

This is the accepted manuscript made available via CHORUS. The article has been published as:

Engineering Excited-State Interactions at Ultracold Temperatures

Michael Mills, Prateek Puri, Ming Li, Steven J. Schowalter, Alexander Dunning, Christian Schneider, Svetlana Kotochigova, and Eric R. Hudson

Phys. Rev. Lett. **122**, 233401 — Published 14 June 2019

DOI: [10.1103/PhysRevLett.122.233401](https://doi.org/10.1103/PhysRevLett.122.233401)

Engineering excited-state interactions at ultracold temperatures

Michael Mills,¹ Prateek Puri,¹ Ming Li,² Steven J. Schowalter,¹ Alexander Dunning,¹ Christian Schneider,¹ Svetlana Kotochigova,² and Eric R. Hudson¹

¹*Department of Physics and Astronomy, University of California, Los Angeles, California 90095, USA*

²*Department of Physics, Temple University, Philadelphia, Pennsylvania 19122, USA*

(Dated: March 26, 2019)

Using a recently developed method for precisely controlling collision energy, we observe a dramatic suppression of inelastic collisions between an atom and ion ($\text{Ca} + \text{Yb}^+$) at low collision energy. This suppression, which is expected to be a universal phenomenon, arises when the spontaneous emission lifetime of the excited state is comparable to or shorter than the collision complex lifetime. We develop a technique to remove this suppression and engineer excited-state interactions. By dressing the system with a strong catalyst laser, a significant fraction of the collision complexes can be excited at a specified atom-ion separation. This technique allows excited-state collisions to be studied, even at ultracold temperature, and provides a general method for engineering ultracold excited-state interactions.

In the last quarter century, the development of techniques for producing ultracold matter led to the ability to observe few and even single partial wave collision events, enabling the observation of quantum threshold behavior and unitarity limited processes [1–3]. It also revealed the impact chemical binding forces, quantum statistics, internal structure, and dimensionality have on collisions, as well as provided the potential for control of chemical reactions [4–10].

The overwhelming majority of these studies were performed with collision partners in their ground electronic state. This is at least partially due to the fact that interactions at ultracold temperatures tend to naturally suppress electronically excited collisions, as pointed out in [11] and demonstrated in [12, 13]. This suppression arises as the long-range interactions between collision partners tend to shift any laser that would electronically excite one of the collision partners out of resonance at very long range. This effect, known as reaction blockading [13], is especially strong in systems with long-range interactions such as atom-ion or molecule-molecule pairs.

Collisions involving electronically excited atoms and molecules play an important role in processes such as combustion [14], explosives, atmospheric chemistry [15], stellar evolution [16], interactions in the interstellar medium [17], and the formation of new molecules [18], yet many studies of these reactions have been limited to high temperature, where quantum effects are often obscured. Here, we demonstrate a general technique to enable the study of such collisions at ultracold temperatures in a prototypical atom-ion system. Building from work on hyperfine-changing collisions in laser-cooled systems [12, 19, 20], we apply a strong laser field that dresses the system and promotes the molecular collision complex to a specified excited state at a specified range. In this way, we engineer the electronic excitation of collision partners at short range and extract the excited channel rate constants. Additionally, by controlling the range at which this laser addresses the reactants, this technique is

sensitive to the features of molecular potentials, enabling a new class of experiments to probe molecular potentials at controlled atom-ion separations. Excitingly, the technique appears to be completely general and can be applied at higher temperatures.

In what follows, we use a recently developed method [21] for precisely controlling collision energy to study the charge-exchange collision between $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ as a function of collision energy from 0.05 K to 0.65 K. From this data, we observe reaction blockading of the charge-exchange rate and measure the dependence of this suppression on collision energy. Finally, we introduce a strong laser to dress the system and observe an increased charge-exchange rate for the $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ channel, effectively eliminating the suppression. A quantum coupled-channels calculation based on ground- and excited-state diabatic potentials, their couplings, and the infinite-order sudden approximation is presented and shows good agreement with the data. The technique is qualitatively explained using a simple semi-classical model based on dressed molecular potentials and a Landau-Zener type transition.

