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## Connecting the Simpler Structures to Topologically Close-Packed Phases

Anirudh Raju Natarajan and Anton Van der Ven

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# Connecting the simpler structures to topologically close-packed phases

Anirudh Raju Natarajan\* and Anton Van der Ven†

Materials Department, University of California, Santa Barbara

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Pathways connecting dissimilar crystal structures are fundamental to our understanding of structural phase transitions. In this letter we report on a new pathway connecting the hexagonal closed packed crystal structure to a hierarchy of topologically close-packed phases consisting of Kagomé and triangular nets. Common intermetallic structure prototypes such as the Friauf-Laves phases,  $\text{CaCu}_5$ ,  $\text{Ce}_2\text{Ni}_7$ ,  $\text{Be}_3\text{Nb}$  and  $\text{Co}_7\text{Gd}_2$  are specific members of this hierarchy. We find that the pathway is facile for compounds with large atomic size differences, which has implications for the nucleation mechanism of these complex phases.

Phase transformations connecting vastly different crystal structures are ubiquitous in pure elements and multicomponent crystals and are exploited in shape-memory[1–6], magnetocaloric[7, 8] and precipitation strengthened alloys[9, 10]. The structural pathways between different crystal structures are a cornerstone of our understanding of common phase transformations in the solid state. Nevertheless, despite almost a century of research on structural phase transitions, only a handful of crystallographic pathways linking a few simple crystal structures are known.

The Bain path[11] links the bcc (body centered cubic) and fcc (face centered cubic) crystal structures through a homogeneous tetragonal deformation of their cubic unit cells. The bcc and hcp (hexagonal close packed) structures can be connected through the Burgers path[12], which combines a macroscopic shape change with an internal atomic shuffle within the hcp unit cell. Pathways have also been identified connecting the hcp and bcc crystal structures to the slightly more complex  $\omega$  phase, made up of alternating triangular and honeycomb layers [13–15]. Pathways connecting bcc to a family of close-packed “R” phases and to the B19’ phase are also known[1, 3–6].

While the structural transformations involving hcp, bcc, fcc,  $\omega$  and other close-packed phases are well understood, a majority of the crystal structure prototypes that are commonly observed in alloys remain disconnected. The topologically close-packed (TCP) Frank-Kasper phases [16, 17], for example, are an especially common[18] class of crystal structures that form in alloy systems whose end members adopt the simpler fcc, hcp and bcc crystal structures [19, 20]. There are many distinct structures that fall into this class of compounds, with the Friauf-Laves phases being the most common and well studied. Many compounds that form topologically close-packed crystal structures display exotic electronic, magnetic and mechanical phenomena[21–29]. Icosahedral arrangements of atoms similar to the Frank-Kasper compounds are also observed in glasses and liquids[30–32]. Frank-Kasper phases are not restricted to metallic alloys, but have also shown promise as photonic crystals [33] and have been observed to form during the crystallization of block co-polymers and colloidal self-assembly

[34–38].

In this letter, we identify a family of new transformation pathways that link a major subset of the Frank-Kasper phases to simpler close packed structures. We show that a simple two-dimensional rearrangement of the triangular lattice can be used to build a series of hierarchical phases starting with the simple close-packed hcp crystal structure. Structures like the Laves phases,  $\text{CaCu}_5$ ,  $\text{Ce}_2\text{Ni}_7$ ,  $\text{Be}_3\text{Nb}$  and  $\text{Co}_7\text{Gd}_2$  structure prototypes are found to be specific members of this hierarchy. Furthermore, we show that the pathways that link the close-packed structures to the TCP phases are facile and will occur spontaneously for particular arrangements of close packed atoms that have very different atomic volumes.

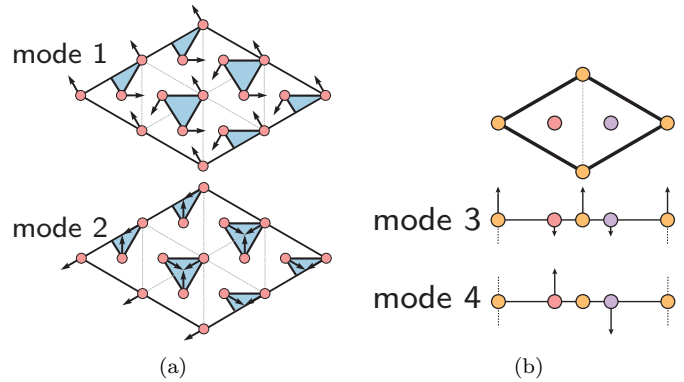


FIG. 1. Symmetry-adapted distortion modes on a  $\sqrt{3} \times \sqrt{3}$  supercell of the triangular lattice: (a) Connects the triangular lattice to the Kagomé net (b) dissociates a triangular layer into three separate triangular layers with one atom in each layer. The SI contains a mathematical description of these modes[39]

