



CHORUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

Shear Softening in a Metallic Glass: First-Principles Local-Stress Analysis

I. Lobzenko, Y. Shiihara, T. Iwashita, and T. Egami

Phys. Rev. Lett. **124**, 085503 — Published 27 February 2020

DOI: [10.1103/PhysRevLett.124.085503](https://doi.org/10.1103/PhysRevLett.124.085503)

Shear softening in a metallic glass: first principle local stress analysis

I. Lobzenko* and Y. Shiihara

*Toyota Technological Institute, Hisakata,
Tempaku-ku, Nagoya 468-8511, Japan*

T. Iwashita

Oita University, Dannoharu, Oita 870-1192, Japan

T. Egami

*University of Tennessee, Knoxville, TN 37996, USA and
Oak Ridge National Laboratory, Oak Ridge, TN 37831, USA*

(Dated: January 28, 2020)

Abstract

Metallic glasses deform elastically under stress. However, the atomic-level origin of elastic properties of metallic glasses remain unclear. In this paper using *ab initio* molecular dynamics simulations of the $\text{Cu}_{50}\text{Zr}_{50}$ metallic glass under shear strain, we show that the heterogeneous stress relaxation results in the increased charge transfer from Zr to Cu atoms, enhancing the softening of the shear modulus. Changes in compositional short-range order and atomic position shifts due to the non-affine deformation are discussed. It is shown that the Zr subsystem exhibits a stiff behavior, whereas the displacements of Cu atoms from their initial positions, induced by the strain, provide the stress drop and softening.

1 Amorphous materials formed by metal atoms, usually referred to as metallic glasses (MG)
2 or glassy metals, have gained significant attention after the procedure of mass production
3 of the bulk form of such materials had been developed in the 1990s [1, 2], made possible by
4 the discovery of glass-forming alloys requiring relatively low cooling rates of $<100 \text{ K}\cdot\text{s}^{-1}$ [3].
5 This enabled broad applications of MGs, for instance, for nanoimprinted technology, bio-
6 implants, and coating, to name a few examples [2, 4].

7 Along with the applications, metallic glasses have received intense scientific interest. One
8 of the striking features of MGs is shear modulus softening (SMS), namely the shear modulus
9 of MGs is significantly lower than that of crystalline counterparts [5]. This phenomenon
10 has been extensively studied theoretically with assumptions being supported by computer
11 simulations in a number of early [6, 7] and most recent [8–11] works, and the microscopic
12 origin has been attributed to non-affine heterogeneous atomic motions during deformation.
13 However, the precise atomic mechanisms of SMS remain elusive. Particularly little quantum
14 level understanding of SMS has been achieved [9].

15 In the case of crystalline materials, inelastic behavior is governed by the well-defined
16 lattice defects in the periodic structure. However, attempts of defining defects in glasses
17 face a formidable conceptual and practical barrier because of structural disorder. A number
18 of theories of deformation in MG were proposed, among which the most widely used relies
19 on the so-called shear transformation zone (STZ) [12–14]. It attributes the deformation
20 in amorphous material to the emergence and development of regions with higher mobility.
21 Despite advances in the STZ theory, a number of its key features remain veiled. For example,
22 many simulations and experiments show contradictory results for STZ sizes, from a few

23 atoms to several hundred, and also the precise physical picture of the STZ is missing [15–
24 18]. Furthermore, the particular mechanism of STZ emergence is yet to be understood.

25 Advances in modern computational methods allow one to obtain insights into the defor-
26 mation behavior of individual atoms. One of the most accurate atomistic modeling methods
27 is based on the density functional theory (DFT), which provides the evaluation of structural
28 parameters with accuracy up to 1% [19]. This approach is effective in defining the starting
29 point of shear transformation zone in metallic glass since relatively small structures (order
30 of 20 atoms) are involved in the STZ [20]. In order to examine the local stress state around
31 the STZ, which is highly heterogeneous at the microscopic scale, we use the atomic-level
32 stresses analysis [21–23], coupled with the DFT calculation [24].

33 In the current work, we study $\text{Cu}_{50}\text{Zr}_{50}$ glass under shear strain. The choice of the
34 system is dictated by the fact that a large amount of data are available for CuZr alloys
35 in both crystalline and glassy structures [24, 25]. It is known that the CuZr glass could
36 be obtained in a wide composition range. Therefore, the influence of the stoichiometry
37 presented in the current work may be verified by future experimental work.