The experiment is performed in the second-generation MOTion trap, sketched in Fig. 1 and described in detail in Refs. [22, 23]. Coulomb crystals of Yb^+ ions are confined in a segmented radio-frequency linear quadrupole trap, while a small Ca oven provides atomic Ca vapor which is loaded into a magneto-optical trap (MOT). By smoothly ramping the voltages of the ion trap end caps, we can shuttle an ion chain through the MOT at a controlled velocity, illustrated in Fig. 1(a), (b), and (c) and described in detail in [21], allowing precise control of the reactant collision energy. By extinguishing the 369 nm Yb^+ cooling laser when the Yb^+ ions are shuttled through the Ca MOT, the ions are prepared in the ground $6s\ ^2S_{1/2}$ state. Using this shuttling method, we measure the charge-exchange rate of $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ as a function of collision energy and observe reaction blockading of the rate, shown in Fig. 1(d).

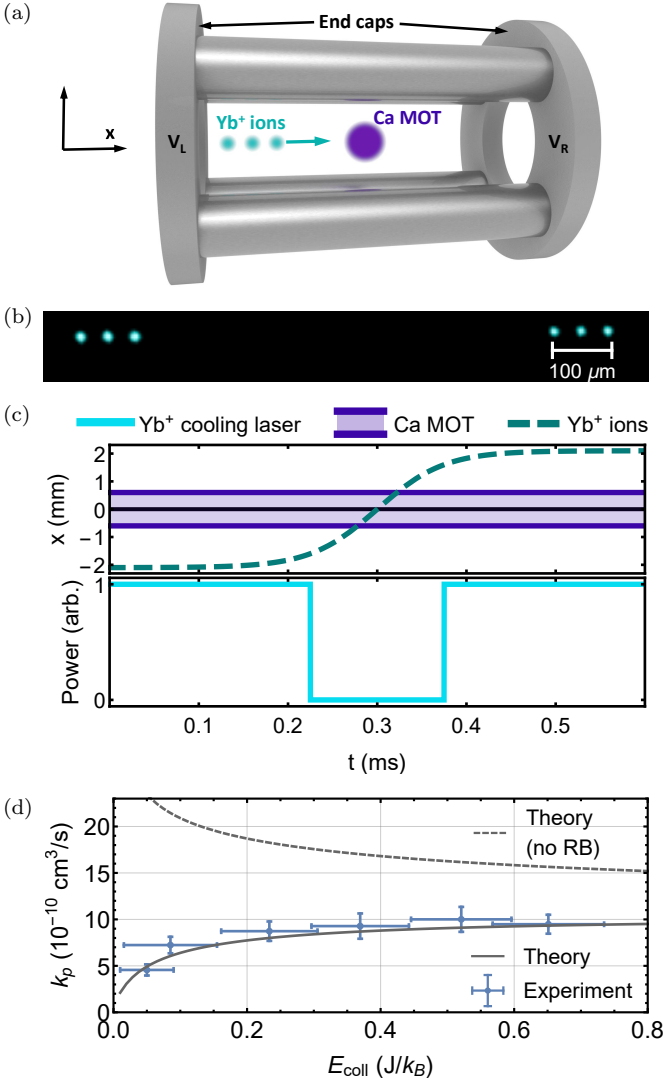


FIG. 1. **Shuttling in the hybrid atom-ion MOTion trap.** (a) Schematic of the MOTion trap. (b) False-color fluorescence image of 3 shuttled Yb^+ ions. As the exposure time is greater than the shuttling period, fluorescence from the 3 ions is concentrated at the positions of the two end points, where the ions spend the most time. (c) Experimental sequence illustrating the shuttling technique. As the Yb^+ ions are shuttled through the Ca MOT, the 369 nm Yb^+ cooling beams are extinguished to prepare the ions in the $6s\ ^2S_{1/2}$ state. (d) Measured charge-exchange rate coefficient (with standard errors) for $Ca(^1P_1) + Yb^+(^2S_{1/2})$ as a function of collision energy using the shuttling technique. Also shown are rate coefficients from coupled-channels calculations, one with (solid line) and one without (dashed line) the effect of reaction blockading (RB).

Specifically, for an ion chain with 50 mK collision energy (defined as $\langle E_{col} \rangle / k_B$, where $\langle E_{col} \rangle$ is the average kinetic energy in the center of mass frame and k_B is the Boltzmann constant), we measure a rate constant of $k_p = (4.6 \pm 0.6) \times 10^{-10}\ cm^3/s$, compared to the no-suppression theoretical prediction of $k_p = 23 \times 10^{-10}\ cm^3/s$, an ob-

served suppression factor of ~ 5 .

