



# CHORUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

## Remarkable thermoelectric performance in BaPdS<sub>2</sub> via pudding-mold band structure, band convergence, and ultralow lattice thermal conductivity

Eric B. Isaacs and Chris Wolverton

Phys. Rev. Materials **3**, 015403 — Published 28 January 2019

DOI: [10.1103/PhysRevMaterials.3.015403](https://doi.org/10.1103/PhysRevMaterials.3.015403)

# Remarkable thermoelectric performance in BaPdS<sub>2</sub> via pudding-mold band structure, band convergence, and ultralow lattice thermal conductivity

Eric B. Isaacs and Chris Wolverton\*

*Department of Materials Science and Engineering,  
Northwestern University, Evanston, Illinois 60208, USA*

(Dated: January 8, 2019)

Efficient thermoelectric materials require a rare and contraindicated combination of materials properties: large electrical conductivity, large Seebeck coefficient, and low thermal conductivity. One strategy to achieve the first two properties is via low-energy electronic bands containing both flat and dispersive parts in different regions of crystal momentum space, known as a pudding-mold band structure. Here, we illustrate that BaPdS<sub>2</sub> successfully achieves the pudding-mold band structure for the valence band, contributing to a large *p*-type thermoelectric power factor, due to its anisotropic crystal structure containing zig-zag chains of edge-sharing square planar PdS<sub>4</sub> units; large power factor is achieved for *n*-type doping as well due to band convergence. In addition, BaPdS<sub>2</sub> exhibits ultralow lattice thermal conductivity, and thus also achieves the third property, due to extremely soft and anharmonic interactions in its transverse acoustic phonon branch. We predict a remarkably large thermoelectric figure of merit, with peak values between 2 and 3 for two of the three crystallographic directions, suggesting BaPdS<sub>2</sub> warrants experimental investigation.

## I. INTRODUCTION

The figure of merit for a thermoelectric material, which can convert a temperature gradient into electrical current, is

$$ZT = \frac{S^2 \sigma}{\kappa_{el} + \kappa_L} T, \quad (1)$$

where  $S$ ,  $\sigma$ ,  $T$ ,  $\kappa_{el}$ , and  $\kappa_L$  are the Seebeck coefficient, electrical conductivity, temperature, electrical contribution to thermal conductivity, and lattice contribution to thermal conductivity, respectively.<sup>1,2</sup> One strategy to achieve a large  $ZT$  is via engineering the electronic band structure, i.e., the electron energy  $\varepsilon$  as a function of crystal momentum  $\mathbf{k}$ .<sup>3</sup> In particular, a band with both flat (small  $\nabla_{\mathbf{k}}\varepsilon$ ) and dispersive (large  $\nabla_{\mathbf{k}}\varepsilon$ ) parts along a direction in  $\mathbf{k}$ -space has been shown to enhance  $\sigma S^2$  if the electron chemical potential  $\mu$  lies at an energy separating the two regions.<sup>4</sup> In such a “pudding-mold” band structure, the band velocity difference leads to large  $S$ , and large  $\sigma$  is achieved due to the dispersive part of the band and a large Fermi surface.<sup>4</sup> More generally, one can consider a broader definition of a pudding-mold band structure in which the flat and dispersive regions can along different directions in  $\mathbf{k}$ -space, which naturally occurs for low-dimensional bulk and nanostructured crystals.<sup>5–7</sup> The pudding-mold band structure has been used to explain the promising thermoelectric performance of Na<sub>*x*</sub>CoO<sub>2</sub>,<sup>4</sup> SnSe,<sup>8–10</sup> and recently-proposed Fe-based Heusler compounds<sup>11</sup> among others.

Recently, the promising thermoelectric performance of several Pd oxide compounds has been attributed in part to the pudding-mold band structure. The layered compound PbPdO<sub>2</sub>, for example, exhibits a large Seebeck coefficient of 175  $\mu\text{V}/\text{K}$  at 600 K when hole doped.<sup>12</sup> Based on first-principles calculations, hole-doped Bi<sub>2</sub>PdO<sub>4</sub> was predicted to exhibit high power factor ( $\sigma S^2$ ) in addition to low  $\kappa_L$ .<sup>13</sup> In both of these compounds, achieving the

pudding-mold band structure appears to be connected to the presence of square planar Pd<sup>2+</sup>. Based on this observation, we previously developed an inverse band structure design approach based on a materials database screening to search for other square planar compounds achieving the pudding-mold band structure.<sup>14</sup> Although several chemistries were considered, we focused on oxide materials such as Ba<sub>2</sub>PdO<sub>3</sub> and La<sub>4</sub>PdO<sub>7</sub> in order to validate the approach.

The thermoelectric efficiency of oxides is often limited by (1) large  $\kappa_L$ , stemming from the low atomic mass of oxygen and rigid chemical bonding, and (2) low carrier mobility.<sup>15,16</sup> Therefore, in this work, we turn our attention to *chalcogenide* materials. Chalcogenides comprise many of the most-investigated and highest-performance thermoelectrics, such as PbTe, Bi<sub>2</sub>Te<sub>3</sub>, SnSe, Cu<sub>2</sub>(S/Se), and La<sub>3–*x*</sub>Te<sub>4</sub>.<sup>17–19</sup> We investigate a chalcogenide material based on square planar Pd, namely BaPdS<sub>2</sub>, with the aim of achieving low  $\kappa_L$  in addition to pudding-mold band structure. BaPdS<sub>2</sub> is the only such square planar chalcogenide identified by our previous inverse band structure design approach<sup>14</sup> other than metallic EuPd<sub>3</sub>S<sub>4</sub><sup>20</sup> and the binaries PdS<sub>2</sub><sup>21</sup> and PdSe<sub>2</sub>.<sup>22</sup> Although BaPdS<sub>2</sub> was first synthesized in 1986,<sup>23</sup> it has not been explored as a thermoelectric material to the best of our knowledge. The presence of heavy Ba, in addition to the lack of oxygen, leads to the potential for low  $\kappa_L$ . Using electronic structure and transport calculations based on the density functional theory, we find that BaPdS<sub>2</sub> exhibits (1) large *p*-type power factor stemming from a pudding-mold valence band structure, (2) large *n*-type power factor due to band convergence in the conduction band, (3) ultralow  $\kappa_L$  due to extremely soft and anharmonic bonding, and (4) highly-anisotropic electronic and thermal transport behavior. We find remarkable predicted thermoelectric performance, with  $ZT$  larger than 2 in certain crystallographic directions at high temperature, assuming optimal doping and using a rea-