The Friauf-Laves phases and related crystal structures can be viewed as stackings of Kagomé nets interleaved by triangular layers [16, 17]. We start by showing how the Kagomé nets and triangular layers of the Laves phases can be formed by a superposition of symmetry adapted collective displacements of the triangular close-packed layers of hcp.

The collective displacements have the periodicity of a  $\sqrt{3}a \times \sqrt{3}a$  supercell of the primitive cell of the tri-

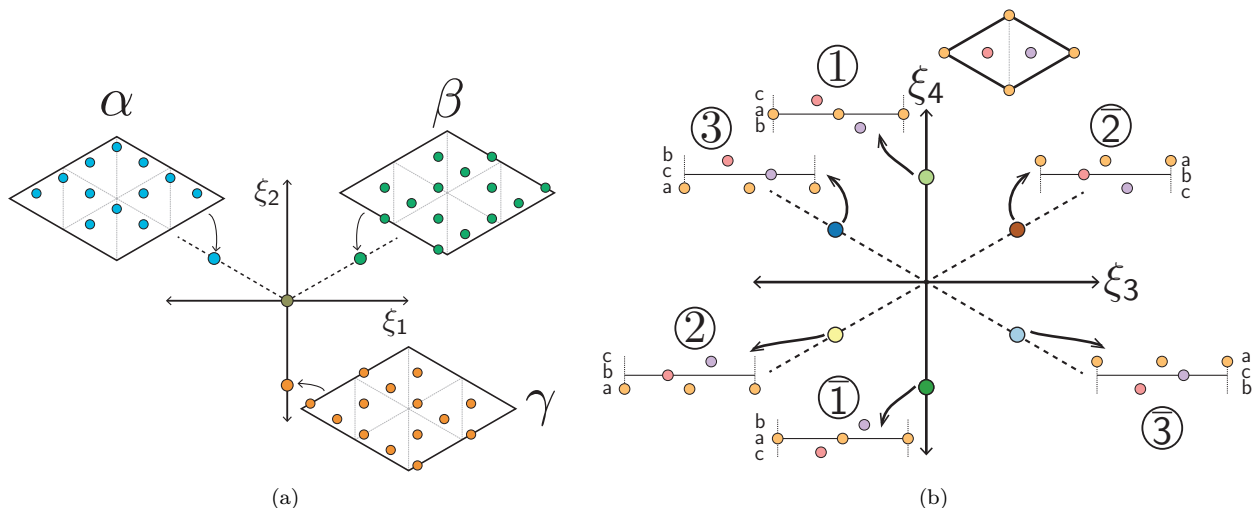


FIG. 2. (a) Amplitudes of symmetry-adapted distortion modes on a triangular lattice that connect them to the Kagomé net (b) Six translational variants of a vertically split triangular lattice.

71 angular lattice (where  $a$  is the lattice parameter). Fig-  
 72 ure 1a shows two symmetry-adapted collective displace-  
 73 ments that link a triangular lattice to a Kagomé net. The  
 74 collective displacements are easily understood when con-  
 75 sidering the periodically repeating triplet clusters on the  
 76 triangular lattice illustrated in fig. 1a. The first collective  
 77 distortion, with amplitude  $\xi_1$ , displaces atoms tangen-  
 78 tially relative to the centers of the periodically repeating  
 79 triplet clusters. The second collective distortion, with  
 80 amplitude  $\xi_2$ , corresponds to a dilation of the periodi-  
 81 cally repeating triplet clusters of sites. Three transla-  
 82 tional variants of the Kagomé net can be generated from  
 83 the triangular lattice in the space spanned by  $\xi_1$  and  $\xi_2$ .  
 84 A negative value of  $\xi_2$  generates one Kagomé net, that  
 85 we label  $\gamma$ . The two other Kagomé nets,  $\alpha$  and  $\beta$ , can be  
 86 generated by a combination of  $\xi_1$  and  $\xi_2$  residing on the  
 87 dashed lines in fig. 2a that are rotated by  $-120^\circ$  and  $120^\circ$   
 88 relative to the negative  $\xi_2$  axis.

