1)	0.654(1)
y	0.036(1)	0.036(1)	0.042(1)
z	0.795(1)	0.793(2)	0.798(1)
U	0.004(1)	0.002(2)	0.001(1)
TM4 (2i)			
x	0.695(1)	0.716(2)	0.710(1)
y	0.829(1)	0.861(2)	0.866(1)
z	0.423(1)	0.391(1)	0.390(1)
U	0.004(1)	0.002(2)	0.001(1)

see how it changes as a function of TM we performed NPD measurements. NPD patterns collected at 100, 20 and 100 K for $\rm Mn_3(SeO_3)_3H_2O$, $\rm Co_3(SeO_3)_3H_2O$ and $\rm Ni_3(SeO_3)_3H_2O$ respectively together with best fit models using the reported triclinic $P\bar{\rm I}$ structure are shown in Fig. 2. As seen the reported structure fits the data well, producing low residuals and visually adequate fits (Table I). Using this structure all observed peaks were fit well with no evidence of any impurity phases within the sensitivity of POWGEN. Selected crystallographic parameters extracted from the Rietveld refinements are reported in Table I.

Despite having the same crystal structure, these three compounds have significant differences in their refined crystallographic parameters. As might be expected from considerations of the ionic radii, the progression from Mn to Co to Ni causes an overall reduction in the unit cell parameters and volume with a nearly 10% volume reduction between $Mn_3(SeO_3)_3H_2O$ and $Co_3(SeO_3)_3H_2O$ and further 4% reduction between $Co_3(SeO_3)_3H_2O$ and

TABLE II. Selected bonding parameters of the $TM4O_6$ octahedra and along the TM1-TM4 quasi-1D sublattice as well as average TM-TM distances. All bond lengths and distances are reported in Å, while bond angles are in degrees.

\overline{TM}	Mn	Со	Ni
TM4-O			
max	2.40(1)	2.37(1)	2.30(1)
\min	2.06(1)	2.00(1)	1.99(1)
O-TM4-O			
max	113.2(1)	109.7(1)	108.5(1)
\min	66.24(1)	69.0(1)	74.0(1)
TM-O- TM	, ,	` ,	, ,
1-4	113.5(1)	95.78(1)	95.34(1)
4-4	95.5(1)	96.84(1)	96.73(1)
TM- TM	. ,	. ,	. ,
avg	3.74(1)	3.62(1)	3.54(1)

Ni₃(SeO₃)₃H₂O[27]. Looking at the individual parameters it is seen that while b, α and γ remain relatively similar across the series (at less than a 2% variation between the compounds) the a, c and β parameters change significantly. Changing the TM from Mn to Ni causes a 4% reduction in both a and β and a 7% in c. Furthermore, between Mn₃(SeO₃)₃H₂O and Co₃(SeO₃)₃H₂O the relative lengths of the lattice parameters change with c > a > b for the Mn compound while c > b > a for both the Co and Ni compounds.

As a measure of these effects, one can compare the unit cell volume V to that of a hypothetical orthogonal cell as a sort of quantification of the triclinic 'distortion' with a ratio of 1 corresponding to no distortion. Doing so results in V_{tri}/V_{ortho} of 0.82, 0.80 and 0.80 respectively for the Mn, Co and Ni compounds. By this measure, the triclinic 'distortion' increases from $Mn_3(SeO_3)_3H_2O$ to $Co_3(SeO_3)_3H_2O$ but remains fairly stable between $Co_3(SeO_3)_3H_2O$ and $Ni_3(SeO_3)_3H_2O$ [28].

With the complex network of corner and edge sharing TMO₆ octahedra previously discussed, such anisotropic structural changes should be expected to have significant impact on the numerous potential exchange pathways between the octahedra. Tuning the anisotropy will change the relative TM-O bond lengths and TM-O-TM bond angles along different directions which may allow for a tuning of the magnetic exchange interactions. Furthermore, the anisotropic application of chemical pressure will effect the crystal field levels in the octahedra possibly changing the ground state spin of the TM. If the structure indicates competing interactions of a similar energy scale as suggested, then small changes to the octahedra and TM-O-TM bond angles which even subtly change orbital overlap and orbital energies could have profound effects on the resulting magnetic structure.

Therefore, despite the inherent difficulty in quantifying such changes in this low symmetry structure, it is worth considering single elements to see the scale and trend of these effects. To start, we consider a single octahedron and measure the distortion via the maximum and minimum TM-O bond lengths and TM-O-TM bond angles. As suggested in Ref. 14 the TM4 octahedra is the most distorted for all of the measured compounds and so we will focus there. As shown in Table II, there are significant changes both in the raw values between the three systems and in the relative differences between the maximal bond measures. In general, one sees a trend towards a less distorted octahedra going from Mn to Co to Ni. In particular, the maximal O-TM-O angles become significantly less dissimilar while the TM-O bond lengths stay relatively stable. If we similarly consider the TM1-TM4 quasi-1D chain structure, it is seen that the TM-O-TMbond angles responsible for any superexchange interaction shift significantly - most notably in the relative value of the two possible exchange paths (TM1-O3-TM4 and TM4-O4-TM4) which flip in relative magnitude between the Mn compound and the Co/Ni compounds. As will be discussed in general terms in the DFT analysis, such structural tunability in a system with expected competing magnetic interactions may give a way to realize different magnetic structures via different TM and allow for tuning between them via doping continuously between the respective TM compounds [29].

