



This is the accepted manuscript made available via CHORUS. The article has been published as:

## Atom-Pair Kinetics with Strong Electric-Dipole Interactions

N. Thaicharoen, L. F. Gonçalves, and G. Raithel

Phys. Rev. Lett. **116**, 213002 — Published 27 May 2016

DOI: 10.1103/PhysRevLett.116.213002

## Atom-pair kinetics with strong electric-dipole interactions

N. Thaicharoen<sup>1\*</sup>, L.F. Gonçalves<sup>1,2</sup>, and G. Raithel<sup>1</sup>

<sup>1</sup>Department of Physics, University of Michigan, Ann Arbor, Michigan 48109, USA and

<sup>2</sup>Instituto de Física de São Carlos, Universidade de São Paulo,

Caixa Postal 369, 13560-970, São Carlos, SP, Brasil

(Dated: April 27, 2016)

Rydberg-atom ensembles are switched from a weakly- into a strongly-interacting regime via adiabatic transformation of the atoms from an approximately non-polar into a highly dipolar quantum state. The resultant electric dipole-dipole forces are probed using a device akin to a field ion microscope. Ion imaging and pair-correlation analysis reveal the kinetics of the interacting atoms. Dumbbell-shaped pair correlation images demonstrate the anisotropy of the binary dipolar force. The dipolar  $C_3$  coefficient, derived from the time dependence of the images, agrees with the value calculated from the permanent electric-dipole moment of the atoms. The results indicate many-body dynamics akin to disorder-induced heating in strongly coupled particle systems.

PACS numbers: 32.80.Ee, 34.20.Cf

Dipolar and van der Waals interactions between atoms and molecules affect the properties of matter on microscopic and macroscopic scales. On the quantum level, the distinctions between van der Waals and electric dipoledipole interactions are in the overall interaction strength, the scaling with the internuclear separation  $(R^{-6})$  for the van der Waals and  $\mathbb{R}^{-3}$  for the electric-dipole interactions), and the (an)isotropy behavior. Highly excited Rydberg atoms present an ideal platform to study these interactions in binary and few-body quantum systems because Rydberg-atom interactions are generally strong and widely tunable between dipole-dipole, van der Waals and other types. In several previous experiments, internal-state changes due to electric dipole-dipole [1-4] and van der Waals [5, 6] interactions between Rydberg atoms have been studied using spectroscopic measurements of level shifts. Another way to study these interactions is through measurements of their mechanical effects on the center-of-mass motion of the atoms. This method has been applied to study van der Waals interactions between Rydberg atoms [7, 8] and magnetic-dipole interactions in Bose-Einstein condensates (which have the same R-scaling as the electric-dipole interactions) [9, 10]. Methods from optical and electron microscopy have been adapted to image Rydberg-atom systems with singleparticle spatial resolution, revealing many-body quantum structures such as Rydberg-atom crystals [11–13] and enabling advanced studies of the Rydberg excitation blockade [14–16].

In our work we employ an adiabatic quantum-state preparation method and a modified field ion microscope [17–19] with single-atom resolution to measure the atom kinetics that result from the dipolar force. We overcome the density limit imposed by the excitation blockade by initially preparing Rydberg atoms under conditions where they are only subject to weak van der Waals interactions. This allows us to prepare Rydberg atom samples with relatively small interatomic separations. In

order to switch on strong dipole-dipole interactions, the atoms are subsequently transferred into a highly dipolar state via a Landau-Zener adiabatic passage through an avoided crossing [20, 21]. After the adiabatic state transformation, the direction of the permanent atomic dipoles is locked to the direction of an external electric field. After initialization, the atoms move under the influence of the strong dipolar forces for a variable interaction time. The center-of-mass positions of the Rydberg atoms are then detected by field ionization [22] and ion imaging [23]. From the recorded images, we calculate the spatial pair correlation functions between the Rydberg atoms [24], which directly show the anisotropic character of the dipolar force between two atoms and allow us to measure the  $C_3$  dipole-dipole interaction strength parameter. The experiment is performed under two conditions, one in which the direction of the atomic dipoles

