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INTRODUCTION

The origin of the charge density wave (CDW) transition, occurring in TiSe₂ around 202 K [1], has been an ongoing debate for decades, with proposed mechanisms including an excitonic insulator phase [8] and the band-type Jahn-Teller effect [9]. For the former, it can arise either in a small band gap semiconductor or a semimetal [8]. Below and its absolute value (positive or negative) is still under debate [10–18]. The latter proposed CDW mechanism is independent of the free carrier concentration [9], and this cannot account for the incommensurate diffraction spots seen in $TiSe_2$ [1]. Recent experimental evidence favors the excitonic insulator scenario [16, 19–23], but theories predict that the exciton condensation can either ́be a superfluid [24], or an insulator [25]. Most recently, њw maraaar maraar Huang *et al.* [27] showed insulating behavior for their

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One problem faced in studying $TiSe_2$ is the inconsis-ى ancy between single-crystalline TiSe₂ grown by I₂ vapor 便置置置置 > 0) is registered in the single-crystalline sample (dashed ፀbools} a seriee de la seriee de ෙ is found in the polycrystalline sample (solid line). To our TiSe₂, and the effect of the synthesis conditions on the plex effects of chemical doping, intercalation, or pressure can be understood.

It is well known that, for TiSe₂ single crystals, the transport agent iodine might partially substitute for Se and dope the system [1, 15]. Se deficiency also serves as a method of self-doping [30]. Both dopants presumably contribute additional density of states near the Fermi sur-

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FIG. 1. A comparison of the resistivity (normalized to room temperature values) for iodine-grown TiSe₂ single crystals with the current $i \parallel ab$ (dashed line) or $i \parallel c$ (open circles), and polycrystalline (solid line). The full triangle is used to identify samples that are 'As Grown, Air Quenched' throughout the text.

METHODS

Polycrystalline samples of TiSe₂ were synthesized by solid state reaction with a Ti:Se ratio of 1:2.02. The excess Se was added to compensate for the partial evaporation inherent during synthesis. The samples were sealed in quartz ampoules under partial Argon atmosphere and heated at 50°C/hr to 650°C, followed by a 48 hour dwell at this temperature. Subsequently, the samples were either cooled at different rates, or annealed at different temperatures or different times under partial Argon atmosphere. TiSe₂ single crystals were grown by chemical vapor transport with I₂ as the transport agent. Ground elemental Ti and Se were sealed in quartz tubes with a ratio of 1:2.02 and 5 mg/cm³ of iodine. The tubes were then placed in a 550°C - 650°C temperature gradient and Structural characterization was done using a Bruker ња чарва The quantitative chemical composition was determined by electron probe microanalysis (EPMA) using a JEOL JXA ل rent, and beam spot size (~ 300 nm). The Se L_{α} and Ti K_{α} X-ray lines were simultaneously measured using counting times of 10 seconds per peak and 5 seconds per each lower and upper background, respectively. Each element was simultaneously measured on two different statistics of the measurement. Se L_{α} was analyzed on two TAP diffracting crystals, and Ti K_{α} was analyzed 止 The standards used to calibrate the spectrometers were Ti = 59.9400 wt. %). For quantification, the ZAF matrix correction was used. The error of analysis, determined after analyzing secondary standards is below 2%. The instrumental standard deviation (1σ) for Se and Ti in each analysis is 0.24% and 0.47%, respectively. The quantitative analyses given in element wt. % were converted to atomic ratios, and then the stoichiometry of the analyzed compound was normalized to one Ti atom.