This reaction blockading can be understood by considering the long-range atom-ion interaction [11, 13]. At long range the atom and ion interact primarily through the charge-induced dipole and charge-quadrupole potentials of the forms $-\frac{\alpha}{2}R^{-4}$ and $-\frac{Q}{2}(3\cos^2(\theta) - 1)R^{-3}$, respectively, where R is the atom-ion separation, α is the neutral atom polarizability, Q is the neutral atom quadrupole moment, and θ is the angle between the quadrupole moment and the internuclear axis. Thus, a laser resonant with two atomic states at long range, which have different polarizabilities and quadrupole moments, is no longer resonant when the atom and ion are in close proximity. For the $Ca\ ^1P_1 \leftarrow ^1S_0$ transition with linewidth Γ and a laser detuning $\delta = -\Gamma = 2\pi \times (-34.6\ MHz)$, the laser becomes resonant at $R \approx 1300\ a_0$ and becomes detuned by 10Γ at $\sim 600\ a_0$. Therefore, for a charge-exchange event to occur, the atom-ion pair must propagate inward without the Ca atom decaying from this distance to distances of $\sim 40\ a_0$, where couplings to other states become significant. For collision temperatures greater than $\gtrsim 10\ K$, the atom-ion pair approaches quickly enough such that the $Ca\ ^1P_1$ state is unlikely to decay before reaching short range, affecting the rate coefficient by $\lesssim 1\%$. For a collision temperature of 1 mK, however, this effect leads to a suppression by a factor of ~ 100 .

To understand this behavior, we first consider charge exchange at low temperatures with no measures taken to overcome reaction blockading. Fig. 2 shows the relevant $CaYb^+$ long-range diabatic potentials, labeled by the projection Ω of the total angular momentum onto the intermolecular axis, as a function of atom-ion separation R . These potentials are the diagonal matrix elements in our diabatic electronic basis, which changes only slowly with R . Asymptotically, these basis states correspond to atomic eigenstates so that non-adiabatic coupling among the potentials are negligible. The variable R describes the separation between the center of mass of the atom and ion. Moreover, the center of mass changes in charge transfer. The couplings originating from these changes are negligibly small for our purposes, justifying our definition for R . The entrance channel to the studied charge-exchange process, $Ca(4s4p\ ^1P_1) + Yb^+(6s\ ^2S_{1/2})$, has both a four-fold-degenerate ($|\Omega| = 1/2, 3/2$) repulsive potential and two-fold-degenerate ($|\Omega| = 1/2$) attractive potential. Substantial non-radiative charge transfer only occurs to the $Ca^+(4s\ ^2S_{1/2}) + Yb(5d6s\ ^3D_2)$ exit channel. The $Ca^+(4s\ ^2S_{1/2}) + Yb(5d6s\ ^3D_3)$ channel is energetically inaccessible to this entrance channel, and the $Ca^+(4s\ ^2S_{1/2}) + Yb(5d6s\ ^3D_1)$ channel is only crossed at short range $R \approx 25\ a_0$, where the estimated couplings between these diabatic potentials, using the Heitler-London method [24], are too large to significantly contribute to the rate coefficient. More details about the potentials, the diabatic couplings, and the cal-