B. Zero-Field Magnetic Structures

Aside from some preliminary magnetic susceptibility measurements on $\rm Mn_3(SeO_3)_3H_2O$, very little work has been performed to actually characterize the magnetic properties of these compounds [16, 30, 31]. As we show here, these materials have a rich magnetic behavior with indications of moderate frustration and a tunability via TM selection.

In Fig. 3 we show the results of magnetic susceptibility measurements performed between 350 and 2 K for $\rm Mn_3(SeO_3)_3H_2O$, $\rm Co_3(SeO_3)_3H_2O$ and $\rm Ni_3(SeO_3)_3H_2O$ along with their inverse susceptibility. The high temperature range of the inverse susceptibility $(1/\chi(T>100~{\rm K}))$ appears linear allowing the use of Curie-Weiss law fitting to characterize the magnetism of these compounds. Using such fitting, we report the effective moment per TM (μ_B/TM) and Curie-Weiss temperature (θ_{CW}) for each compound in Fig. 3.

Starting with Mn₃(SeO₃)₃H₂O (Fig. 3(a)), we find the magnetic susceptibility to increase with decreasing temperature until ~ 15 K where there is a sudden downturn likely indicating the onset of AFM order (with $T_N \sim 15$ K). From the Curie-Weiss Law fit, we find $\theta_{CW} = -52$ K which both indicates AFM interactions between the magnetic sites and moderate magnetic frustration (with a frustration index $f = |\theta_{CW}|/T_N$ of 3.5). These results agree reasonably well with those reported in Ref. 16. From our fitting we find an effective moment of 5.6 $\mu_B/{\rm Mn}$ which is consistent with a Mn²⁺ ion in a high spin state.

For Ni₃(SeO₃)₃H₂O we see similar behavior with χ increasing until ~ 21 K and a θ_{CW} of -85 K. This again

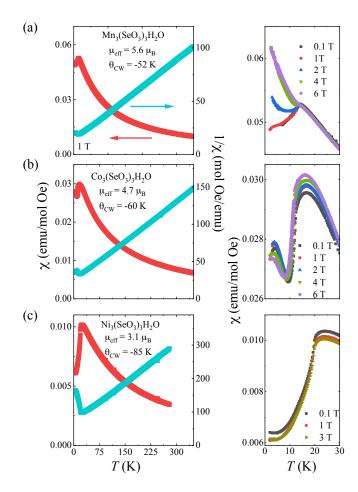


FIG. 3. Magnetic susceptibility χ and the inverse susceptibility $(1/\chi)$ for (a) Mn₃(SeO₃)₃H₂O, (b) Co₃(SeO₃)₃H₂O and (c) Ni₃(SeO₃)₃H₂O. The scale for χ is shown on the left axis while the scale for $1/\chi$ is shown on the right axis. All samples where measured under an applied field of 1 T. Plotted in the right panels are the low temperature data together with measurements taken under different applied fields.

indicates AFM interactions/order and a moderate frustration index of 3.8 similar to $\rm Mn_3(SeO_3)_3H_2O$. The Curie-Weiss fit suggests an effective moment per Ni of $3.1\mu_B$ which is close to that expected for Ni²⁺.

 ${\rm Co_3(SeO_3)_3H_2O}$ shows similar high temperature behavior with a sudden downturn at ~ 20 K, θ_{CW} indicative of AFM interactions, frustration index of 3 and an effective moment consistent with the ${\rm Co^{2+}}$ high spin state. However, below the apparent AFM transition, the susceptibility does not exhibit the monotonic decrease with temperature seen in the other compounds. Instead, χ_{Co} oscillates reaching a local minimum at 10 K then again increasing with decreasing temperature before starting to decrease with temperature at the lowest measured temperatures. These multiple local extrema may indicate several competing ground states and multiple magnetic transitions.

These results encourage low temperature neutron

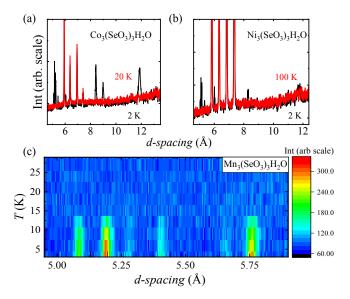


FIG. 4. Comparisons of the high d-spacing regions of NPD patterns collected in the paramagnetic state and below the putative magnetic transition for (a) $\text{Co}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ and (b) $\text{Ni}_3(\text{SeO}_3)_3\text{H}_2\text{O}$. (c) A NPD diffractogram for $\text{Mn}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ collected on warming from base temperature (2 K) to above the cusp feature seen in the magnetic susceptibility.

diffraction experiments to more fully characterize the potential magnetic orders as well as learn how they change between the three compounds. Therefore, we collected NPD patterns on all three samples between 100 and 2 K to look for magnetic Bragg peaks expected from the susceptibility measurements and solve any such magnetic structure.