38 This paper is organized as follows. At first, the details on the structures under study and
39 settings for the first principle calculations are given. Next, the main results are presented
40 and discussed with emphasis on the fundamental properties of the structure leading to shear
41 modulus softening.

42 Initially, a random atomic configuration of 96 atoms was prepared at a density of
43 57.1 nm^{-3} [26]. The first principle molecular dynamics (FPMD) simulation at 3000 K
44 for 2 ps was performed under the NVE ensemble with periodic boundary conditions to
45 thermalize the original structure, thus obtaining an equilibrium liquid structure. The time
46 step of the simulations was 2 fs. The system was quenched to a supercooled state at 1200 K,
47 which led to the relatively stable glass structure, then FPMD was performed for 1 ps. Next
48 the system was gradually cooled down to a glass state at 700 K with a cooling rate of
49 0.5 K/fs, and in addition, another thermalization at 300 K was performed for 1 ps to stabi-
50 lize the glass structure. Finally, the glass structure was relaxed to 0 K by applying energy
51 minimization using the conjugate gradient method under zero macroscopic stress, allowing
52 the box size and shape to vary during the iterations. The described procedure provides us
53 with an equilibrium structure of the metallic glass ready for subsequent analyses.

54 In our study, the quantum mechanical approach in the framework of density func-

55 tional theory was used, which is implemented in the Vienna Ab initio Simulation Package
56 (VASP) [27]. The generalized gradient approximation was used for the exchange-correlation
57 energy, which is essential for the achievement of high accuracy. The energy cutoff regulating
58 the number of basis functions was set to 410 eV. Because of the relatively large size of
59 the system, only the Γ -point was used in the reciprocal space. The convergence of the
60 self-consistent calculation was enhanced by the Methfessel-Paxton method [28] with the
61 smearing of 0.1 eV. In the energy minimization process, structures were optimized until the
62 atomic forces become less than 0.01 eV/Å.

63 To model the response of the CuZr system to strain, the athermal quasi-static shear (AQS)
64 simulations, which are usually used in the framework of classical molecular dynamics, was
65 performed at the quantum mechanics level using the FPMD. Firstly the glass structure at 0 K
66 obtained by cooling is uniformly deformed with a simple shear strain ε_{xy} (affine deformation).
67 Next, atomic positions optimization is performed with the box geometry fixed. That results
68 in the stress relaxation from the affine state during the energy minimization. To improve
69 statistics, four different original structures, each of which is deformed independently at the
70 same strain in six different directions, were prepared and the results were averaged over 24
71 samples ($= 4 \times 6$) in total.

72 In the process of analyzing the mechanical properties of materials under strain, local stress
73 calculation is a standard tool to unveil atomic-level correlations. Nevertheless, until recently
74 that tool was available only in the classical approximation. Lately, several approaches were
75 proposed allowing the calculation of atomic level stress in a quantum-mechanical framework
76 (see [24] for the detailed discussion of differences between those approaches). In particular,
77 in this technique, the local stress is obtained as the strain derivative of local energy assigned
78 to a single atom, as it is done in [29] and [24]. The calculations were performed using
79 the Open source package for Material eXplorer (OpenMX), which utilizes the orbital-based
80 energy decomposition scheme [30]. Such approach allows direct calculation of the derivatives
81 of atomic energy with respect to the strain tensor.

82 Figure 1 (a) shows stress as a function of strain for several shear strains (0.005, 0.01,
83 0.02, 0.04, 0.08). To capture the effect of shear modulus softening [9], under particular strain
84 we analyzed two structures: the *affine* one (obtained directly from the original structure
85 by applying affine strain) and the *relaxed* one (obtained after performing the optimization
86 of atomic positions in the deformed structure). We see that stress is linearly proportional