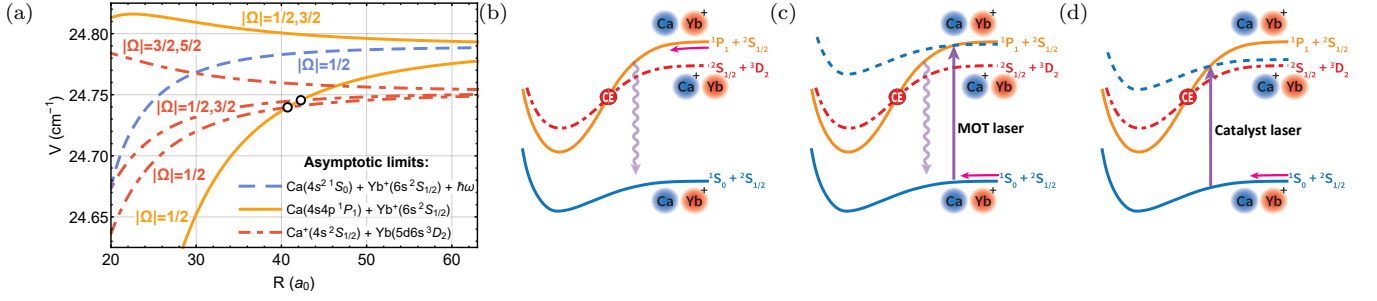


FIG. 2. **Long-range diabatic potential energy curves.** (a) Relevant long-range molecular potentials. The two crossings between potentials relevant for charge exchange are indicated with black circular markers. The potential energy zero is located at the $\text{Ca}(4s^2\ ^1S_0) + \text{Yb}^+(6s\ ^2S_{1/2})$ dissociation limit. (b) The first pathway corresponds to a collision between an excited $4s4p\ ^1P_1$ Ca atom with a ground-state $6s\ ^2S_{1/2}$ Yb^+ ion. The charge-exchange (CE) crossing is shown by a red circle. The vertical wavy line represents spontaneous emission to the ground $\text{Ca}(4s^2\ ^1S_0) + \text{Yb}^+(6s\ ^2S_{1/2})$ channel. (c) The second pathway corresponds to a collision between a ground-state $4s^2\ ^1S_0$ Ca atom with a $6s\ ^2S_{1/2}$ Yb^+ ion in the presence of a photon of the MOT laser. The dashed blue curve corresponds to the dressed-state potential for this entrance channel. It has an avoided crossing with the excited $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ potential. (d) In the presence of a catalyst laser, the incoming $\text{Ca}(4s^2\ ^1S_0) + \text{Yb}^+(6s\ ^2S_{1/2})$ state is coupled to the reactive $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ state at short range, where spontaneous emission is unlikely before reaction.

culcation are given in our accompanying paper of Ref. [25].

Therefore, the non-radiative charge transfer is primarily driven by coupling of the $|\Omega| = 1/2$ entrance channel diabats at their crossings with the exit channels. In the diabatic representation, this coupling arises from the molecular electrostatic interaction and therefore conserves Ω , implying that only charge transfer to the $|\Omega| = 1/2$ exit channel diabats, at crossing points $R_c = 40.7\ a_0$ and $42.3\ a_0$, is relevant. Since the electronic basis functions are very different for the two channels, the non-adiabatic coupling is localized and approximated by identical Lorentzians centered at each R_c . The half width of this Lorentzian, R_0 , is chosen to match the experimentally determined charge transfer rates. In the absence of any additional means to overcome reaction blockading, the atom-ion pair can reach these crossing points and undergo a charge-exchange reaction via two pathways. The first pathway is directly on the $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ entrance channel, where we determine the population of Ca atoms in the 1P_1 state by solving a rate equation derived from the optical Bloch equations, which includes the distance-dependent detuning of the MOT beams [26]. The second pathway describes a collision on the photon-dressed $\text{Ca}(4s^2\ ^1S_0) + \text{Yb}^+(6s\ ^2S_{1/2})$ state, which is coupled to the $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ state via the MOT laser. Because the MOT laser is tuned $2\pi \times 34.6$ MHz below the asymptotic transition energy, as the atom and ion collide the laser is shifted into resonance. At this point, there is a resonant amplification of the coupling to the $\text{Ca}(4s4p\ ^1P_1) + \text{Yb}^+(6s\ ^2S_{1/2})$ state by the molecular interaction due to the large density of states near the threshold.

Using the infinite-order sudden approximation (IOSA) [27–30], a coupled-channels calculation is performed on these potentials to determine the charge

transfer cross-section, $\sigma(E, \ell)$. The effect of spontaneous emission is included by classically computing the collision time on the entrance channel and determining the probability, $p(E, \ell)$, for a colliding pair to survive to R_c without spontaneously emitting. The charge transfer rate constant is then determined as

$$k = \sum_{\ell=0}^{\infty} (2\ell + 1) p(E, \ell) \sigma(E, \ell) \quad (1)$$

where ℓ is the average orbital angular momentum quantum number used in the IOSA and E is the collision energy. The resulting rate constant is displayed alongside the data in Fig. 1(d) for $R_0 = 0.39\ a_0$, with and without the inclusion of $p(E, \ell)$. A detailed description of the excited-state potentials and charge transfer can be found in [25].