Fig. 4 compares NPD patterns collected above and below the features seen in the susceptibility measurements, focusing on the high d-spacing range where magnetic scattering is generally strongest. In all three samples new Bragg reflections are seen upon cooling below the susceptibility downturn. For the Mn₃(SeO₃)₃H₂O sample, the large size of the Mn²⁺ moment allowed for quicker collection times and so the new Bragg peaks could be followed pseudo-continuously through the supposed T_N , the result is plotted as a diffractogram in Fig. 4(c). As seen the appearance of the new reflections coincide with the temperature of the cusp in the susceptibility ($\sim 15 \text{ K}$), indicating that this feature is indeed a magnetic transition. As new peaks appear at d-spacings not allowed by the nuclear structure we can characterize this as an AFM transition as suggested by the downturn seen in the susceptibility data. We note that no additional features are seen in the NPD patterns of Co₃(SeO₃)₃H₂O, which correspond to the lower temperature non-monotonic behavior observed in the susceptibility. As will be confirmed later, we take this as indicating only a single magnetic transition in this compound.

In order to further characterize the magnetic structure, we turn to identifying a magnetic ordering vector

TABLE III. Irreducible representations (Γ), Mn sites active for the Γ , number of basis vectors per TM site (ψ) and magnetic space groups for the $P\overline{1}$ nuclear symmetry with $k = (0, \frac{1}{2}, \frac{1}{2})$ or $(\frac{1}{2}, \frac{1}{2}, 0)$.

Γ	Mn site	ψ	Magnetic space group
Γ_1	1	3	$P\overline{1}$
	2	3	
	3	3	
	4	3	
Γ_2	3	3	$P\overline{1}'$
	4	3	

(k). Considering the d-spacings of the new peaks and the nuclear unit cell it is possible to index all the new low temperature peaks by applying factors of two to the nuclear unit cell. Doing so we find two k vectors with $k=(0,\frac{1}{2},\frac{1}{2})$ for $\mathrm{Mn_3(SeO_3)_3H_2O}$, and $k=(\frac{1}{2},\frac{1}{2},0)$ for both $\mathrm{Co_3(SeO_3)_3H_2O}$ and $\mathrm{Ni_3(SeO_3)_3H_2O}$. It is notable that these three materials are bipartite and do not all realize a single k-vector. We tentatively note that this correlates to several previously discussed structural differences which grouped these materials similarly such as the relative lengths of the lattice parameters with a>b for the Mn compound but b>a for both Co and Ni.

Shown in Table III are the results of representational analysis using the determined k vectors, the P1 nuclear symmetry and the positions of the four TM ions for $\vec{k} = (0, \frac{1}{2}, \frac{1}{2})$ and $\vec{k} = (\frac{1}{2}, \frac{1}{2}, 0)$. We note that representational analysis using either k and the $P\overline{1}$ symmetry results in identical constituent irreducible representations and so we only show a single table. The difference then between the resulting structures is found in the direction of the AFM correlations as determined by the distinct kvectors. For each k we find only two possible irreducible representations (Γ): Γ_1 and Γ_2 . Each Γ has three basis vectors (ψ) per TM site, with each basis vector corresponding to a magnetic moment component along one of the crystallographic axes. For Γ_1 this essentially allows each site to have a freely refinable magnetic moment which can have components along all three of the crystal axes with the k-vector describing the nature (AFM or FM) of the intrasite correlations. On the other hand, the Γ_2 irrep is only active for the TM3 and TM4 sites, enforcing a magnetic moment of zero on the TM1 and TM2 sites by symmetry. We find that this model produces poor fits of the data in addition to being physically unlikely, and so focus only on the Γ_1 structure.

Shown in Fig. 5 are the NPD data collected at 2 K for each compound together with the calculated diffraction pattern for the best fit model determined from Rietveld refinements using the Γ_1 irrep. As seen (and reported quantitatively in Table IV), the models using Γ_1 are capable of reproducing well the positions and intensities of all the new low temperature reflections for all three compounds. Comparing the crystallographic parameters extracted from the low temperature fits we generally find

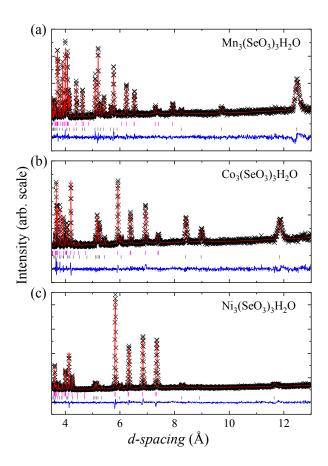


FIG. 5. Best model fits using the reported triclinic nuclear structure and the Γ_1 magnetic structures to neutron powder diffraction patterns collected on POWGEN at 2 K for (a) Mn₃(SeO₃)₃H₂O, (b) Co₃(SeO₃)₃H₂O and (c) Ni₃(SeO₃)₃H₂O respectively. Peaks belonging to the nuclear and magnetic structures are indicated by magenta and black tick marks respectively. As in Fig. 5, the black '×' markers, red lines and blue lines represent the data, model and difference curves respectively.