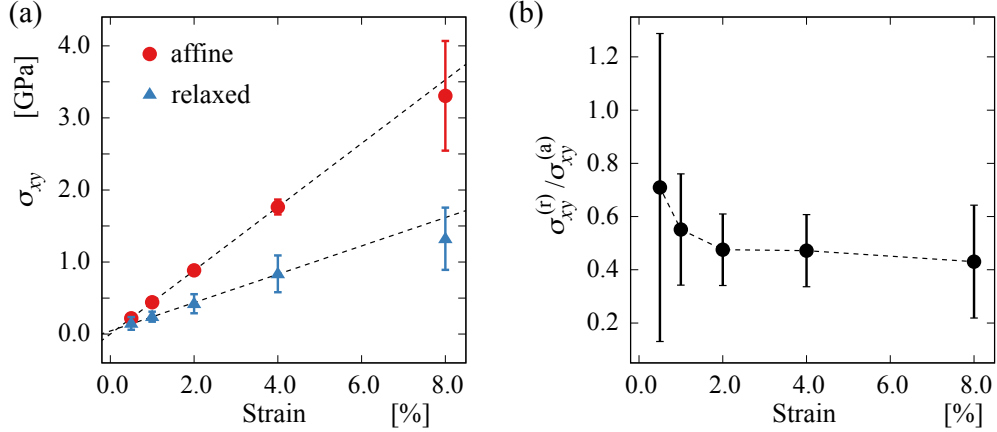


FIG. 1. (a) Stress - strain curves of $\text{Cu}_{50}\text{Zr}_{50}$ glass for affine and relaxed structures. (b) Ratio of relaxed stress $\sigma_{xy}^{(r)}$ to affine stress $\sigma_{xy}^{(a)}$ for various shear strains. Vertical bars show errors due to averaging (see discussion in text).

87 to applied strain for both cases of deformation. The estimated shear modulus is 44 GPa
 88 for the affine deformation and 21 GPa for the relaxed deformation. That is to say, the
 89 stress decreases during the relaxation process from the affine state by about 50%, as shown
 90 in Fig 1 (b), which is consistent with previous results based on classical AQS simulations
 91 (see [31] as an example).

92 In order to study the origin of the stress drop, we decomposed the macroscopic stress
 93 into the contributions of individual atoms and investigated the responses of the atomic-level
 94 stress under the shear for Cu and Zr. As shown in Fig. 2 (a), the local stress analysis
 95 revealed a fascinating behavior. The local stress of Zr increases positively, as expected, with
 96 the strain for affine and relaxed structures, whereas the local stress of Cu becomes more
 97 negative with the strain, which is a quite unusual opposite behavior. Also, it can be seen
 98 that the relaxation decreases the magnitude of the stress on each atom, and the deviation,
 99 $\delta = (\sigma_{relaxed} - \sigma_{affine}) / |\sigma_{affine}|$ of the stress from the affine stress is plotted in Fig. 2 (b).
 100 Interestingly Zr atoms show a negative contribution to the stress drop, while Cu atoms have
 101 a positive contribution. The origin of this behavior will be discussed later. At the highest
 102 strain of 8%, the change in the stress corresponds to 1.38 GPa for Cu atoms and -3.31 GPa
 103 for Zr atoms, thus leading to the stress drop of -1.93 GP in total. Therefore, we see that
 104 the change in the local environment around Zr atoms should be mainly contributing to the
 105 total stress drop.