Given that this reaction blockading is expected to occur in all low-temperature excited-state collisions, it is desirable to develop a method to remove it. Here, we demonstrate one such means. Building on ideas developed for control of hyperfine-changing collisions [12, 20, 31], we apply a strong laser, dubbed the catalyst laser, that couples the ground-state with an excited state at short range. This allows selection of the excited-state reaction channel and may, in principle, be used to select a desired reaction product in polyatomic systems.

The operation of the technique is sketched in Fig. 2(d), where the CaYb^+ molecular potentials dressed by the photon energy of the applied laser are shown. Near the catalyst laser avoided crossing distance R_{CL} , the catalyst beam couples the upper and lower states, promoting the complex to the $\text{Ca}(^1P_1) + \text{Yb}^+(^2S_{1/2})$ state at short range. The probability of promotion can be estimated from Landau-Zener transition theory as $P(\Omega_R) = 1 - \text{Exp}[-\pi(\hbar\Omega_R)^2 / (2\hbar v \frac{\partial}{\partial R} \Delta E)]$, where Ω_R is the Rabi

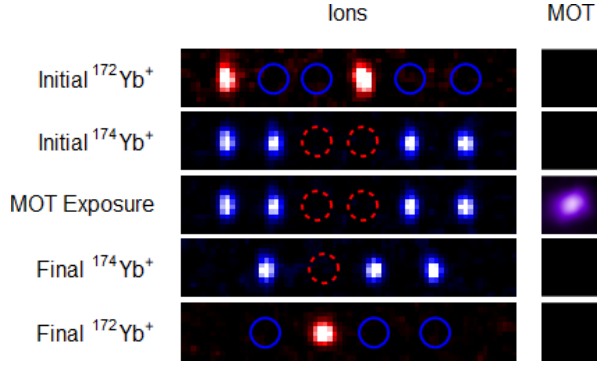


FIG. 3. **Dual-isotope technique.** False-color fluorescence images of the Yb^+ ions and the Ca MOT (not to scale) illustrating the dual-isotope method used to measure the low decay rate of $^{172}\text{Yb}^+(^2\text{S}_{1/2})$. We first trap $^{172}\text{Yb}^+$ and $^{174}\text{Yb}^+$, while laser-cooling only $^{172}\text{Yb}^+$ ions (shown in red), while $^{174}\text{Yb}^+$ ions (shown as blue circles) remain dark. We then switch the 369 nm cooling laser frequency to cool $^{174}\text{Yb}^+$ ions (shown in blue), while the $^{172}\text{Yb}^+$ ions (shown as red dashed circles) remain dark. We then overlap the MOT with the laser-cooled $^{174}\text{Yb}^+$ ions as well as the ground-state $^{172}\text{Yb}^+(^2\text{S}_{1/2})$ ions for a variable amount of time. Finally, we cool and measure the final number of $^{172}\text{Yb}^+$ ions.

frequency of the catalyst beam, v is the radial velocity, and ΔE is the energy difference between the diabatic potentials [32]. Thus, for a scattering event with rate constant, k , this technique yields an experimentally observable rate $k_o = kP(\Omega_R)e^{-\Delta t(R_{CL})/\tau_P}$, where $\Delta t(R_{CL})$ is the time required for the atom-ion pair to propagate from R_{CL} to short range and τ_P is the lifetime of the excited state.

In order to test the catalyst laser technique at the lowest possible collision energy, where the suppression is strongest, the ions cannot be shuttled but must be arranged in a stationary linear ion chain overlapped with the MOT. Due to collisional heating effects [23, 33, 34], the linear ion chain cannot be maintained during MOT exposure without active laser cooling. However, if the ions are laser cooled, the charge-exchange rates include collisions originating in the $\text{Yb}^+(^2\text{P}_{1/2})$ and $\text{Yb}^+(^2\text{D}_{3/2})$ states [35]. Therefore, in order to isolate the $\text{Ca}(^1\text{P}_1) + \text{Yb}^+(^2\text{S}_{1/2})$ charge-exchange rate without the shuttling method, we develop and implement a dual-isotope technique (see Fig. 3) for collision rate measurement. Specifically, we simultaneously trap both $^{172}\text{Yb}^+$ and $^{174}\text{Yb}^+$ ions, while laser-cooling only the $^{174}\text{Yb}^+$ ions, which, in turn, sympathetically cool the $^{172}\text{Yb}^+$ ions. As the $^{172}\text{Yb}^+$ ions are only sympathetically cooled, they remain in the $6s\ ^2\text{S}_{1/2}$ state. Due to off-resonant scattering of the cooling laser for the $^{174}\text{Yb}^+$ ions, it is necessary to apply a repumping laser for the $^{172}\text{Yb}^+$ ions to prevent population from accumulating in the $5d\ ^2\text{D}_{3/2}$ state. Therefore, by monitoring the number of $^{172}\text{Yb}^+$ ions with time, we isolate and measure the charge exchange of $\text{Ca}(^1\text{P}_1) + \text{Yb}^+(^2\text{S}_{1/2})$.