little change between the PM and AFM states in either the lattice parameters or the TM sites' fractional coordinates. Considering the low temperatures of the PM data this is not surprising as little thermal expansion should be expected below 100 K. It does suggest however, that either there is little magnetoelastic coupling or, as our DFT results will suggest, that magnetic frustration may already be affecting the nuclear structure even above T_N .

Looking at the extracted magnetic moment magnitude, we see a reduction from the effective moments found in the Curie-Weiss fitting of the magnetic susceptibility in the PM state. The magnetic moments determined from the Rietveld refinements are $\sim 75,57$ and 74% the values found from the Curie-Weiss fitting for $\rm Mn_3(SeO_3)_3H_2O$, $\rm Co_3(SeO_3)_3H_2O$ and $\rm Ni_3(SeO_3)_3H_2O$, respectively. A similar reduction of the ordered moment from the expected free ion moment has been seen in the rare-earth pyrochlores where it was a sign of magnetic frustration, though we note that there is some expectation that the

ordered moment should be smaller than the effective moment by definition [32, 33]. However, if we instead compare the refined moment to the calculated moment for TM with d^5 , d^7 and d^8 (i.e. Mn^{2+} , Co^{2+} and Ni^{2+}) with quenched orbital angular momentum $(M = gS\mu_B)$ we find differences of < 16% of the free ion moment (5, 3) and 3 μ_B for Mn, Co and Ni respectively) and the average moment per TM obtained from our refinements (4.2, 2.7 and 2.3 μ_B for Mn, Co and Ni respectively). Considering the differences between the effective and refined moments, it is interesting that $Co_3(SeO_3)_3H_2O$ has the largest reduction in the observed magnetic moment in the ordered state. This compound also exhibits a nonmonotonic behavior in its low temperature susceptibility - these observations together may indicate competing grounds states leading to strong frustration. However, further work fully explicating the spin-Hamiltonian is needed to elucidate potential competing interactions in these materials.

We note that while we do report variations between the moments on the four distinct TM sites in each compound which could ostensibly indicate different valences or spin states due to different crystal field levels seen by different ions, the errors on our reported moments is relatively large and prevents us from ascribing a strong certainty to this variation. This is also why we considered only the average moment per TM in the above discussion. However, as will be discussed later, our DFT results suggest that this is certainly a possibility, especially in the Co compound. More work using single crystals for neutron diffraction measurements could help reduce these uncertainties and determine whether the distinct sites have different spin states.

The magnetic models obtained from the best fit Rietveld refinements are shown in Fig. 6. Here they have been plotted using the expanded magnetic unit cells (transformed unit cells of (-c, -a + b, 2b) for $\text{Mn}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ and (a, b - c, 2c) for $\text{Co}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ and $\text{Ni}_3(\text{SeO}_3)_3\text{H}_2\text{O})$. To allow for ease of comparison despite the different ordering vectors the unit cells have been orientated along directions, which allow all three compounds to be compared to the figures of the nuclear unit cell shown in Fig. 1(c) and (d).

 $\mathrm{Mn_3(SeO_3)_3H_2O}$ has the simplest magnetic structure, only requiring two of the three possible ψ per TM leading to the co-linear order shown in Fig. 6(a). Here the Mn atoms form corrugated planes at the $z=0,\frac{1}{2}$ positions which are parallel to the magnetic cell's ab plane and stack with AFM correlations. However, since this magnetic unit cell is unique to $\mathrm{Mn_3(SeO_3)_3H_2O}$ it does not compare readily to the Co and Ni compounds. Therefore we also show $\mathrm{Mn_3(SeO_3)_3H_2O}$'s magnetic structure along the crystallographic directions established in Fig 1 in which it is easier to discern the previously described TM sublattices. In this view, we find AFM correlations along the Mn1–Mn4 chain structure. This chain is then linked AFM to the neighboring Mn1–Mn4 chain through the Mn3 which is FM to the Mn4 site and AFM to the