89 Out of plane collective displacements having the trans-  
 90 lational periodicity of the  $\sqrt{3}a \times \sqrt{3}a$  supercell can be  
 91 defined that split a triangular lattice into three parallel  
 92 triangular lattices with an "abc" stacking, each having  
 93 one third of the sites of the original triangular lattice. Six  
 94 translational variants of a vertically split triangular lat-  
 95 tice can be generated with two symmetry adapted collec-  
 96 tive displacements. One collective displacement with am-  
 97 plitude  $\xi_3$  moves two atoms within a  $\sqrt{3}a \times \sqrt{3}a$  supercell  
 98 down and the third atom up as illustrated in fig. 1b. The  
 99 other collective displacement with amplitude  $\xi_4$  "dissoci-  
 100 ates" the layer, in that one of the atoms in the  $\sqrt{3}a \times \sqrt{3}a$   
 101 supercell moves up, another moves down and the third  
 102 atom remains fixed. The six translationally equivalent  
 103 split configurations reside on a hexagon in  $\xi_3$ - $\xi_4$  space as  
 104 illustrated in fig. 2b.

105 The collective distortions illustrated in figs. 1a and 1b

106 can be used to generate a hierarchy of crystal structures  
 107 starting from hcp that consist of different stackings of  
 108 Kagomé nets and triangular layers. The hierarchy can  
 109 be generated by converting the A layers of an ABAB  
 110 stacked hcp crystal to a Kagomé net upon activation of  
 111 the  $\xi_1 - \xi_2$  distortion amplitudes of fig. 2a, and by disso-  
 112 ciating a subset of the B layers into three new triangular  
 113 layers by activating the  $\xi_3 - \xi_4$  amplitudes. The particu-  
 114 lar distortion of each A layer will be represented with  
 115  $t, \alpha, \beta, \gamma$ , where  $t$  refers to the undistorted layer, and the  
 116 other three letters correspond to the Kagomé variants.  
 117 The dissociation of the B layer will be represented as  
 118  $1, 2, 3, \bar{1}, \bar{2}, \bar{3}$  as shown in fig. 2b, with the undistorted  
 119 layer denoted as 0. With this notation the hcp structure  
 120 becomes " $t0$ ".

121 The first generation of hierarchical phases derived from  
 122 hcp is  $\{\alpha 0\}$ .  $\alpha 0$  consists of identical Kagomé nets gener-  
 123 ated from the A layers interleaved by undissociated tri-  
 124 angular B layers. The resulting crystal structure corre-  
 125 sponds to the  $\text{CaCu}_5$  prototype. The second generation  
 126 results in crystal structures that are periodic over four  
 127 layers and include:  $\{t0\alpha 0, \alpha 2\gamma\bar{2}\}$ . The first member can  
 128 be formed by combining the zeroth generation (i.e. hcp)  
 129 with the first generation. The second distortion,  $\alpha 2\gamma\bar{2}$ ,  
 130 is new, and results in the formation of the well-known  
 131  $\text{C14-MgZn}_2$  Laves crystal structure.

132 Several other well-known structure prototypes appear  
 133 at higher generations. Examples are listed in table I. The  
 134 two other Laves phase prototypes emerge as distortions  
 135 in the third and fourth generations. The structure proto-  
 136 types  $\text{Ce}_2\text{Ni}_7$ ,  $\text{Be}_3\text{Nb}$  and  $\text{Co}_7\text{Gd}_2$  form in the sixth and  
 137 ninth generations. Table I shows that the structural dis-  
 138 tortions of fig. 2a and fig. 2b are capable of connecting a  
 139 significant fraction of common intermetallic compounds  
 140 to hcp. Similar relationships may be drawn between hcp

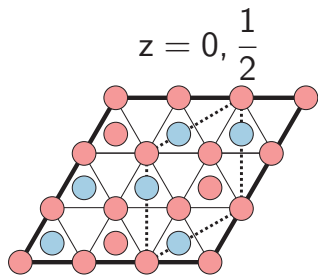


FIG. 3. Ordering on hcp related to  $\text{MgZn}_2$  (C14). The atoms shown in red are the smaller majority specie, while the blue atoms are the minority specie. Atoms on a grid point are in the A layer while atoms in the middle of a triangle are on the B layer.