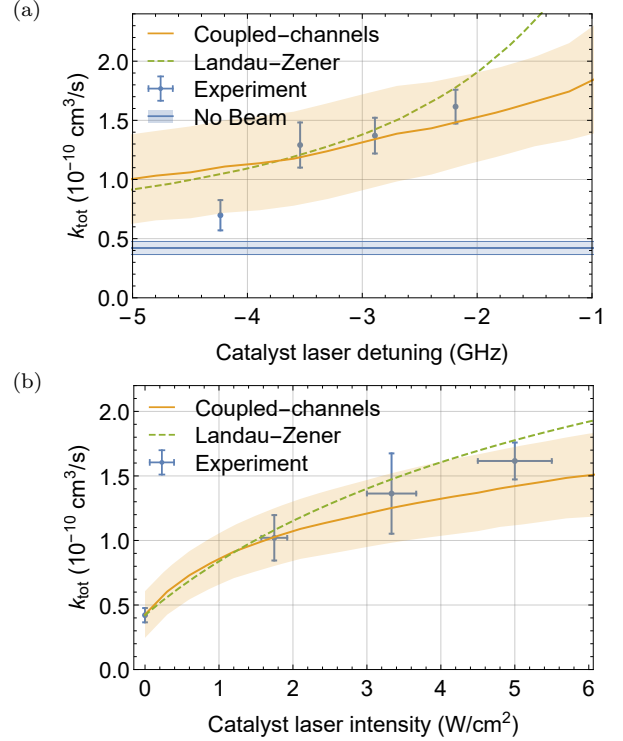


FIG. 4. **Removing suppression with addition of a catalyst laser.** Total charge-exchange rate coefficient as a function of catalyst laser (a) frequency and (b) intensity. Plotted alongside experimental data are the results of a coupled-channels calculation and an estimate using the Landau-Zener approximation. For reference, the experimental rate with no catalyst beam is shown. Error bars correspond to the standard error in experimental measurements and error bands include uncertainties from the theoretical simulations and experimental parameters. Horizontal error bars in (a) are smaller than the plot marker. Details are found in the Supplementary Material.

Figs. 4(a) and (b) show the results of using this dual-isotope technique to monitor $\text{Ca}(^1\text{P}_1) + \text{Yb}^+(^2\text{S}_{1/2})$ charge-exchange reactions as a function of the detuning and intensity of the catalyst laser, respectively, at a collision temperature of ~ 50 mK. For large detunings, although the atom-ion pair is promoted at a small value of R_{CL} , increasing the likelihood of reaching short range before spontaneous emission, the large value of $\frac{\partial}{\partial R}\Delta E$ and the high velocity of the reactants leads to a lower probability of promotion to the reactive state from Landau-Zener transition theory. For the given experimental intensity of 5 W/cm^2 , the catalyst beam cannot be closer to resonance than $\sim -2\text{ GHz}$ due to adverse effects on the MOT.

The dependence of the measured rate on the catalyst laser intensity can be understood by the increased probability of promotion to the reactive state given by Landau-Zener transition theory for increasing Rabi frequencies Ω_R . Also shown are the results of a coupled-channels

calculation. Here, the rates are calculated by allowing for, in addition to the two previously discussed pathways from the MOT laser, a catalyst-laser-enhanced charge-exchange pathway, coupling the $\text{Ca}(^1\text{S}_0) + \text{Yb}^+(^2\text{S}_{1/2})$ entrance channel to the $\text{Ca}^+(^2\text{S}_{1/2}) + \text{Yb}(^3\text{D}_2)$ exit channel via the intermediate $\text{Ca}(^1\text{P}_1) + \text{Yb}^+(^2\text{S}_{1/2})$ channel. The experimental data shows good agreement with both the coupled-channels calculations and the simple Landau-Zener model, supporting this interpretation of the results.