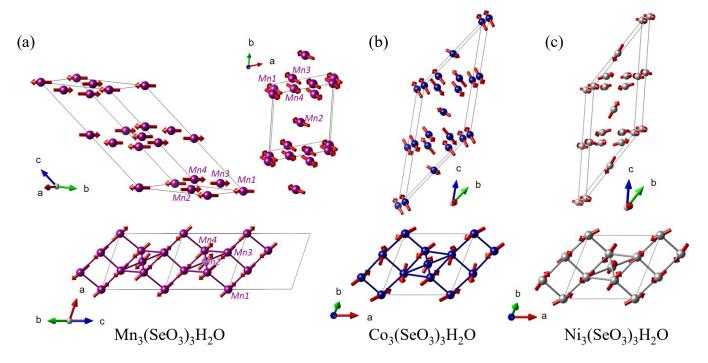


FIG. 6. Magnetic structures determined from the best fit models found in Rietveld refinments using the POWGEN NPDs collected at 2 K for (a) $Mn_3(SeO_3)_3H_2O$, (b) $Co_3(SeO_3)_3H_2O$ and (c) $Ni_3(SeO_3)_3H_2O$. All structures are plotted in their magnetic unit cells which are different from the nuclear unit cell due to the non-zero k vectors. To ease in comparison, all structures have been plotted along the equivalent crysatllographic directions to those shown in Fig. 1. For $Mn_3(SeO_3)_3H_2O$, an additional viewing direction was added to discuss the structures unique co-planer behavior.

neighboring chain's Mn1. This Mn1–Mn4–Mn3 structure creates the planes which are then linked through the Mn2 site. Using this description, we find the layers to be FM correlated along the stacking direction.

For $\text{Co}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ and $\text{Ni}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ no simple picture of the magnetic structure emerges as it does for $\text{Mn}_3(\text{SeO}_3)_3\text{H}_2\text{O}$. In both compounds each symmetry independent TM site requires non-zero coefficients on all three available ψ to produce adequate fits of the data. The resulting structures are neither purely co-linear nor co-planer and are most easily described via the quasi-1D and quasi-2D sublattices defined earlier.

Starting with the quasi-1D chains of $Co_3(SeO_3)_3H_2O$, we find FM correlations along the Co1-Co4 links and AFM correlations along the Co4–Co4 link with the moments pointing approximately perpendicular to the Co4–Co4 direction. This creates chain segments of three FM aligned Co4-Co1-Co4 coupled AFM to each other through a Co4-Co4 link (note that because of the reorientation of the magnetic unit cell the Co1 site at the bottom right corner of Fig. 6(b) is not symmetry equivalent to the Co1 at the top left corner - it only appears that way do to projecting along the c axis, this is clear in the perpendicular view.) The Co1–Co4 chains are then linked as before through nominally FM correlations along the short Co4–Co3 direction creating the Co1–Co4–Co3 planes. Within these planes the Co moments are nearly co-linear and canted out of the plane. The Co2 sites which link these planes are less canted from this stacking direction and are AFM correlated from plane to plane.

For $Ni_3(SeO_3)_3H_2O$, the interactions along the Ni4–Ni1–Ni3 planes and between planes can generally be described similarly to $Co_3(SeO_3)_3H_2O$. This is most easily seen in the projection along the magnetic unit cell's c-axis (Fig. 6(c)) where we see similar FM Ni4–Ni1–Ni4 segments AFM aligned through a Ni4–Ni4 link and FM connected to each other through the short Ni4–Ni3. However, as seen in the perpendicular view, the Ni1–Ni4–Ni3 moments show much less canting than in $Co_3(SeO_3)_3H_2O$ and are nearly co-linear along a chain made of the Ni1–Ni3–Ni4 sites.

As described, there are significant differences in the magnetic structures realized in these three compounds. $\rm Mn_3(SeO_3)_3H_2O$ exhibits a co-linear structure where all four Mn sites align either up or down along a globally determined direction (the crystallographic $(1\bar{1}0)$ direction in the magnetic unit cell). In the Co compound this is clearly not the case as the Co2 and Co1–Co4–Co3 planes have different moment directions. Even in the Co1–Co4–Co3 plane we find no single shared moment direction, with the Co1 and Co4 sites appearing co-linear while the Co3 site is rotated with respect to the rest of the plane. The description also breaks down for the Ni compound where no two sites share a moment direction. While in the Ni1–Ni4–Ni3 plane the moments loosely point along the Ni1–Ni3 and Ni–Ni4 directions,

TABLE IV. Crystallographic parameters of the magnetic phases of $\mathrm{Mn_3(SeO_3)_3H_2O}$, $\mathrm{Co_3(SeO_3)_3H_2O}$ and $\mathrm{Ni_3(SeO_3)_3H_2O}$ at 2 K. Parameters determined from Rietveld refinements performed using NPD collected on the POWGEN diffractometer with the frame centered at $\lambda=2.665$ Å. The atomic displacement parameters are reported in units of Ų and magnetic moments M are reported in μ_B/TM . The unit cell and atomic positions are reported in the nuclear unit cell setting for ease of comparison. The magnetic space group $P\overline{1}$ is listed in the Belov-Neronova-Smirnova notation and corresponds to magnetic space group #2.4.