141 and structures such as the  $\sigma$  and  $\mu$  phases. However,  
142 these transformations are slightly more complicated as  
143 they require additional interstitial atoms.

Structure Prototype	Structure Label
$\text{CaCu}_5$	$\alpha 0$
$\text{MgZn}_2$ (C14)	$\beta 1\gamma\bar{1}$
$\text{MgCu}_2$ (C15)	$\beta 1\gamma\bar{2}\alpha 3$
$\text{MgNi}_2$ (C36)	$\beta 1\gamma\bar{1}\beta\bar{3}\alpha 3$
$\text{Ce}_2\text{Ni}_7$	$\beta\bar{3}\alpha 0\alpha 0\alpha 3\beta 0\beta 0$
$\text{Be}_3\text{Nb}$	$\alpha 2\gamma 0\beta 1\gamma 0\gamma\bar{2}\alpha 0$
$\text{Co}_7\text{Gd}_2$	$\alpha 0\alpha 3\beta 0\beta 0\beta 1\gamma 0\gamma 0\gamma\bar{2}\alpha 0$

TABLE I. Known crystal structure prototypes their corresponding labels within the family of structures that can be derived from hcp.

144 Most topologically close-packed phases are compounds  
145 containing two or more chemical species. The majority  
146 element X is usually smaller than the minority element  
147 Y, with both having strong site preferences within the  
148 crystal structure of the topologically close-packed phase.  
149 There are, therefore, very specific orderings in the pre-  
150 cursor hcp phase that result in different topologically  
151 close-packed phases upon application of the distortions  
152 of fig. 2a and fig. 2b. Figure 3 illustrates the decoration  
153 of Mg and Zn on an hcp crystal structure that is necessary  
154 to form perfect  $\text{MgZn}_2$  having the C14 crystal structure.  
155 The triangular layer of hcp that converts to a Kagomé  
156 net only contains the smaller Zn atoms (i.e. X) while  
157 the other layer that dissociates into three new triangular  
158 layers contains two Mg atoms (i.e. Y) for every Zn atom.  
159 The precursor ordering in hcp has a  $\sqrt{3}a \times \sqrt{3}a$  super cell  
160 on the triangular lattice. Inspection of known topologi-  
161 cally close-packed phases reveals that the precursor hcp  
162 layers that transform into a Kagomé net will invariably  
163 only consist of the smaller X. The layers that dissociate  
164 or remain triangular, are found to be either pure X or  
165 exhibit a similar  $\sqrt{3}a \times \sqrt{3}a$  ordering of X and Y as in  
166 the precursor hcp phase of  $\text{MgZn}_2$  illustrated in fig. 3.  
167 First-principles electronic structure[40] calculations

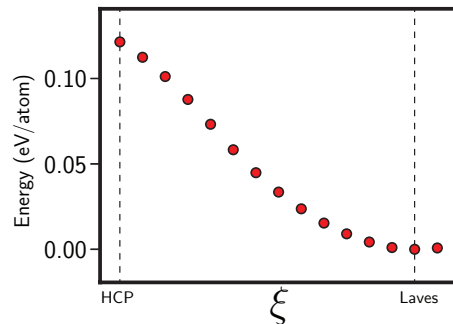


FIG. 4. Energy landscape along the path connecting the ordering of fig. 3 to C14 in the Mg-Zn binary alloy.

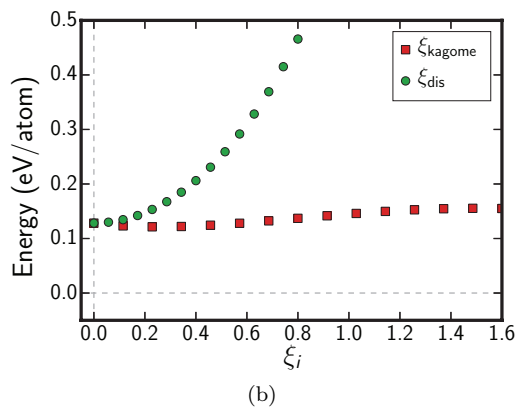
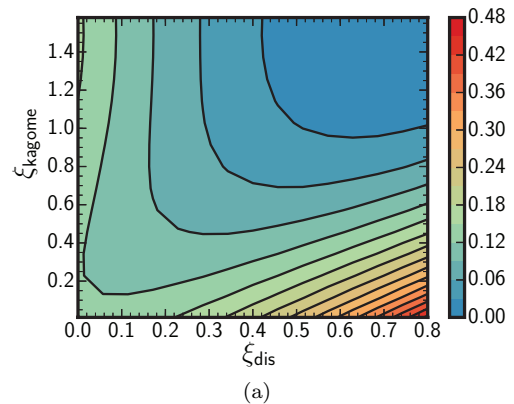