In summary, we have investigated and engineered electronically excited-state collisions of Ca with Yb^+ at low collision energy. Using a method for precise control of collision energy, we find that the interaction of the atom with the ion leads to a strong shift between the ground and excited atomic states, causing any laser addressing the bare atomic transition frequency to be shifted from resonance, even at long range. Thus, at low collision energy, an atomic excited state is likely to undergo spontaneous emission before reaching short range. This leads to a strong suppression of scattering events that occur via molecular states corresponding to an atomic excited state. These features are expected to be universal at low temperature for systems with short-lived electronic excitations and long-ranged interactions. To overcome this suppression, we demonstrate a technique using a catalyst laser, which selectively excites colliding molecular complexes at short range. This technique removes the observed reaction blockading, allowing excited-state collisions to be studied even at ultracold temperatures. As low-temperature techniques provide precise information about the underlying dynamics, this technique should find use as a general tool for studying excited-state collisions. Further, because the technique selectively excites the colliding pair to a chosen state it may be used as a means to select a desired product outcome in polyatomic chemical reactions.

Finally, the reaction blockading effect observed and controlled here is extremely important for the growing field of hybrid atom-ion trapping, where sympathetic cooling of ions with laser-cooled atoms is being pursued [22, 36, 37]. The existence of this reaction blockading effect means that detrimental chemical reactions from excited atomic states, which are energetically unavoidable, will not occur during the sympathetic cooling process. Thus, a large variety of molecular ions can be cooled by laser-cooled atoms without loss to unwanted chemical reactions.

We thank Wesley Campbell for helpful discussions. This work was supported in part by the National Science Foundation and the Army Research Office. Work at the Temple University is supported by the Army Research Office, the U.S. Air Force Office of Scientific Research, and the National Science Foundation.