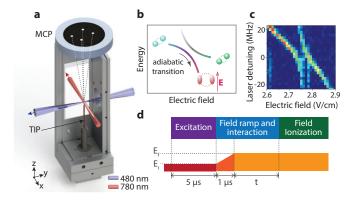


FIG. 1. (a) Experimental set-up. (b) Rydberg atoms initially prepared in an S-like state (blue circle) become adiabatically transferred into a highly dipolar state (pink oval), when passing through an avoided crossing in the Stark energy level diagram in the applied electric field  ${\bf E}$ . The diameters of the dashed circles indicate binary-interaction strengths. (c) Measured Stark spectra at the  $6^{\rm th}$  crossing between the  $50S_{1/2}$  state and n=47 hydrogen-like states. (d) Timing sequence.

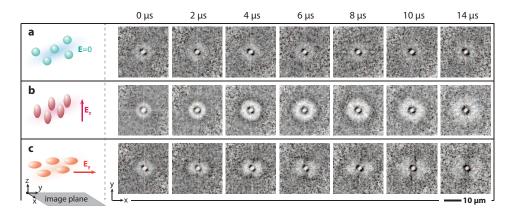


FIG. 2. Experimental pair correlation images for the indicated wait times. The linear grayscale ranges from 0.4 (white) to 1.5 (black), where values of 1, <1, and >1 indicate no correlation, anticorrelation, and positive correlation, respectively. The illustrations on the left show the electric-field and dipolar alignments relative to the object/image planes for the three data rows. (a) Zero field. Atoms undergo weak van der Waals interaction. (b) Applied electric field in z-direction. The atomic dipoles are aligned along z, perpendicular to the image (xy) plane. Binary interactions are azimuthally symmetric about z and primarily repulsive. (c) Applied electric field in y-direction. The atomic dipoles are aligned along y, in-plane with the image plane. Binary interactions are mixed attractive / repulsive, leading to anisotropic images showing repulsion along x and attraction along y. The dark rings immediately around the centers are experimental artifacts.

is transverse and another in which it is perpendicular to the observation plane.

The experimental setup is shown in Fig. 1(a). Cold  $^{85}{\rm Rb}$  atoms in the  $5S_{1/2}$  state, prepared at a density of  $\gtrsim 10^{10}~{\rm cm^{-3}}$  and a temperature of  $\sim 100~\mu{\rm K}$ , undergo two-photon Rydberg excitation into the  $50S_{1/2}$  state by simultaneously applying 780- and 480-nm laser pulses with 5  $\mu{\rm s}$  duration and  $\approx 1~{\rm GHz}$  red-detuning from the  $5P_{3/2}$  intermediate state. The 780-nm beam propagates along the x-direction (coordinate frame defined in Fig. 1(a)) and has a Gaussian beam parameter of  $w_0\approx 70~\mu{\rm m}$ . The 480-nm beam propagates in the xy plane, forms an angle of approximately 70° with the 780 nm beam, and has  $w_0\approx 8~\mu{\rm m}$ . Therefore, the excitation volume is cylindrical and aligned approximately along the y-direction. The number of Rydberg excitations in each sample is on the order of ten.

To introduce strongly dominant dipole-dipole forces into the Rydberg-atom sample, after excitation into the  $50S_{1/2}$  state the atoms are adiabatically transferred into a linear Stark state by sweeping the applied electric field through the avoided crossing between those two states [20, 21]. The permanent electric dipole moment of the Stark state is about 16 times larger than that of the initial (perturbed)  $50S_{1/2}$  state, leading to much stronger, longer-range and anisotropic interactions, as sketched in Fig. 1(b). We perform the experiment at the 6<sup>th</sup> avoided crossing (counted from zero applied field) of rubidium  $50S_{1/2}$  with the manifold of n=47 hydrogenlike dipolar states, located at a calculated electric-field value of 2.757 V/cm. We use Stark spectroscopy to verify the position of this avoided crossing, as shown in Fig. 1(c). This particular avoided crossing is chosen because its gap size is large enough that we are able to experimentally resolve it in the Stark spectra, and it is large enough to ensure near-perfect adiabatic passage behavior.