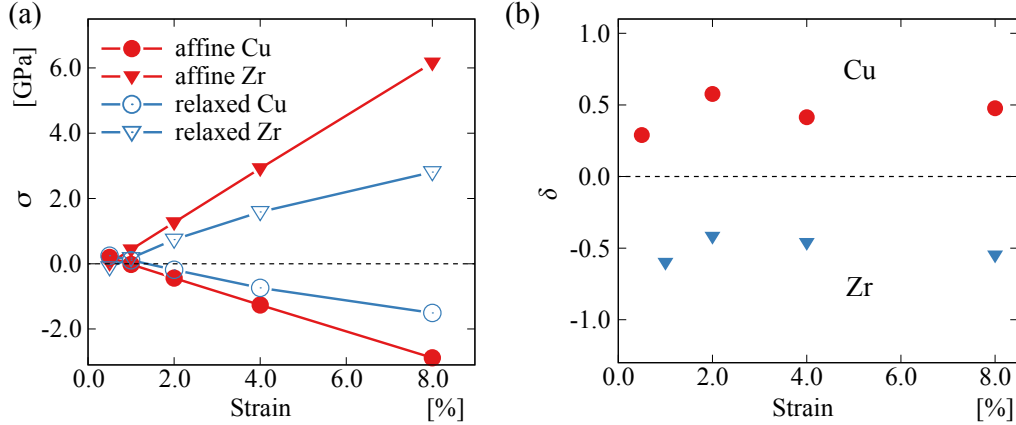


FIG. 2. (*Color online*) (a) Shear stress in the affine and relaxed structures for Cu₅₀Zr₅₀ under shear strain. Red filled circles (triangles) show Cu (Zr) contribution to the stress in the affine structure, while blue empty circles (triangles) show Cu (Zr) contribution to stress in the relaxed structure. (b) Deviation ($\delta = (\sigma_{relaxed} - \sigma_{affine})/|\sigma_{affine}|$) of the xy component of stress in the relaxed system from the xy component of stress in the affine system. Red circles (blue triangles) represent Cu (Zr) atoms.

106 We investigate the role of charge on the stress drop under deformation. Figure 3 shows
 107 the strain dependence on the charge transfer, δQ , between the affine and the relaxed states
 108 for Cu and Zr atoms. Units of δQ are number of electrons per atom. At small strains ($\epsilon \leq$
 109 4%) Cu atoms gain more charge during relaxation, whereas Zr atoms lose it. These results
 110 indicate that the non-affine relaxation enhances charge transfer from Zr to Cu, which is
 111 explained by the fact that the original structure obtained by rapid quench is unstable and
 112 deformation leads to the formation of more bonds between Cu and Zr, resulting in a more
 113 stable structure. At the highest strain of 8%, the sign of δQ becomes opposite, which may
 114 be due to the plastic flow far from the elastic region which tends to increase the effective
 115 temperature of the system and rejuvenate it [32].

116 Finally, we show a clear correlation between the local shear stress deviation ($\delta = (\sigma_{relaxed} -$
 117 $\sigma_{affine})/|\sigma_{affine}|$) and charge difference δQ for small strains in Fig. 4 (a). The data for the
 118 highest 8% strain beyond the elastic region was not included in the figure. Particular
 119 example of the structure rearrangement and associated charge change of an atom can be
 120 found in the Supplemental Material. We can conclude that, during the relaxation process,
 121 Cu gains charge, which positively increases the stress, whereas Zr loses charge, which makes

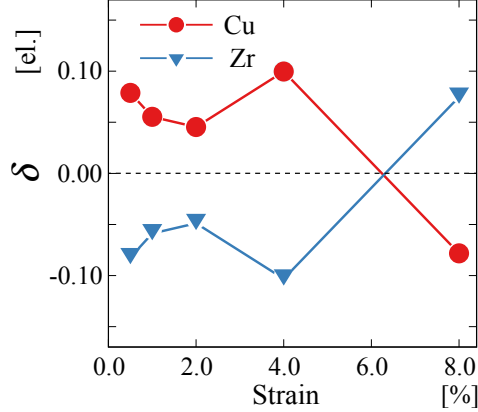


FIG. 3. (*Color online*) Averaged total charge difference (δQ , relaxed to affine) expressed in number of electrons for Cu (red circles) and Zr (blue triangles) atoms with shear strain.

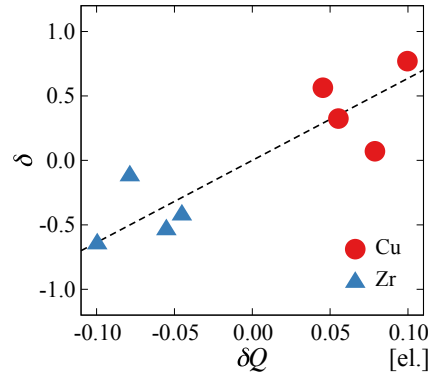


FIG. 4. (*Color online*) Correlation between the local stress deviation ($\delta = (\sigma_{relaxed} - \sigma_{affine})/|\sigma_{affine}|$) and the change in atomic charge ($\delta Q = Q_{relaxed} - Q_{affine}$) for $\text{Cu}_{50}\text{Zr}_{50}$ under elastic deformation. Dashed line is guide for eyes.