TM	Mn	Со	Ni
Mag. Space Group	$P\overline{1}$	$P\overline{1}$	$P\overline{1}$
T	2 K	$2 \mathrm{K}$	$2~\mathrm{K}$
R_{wp}	2.5~%	2.7~%	3.2~%
a (Å)	8.3219(3)	8.0913(5)	7.9799(3)
b (Å)	8.2636(3)	8.2165(4)	8.1153(3)
c (Å)	8.9881(4)	8.5402(5)	8.4048(3)
$\alpha \text{ (deg)}$	68.670(1)	69.191(1)	69.610(1)
β (deg)	65.379(1)	62.823(1)	62.570(1)
$\gamma \text{ (deg)}$	67.830(1)	67.215(1)	67.725(1)
$V(Å^3)$	504.47(1)	454.53(1)	436.93(1)
TM1 $(1a)$			
x	0	0	0
y	0	0	0
z	0	0	0
U	0.003(2)	0.003(2)	0.002(2)
M	4.2(2)	3.0(3)	2.3(5)
TM2 (1h)			
x	0.5	0.5	0.5
y	0.5	0.5	0.5
z	0.5	0.5	0.5
U	0.003(2)	0.003(2)	0.002(2)
M	4.3(2)	2.3(4)	2.0(5)
TM3 (2i)			
x	0.658(2)	0.651(1)	0.654(1)
y	0.035(1)	0.033(1)	0.039(1)
z	0.795(2)	0.793(2)	0.800(1)
U	0.003(2)	0.003(2)	0.002(2)
M	3.9(2)	2.7(3)	2.9(4)
TM4 (2i)			
x	0.698(2)	0.718(2)	0.709(1)
y	0.826(2)	0.863(2)	0.867(1)
z	0.422(2)	0.389(2)	0.389(1)
U	0.003(2)	0.003(2)	0.002(2)
M	4.5(2)	2.8(3)	1.9(5)

the Ni2 site is rotated significantly away from this direction.

Correlating these changes to the previously discussed structural changes is quite difficult in such a low symmetry structure. Previously, we defined a potential measure of the triclinic distortion as a volume ratio and found that $\rm Mn_3(SeO_3)_3H_2O$ had the smallest such distortion while $\rm Co_3(SeO_3)_3H_2O$ and $\rm Ni_3(SeO_3)_3H_2O$ were more distorted and similar to one another. Considering the magnetic structure, we see that the least distorted structure (by this measure) has the most regular magnetic order. On the other hand, the bonding parame-

ters of the TM octahedra were found to be the most highly distorted in Mn₃(SeO₃)₃H₂O when compared to Co₃(SeO₃)₃H₂O and Ni₃(SeO₃)₃H₂O and so such simple considerations are not too illuminating here (Table II). Similarly, if we try to apply the Goodenough-Kanamori rules based off the TM-O-TM bond angles reported in Table II we generally do not find good agreement finding for instance AFM correlations along bond angles close to 90° . Looking at the average TM-TM distance we find that the magnetic structure becomes less co-linear as the distance decreases. This may indicate an increase in orbital overlap in the Ni and Co compounds, which could strengthen both direct and superexchange interactions relative to single ion physics and cause the non co-linear structures we observe. This would suggest that single ion anisotropies may ostensibly be stronger in the Mn compound and help establish the co-linear structure assuming they may arise from easy-axis physics. However, a more complete understanding of what leads to these changes would require careful inelastic neutron scattering experiments to determine both the exchange interactions and local anisotropies of the magnetic Hamiltonian - we leave this to future neutron scattering work.

C. First Principles Analysis of the Magnetic Structures

In an effort to shed light on the observed magnetic behavior, we performed DFT calculations studying the Co and Ni compounds. For the crystal parameters we used the experimentally determined lattice parameters and angles, but relaxed the internal parameters in a nonmagnetic configuration. This produces internal bonding parameters somewhat different than those determined experimentally, in terms of TM-O distances. This suggests either the influence of magnetoelastic coupling in determining these structures, or potentially the difficulty in applying standard DFT approaches to these compounds. Note that recent work [34, 35] has found that for magnetically frustrated materials as considered here, the magnetic order can play an important role in determining the crystal structure, even well above the ordering temperature, and it is quite possible that similar effects are relevant here. The results presented here are to be considered in this context.

For both compounds, FM and AFM configurations were considered with all magnetic states falling more than 200 meV or 450 meV per 3d atom (for the Co and Ni compounds respectively) below the non-magnetic configurations. Therefore, despite the relatively low ordering temperatures, the magnetism can be considered as local moment magnetism. For both compounds the AFM states were lower in energy than the FM states in agreement with the experimentally determined AFM structures. However, it is important to note that we only considered co-linear AFM states in our calculations due to the computational challenge of handing these low sym-

metry nuclear structures with canted magnetism.