The timing sequence of the experiment is shown in Fig. 1(d). The  $50S_{1/2}$ -like atoms, prepared at 0.1 V/cm below the avoided crossing, have a small electric dipole moment of  $1.26 \times 10^{-27}$  Cm. They are adiabatically transferred into the dipolar target state by linearly increasing the electric field by 0.2 V/cm within 1  $\mu$ s. The dipolar state has a dipole moment of  $20.3 \times 10^{-27}$  Cm, corresponding to an increase of the dipolar interaction strength by a factor of 260. The sweep duration is slow enough that the adiabatic transfer from the  $50S_{1/2}$ -like into the dipolar state has an efficiency very close to unity [25]. The sweep duration is also short enough that during the adiabatic transfer the atoms do not move significantly. Hence, the state switching is practically instantaneous with respect to the center-of-mass motion of the atoms. After the switch, the electric field is kept constant for the duration of the interaction time, t. We study the atom kinetics that follow from the dipolar force as a function of t.

To perform atom imaging, Rydberg atoms are ionized by sudden application of a positive high voltage to a tip imaging probe (TIP). Ions are accelerated by the TIP electric field towards a microchannel plate (MCP). Every detected ion results in a blip produced by the MCP-phosphor detector assembly, which reveals the center-of-mass position of its parent Rydberg atom at the time of ionization. The excitation region (object plane) is about 300  $\mu{\rm m}$  above the TIP. For this distance, the magnification is calibrated to be 200  $\pm$  10% [25]. We take 10 000

images in each data set. Using a peak-detection algorithm [24], the images are processed into a data structure in which each record contains the total number of detected ions and the ion impact positions in an individual image. The 5000 records with the highest ion numbers are processed into a sample-averaged pair correlation image, which is normalized such that at large distances it approaches the value of one. The average pair correlation images yield the information on the atom-pair kinetics.

For two identical dipoles pointing along the same direction, given by the direction of the applied electric field, the dipolar force has radial and angular components

$$F_{\rm R} = \frac{3p^2}{R^4} [1 - 3\cos^2(\Theta)]$$

$$F_{\Theta} = \frac{-3p^2}{R^4} [2\cos(\Theta)\sin(\Theta)]$$
(1)

where  $\Theta$  is the angle between the internuclear separation vector,  $\mathbf{R}$ , and the dipole vectors,  $\mathbf{p}$ .

When the applied electric field is perpendicular to the detection (xy) plane, the atomic dipoles point in the z-direction. Since the excitation blockade radius is on the order of the diameter of the 480-nm excitation beam, for most dipole-dipole-interacting atom pairs the angle  $\Theta$  in Eq. 1 is  $\approx \pi/2$ . Hence, for the vast majority of atom pairs the dipole force is repulsive, with  $F_{\rm R} \approx 3p^2/R^4$ , and azimuthally symmetric about the line of sight. The pair correlation images at different interaction times, presented in Fig. 2(b), are azimuthally symmetric at all times. With increasing interaction time, the (projected) radius of the region that is largely devoid of pair-correlation events increases, reflecting an increase of the interatomic separation due to repulsive dipole-dipole interactions. We also find that the pair correlation is enhanced at a certain radius that increases in time; this radius approximately doubles over a time of 14  $\mu$ s.

The expansion in Fig. 2(b), which is due to dipole-dipole interaction, is considerably faster than that due to repulsive van der Waals interactions between  $50S_{1/2}$  atoms at zero electric field, shown in Fig. 2(a). The van der Waals interaction does not cause any significant expansion over the time scale in Fig. 2. In previous work [7] it was found that even  $70S_{1/2}$  Rb Rydberg atoms, which interact about 50 times more strongly than  $50S_{1/2}$  Rb Rydberg atoms (present case), exhibit significant repulsion effects only after about 30  $\mu$ s. Hence, a cursory comparison of the pair correlation data in Figs. 2(a) and 2(b) already shows that the interaction between the dipolar atoms must be one to two orders of magnitude stronger than the interaction between the non-polar, van der Waals-interacting atoms.