sonable value of electronic relaxation time  $\tau$  (5 fs).<sup>11,24–28</sup> Based on these results, we suggest BaPdS<sub>2</sub> warrants experimental investigation.

## II. COMPUTATIONAL DETAILS

We perform density functional theory<sup>29,30</sup> calculations using the generalized gradient approximation<sup>31</sup> and the projector augmented wave method (Ba,sv, Pd, and S potentials)<sup>32,33</sup> in the Vienna *ab initio* simulation package (VASP).<sup>34–37</sup> A plane wave basis with 500 eV kinetic energy cutoff and uniform  $k$ -meshes with  $\geq 700$   $k$ -points per  $\text{\AA}^{-3}$  are employed. The ionic forces and total energy are converged to  $10^{-3}$  eV/ $\text{\AA}$  and  $10^{-6}$  eV, respectively. High-symmetry  $k$ - and  $q$ -point paths are based on the conventions of Setyawan *et al.*<sup>38</sup>

Semiclassical electronic transport calculations within the constant relaxation time ( $\tau$ ) approximation are performed using BOLTZTRAP<sup>39</sup> with  $k$ -point density of  $3,500/\text{\AA}^{-3}$ . Although the use of an isotropic relaxation time is an approximation, previous studies suggest only minor anisotropy in  $\tau$  even in highly anisotropic crystal structures.<sup>39</sup> Phonon calculations are performed using the direct approach in PHONOPY<sup>40</sup> with a 192-atom supercell chosen to be approximately cubic based on the algorithm of Erhart *et al.*<sup>41</sup> as implemented in ASE.<sup>42</sup> The mode Grüneisen parameter is computed using a  $\pm 0.3\%$  volume difference.  $\kappa_L$  is computed via the Debye-Callaway model<sup>43–45</sup> and compared to that of the minimum thermal conductivity model of Cahill *et al.*<sup>46</sup>  $\kappa_{el}$  is computed via the Wiedemann-Franz law. Additional details on the electronic and thermal transport calculations are included in the Supplemental Material.<sup>47</sup>

## III. RESULTS AND DISCUSSION

BaPdS<sub>2</sub>, shown in Fig. 1(a) and Fig. 1(b), has a C-centered orthorhombic unit cell with space group  $Cmcm$ . Isostructural to BaNiO<sub>2</sub>, it consists of 1D chains of edge-sharing PdS<sub>4</sub> square planar units (corresponding to a PdS<sub>2</sub> composition) connected in a zig-zag pattern. The computed lattice parameters, atomic positions, and formation energy are in excellent agreement with experimental values, as discussed in the Supplemental Material. The electronic band structure of BaPdS<sub>2</sub> is shown in Fig. 1(c). The valence band, which consists primarily of Pd  $d$  states, exhibits a pudding-mold band structure. It is dispersive in the direction along the 1D chains (e.g.,  $\Gamma$ -Z) and flat along other directions (e.g.,  $\Gamma$ -X and  $\Gamma$ -Y). In other words, the pudding-mold band structure is closely related to the highly-anisotropic crystal structure of BaPdS<sub>2</sub>. We do not find a pudding-mold band structure for the conduction band, which consists of a hybridization of Pd  $d$  and S  $p$  orbitals. We note that an additional conduction band minimum between S and R is close in energy ( $\sim 130$  meV) to the band edge at  $\Gamma$ .