Starting with Co₃(SeO₃)₃H₂O, we find the magnitude of the ordered moments for the Co1, Co2 and Co3 sites in the ground-state AFM configuration are 2.45, 2.31 and $2.45 \mu_B$ respectively in reasonable agreement with the refined values. On the other hand, we find a moment on the Co4 site of 0.28 μ_B - significantly less than the 2.8 μ_B found experimentally. This is surprising, while it is possible that we observe a reduction in one of the Co sites experimentally, it is the Co2 site not the Co4 site and the effect, if real, is significantly less than seen here. It is likely that this reduction originates from the substantial structural differences among the various Co sites which will produce different local environments and possibly spin-states. It is possible that the internal parameter relaxation applied in our calculations gives rise to this discrepancy with experiment. However an important take away is then that the local bonding can lead to such a change in the local moment on ostensibly isovalent sites and that the employed model is likely missing some interaction (such as magnetoelastic coupling) as suggested previously and seen in other similar systems [34, 35]. We also note, that in our calculations we found a significant reduction in energy when the Co3 and (moreso) Co4 2i sites were allowed to split. This led to differing moments on Co₃/Co₃' and Co₄/Co₄' sites. However, since this is not observed experimentally we compare to a single value on either of these two sites.

For Ni₃(SeO₃)₃H₂O we also find an AFM ground state. Here the magnitude of the ordered moments on the Ni1–4 sites, respectively, are 1.53, 1.54, 1.54 and 1.48 μ_B . While these are smaller than the observed values, which range from 1.9 to 2.9 μ_B , we note that the calculated values are understated due to the small muffin tin radius. A fair estimate of the full ordered moment can be made from the FM calculation, which finds a total moment of 12 μ_B per cell, or 2 μ_B , which is significantly closer to the 1.9 to 2.9 μ_B . This presents a substantial difference from the Co₃(SeO₃)₃H₂O results, where the Co4 moment (in the ferromagnetic configuration) is just 0.38 μ_B .

The implication of these results is that, as found in neutron diffraction, there are substantial differences in the magnetic order of the two compounds, despite a nominally similar physical structure and that small changes to the internal bonding parameters can have vast effects on the magnetic behavior. In the calculated ground states both compounds are insulating, so that simple charge counting would yield a divalent state for both 3d atoms, with the extra paired 3d electron of Ni accounting for the reduced moment in the Ni compound. This extra electron of the Ni element clearly drives the substantial differences in magnetic order between the two compounds, with the magnetic order already complex due to the low-symmetry crystal structure in a quaternary oxide.

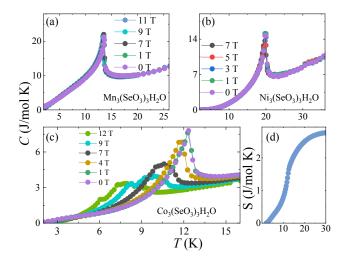


FIG. 7. Heat capacity collected under applied magnetic fields on pelletized powder samples of (a) $\rm Mn_3(SeO_3)_3H_2O$, (b) $\rm Ni_3(SeO_3)_3H_2O$ and (c) $\rm Co_3(SeO_3)_3H_2O$. (d) Magnetic entropy of $\rm Co_3(SeO_3)_3H_2O$ for zero applied field (see text for details).

D. Field-Dependent Magnetism

To better characterize the strength of the magnetic interactions, we performed field dependent magnetic susceptibility and heat capacity measurements (Figs. 3 and 7). Such measurements may allow us to identify the relative scales of magnetic interactions via observing metamagnetic transitions and the field strengths needed to influence the magnetic order

For Mn₃(SeO₃)₃H₂O we find a metamagnetic transition in the magnetic susceptibility which occurs between 2 and 4 T. Here we see the temperature of the susceptibility feature does not change significantly, but the behavior below T_N goes from a down turn indicating AFM order to an upturn indicating FM type order. Looking at the heat capacity (Fig. 7(a)) we can be sure this is not a change to PM as the sharp cusp feature observed at zero field persists up to the highest measured field of 11 T. While we cannot be sure from our current measurements, an explanation consistent with these data is that the Mn sites' main interaction is single-ion anisotropies which determine locally the moment direction. As the field is increased these moments may undergo a spin flip transition (as we recently reported for a local-Ising type system (36 and 37)), which would allow them to align somewhat with the field while remaining distinct from a polarized PM state as indicated by the sharp feature in the heat capacity.

On the other hand, $\text{Co}_3(\text{SeO}_3)_3\text{H}_2\text{O}$ shows relatively little change in its susceptibility under fields up to 6 T (Fig. 3(b)). No upturn indicating FM like behavior is observed. Instead, the peak indicative of T_N and the previously discussed lower temperature features are slightly suppressed to lower temperatures with increas-

ing field. However, the heat capacity tells a different story. At zero field a sharp transition is observed (as for $Mn_3(SeO_3)_3H_2O$) with no evidence of the second transition tentatively identified in the susceptibility data (Fig. 7(c)). As a magnetic field is applied and increased this peak broadens and decreases in temperature, indicating a suppression of T_N with field. However, by 9 T this heat capacity feature has broadened considerably and appears to have split into two peaks. Upon increasing the field further (to 12 T), this behavior continues with the both peaks being pushed to lower temperatures and the two peak feature persisting.