The angular integrals I(R) of the pair correlation images from Fig. 2(b) are shown in Fig. 3(a). The most probable separation between Rydberg-atom pairs,  $R_c(t)$ , is determined by local parabolic fits to I(R) in the vicinity of the peaks found within the shaded region in

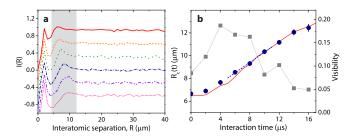


FIG. 3. (a) Angular integrals I(R) of the pair correlation images in Fig. 2(b) at wait times (from top to bottom): 0  $\mu$ s, 2  $\mu$ s, 4  $\mu$ s, 6  $\mu$ s, 8  $\mu$ s, and 10  $\mu$ s. The y axis is for the 0  $\mu$ s curve; for clarity, the other curves are shifted down in equidistant steps of 0.3. (b) Interatomic separations  $R_c(t)$  between Rydberg-atom pairs as a function of interaction time (left axis) from experiment (blue circles) and simulation (red line). The blue dashed line shows a linear fit to the experimental data between 6  $\mu$ s to 10  $\mu$ s. The gray squares show the visibility of the experimental pair correlation enhancement (right axis).

Fig. 3(a). The blue circles in Fig. 3(b) show the resulting  $R_c(t)$  values that represent the most probable radial atom-pair trajectory projected into the xy-plane. It is seen that atom pairs are initially prepared at a preferential separation  $R_c(0) = 6.7 \pm 0.7 \ \mu \text{m}$ , controlled by the excitation-laser detuning and the atomic interaction strength before the adiabatic state transformation [24]. The large positive acceleration observed subsequent to the state transformation is due to the strong repulsive dipole-dipole interaction that occurs for angles  $\Theta$  near  $\pi/2$ . The acceleration apparently diminishes and changes sign from positive to negative at an interaction time near 8  $\mu$ s, where  $d^2R_c(t)/dt^2 \sim 0$ .

To verify that the observed rapid expansion in the xyplane is consistent with the known permanent electric
dipole moment of the atoms after adiabatic transformation, we extract the  $C_3$  coefficient using conservation of
energy between interaction times of 0 and t,

$$\frac{C_3}{R_c(0)^3} = \frac{C_3}{R_c(t)^3} + \frac{1}{2}\mu V(t)^2 \quad . \tag{2}$$

There,  $\mu$  and V(t) are the reduced mass of a pair of <sup>85</sup>Rb atoms and the relative pair velocity, respectively. We choose t=8  $\mu$ s because at that time the atom pairs have an approximately constant velocity, which can be extracted well from a local linear fit within the range 6  $\mu$ s  $\leq t \leq 10$   $\mu$ s, indicated by the blue dashed line in Fig. 3(b). We obtain  $V(8 \mu s) = 0.45 \pm 0.06$  m/s and  $R_c(8 \mu s) = 9.3 \pm 0.9$   $\mu$ m, where the uncertainties include the statistical fit uncertainty and the magnification uncertainty. With  $R_c(0) = 6.7 \pm 0.7$   $\mu$ m from above, the resulting  $C_3$  value becomes  $(3.3 \pm 1.8) \times 10^{-42}$  Jm<sup>3</sup>. This value agrees with the calculated  $C_3$  value,  $p^2/(4\pi\epsilon_0) = 3.72 \times 10^{-42}$  Jm<sup>3</sup>. This agreement also implies that the interacting entities are individual atoms and not superatoms, as has been predicted in [29] and experimentally

observed for van-der-Waals-interacting Rydberg excitations in [7].