FIG. 5. Energy landscape connecting the hcp crystal structure with the C14 Laves phase in Mg-Zn. (a) Energy as a function of the dissociative and Kagomé modes. (b) Sections of the energy landscape showing that the hcp structure is unstable with respect to a transformation towards Kagomé nets.

168 can reveal how facile the pathways described above are.  
169 Figure 3 shows the Mg-Zn ordering in the hcp precursor  
170 phase. It is evident from fig. 4 that there is no barrier  
171 separating the two crystal structures, with the hcp form  
172 of  $\text{MgZn}_2$  being dynamically unstable. Any local ordering  
173 on hcp that is similar to fig. 3 will experience a driving  
174 force to spontaneously collapse into C14- $\text{MgZn}_2$ .

175 Figure 5a shows the energy as a function of an ampli-  
176 tude  $\xi_{\text{Kagomé}}$  that converts the A layers of hcp  $\text{MgZn}_2$

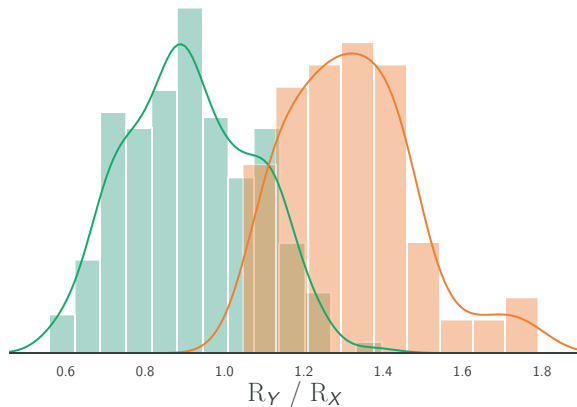


FIG. 6. Density distributions of  $X_2Y$  orderings on hcp that are stable in hcp (green) and collapse to form the C15 Laves phase (orange) across a range of chemistries and radius ratios. Metallic radii are used for each element[39, 41].

(consisting only of Zn) into Kagomé nets and a second amplitude  $\xi_{\text{dis}}$  that dissociates the B layers (consisting of Mg and Zn in a 2:1 ratio) into three new triangular layers. As is clearly revealed by fig. 5a a combination of both  $\xi_{\text{Kagomé}}$  and  $\xi_{\text{dis}}$  is necessary to achieve the steepest descent from hcp to C14. However, fig. 5b, which plots the energy as a function of either  $\xi_{\text{Kagomé}}$  or  $\xi_{\text{dis}}$ , shows that distortions along  $\xi_{\text{dis}}$  do not lead to an instability of the hcp crystal. Rather, it is a distortion of the A layer to form Kagomé nets that induces the spontaneous formation of the Laves phase. This can be attributed to the size mismatch between the Mg and Zn atoms. The larger Mg atoms reside in the B layers of hcp and require the formation of Kagomé nets containing large hexagonal openings before they can dissociate upwards and downwards. The formation of the Kagomé nets are essential to reduce the overcrowding that exists when arranging two species with very different sizes within the close-packed hcp crystal structure.