-
- [1] N. R. Thomas, N. Kjærgaard, P. S. Julienne, and A. C. Wilson, *Phys. Rev. Lett.* **93**, 173201 (2004).
 - [2] S. Ospelkaus, K.-K. Ni, D. Wang, M. De Miranda, B. Neyenhuis, G. Quémener, P. Julienne, J. Bohn, D. Jin, and J. Ye, *Science* **327**, 853 (2010).
 - [3] K.-K. Ni, S. Ospelkaus, D. Wang, G. Quémener, B. Neyenhuis, M. De Miranda, J. Bohn, J. Ye, and D. Jin, *Nature* **464**, 1324 (2010).
 - [4] B. DeMarco, J. L. Bohn, J. P. Burke, M. Holland, and D. S. Jin, *Phys. Rev. Lett.* **82**, 4208 (1999).
 - [5] K. M. Jones, E. Tiesinga, P. D. Lett, and P. S. Julienne, *Rev. Mod. Phys.* **78**, 483 (2006).
 - [6] F. Lang, K. Winkler, C. Strauss, R. Grimm, and J. H. Denschlag, *Phys. Rev. Lett.* **101**, 133005 (2008).
 - [7] E. R. Hudson, C. Ticknor, B. C. Sawyer, C. A. Taatjes, H. J. Lewandowski, J. R. Bochinski, J. L. Bohn, and J. Ye, *Phys. Rev. A* **73**, 063404 (2006).
 - [8] L. Ratschbacher, C. Zipkes, C. Sias, and M. Köhl, *Nature Phys.* **8**, 649 (2012).
 - [9] T. Yang, A. Li, G. K. Chen, C. Xie, A. G. Suits, W. C. Campbell, H. Guo, and E. R. Hudson, *J. Phys. Chem. Lett.* **9**, 3555 (2018), pMID: 29893569.
 - [10] G. K. Chen, C. Xie, T. Yang, A. Li, A. G. Suits, E. R. Hudson, W. C. Campbell, and H. Guo, *Phys. Chem. Chem. Phys.*, Advance Article (2019).
 - [11] P. S. Julienne and F. H. Mies, *J. Opt. Soc. Am. B* **6**, 2257 (1989).
 - [12] C. D. Wallace, V. Sanchez-Villicana, T. P. Dinneen, and P. L. Gould, *Phys. Rev. Lett.* **74**, 1087 (1995).
 - [13] P. Puri, M. Mills, I. Simbotin, J. A. Montgomery, R. Côté, E. P. West, C. R. Schneider, A. G. Suits, and E. R. Hudson, to be published (2018).
 - [14] V. V. Smirnov, O. M. Stelmakh, V. I. Fabelinsky, D. N. Kozlov, A. M. Starik, and N. S. Titova, *J. Phys. D: Appl. Phys.* **41**, 192001 (2008).
 - [15] J. R. Wiesenfeld, *Accounts Chem. Res.* **15**, 110 (1982).
 - [16] C. Matei, L. Buchmann, W. Hanne, D. Hutcheon, C. Ruiz, C. Brune, J. Caggiano, A. Chen, J. DAuria, A. Laird, *et al.*, *Phys. Rev. Lett.* **97**, 242503 (2006).
 - [17] M. Agúndez, J. Goicoechea, J. Cernicharo, A. Faure, and E. Roueff, *Astrophys. J.* **713**, 662 (2010).
 - [18] P. Puri, M. Mills, C. Schneider, I. Simbotin, J. A. Montgomery, R. Côté, A. G. Suits, and E. R. Hudson, *Science* (2017).
 - [19] D. Sesko, T. Walker, C. Monroe, A. Gallagher, and C. Wieman, *Phys. Rev. Lett.* **63**, 961 (1989).
 - [20] V. Sanchez-Villicana, S. D. Gensemer, K. Y. N. Tan, A. Kumarakrishnan, T. P. Dinneen, W. Süptitz, and P. L. Gould, *Phys. Rev. Lett.* **74**, 4619 (1995).
 - [21] P. Puri, M. Mills, E. P. West, C. Schneider, and E. R. Hudson, *Rev. Sci. Instrum.* **89**, 083112 (2018).
 - [22] W. Rellergert *et al.*, *Nature* **495**, 490 (2013).
 - [23] S. Schowalter *et al.*, *Nature Comm.* **7**, 124448 (2016).
 - [24] K. Tang, J. P. Toennies, and C. L. Yiu, *Int. Rev. Phys. Chem.* **17**, 363 (1998).
 - [25] M. Li, M. Mills, P. Puri, A. Petrov, E. R. Hudson, and S. Kotochigova, to be published (2019).
 - [26] M. Mills, P. Puri, Y. Yu, A. Derevianko, C. Schneider, and E. R. Hudson, *Phys. Rev. A* **96**, 033402 (2017).
 - [27] R. T. Pack, *J. Chem. Phys.* **60**, 633 (1974).
 - [28] D. Secrest, *J. Chem. Phys.* **62**, 710 (1975).

- [29] L. W. Hunter, J. Chem. Phys. **62**, 2855 (1975).
- [30] D. J. Kouri, “Rotational excitation ii: Approximation methods,” in *Atom - Molecule Collision Theory: A Guide for the Experimentalist*, edited by R. B. Bernstein (Springer US, Boston, MA, 1979) pp. 301–358.
- [31] S. C. Zilio, L. Marcassa, S. Muniz, R. Horowicz, V. Bagnato, R. Napolitano, J. Weiner, and P. S. Julienne, Phys. Rev. Lett. **76**, 2033 (1996).
- [32] C. Wittig, J. Phys. Chem. B **109**, 8428 (2005), pMID: 16851989.
- [33] K. Chen, S. T. Sullivan, W. G. Rellergert, and E. R. Hudson, Phys. Rev. Lett. **110**, 173003 (2013).
- [34] K. Chen, S. T. Sullivan, and E. R. Hudson, Phys. Rev. Lett. **112**, 143009 (2014).
- [35] W. G. Rellergert, S. T. Sullivan, S. Kotochigova, A. Petrov, K. Chen, S. J. Schowalter, and E. R. Hudson, Phys. Rev. Lett. **107**, 243201 (2011).
- [36] E. R. Hudson, EPJ Tech. Instrum. **3**, 8 (2016).
- [37] E. R. Hudson and W. C. Campbell, Phys. Rev. A **98**, 040302 (2018).