In Fig. 3(b) it is evident from the experimental data (circles) and the result of a semi-classical simulation [25] (red line) that the acceleration is negative for  $t \gtrsim 10 \ \mu s$ . The late-time deceleration appears to be due to repulsion from initially farther-away atoms, indicating manybody dynamics that involve more than two atoms. A related conclusion can be drawn from considering the visibility of the pair-correlation enhancement as a function of time, calculated from the I(R) curves in Fig. 3(a) as  $(I(R_c(t)) - \bar{I})/\bar{I}$ , where the asymptotic values  $\bar{I}(t)$ are obtained by averaging I(R) curves over the range  $R > 15 \mu \text{m}$ . The visibility values are shown as gray squares in Fig. 3(b). The visibility rapidly increases at early times,  $t \lesssim 4 \mu s$ , and passes through a broad maximum between 4 and 8  $\mu$ s. Hence, the significance of the pair correlation enhancement at  $R_c$ , equivalent to the degree of short-range order, and the kinetic energy in the sample, equivalent to the slope in  $R_c(t)$ , both become maximal approximately at the same time. These evidences are reminiscent of those in disorder-induced heating, which has been observed in the strongly-coupled ion component of an ultracold plasma [30]. In both cases, particles initially repel each other due to dominant nearest-neighbor forces, before encountering repulsive forces from initially more distant particles. At wait times  $t \gtrsim 10 \ \mu s$  the correlation enhancement disappears, which is in part due to the initial thermal atom velocity (the magneto-optical trap has an atom temperature of  $\sim 100 \ \mu \text{K}$ ). The "coupling parameter" of the dipolar system can be defined as the ratio between the initial dipole-dipole interaction energy and the thermal energy at 100  $\mu$ K. This ratio is  $\Gamma = 9$ , which is sufficiently large that the system should indeed develop (transient) shortrange order.

When the electric field is applied along y, the atomic permanent electric dipole moments are oriented along y. In that case, the angular dependence of the dipole force can be observed within the xy plane, encompassing maximally-repulsive interactions ( $\Theta = \pi/2$ ) and maximally-attractive interactions ( $\Theta = 0$ ), as well as all intermediate cases. The interaction then leads to the characteristic anisotropic patterns in the pair correlation images shown in Fig. 2(c).

The angular force in Eq. 1 is maximal at  $\Theta=\pi/4$  and  $3\pi/4$ , zero at  $\Theta=0$ ,  $\pi/2$  and  $\pi$ . It always points towards the "poles" ( $\Theta=0$  or  $\pi$  directions, see Fig. 4(a)). Hence, the angular force leads to an accumulation of atom pairs lined up close to the electric-field direction (the same direction as **p**). These atom pairs then become pulled close to each other due to the radial component of the force, which is attractive for  $\Theta \leq 55^{\circ}$  and  $\Theta \geq 125^{\circ}$ . These atom pairs form the prominent vertical dark strip across the center of the images at interaction times  $\gtrsim 4~\mu \text{s}$  in Fig. 2(c). Conversely, atom

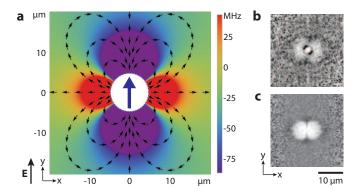


FIG. 4. (a) Color map of the dipolar potential and corresponding force vectors as a function of  $\mathbf{R}$  for a pair of dipoles pointing along y. Atoms repel each other in the equatorial direction, x, and attract each other in the polar direction, y. Due to the combination of radial and angular forces, the atom pairs become funneled into narrow conical sections at the poles. This leads to characteristic pair correlation images as shown in (b) (from the experiment) and (c) (from a simulation). The images in (b) and (c) are for t=4  $\mu$ s.

pairs initially positioned at  $\Theta \approx \pi/2$  will keep repelling each other, while being diverted towards the poles by the angular force (see Fig. 4(a)). Within the experimental uncertainty, the most probable pair separations  $R_c(t)$ in Fig. 2(b), along any direction in the xy plane, and in Fig. 2(c), along the x direction, are the same. This is expected because all these cases correspond to  $\Theta = \pi/2$ in Eq. 1. The "funneling effect" pointing towards the poles eventually leads to dumbbell-shaped pair correlation images that are void of signal in a volume extending along x and that possess enhanced signals along y. These characteristics are seen in the experiment (Fig. 2(c) and Fig. 4(b) and in our simulations (Fig. 4(c)). The small deviation of the enhancement cones in Fig. 2(c) from the y-direction is attributed to a slight deviation of the electrode arrangement from perfect symmetry.