The instability of hcp with respect to a spontaneous collapse to a Laves phase is not restricted to  $\text{MgZn}_2$ , but is predicted to occur quite uniformly among a large number of  $X_2Y$  compounds when Y is substantially larger than X. We investigated the stability of 511  $X_2Y$  compounds in the hcp crystal structure with respect to a spontaneous collapse to C15 with DFT (for more details on the chemistries of these compounds see SI[39]). Figure 6 shows normalized histograms of the compounds that remain stable in hcp and those that spontaneously collapse to C15, plotted as a function of the ratio of the metallic radii[41] of Y and X, i.e.  $R_Y/R_X$ . As is clearly revealed by fig. 6, compounds with large  $R_Y/R_X$  radius ratios are found to undergo a barrierless structural phase transition to C15 when ordered on hcp, while those with smaller radius ratios are stable in hcp. There is some

overlap between the two distributions around  $R_Y/R_X \approx 1.1$ , however, a large number of these compounds consist of refractory elements, which are likely to adopt radii within compounds that differ from their metallic radii due to charge transfer [42]. As shown in the SI [39], the qualitative trends of fig. 6 remain unchanged when using different definitions for the atomic radii[43, 44]. These results suggest that atoms with large size mismatch are unlikely to order on close-packed parent crystal structures due to the facile pathways discovered here that will convert hcp into the more favorable Frank Kasper phases.

The discovery of the Bain and Burgers paths almost a century ago played a crucial role in rationalizing a large number of phase transformations in the solid state. The facile crystallographic pathways reported here will similarly enable a deeper understanding of phase transformation mechanisms between close-packed structures and Frank-Kasper phases, which rank among the most common structures adopted by intermetallic compounds[18]. Frank-Kasper phases have traditionally been viewed as crystallographically very distinct from the common close-packed structures. We have demonstrated that hcp can be converted into a large number of the Frank-Kasper phases with minimal structural distortions. This facile pathway suggests that the nucleation of Frank-Kasper phases within supersaturated hcp solid solutions does not require large structural rearrangements but rather can occur coherently and in a continuous manner. Composition and ordering fluctuations within the hcp parent crystal structure can enable the formation of local precursor orderings, that can then collapse into a coherent nucleus of the incipient Frank-Kasper phase through a dynamical instability. Nucleation by this mechanism, while possibly accompanied by coherency strains, would bypass the need for energetically costly incoherent interfaces and will result in well defined orientational relationships, at least in the early stages when coherency strains play less of a role. A survey of binary alloy phase diagrams suggests several material systems where this mechanism may be exploited[45]. In fact, the observed orientation relationship for Laves phases in magnesium alloys[23, 46–52] ( $(0001)_{\text{Laves}} \parallel (0001)_{\text{hcp}}, [1\bar{1}00]_{\text{Laves}} \parallel [11\bar{2}0]_{\text{hcp}}$ ) is in agreement with our prediction of the basal plane of the Laves structure being aligned with that of hcp and the  $\sqrt{3} \times \sqrt{3}$  direction of hcp being along one of the basal lattice vectors of the Laves phase. Instead of treating the nucleation of Frank-Kasper phases within hcp as a singularity, as conventionally done for reconstructive phase transitions, the simple hcp to Frank-Kasper phase pathway along with the predicted instabilities of the precursor ordering within hcp suggests a continuous nucleation mechanism. These insights are expected to unlock new design routes to either encourage (or suppress) the formation of topologically close-packed phases in a wide variety of metallic, polymeric and colloidal systems.