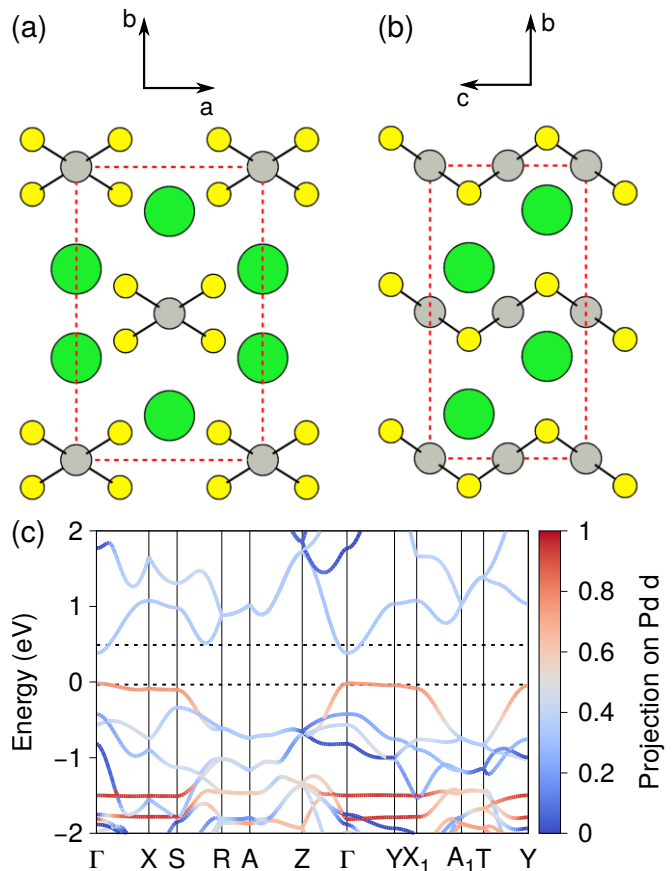


FIG. 1. Orthographic projection of the crystal structure of BaPdS<sub>2</sub> along the (a)  $c$  and (b)  $a$  axes of the conventional unit cell (red dashed lines). The green, gray, and yellow circles represent Ba, Pd, and S atoms, respectively. Black lines indicate the Pd-S bonds of the square planar units. (c) The electronic bands of BaPdS<sub>2</sub> colored by the Pd  $d$  character of the electronic states. The zero of energy is set to the valence band maximum, and the black horizontal dotted lines indicate the  $\pm 10^{20}$  cm<sup>-3</sup> doping levels.

The band structure of BaPdS<sub>2</sub> leads to notable electronic transport properties. Figure 2(a) illustrates the behavior of the average (over Cartesian directions) of  $\sigma S^2/\tau$  as a function of carrier concentration for several temperatures. Results for BaPdS<sub>2</sub> are shown in comparison to those of SnSe, a high-performing thermoelectric material.<sup>8,48–50</sup> BaPdS<sub>2</sub> achieves comparable  $p$ -type power factor behavior to SnSe, albeit at larger (though still reasonable<sup>17</sup>) doping values. For example, at 700 K, the maximum predicted  $p$ -type  $\sigma S^2/\tau$  is  $\sim 3 \times 10^{11}$  W/(m·K<sup>2</sup>·s) for both BaPdS<sub>2</sub> (for  $7 \times 10^{20}$  holes/cm<sup>3</sup>) and SnSe (for  $3 \times 10^{20}$  holes/cm<sup>3</sup>). We also note that the  $p$ -type power factor behavior exhibits significantly less temperature dependence for BaPdS<sub>2</sub> than for SnSe. Although the conduction band does not exhibit pudding-mold qualities, we find even larger  $\sigma S^2/\tau$  for  $n$ -type doping, though smaller than the corresponding SnSe behavior. For example, the peak  $n$ -type  $\sigma S^2/\tau$

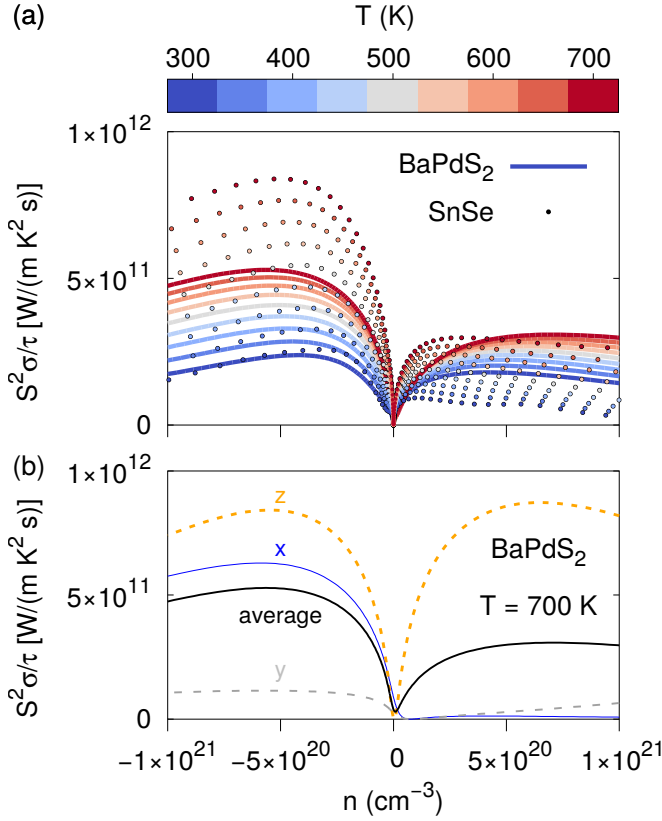


FIG. 2. (a) Power factor divided by electronic relaxation time as a function of carrier concentration for BaPdS<sub>2</sub> (lines) and SnSe (circles), averaged over the  $x$ ,  $y$ , and  $z$  directions, for several values of  $T$ . (b)  $T = 700$  K power factor divided by electronic relaxation time versus carrier concentration in each direction, in addition to the average, for BaPdS<sub>2</sub>.

for BaPdS<sub>2</sub> at 700 K is  $5.3 \times 10^{11}$  W/(m·K<sup>2</sup>·s), occurring at a doping value of  $\sim 5.5 \times 10^{20}$  electrons/cm<sup>3</sup>; SnSe exhibits a significantly larger peak  $n$ -type  $\sigma S^2/\tau$  of  $8.4 \times 10^{11}$  W/(m·K<sup>2</sup>·s) at around the same doping magnitude. The appreciable  $n$ -type power factor behavior for BaPdS<sub>2</sub> likely stems from the band convergence,<sup>51–55</sup> i.e., the small energy separation between the conduction band minima (1) at  $\Gamma$  and (2) between S and R. Although the conduction band minimum at  $\Gamma$  is singly degenerate, the band minimum between S and R has a higher degeneracy of 4, contributing to a large Seebeck coefficient.