122 the stress more negative. It is worth mentioning that there is no direct correlation between
 123 the atomic-level pressure deviation and δQ , although the data are not shown here.

124 To show the difference in the behavior of the Cu and Zr subsystems under strain, let
 125 us discuss the 3-atoms angle change induced by the relaxation of atom positions from the
 126 affine structure. For each atom, we found closest neighbors of the same type. Next, for that
 127 group of atoms, the change of all unique 3-atoms angles was calculated with the targeted
 128 atom being the vertex. All angle change values were averaged and associated with each
 129 atom type(see Fig. 5a). The optimization process for the system under the 8% shear strain
 130 changes angles of the Cu subsystem by the value of 7.06 degrees, while for the Zr subsystem

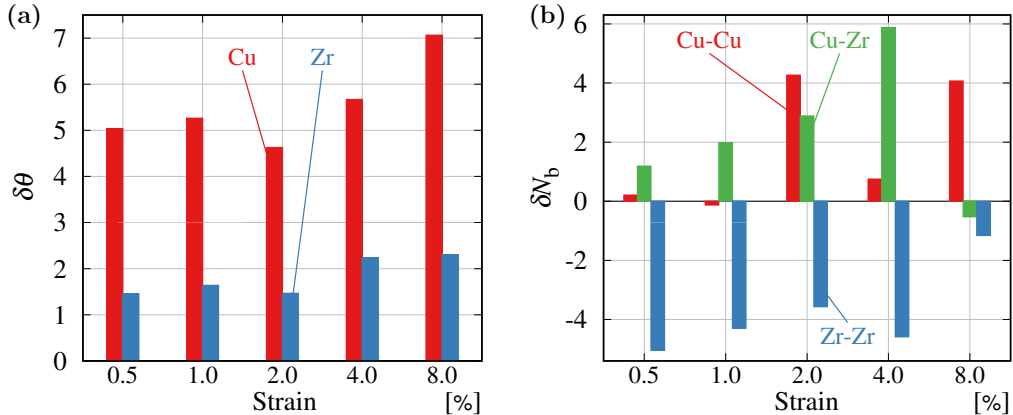


FIG. 5. (*Color online*) Relaxation induced changes of Cu and Zr subsystems for studied shear strain values. (a) Change in same type angles (see discussion in text). (b) Change in the number of chemical bonds.

131 that parameter equals only to 2.24 degrees. The Cu *d*-electron states are full and do not
 132 participate in bonding, whereas the Zr *d*-states are only partially filled and form covalent
 133 bonds [33]. The significant difference in the changes in angle induced by relaxation between
 134 Cu and Zr is consistent with this difference in the *d*-states occupation.

135 Taking only the neighbors of the same type, we can analyze the Cu subsystem and the Zr
 136 subsystem separately. However, the interplay between the Cu and Zr subsystems can only
 137 be understood if we take into account the types of neighboring atoms (see Supplemental
 138 Material for more data). The Cu subsystem is rearranged significantly during the relaxation
 139 process, resulting in the stress of an opposite sign, whereas the Zr-Zr angles are affected
 140 only slightly. At the same time, the Zr/Cu compositional short-range order (CSRO) is
 141 changed by deformation (see Fig. 5b). The fraction of the Zr-Cu pairs is increased at the
 142 expense of the Zr-Zr pairs. The increased CSRO induces more charge transfer, and the
 143 lowering of the potential energy, resulting in softening. However, the CSRO of the system
 144 before deformation depends on the condition of the system preparation through the fictive
 145 temperature. When we start with the system with a high degree of CSRO, deformation
 146 could degrade the CSRO, rather than to enhance it as observed here, resulting in a lesser
 147 amount of softening. Such a possibility remains to be studied.

148 Compelling, we found that the Cu subsystem is affected more strongly by deformation
 149 than the Zr subsystem. However, deformation decreases the number of Zr-Zr bonds and

150 increases the Zr-Cu neighbors, thus increasing the charge transfer. Consequently, the stress
151 on Cu is reduced by deformation, leading to the shear softening. Thus, in a $\text{Cu}_{50}\text{Zr}_{50}$ system,
152 not only a geometrical rearrangement of atoms but also a change in a CSRO contributes to
153 the softening of shear modulus.