RESULTS AND DISCUSSION

Post synthesis cooling rate r

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The first experiment was dedicated to testing the effect of the cooling rate r post synthesis on the electrical resistivity. Three samples were synthesized as described in the Methods. Sample A was air quenched $(r > 2000^{\circ} C/hr)$, sample B was fast-cooled to room temperature at a rate $r = 20^{\circ}$ C/hr, and sample C was slow-cooled at r =4°C/hr. The scaled semi-log $\rho(T)/\rho(300 \text{ K})$ plot is displayed in Fig. 2(a). While all three samples show a nearly 5 time increase in $\rho/\rho(300 \text{ K})$ on cooling to 150 K, the air-quenched sample A displays a broad local minimum centered around 60 K, while both samples B and C exhibit nearly two orders of magnitude resistivity increase down to 2 K. Hall coefficient values (not shown) are negative This rules out the possibility of a change in dominant carrier type as the cause of change in the low temperature resistivity. The large change in the resistivity as a function of cooling rate prompted the need to check sample composition for possible non-stoichiometry. The results of the EPMA analysis, displayed in Table I, indicate that all three samples are stoichiometric (to within a 1% error). This does not rule out that the resistivity ل position variations below the EPMA resolution limit, or, as discussed below, conductive grain boundaries and WL 3) does not show any measurable change in either the peak position or peak shape among the three samples, consistent with invariable lattice parameters.

Cooling rate r (°C/hr)	Se
A: > 2000 (air quench)	2.02 ± 0.01
B: 20	2.01 ± 0.01
C: 4	2.00 ± 0.01

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For comparison, the single crystal sample with iodine inclusions does not show WL behavior either in $\rho(T)$ or in MR (Fig. 2(d)). EPMA reveals a 1% iodine impurity per formula unit in the single crystalline samples. In our single crystal sample, the iodine inclusions might dope



the system and dominate the transport property which leads to a suppression of WL behavior. A recent electrical transport study on iodine-free TiSe₂ single crystals does show a large increase in electrical resistivity on cooling, qualitatively consistent with what is seen in our polycrystalline Samples B and C [28]. It will be informative to investigate the magnetic field effects on the transport properties in these iodine-free single crystals to quantitatively analyze the characteristic parameters from the WL correction. The WL effect noted here for the first time in TiSe₂ had been previously reported in another TMDC, VSe₂[40].

Cooling samples slowly after synthesis was expected to decrease the extent of disorder in the crystals and increase the average grain size. In an attempt to characterized disorder, we turn again to the X-ray refinements. There are at least four contributions to peak width in powder X-ray diffraction [33]: instrumental broadening, thermal vibrations, grain size, and microstrain. Instrumental Thermal vibrations increase the peak width with increasing temperature. Peak width increases with reduced grain size and increasing microstrain. No variations in the X-ray peak widths are measured in the current pollycrystalline samples (inset of Fig. 3). Differential instrumental or thermal peak broadening can be ruled out, since all samples were prepared and measured at room temperature on the same instrument. Because all peaks are of similar width, no difference due to grain size or microstrain can be resolved between samples A, B and C.

Post synthesis annealing time t

The next set of experiments focuses on the effect of annealing time t. Different pieces of sample A were annealed at $T = 200^{\circ}C$, for times t ranging from 1 to 6 days, followed by air quenching. The low anneal temperature was chosen to relieve quenched-in disorder without adding more disorder from quenching at a high temperature. Resistivity shows a general upward, albeit small trend at low temperatures for increasing t (Fig. 4a). As before, no change is recorded in the X-ray peak width and lattice parameters (not shown). By comparison with the cooling rate r (Fig. 2 and Table I), the change in the low temperature resistivity is much smaller when varying the annealing time t at $T = 200^{\circ}C$: at the lowest measured temperature, the relative change in ρ as a etry changes less than 1% (Table I). The corresponding (Fig. 4) is $\rho(6days)/\rho_A \sim 1.5$ with larger composition possibly intrinsic semiconductor state in TiSe₂, which is of stoichiometry variations, while unclear, appears to be

minimal compared to the disorder and WL effects.

Post synthesis annealing temperature T

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Anneal t	Polycrystal	Single Crystal
(days)	Se	Se
As Grown	2.00 ± 0.02	2.013 ± 0.02
2	2.03 ± 0.02	
3	2.03 ± 0.03	2.017 ± 0.02
4	2.01 ± 0.02	
5	1.98 ± 0.02	
6	2.00 ± 0.03	2.012 ± 0.02

TABLE II. Se normalized to 1 Ti in $TiSe_2$ samples annealed at 200°C in one day increments up to six days corresponding to Fig. 4.

In summary, our results on polycrystalline TiSe₂ are

CONCLUSIONS

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