Due to the structural anisotropy, the power factor behavior for BaPdS<sub>2</sub> is highly anisotropic. We illustrate the power factor behavior at 700 K, for example, in Fig. 2(b) for each direction. For hole doping,  $\sigma S^2/\tau$  is only appreciable along the  $z$  direction (i.e., along the 1D chains), which is consistent with the pudding-mold band structure. In contrast, for electron doping, large  $\sigma S^2/\tau$  is observed in both the  $z$  and  $x$  directions. A rationalization for the lower  $n$ -type power factor behavior in the  $y$  direction is discussed in the Supporting Information.

Having established the promising predicted electronic

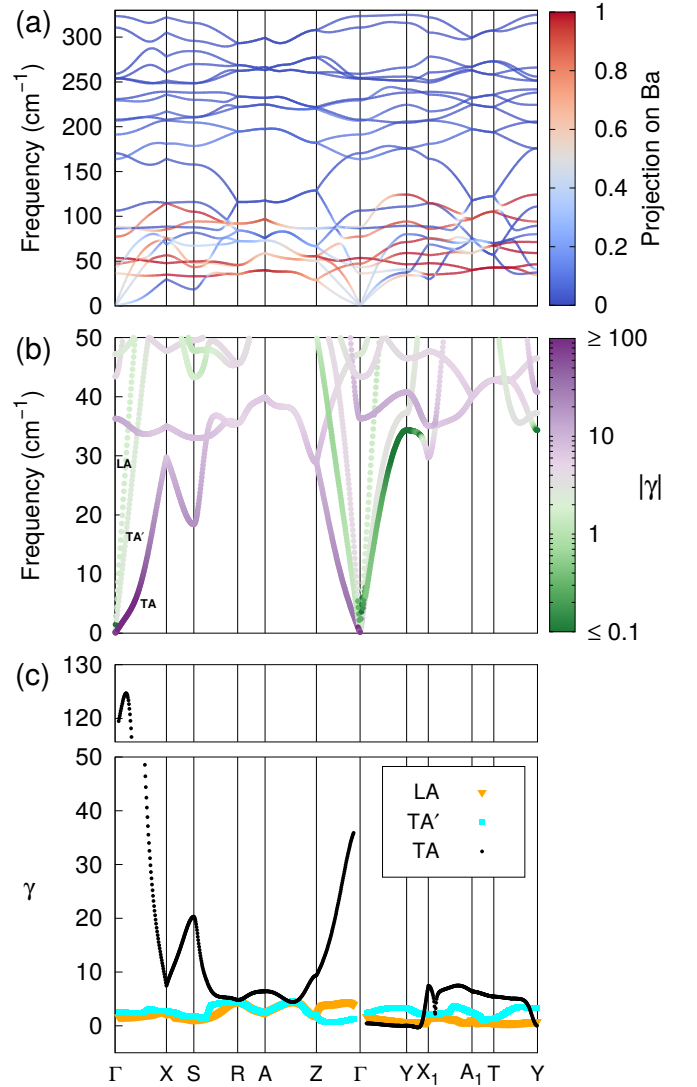


FIG. 3. (a) BaPdS<sub>2</sub> phonon bands colored by Ba contribution to the phonon eigenvector. (b) The low-frequency portion of the phonon dispersion with bands colored by the magnitude of the mode Grüneisen parameter. (c) Dispersion of the mode Grüneisen parameter  $\gamma_{\mathbf{q}\nu}$  for the acoustic branches. Due to the large  $\gamma_{\mathbf{q}\nu}$  values for the TA branch along  $\Gamma$ -X, we use a broken  $\gamma_{\mathbf{q}\nu}$  axis for clarity.

Branch	$v_x$	$v_y$	$v_z$	$\Theta_{\Gamma X}$	$\Theta_{\Gamma Y}$	$\Theta_{\Gamma Z}$
TA	0.5	1.9	0.8	43	49	42
TA'	2.4	2.4	1.9	82	54	72
LA	4.1	3.7	3.6	99	115	72

TABLE I. Group velocity  $v$  at  $\Gamma$  (in km/s) and Debye temperature  $\Theta$  (in K) of BaPdS<sub>2</sub> for each acoustic branch and direction.

transport properties of BaPdS<sub>2</sub>, we now turn our attention to thermal transport. The phonon dispersion of BaPdS<sub>2</sub> is shown in Fig. 3(a). Low-frequency optical modes, primarily involving vibration of the heavy Ba atoms, are observed in the  $\sim 30$ –110 cm<sup>-1</sup> frequency range. For each acoustic branch, we compute the group velocity in the long-wavelength limit ( $v_x$ ,  $v_y$ , and  $v_z$ ) and Debye temperature ( $\Theta_{\Gamma X}$ ,  $\Theta_{\Gamma Y}$ , and  $\Theta_{\Gamma Z}$ ) for each direction, as shown in Table I. Here, we define  $\Theta$  as the acoustic phonon frequency at the zone boundary, and the acoustic branch definitions are given in the Supplemental Material. BaPdS<sub>2</sub> is a very soft material, exhibiting low group velocities (0.5–4.1 km/s) and Debye temperatures (42–115 K). For comparison,  $v$  of 1.0–2.9 and  $\Theta$  of 19–72 are found in SnSe.<sup>8</sup> The transverse acoustic (TA) branch of BaPdS<sub>2</sub> is particularly soft, with  $v = 0.5$  km/s (0.8 km/s) and  $\Theta = 43$  K (42 K) along the  $x$  ( $z$ ) direction. Animations of the TA phonons along X (featuring rigid PdS<sub>2</sub> chains sliding in the  $z$  direction with respect to the others) and along Z (featuring oscillations of the individual PdS<sub>2</sub> chains in the  $x$  direction) are included in the Supplemental Material. We note that the behavior of the TA branch is significantly harder along the  $y$  direction in terms of the group velocity (1.9 km/s), though not the Debye temperature (49 K). Corresponding to PdS<sub>2</sub> chains sliding in the  $z$  direction, the TA branch along Y corresponds to a similar atomic motion to that along X; the significantly larger inter-chain distance in the  $y$  direction (10.8 Å) compared to that in  $x$  direction (6.9 Å) may be responsible for the very different phonon properties.