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\* anirudh@ucsb.edu

† avdv@ucsb.edu

- [1] X. Huang, G. J. Ackland, and K. M. Rabe, *Nature materials* **2**, 307 (2003).
- [2] Y. Ogawa, D. Ando, Y. Sutou, and J. Koike, *Science* **3**, 30 (2016).
- [3] N. A. Zarkevich and D. D. Johnson, *Physical Review Letters* **113** (2014), 10.1103/PhysRevLett.113.265701.
- [4] K. Otsuka and X. Ren, *Progress in Materials Science* **50**, 511 (2005).
- [5] S. Sarkar, X. Ren, and K. Otsuka, *Physical Review Letters* **95** (2005), 10.1103/PhysRevLett.95.205702.
- [6] Y. Ji, D. Wang, X. Ding, K. Otsuka, and X. Ren, *Physical Review Letters* **114** (2015), 10.1103/PhysRevLett.114.055701.
- [7] B. Dutta, A. Çakır, C. Giacobbe, A. Al-Zubi, T. Hickel, M. Acet, and J. Neugebauer, *Physical Review Letters* **116** (2016), 10.1103/PhysRevLett.116.025503.
- [8] J. Liu, T. Gottschall, K. P. Skokov, J. D. Moore, and O. Gutfleisch, *Nature Materials* **11**, 620 (2012).
- [9] T. M. Pollock, *Science* **328**, 986 (2010).
- [10] J. H. Chen, E. Costan, M. a van Huis, Q. Xu, and H. W. Zandbergen, *Science (New York, N.Y.)* **312**, 416 (2006).
- [11] E. Bain and N. Dunkirk, *Transactions of the American Institute of Mining and Metallurgical Engineers* **70** (1924).
- [12] W. Burgers, *Physica* **1**, 561 (1934).
- [13] J. C. Jamieson, *Science* **140**, 72 (1963).
- [14] D. de Fontaine, *Acta Metallurgica* **18**, 275 (1970).
- [15] D. R. Trinkle, R. G. Hennig, S. G. Srinivasan, D. M. Hatch, M. D. Jones, H. T. Stokes, R. C. Albers, and J. W. Wilkins, *Physical Review Letters* **91** (2003), 10.1103/PhysRevLett.91.025701.
- [16] F. Frank and J. S. Kasper, *Acta Crystallographica* **11**, 184 (1958).
- [17] F. Frank and J. S. Kasper, *Acta Crystallographica* **12**, 483 (1959).
- [18] R. Ferro and A. Saccone, *Intermetallic Chemistry* (Pergamon, 2008).
- [19] A. K. Sinha, *Progress in Materials Science* **15**, 81 (1972).
- [20] F. Stein, M. Palm, and G. Sauthoff, *Intermetallics* **12**, 713 (2004).
- [21] J. C. Duthie and D. G. Pettifor, *Physical Review Letters* **38**, 564 (1977).
- [22] W. Zhang, R. Yu, K. Du, Z. Cheng, J. Zhu, and H. Ye, *Physical Review Letters* **106** (2011), 10.1103/PhysRevLett.106.165505.
- [23] H. Xie, H. Pan, Y. Ren, L. Wang, Y. He, X. Qi, and G. Qin, *Physical Review Letters* **120** (2018), 10.1103/PhysRevLett.120.085701.
- [24] M. F. Chisholm, S. Kumar, and P. Hazledine, *Science* **307**, 701 (2005).
- [25] S. Khmelevskiy, P. Mohn, J. Redinger, and M. Weinert, *Physical Review Letters* **94** (2005), 10.1103/PhysRevLett.94.146403.
- [26] J. Feng, N. W. Ashcroft, and R. Hoffmann, *Physical Review Letters* **98** (2007), 10.1103/PhysRevLett.98.247002.
- [27] T. Matsuoka, M. Debessai, J. J. Hamlin, A. K. Gangopadhyay, J. S. Schilling, and K. Shimizu, *Physical Review Letters* **100** (2008), 10.1103/PhysRevLett.100.197003.
- [28] A. Alam and D. D. Johnson, *Physical Review Letters* **107** (2011), 10.1103/PhysRevLett.107.206401.
- [29] M.-C. Marinica, F. Willaime, and J.-P. Crocombette, *Physical Review Letters* **108** (2012), 10.1103/PhysRevLett.108.025501.
- [30] D. R. Nelson and F. Spaepen, in *Solid State Physics*, Vol. 42 (Elsevier, 1989) pp. 1–90.
- [31] J. Fransaer, A. V. Wagner, and F. Spaepen, *Journal of Applied Physics* **87**, 1801 (2000).
- [32] J. Zemp, M. Celino, B. Schonfeld, and J. Loffler, *Physical Review Letters* **115** (2015), 10.1103/PhysRevLett.115.165501.
- [33] A.-P. Hynninen, J. H. J. Thijssen, E. C. M. Vermolen, M. Dijkstra, and A. van Blaaderen, *Nature Materials* **6**, 202 (2007).
- [34] S. Lee, C. Leighton, and F. S. Bates, *Proceedings of the National Academy of Sciences* **111**, 17723 (2014).
- [35] C. X. Du, G. van Anders, R. S. Newman, and S. C. Glotzer, *Proceedings of the National Academy of Sciences* **114**, E3892 (2017).
- [36] K. Kim, A. Arora, R. M. Lewis, M. Liu, W. Li, A.-C. Shi, K. D. Dorfman, and F. S. Bates, *Proceedings of the National Academy of Sciences* **115**, 847 (2018).
- [37] S. Hajiw, B. Pansu, and J.-F. Sadoc, *ACS Nano* **9**, 8116 (2015).
- [38] B. Cabane, J. Li, F. Artzner, R. Botet, C. Labbez, G. Bareigts, M. Sztucki, and L. Goehring, *Physical Review Letters* **116** (2016), 10.1103/PhysRevLett.116.208001.
- [39] See Supplemental Material at [URL will be inserted by publisher] for a detailed description of the transformation pathways and calculations of the correlation between radius ratio and the onset of structural transformation. ( ).
- [40] Total energies were calculated within the Perdew-Burke-Ernzerhof (PBE) parameterization of the generalized gradient approximation (GGA)[53] to density functional theory (DFT)[54, 55]. The *Vienna Ab-Initio Simulation Package* (VASP)[56–59] was used to relax configurations. All degrees of freedom were relaxed to test the stability of specific orderings across a range of chemistries. The energy landscape along the pathway connecting hcp to the C14 Laves phase was calculated by minimizing the energy with respect to the volume while fixing all other degrees of freedom. An energy cutoff of 600 eV was used for the plane-wave basis set. The k-point grids were chosen to contain 71 k-points per Å<sup>-1</sup>. Structure interpolations and symmetry analysis of the pathways were carried out with the CASM[60] and ISODISPLACE[61] software packages.