154 Summarizing, the effect of elastic deformation on the electronic states in metallic glass is
155 studied under shear strain by means of ab initio calculations. The simulations of $\text{Cu}_{50}\text{Zr}_{50}$
156 glassy alloy under shear strain show that atoms undergo non-affine deformation even in the
157 elastic regime, accompanied by substantial charge transfer. In particular the Cu subsystem
158 is severely rearranged under the shear strain, surprisingly resulting in the shear stress of
159 the opposite sign on Cu atoms. On the other hand the Zr subsystem deforms in a nearly
160 affine manner. Deformation produces increased Zr/Cu CSRO, decreasing Zr-Zr bonds, which
161 leads to increased charge transfer. This work demonstrates that the change in the CSRO
162 contributes to the softening of the shear modulus through increased charge transfer, along
163 with the geometrical transformation. The evidence from our results points toward the need
164 of considering the effect of the deformation on the CSRO in metallic glasses.

165 **ACKNOWLEDGEMENT**

166 YS is grateful to M. Kohyama and T. Ozaki for discussion and comments. This work
167 was partly supported by Grant-in-Aid for Scientific Research in Innovative Areas to YS
168 (26109705 and 19H05177). TE was supported by the U.S. Department of Energy, Office
169 of Science, Basic Energy Sciences, Materials Science and Engineering Division. TI was
170 supported by JPSJ KAKENHI Grant Number JP19K03771.

* corresponding author ivan.lobzenko@gmail.com

- [1] A. Inoue, T. Zhang, and T. Masumoto, Production of amorphous cylinder and sheet of $\text{La}_{55}\text{Al}_{25}\text{Ni}_{20}$ alloy by a metallic mold casting method, *Materials Transactions, JIM* **31**, 425 (1990).
- [2] A. Inoue and A. Takeuchi, Recent development and application products of bulk glassy alloys, *Acta Materialia* **59**, 2243 (2011).

- [3] W. Johnson, Bulk glass-forming metallic alloys: Science and technology, *MRS Bulletin* **24**, 42 (1999).
- [4] G. Kumar, A. Desai, and J. Schroers, Bulk metallic glass: The smaller the better, *Advanced Materials* **23**, 461.
- [5] L. Davis, Mechanics of metallic glasses, *Rapidly Quenched Metals*. Massachusetts Institute of Technology, Cambridge. 1976, 369-391 (1976).
- [6] D. Weaire, M. Ashby, J. Logan, and M. Weins, On the use of pair potentials to calculate the properties of amorphous metals, *Acta Metallurgica* **19**, 779 (1971).
- [7] Y. Suzuki and T. Egami, Shear deformation of glassy metals: Breakdown of cauchy relationship and anelasticity, *Journal of Non-Crystalline Solids* **75**, 361 (1985), proceedings of the international conference on the theory of the structures of non-crystalline solids.
- [8] W. Dmowski, T. Iwashita, C.-P. Chuang, J. Almer, and T. Egami, Elastic heterogeneity in metallic glasses, *Phys. Rev. Lett.* **105**, 205502 (2010).
- [9] T. Egami, T. Iwashita, and W. Dmowski, Mechanical properties of metallic glasses, *Metals* **3**, 77 (2013).
- [10] S. Saw and P. Harrowell, Rigidity in condensed matter and its origin in configurational constraint, *Phys. Rev. Lett.* **116**, 137801 (2016).
- [11] H.-B. Yu, R. Richert, and K. Samwer, Correlation between viscoelastic moduli and atomic rearrangements in metallic glasses, *The Journal of Physical Chemistry Letters*, *The Journal of Physical Chemistry Letters* **7**, 3747 (2016).
- [12] A. Argon, Plastic deformation in metallic glasses, *Acta Metallurgica* **27**, 47 (1979).
- [13] A. Argon and H. Kuo, Plastic flow in a disordered bubble raft (an analog of a metallic glass), *Materials Science and Engineering* **39**, 101 (1979).
- [14] J. S. Langer, Dynamics of shear-transformation zones in amorphous plasticity: Formulation in terms of an effective disorder temperature, *Phys. Rev. E* **70**, 041502 (2004).
- [15] J. D. Ju, D. Jang, A. Nwankpa, and M. Atzmon, An atomically quantized hierarchy of shear transformation zones in a metallic glass, *Journal of Applied Physics* **109**, 053522 (2011).
- [16] I.-C. Choi, Y. Zhao, B.-G. Yoo, Y.-J. Kim, J.-Y. Suh, U. Ramamurty, and J. il Jang, Estimation of the shear transformation zone size in a bulk metallic glass through statistical analysis of the first pop-in stresses during spherical nanoindentation, *Scripta Materialia* **66**, 923 (2012), viewpoint set no. 49: Strengthening effect of nano-scale twins.