Branch	BaPdS <sub>2</sub>				SnSe		
	$\gamma_{all}$	$\gamma_{\Gamma X}$	$\gamma_{\Gamma Y}$	$\gamma_{\Gamma Z}$	$\gamma_{\Gamma X}$	$\gamma_{\Gamma Y}$	$\gamma_{\Gamma Z}$
TA	25.5	80.3	0.3	22.8	5.1	2.7	1.0
TA'	2.8	2.6	3.1	1.0	1.1	2.4	2.7
LA	2.3	2.1	1.2	4.0	5.9	1.3	3.3
Avg.	10.2	28.3	1.5	9.3	4.1	2.1	2.3

TABLE II. Averages (computed as  $\sqrt{\langle \gamma^2 \rangle}$ ) of mode Grüneisen parameter  $\gamma_{\mathbf{q}\nu}$  for BaPdS<sub>2</sub> and SnSe<sup>8</sup> for different acoustic branches and directions. For BaPdS<sub>2</sub>, the average over the entire high-symmetry path in the Brillouin zone is also given as  $\gamma_{all}$ .

To acquire a baseline assessment for the overall magnitude of the anharmonicity, i.e., the phonon-phonon scattering strength, we compute the mode Grüneisen parameter

$$\gamma_{\mathbf{q}\nu} = -\frac{\partial \omega_{\mathbf{q}\nu} / \omega_{\mathbf{q}\nu}}{\partial V / V}, \quad (2)$$

where  $V$  is the volume and  $\omega_{\mathbf{q}\nu}$  is the phonon frequency for crystal momentum  $\mathbf{q}$  and branch  $\nu$ . The low-frequency phonons of BaPdS<sub>2</sub>, colored by  $|\gamma|$ , are shown in Fig. 3(b), whereas the full dispersion of  $\gamma_{\mathbf{q}\nu}$  for the acoustic branches is included in Fig. 3(c). The most prominent feature is that BaPdS<sub>2</sub> has gigantic  $\gamma_{\mathbf{q}\nu}$  for

the TA branch along  $\Gamma$ -X (values up to  $\sim 125$ ) and  $\Gamma$ -Z (values up to  $\sim 35$ ), which indicates extremely anharmonic interactions for this acoustic branch in BaPdS<sub>2</sub>. We perform tests to confirm the large computed  $\gamma_{\mathbf{q}\nu}$  are not a numerical artifact, as discussed in the Supplemental Material. Averages of the mode Grüneisen parameter for each acoustic branch of BaPdS<sub>2</sub>, averaged over  $\Gamma$ -X,  $\Gamma$ -Y,  $\Gamma$ -Z, and the full high-symmetry path in the Brillouin zone, are shown in Table II, in which we also compare to literature results for SnSe.<sup>8</sup> The average  $\gamma_{\mathbf{q}\nu}$  for the TA branch of BaPdS<sub>2</sub> is 80.3 in the  $x$  direction and 22.8 in the  $z$  direction, compared to much smaller (less anharmonic) values of 5.1 and 1.0 for SnSe in the corresponding directions. We note that a key exception to the large anharmonicity of the TA branch of BaPdS<sub>2</sub> is along the  $y$  direction, for which the average  $\gamma_{\mathbf{q}\nu}$  is only 0.3, as compared to 2.7 for SnSe. In other words, the TA branch of BaPdS<sub>2</sub> is not only harder in the  $y$  direction than the other directions, but also much less anharmonic. Although the most remarkable finding is the extremely large  $\gamma_{\mathbf{q}\nu}$  for the TA branch, we also find significant anharmonicity for other phonon branches. For example, the TA' (LA) branch exhibits appreciable  $\gamma_{\mathbf{q}\nu}$  of 1.0–3.1 (1.2–4.0), as compared to 1.1–2.7 (1.3–5.9) for SnSe. Large anharmonicity is also found in the lowest-frequency optical branch, with mode Grüneisen parameter values as large as  $\sim 12$ , as shown in Fig. 3(b). This branch corresponds to oscillation of the Ba sublattice with respect to the PdS<sub>2</sub> chains in the  $z$  direction.

We employ the Debye-Callaway model, an approximate model taking into account normal and Umklapp acoustic phonon-phonon scattering via the mode Grüneisen parameter, as a means to estimate the lattice thermal conductivity of BaPdS<sub>2</sub>.<sup>43–45</sup> We note that the low-frequency optical phonons with large  $\gamma_{\mathbf{q}\nu}$ , not taken into account in the Debye-Callaway model, may further lower  $\kappa_L$  compared to the values reported here. The contributions of each acoustic phonon branch to  $\kappa_L$  as a function of  $T$  in the  $x$ ,  $y$ , and  $z$  directions are shown in Fig. 4(a)–4(c), respectively. BaPdS<sub>2</sub> exhibits ultralow lattice thermal conductivity ( $\kappa_L$  less than 1 W/m-K) in the  $x$  and  $z$  directions. In these directions, the predicted  $\kappa_L$  are of the same magnitude as that of SnSe along its smallest- $\kappa_L$  direction (values of  $\sim 0.25$ –0.7 W/m-K).<sup>8</sup> For BaPdS<sub>2</sub>, the main contributor to  $\kappa_L$  in the  $x$  direction is the LA branch, which exhibits the largest group velocity; the main contributor in the  $z$  direction is the TA' branch, which is the least anharmonic. Like the power factor, the thermal transport is highly anisotropic: in the  $y$  direction,  $\kappa_L$  is much larger (values of 4–8 W/m-K) than in the other directions, stemming primarily from the small  $\gamma_{\Gamma Y}$  of 0.3 for the TA branch, as discussed above. In the  $x$  and  $z$  (but not  $y$ ) directions, BaPdS<sub>2</sub> achieves the minimum possible  $\kappa_L$ , shown as red lines in Fig. 4(a)–4(c), estimated from the model of Cahill *et al.* Overall, BaPdS<sub>2</sub> is predicted to exhibit remarkably poor thermal transport in directions other than  $y$ .