- 331 [41] N. N. Greenwood and A. Earnshaw, *Chemistry of the Elements*, 2nd ed. (Butterworth-Heinemann, 1997).
- 332 [42] A. Dwight, Transactions of the American Society of Metals **53**, 479 (1961).
- 333 [43] E. Clementi, D. L. Raimondi, and W. P. Reinhardt, The Journal of Chemical Physics **47**, 1300 (1967).
- 334 [44] J. C. Slater, The Journal of Chemical Physics **41**, 3199 (1964).
- 335 [45] A survey of the Landolt-Bornstein database[62] and the ASM binary phase diagram handbook[63] suggests that the
- 336 following binary alloys contain two-phase regions between a disordered hcp solid solution and a structure belonging to
- 337 the family of orderings described in this study: Be-{Ag,Ti},Ca-Mg,Co-{Mg,Sc},Cr-{Hf,Ti,Zr},Hf-{Mo,V,W},Mg-Yb,Mn-
- 338 {Y,Zr},Mo-Zr,Re-{Sc,Y},{V,W}-Zr.
- 339 [46] J.-F. Nie, Metallurgical and Materials Transactions A **43**, 3891 (2012).
- 340 [47] C. Mendis, K. Oh-ishi, and K. Hono, Materials Science and Engineering: A **527**, 973 (2010).
- 341 [48] B. Langelier, A. Korinek, P. Donnadiou, and S. Esmaili, Materials Characterization **120**, 18 (2016).
- 342 [49] A. Suzuki, N. Saddock, J. Jones, and T. M. Pollock, Acta Materialia **53**, 2823 (2005).
- 343 [50] T. Homma, S. Nakawaki, K. Oh-ishi, K. Hono, and S. Kamado, Acta Materialia **59**, 7662 (2011).
- 344 [51] X. Gao and J. Nie, Scripta Materialia **56**, 645 (2007).
- 345 [52] T. Homma, S. Nakawaki, and S. Kamado, Scripta Materialia **63**, 1173 (2010).
- 346 [53] J. P. Perdew, K. Burke, and M. Ernzerhof, Physical Review Letters **77**, 3865 (1996).
- 347 [54] P. Hohenberg and W. Kohn, Physical Review **136**, 864 (1964).
- 348 [55] W. Kohn and L. Sham, Physical Review **140**, 1133 (1965).
- 349 [56] G. Kresse and J. Hafner, Physical Review B **47**, 558 (1993).
- 350 [57] G. Kresse and J. Furthmüller, Physical Review B **54**, 11169 (1996).
- 351 [58] G. Kresse and J. Furthmüller, Computational Materials Science **6**, 15 (1996).
- 352 [59] G. Kresse and J. Hafner, Physical Review B **49**, 14251 (1994).
- 353 [60] CASM Developers, “CASM: A Clusters Approach to Statistical Mechanics,” (2016).
- 354 [61] B. J. Campbell, H. T. Stokes, D. E. Tanner, and D. M. Hatch, Journal of Applied Crystallography **39**, 607 (2006).
- 355 [62] *Landolt-Börnstein Database* (Springer).
- 356 [63] *Alloy Phase Diagrams*, 10th ed., ASM Handbook, Vol. 3 (1992).