- [17] C. A. Schuh, A. C. Lund, and T. Nieh, New regime of homogeneous flow in the deformation map of metallic glasses: elevated temperature nanoindentation experiments and mechanistic modeling, *Acta Materialia* **52**, 5879 (2004).
- [18] D. Pan, A. Inoue, T. Sakurai, and M. W. Chen, Experimental characterization of shear transformation zones for plastic flow of bulk metallic glasses, *Proceedings of the National Academy of Sciences* **105**, 14769 (2008).
- [19] W. Kohn, Nobel lecture: Electronic structure of matter—wave functions and density functionals, *Rev. Mod. Phys.* **71**, 1253 (1999).
- [20] Y. Fan, T. Iwashita, and T. Egami, Crossover from localized to cascade relaxations in metallic glasses, *Phys. Rev. Lett.* **115**, 045501 (2015).
- [21] T. Egami, K. Maeda, and V. Vitek, Structural defects in amorphous solids a computer simulation study, *Philosophical Magazine A* **41**, 883 (1980).
- [22] T. Egami, Elementary excitation and energy landscape in simple liquids, *Modern Physics Letters B* **28**, 1430006 (19pp) (2014).
- [23] T. Egami, Y. Fan, and T. Iwashita, Mechanical deformation in metallic liquids and glasses: From atomic bond-breaking to avalanches, in *Avalanches in Functional Materials and Geophysics*, edited by E. K. Salje, A. Saxena, and A. Planes (Springer International Publishing, Cham, 2017) pp. 199–225.
- [24] D. M. Nicholson, M. Ojha, and T. Egami, First-principles local stress in crystalline and amorphous metals, *Journal of Physics: Condensed Matter* **25**, 435505 (2013).
- [25] N. Mattern, A. Schops, U. Kuhn, J. Acker, O. Khvostikova, and J. Eckert, Structural behavior of $\text{Cu}_x\text{Zr}_{100-x}$ metallic glass ($x = 35-70$), *Journal of Non-crystalline Solids* **354**, 1054 (2008).
- [26] N. Mattern, A. Schöps, U. Kühn, J. Acker, O. Khvostikova, and J. Eckert, Structural behavior of $\text{Cu}_x\text{Zr}_{100-x}$ metallic glass ($x=35-70$), *Journal of Non Crystalline Solids* **354**, 1054 (2008).
- [27] The vasp web-site, <https://www.vasp.at>, .
- [28] M.Methfessel and A.T.Paxton, High-precision sampling for brillouin-zone integration in metals, *Physical Review B* **40**, 3616 (1989).
- [29] Y. Shihara, M. Kohyama, and S. Ishibashi, Ab initio local stress and its application to al (111) surfaces, *Phys. Rev. B* **81**, 075441 (2010).
- [30] The openmx web-site, <http://www.openmx-square.org/>, .
- [31] A. Nakamura, Y. Kamimura, K. Edagawa, and S. Takeuchi, Elastic and plastic characteristics

- of a model cuzr amorphous alloy, *Materials Science and Engineering: A* **614**, 16 (2014).
- [32] W. Dmowski, Y. Yokoyama, A. Chuang, Y. Ren, M. Umemoto, K. Tsuchiya, A. Inoue, and T. Egami, Structural rejuvenation in a bulk metallic glass induced by severe plastic deformation, *Acta Materialia*, **58**, 429 (2010).
- [33] W. Dmowski, S. Gierlotka, Z. Wang, Y. Yokoyama, B. Palosz, and T. Egami, Pressure induced liquid-to-liquid transition in zr-based supercooled melts and pressure quenched glasses, *Scientific Reports* **7**, 6564 (2017).