As illustrated in Fig. 4(d)–4(f), the combination of

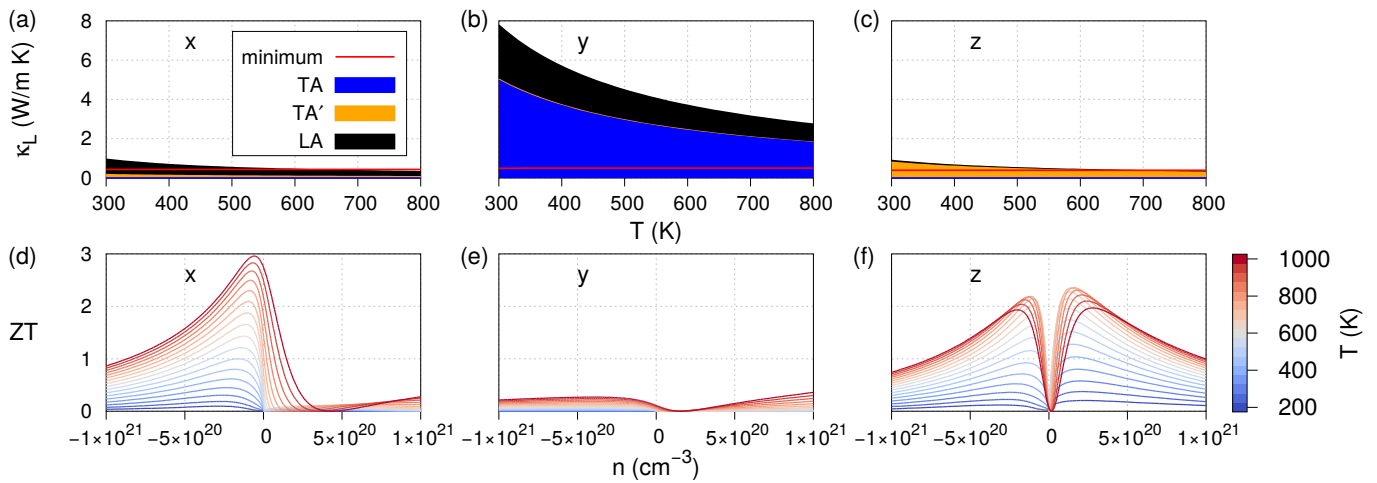


FIG. 4. (a)–(c) Stacked plots of BaPdS<sub>2</sub> acoustic branch contributions to  $\kappa_L$  versus  $T$  computed via the Debye-Callaway model for the (a)  $x$ , (b)  $y$ , and (c)  $z$  directions, respectively. Red lines indicate the minimum  $\kappa_L$  of the Cahill model. (d)–(f)  $ZT$  as a function of carrier concentration for different  $T$  in the (d)  $x$ , (e)  $y$ , and (f)  $z$  directions, respectively. Values are shown for  $\tau = 5$  fs.

large power factor and low  $\kappa_L$  leads to impressive figure of merit for BaPdS<sub>2</sub>, particularly in the  $x$  direction for electron doping and in the  $z$  direction for both electron and hole doping. At high temperature, we find peak  $ZT$  values of nearly 3 in the  $x$  direction (for  $6.0 \times 10^{19}$  electrons/cm<sup>3</sup>) and values of nearly 2.3 in the  $z$  direction (for  $1.1 \times 10^{20}$  electrons/cm<sup>3</sup> or  $1.6 \times 10^{20}$  holes/cm<sup>3</sup>). In contrast, much smaller peak  $ZT$  ( $< 1$ ) is observed for BaPdS<sub>2</sub> in the  $y$  direction, due to the smaller power factor and larger  $\kappa_L$ . In order to obtain the predicted  $ZT$  values discussed above, we have chosen an electronic relaxation time ( $\tau$ ) of 5 fs. While an *ab initio* estimation of  $\tau$  (e.g., from electron-phonon scattering calculations) will be important future work, we note that 5 fs is reasonable in terms of the order of magnitude<sup>56,57</sup> and can be considered a conservative estimate based on previous works.<sup>11,24–28</sup> Even with a less optimistic guess for the electronic scattering time, we still find quite large values for the thermoelectric figure of merit, albeit at larger doping. For example, assuming  $\tau = 1$  fs, a peak  $ZT$  of 1.3 is achieved in the  $x$  direction for  $2.0 \times 10^{20}$  electrons/cm<sup>3</sup>. Therefore, even with the uncertainty in  $\tau$ , our results strongly suggest BaPdS<sub>2</sub> merits experimental investigation.

#### IV. CONCLUSIONS

BaPdS<sub>2</sub>, previously identified by our inverse band structure design approach, is a square planar sulfide ma-

terial with remarkable, anisotropic thermoelectric properties. Due to a pudding-mold valence band structure and multiple conduction bands, BaPdS<sub>2</sub> exhibits impressive power factor behavior in two of the crystallographic directions. With heavy Ba atoms and extremely soft and anharmonic bonding, BaPdS<sub>2</sub> achieves ultralow lattice thermal conductivity in the  $x$  and  $z$  directions. We predict peak  $ZT$  values of 2–3 in the  $x$  direction ( $n$ -type) and the  $z$  direction ( $n$ - and  $p$ -type) for an electronic relaxation time of 5 fs. Our results strongly suggest BaPdS<sub>2</sub> warrants experimental investigation for its remarkable electronic, thermal transport, and thermoelectric properties.