Such behavior is difficult to interpret in light of the susceptibility data. With no field dependent upturn to the susceptibility we can suppose that the field is not driving a spin-flip transition (even of some part of the Co sublattice.) However, the two peak feature indicates the applied field must be inducing some second transition, whether it is due to the random field orientation with respect to individual crystallites in the measured powder or is intrinsic of the whole sample undergoing multiple consecutive transitions is difficult to say. However, due to the already low crystal symmetry $(P\overline{1})$ and low temperatures involved we argue that this is likely an induced magnetic transition rather than a structural one. Therefore, we suggest that such results indicate the presence of multiple AFM states accessible with relatively low fields and so merit future neutron diffraction work with single crystal samples.

As another check for the low temperature upturn seen in Co₃(SeO₃)₃H₂O's susceptibility we estimated the magnetic entropy of Co, using the zero field heat capacity data and extracting the phonon background by fitting the curve above the transition with the Debye model (Fig. 7(d)). As in the heat capacity, no second transition is observed in the magnetic entropy. However, interestingly we find the integrated magnetic entropy (S)to be only half of the expected Rln2 value at 2.8 J/mol K. This result is not uncommon in Co compounds and has been suggested variously to indicate the possibility of the residual entropy releasing below 2 K, more entropy releasing above the transition temperature due to spin fluctuations or a transition from a low-dimensional ordered state to full long range order in three dimensions [38, 39]. Nonetheless, the temperature-dependent heat capacity and magnetic entropy data present no evidence of the second transition tentatively identified in the susceptibility data, confirming our analysis of the NPD data (Fig. 7(c)) but leaving the upturn in the susceptibility unexplained.

Finally, the field dependent measurements of $Ni_3(SeO_3)_3H_2O$ indicate behavior different than either of these samples. In the susceptibility measurements nearly no change in the sample response is found with neither the FM behavior of $Mn_3(SeO_3)_3H_2O$ nor the suppression of T_N seen in $Co_3(SeO_3)_3H_2O$. A similar result is obtained in the heat capacity measurements with a single sharp feature observed at O T which

persists up to the highest measured field of 7 T with no observable change in temperature.

IV. CONCLUSIONS

Our results show $TM_3(SeO_3)_3H_2O$ to be an interesting system to study the use of non-magnetic linker structural elements to form tunable magnetic sublattices. Studying the TM = Mn, Ni and Co compounds, we first showed that all three crystallize in the P1 space group with a complex network of highly distorted edge and corner sharing TMO₆ octahedra. Simplistic quantifications of the structural triclinic 'distortion' lead to the identification of the unit cell structure of Mn₃(SeO₃)₃H₂O being the least distorted while having the most distorted local environments for the TM sites. This was in contrast to both Co₃(SeO₃)₃H₂O and Ni₃(SeO₃)₃H₂O, which were relatively similar with larger triclinic distortions but more regular TMO_6 octahedra. Magnetic susceptibility measurements showed all three compounds to have AFM like transitions at temperatures < 20 K which Curie Weiss fitting revealed as moderately frustrated with frustration indexes around 3.5. Using neutron powder diffraction we confirmed these susceptibility features indicated the onset of long range AFM order with $k = (0, \frac{1}{2}, \frac{1}{2})$ for Mn₃(SeO₃)₃H₂O and $k = (\frac{1}{2}, \frac{1}{2}, 0)$ for both $Co_3(SeO_3)_3H_2O$ and $Ni_3(SeO_3)_3H_2O$.

Using Rietveld refinements and representational analysis, the low temperature magnetic structures were solved. showing an evolution from a relatively simple co-linear structure for Mn₃(SeO₃)₃H₂O to more complex canted structures for both $Co_3(SeO_3)_3H_2O$ and $Ni_3(SeO_3)_3H_2O$. Comparisons of the extracted magnetic moment to the effective moment of the Curie Weiss fits and calculated TM^{2+} moment reveal disparities of ~ 20 and $\sim 16~\%$ respectively, also potentially indicating frustrated physics. Field dependent heat capacity and magnetic susceptibility measurements, suggest possible metamagnetic transitions for both $Mn_3(SeO_3)_3H_2O$ and $Co_3(SeO_3)_3H_2O$. These latter observations suggest follow-up work using field dependent single crystal diffraction data to identify the metamagnetic states and inelastic neutron scattering to solve the spin-Hamiltonians. From there interesting physics may be realized through applying hydrostatic or uniaxial pressure to tune the structural and consequently magnetic properties. Additional work creating doping series may also be interesting as one tunes between the co-linear state of Mn₃(SeO₃)₃H₂O to the canted state of $Co_3(SeO_3)_3H_2O$. Doing so may increase the frustration in these materials and lead to a quantum critical point as the material crosses from $k = (0, \frac{1}{2}, \frac{1}{2})$ to $k = (\frac{1}{2}, \frac{1}{2}, 0)$. Of further interest would be attempting to grow similar other structures using the SeO₃ non-magnetic linkers, or even larger such linkers, particularly if the dimensionality of the TM sublattice could be decreased. Such work may lead to quantum effects as the magnetism is confined and the structural variability through TM site allows for the different interaction pathways to be tuned.

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