#### ACKNOWLEDGMENTS

We acknowledge support from the U.S. Department of Energy under Contract DE-SC0014520. Computational resources were provided by the National Energy Research Scientific Computing Center (U.S. Department of Energy Contract DE-AC02-05CH11231) and the Extreme Science and Engineering Discovery Environment (National Science Foundation Contract ACI-1548562).

\* c-wolverton@northwestern.edu

<sup>1</sup> H. J. Goldsmid, *Introduction to Thermoelectricity*, Springer Series in Materials Science, Vol. 121 (Springer,

- Berlin, Heidelberg, 2010).
- <sup>2</sup> A. Zevalkink, D. M. Smiadak, J. L. Blackburn, A. J. Ferguson, M. L. Chabinyk, O. Delaire, J. Wang, K. Kovnir, J. Martin, L. T. Schelhas, T. D. Sparks, S. D. Kang, M. T. Dylla, G. J. Snyder, B. R. Ortiz, and E. S. Toberer, *Appl. Phys. Rev.* **5**, 021303 (2018).
  - <sup>3</sup> Y. Pei, H. Wang, and G. J. Snyder, *Adv. Mater.* **24**, 6125 (2012).
  - <sup>4</sup> K. Kuroki and R. Arita, *J. Phys. Soc. Jpn.* **76**, 083707 (2007).
  - <sup>5</sup> L. D. Hicks and M. S. Dresselhaus, *Phys. Rev. B* **47**, 12727 (1993).
  - <sup>6</sup> L. D. Hicks and M. S. Dresselhaus, *Phys. Rev. B* **47**, 16631 (1993).
  - <sup>7</sup> H. Usui and K. Kuroki, *J. Appl. Phys.* **121**, 165101 (2017).
  - <sup>8</sup> L.-D. Zhao, S.-H. Lo, Y. Zhang, H. Sun, G. Tan, C. Uher, C. Wolverton, V. P. Dravid, and M. G. Kanatzidis, *Nature* **508**, 373 (2014).
  - <sup>9</sup> K. Kutorasinski, B. Wiendlocha, S. Kaprzyk, and J. Tobola, *Phys. Rev. B* **91**, 205201 (2015).
  - <sup>10</sup> Z. Wang, C. Fan, Z. Shen, C. Hua, Q. Hu, F. Sheng, Y. Lu, H. Fang, Z. Qiu, J. Lu, Z. Liu, W. Liu, Y. Huang, Z.-A. Xu, D. W. Shen, and Y. Zheng, *Nat. Commun.* **9**, 47 (2018).
  - <sup>11</sup> D. I. Birc, G. Hautier, D. Waroquier, G.-M. Rignanese, and P. Ghosez, *Phys. Rev. Lett.* **114**, 136601 (2015).
  - <sup>12</sup> L. K. Lamontagne, G. Laurita, M. W. Gaultois, M. Knight, L. Ghadbeigi, T. D. Sparks, M. E. Gruner, R. Pentcheva, C. M. Brown, and R. Seshadri, *Chem. Mater.* **28**, 3367 (2016).
  - <sup>13</sup> J. He, S. Hao, Y. Xia, S. S. Naghavi, V. Ozoliņš, and C. Wolverton, *Chem. Mater.* **29**, 2529 (2017).
  - <sup>14</sup> E. B. Isaacs and C. Wolverton, *Chem. Mater.* **30**, 1540 (2018).
  - <sup>15</sup> J. He, Y. Liu, and R. Funahashi, *J. Mater. Res.* **26**, 1762 (2011).
  - <sup>16</sup> K. F. Garrity, *Phys. Rev. B* **94**, 045122 (2016).
  - <sup>17</sup> G. J. Snyder and E. S. Toberer, *Nat. Mater.* **7**, 105 (2008).
  - <sup>18</sup> X. Zhang and L.-D. Zhao, *J. Materiomics* **1**, 92 (2015).
  - <sup>19</sup> G. Tan, L.-D. Zhao, and M. G. Kanatzidis, *Chem. Rev.* **116**, 12123 (2016).
  - <sup>20</sup> K. L. Yao, L. Q. Wang, and Z. L. Liu, *Solid State Comm.* **130**, 741 (2004).
  - <sup>21</sup> R. Bhatt, S. Bhattacharya, R. Basu, A. Singh, U. Deshpande, C. Surger, S. Basu, D. K. Aswal, and S. K. Gupta, *Thin Solid Films* **539**, 41 (2013).
  - <sup>22</sup> J. Sun, H. Shi, T. Siegrist, and D. J. Singh, *Appl. Phys. Lett.* **107**, 153902 (2015).
  - <sup>23</sup> J. Huster and W. Bronger, *J. Less Common Met.* **119**, 159 (1986).
  - <sup>24</sup> G. K. H. Madsen, *J. Am. Chem. Soc.* **128**, 12140 (2006).
  - <sup>25</sup> A. N. Gandi and U. Schwingenschlögl, *Chem. Mater.* **26**, 6628 (2014).
  - <sup>26</sup> F. Q. Wang, S. Zhang, J. Yu, and Q. Wang, *Nanoscale* **7**, 15962 (2015).
  - <sup>27</sup> S. Hao, F. Shi, V. P. Dravid, M. G. Kanatzidis, and C. Wolverton, *Chem. Mater.* **28**, 3218 (2016).
  - <sup>28</sup> K. Pal, J. He, and C. Wolverton, *Chem. Mater.* **30**, 7760 (2018).
  - <sup>29</sup> P. Hohenberg and W. Kohn, *Phys. Rev.* **136**, B864 (1964).
  - <sup>30</sup> W. Kohn and L. J. Sham, *Phys. Rev.* **140**, A1133 (1965).
  - <sup>31</sup> J. P. Perdew, K. Burke, and M. Ernzerhof, *Phys. Rev. Lett.* **77**, 3865 (1996).
  - <sup>32</sup> P. E. Blöchl, *Phys. Rev. B* **50**, 17953 (1994).
  - <sup>33</sup> G. Kresse and D. Joubert, *Phys. Rev. B* **59**, 1758 (1999).
  - <sup>34</sup> G. Kresse and J. Hafner, *Phys. Rev. B* **49**, 14251 (1994).
  - <sup>35</sup> G. Kresse and J. Hafner, *Phys. Rev. B* **47**, 558 (1993).
  - <sup>36</sup> G. Kresse and J. Furthmüller, *Phys. Rev. B* **54**, 11169 (1996).
  - <sup>37</sup> G. Kresse and J. Furthmüller, *Comput. Mater. Sci.* **6**, 15 (1996).
  - <sup>38</sup> W. Setyawan and S. Curtarolo, *Comput. Mater. Sci.* **49**, 299 (2010).
  - <sup>39</sup> G. K. H. Madsen and D. J. Singh, *Comput. Phys. Commun.* **175**, 67 (2006).
  - <sup>40</sup> A. Togo, F. Oba, and I. Tanaka, *Phys. Rev. B* **78**, 134106 (2008).
  - <sup>41</sup> P. Erhart, B. Sadigh, A. Schleife, and D. Åberg, *Phys. Rev. B* **91**, 165206 (2015).
  - <sup>42</sup> A. H. Larsen, J. J. Mortensen, J. Blomqvist, I. E. Castelli, R. Christensen, M. Dułak, J. Friis, M. N. Groves, B. Hammer, C. Hargus, E. D. Hermes, P. C. Jennings, P. B. Jensen, J. Kermode, J. R. Kitchin, E. L. Kolsbjerg, J. Kubal, K. Kaasbjerg, S. Lysgaard, J. B. Maronsson, T. Maxson, T. Olsen, L. Pastewka, A. Peterson, C. Rostgaard, J. Schiøtz, O. Schütt, M. Strange, K. S. Thygesen, T. Vegge, L. Vilhelmsen, M. Walter, Z. Zeng, and K. W. Jacobsen, *J. Phys. Condens. Matter* **29**, 273002 (2017).
  - <sup>43</sup> J. Callaway, *Phys. Rev.* **113**, 1046 (1959).
  - <sup>44</sup> D. T. Morelli, J. P. Heremans, and G. A. Slack, *Phys. Rev. B* **66**, 195304 (2002).
  - <sup>45</sup> Y. Zhang, *J. Materiomics* **2**, 237 (2016).
  - <sup>46</sup> D. G. Cahill, S. K. Watson, and R. O. Pohl, *Phys. Rev. B* **46**, 6131 (1992).
  - <sup>47</sup> See Supplemental Material for additional details on electronic and thermal transport methods, phonon and anharmonicity calculation convergence, animations of several low-frequency phonon modes, comparison to experimental crystal structure and formation energy, and the electronic band structure for a different high-symmetry crystal momentum path.
  - <sup>48</sup> L.-D. Zhao, G. Tan, S. Hao, J. He, Y. Pei, H. Chi, H. Wang, S. Gong, H. Xu, V. P. Dravid, C. Uher, G. J. Snyder, C. Wolverton, and M. G. Kanatzidis, *Science* **351**, 141 (2015).
  - <sup>49</sup> A. T. Duong, V. Q. Nguyen, G. Duvjir, V. T. Duong, S. Kwon, J. Y. Song, J. K. Lee, J. E. Lee, S. Park, T. Min, J. Lee, J. Kim, and S. Cho, *Nat. Commun.* **7**, 13713 (2016).
  - <sup>50</sup> C. Chang, M. Wu, D. He, Y. Pei, C.-F. Wu, X. Wu, H. Yu, F. Zhu, K. Wang, Y. Chen, L. Huang, J.-F. Li, J. He, and L.-D. Zhao, *Science* **360**, 778 (2018).
  - <sup>51</sup> Y. Pei, X. Shi, A. LaLonde, H. Wang, L. Chen, and G. J. Snyder, *Nature* **473**, 66 (2011).
  - <sup>52</sup> W. Liu, X. Tan, K. Yin, H. Liu, X. Tang, J. Shi, Q. Zhang, and C. Uher, *Phys. Rev. Lett.* **108**, 166601 (2012).
  - <sup>53</sup> Y. Tang, Z. M. Gibbs, L. A. Agapito, G. Li, H.-S. Kim, M. B. Nardelli, S. Curtarolo, and G. J. Snyder, *Nat. Mater.* **14**, 1223 (2015).
  - <sup>54</sup> W. G. Zeier, A. Zevalkink, Z. M. Gibbs, G. Hautier, M. G. Kanatzidis, and G. J. Snyder, *Angew. Chem. Int. Ed.* **55**, 6826 (2016).
  - <sup>55</sup> W. G. Zeier, *Curr. Opin. Green Sustain. Chem.* **4**, 23 (2017).
  - <sup>56</sup> N. W. Ashcroft and N. D. Mermin, *Solid State Physics* (Saunders College Publishing, 1976).
  - <sup>57</sup> P. Y. Yu and M. Cardona, *Fundamentals of Semiconductors: Physics and Material Properties* (Springer-Verlag, 1996).