Atoms pulled close to each other along the polar direction (y) will likely undergo Penning-ionizing collisions [31], when the interatomic separation R drops below about 0.5  $\mu$ m in our case. This distance is below the image resolution and is not directly observed in the experiment. It is, however, noticed that the amount of signal within the enhancement cones near  $\Theta=0$  and  $\Theta=\pi$  in Fig. 2(c) plateaus at  $t\gtrsim 4~\mu s$  and eventually drops. This observation is consistent with atom-pair loss within the polar cones due to Penning ionization.

In summary, we have employed an adiabatic state transformation method to prepare dense samples of about ten Rydberg atoms with large permanent electric-dipole moments. We have determined the dipolar dispersion coefficient,  $C_3$  by probing the dynamics of the center-of-mass motion of the atoms. The work differs

from [3], where the  $C_3$  coefficient of a Föster resonance was spectroscopically measured (with both atoms held at fixed positions). The measured pair correlation images portray the anisotropic character of dipolar atompair kinetics in a forceful, intuitive manner. Results of a semi-classical trajectory model agree well with our experimental observations. We have observed dynamics reminiscent of disorder-induced heating, as found elsewhere in strongly-coupled plasmas. Further insight may be gained by varying the ratio between the time scales governing the atoms' center-of-mass motion and the dipolarization (the ramp speed used during adiabatic passage), in a way that the atom positions have sufficient time to respond to the change of the potential landscape (in the manybody nuclear configuration space) that occurs during the adiabatic passage. This may lead to adiabatic formation of ordered structures akin to Wigner crystals in strongly coupled ion systems [32].

This work was supported by NSF Grant No. 1205559 and 1506093, and AFOSR Grant No. FA9550-10-1-0453. N.T. acknowledges support from DPST of Thailand. L.F.G. acknowledges support from FAPESP Grant No. 2014/09369-0.

- C. van Ditzhuijzen, A. Koenderink, J. Hernández, F. Robicheaux, L. Noordam, and H. van den Heuvell, Phys. Rev. Lett. 100, 243201 (2008).
- [2] E. Altiere, D. P. Fahey, M. W. Noel, R. J. Smith, and T. J. Carroll, Phys. Rev. A 84, 053431 (2011).
- [3] S. Ravets, H. Labuhn, D. Barredo, L. Béguin, T. Lahaye, and A. Browaeys, Nat Phys 10, 914 (2014).
- [4] S. Ravets, H. Labuhn, D. Barredo, T. Lahaye, and A. Browaeys, Phys. Rev. A 92, 020701 (2015).
- [5] A. Reinhard, K. C. Younge, T. C. Liebisch, B. Knuffman, P. R. Berman, and G. Raithel, Phys. Rev. Lett. 100, 233201 (2008).
- [6] L. Béguin, A. Vernier, R. Chicireanu, T. Lahaye, and A. Browaeys, Phys. Rev. Lett. 110, 263201 (2013).
- [7] N. Thaicharoen, A. Schwarzkopf, and G. Raithel, Phys. Rev. A 92, 040701 (2015).
- [8] R. Faoro, C. Simonelli, M. Archimi, G. Masella, M. M. Valado, E. Arimondo, R. Mannella, D. Ciampini, and O. Morsch, Phys. Rev. A 93, 030701 (2016).
- [9] T. Lahaye, J. Metz, B. Fröhlich, T. Koch, M. Meister, A. Griesmaier, T. Pfau, H. Saito, Y. Kawaguchi, and M. Ueda, Phys. Rev. Lett. 101, 080401 (2008).
- [10] T. Koch, T. Lahaye, J. Metz, B. Fröhlich, A. Griesmaier,

- and T. Pfau, Nat Phys 4, 218 (2008).
- [11] T. Pohl, E. Demler, and M. D. Lukin, Phys. Rev. Lett. 104, 043002 (2010).
- [12] R. M. W. v. Bijnen, S. Smit, K. A. H. v. Leeuwen, E. J. D. Vredenbregt, and S. J. J. M. F. Kokkelmans, J. Phys. B: At. Mol. Opt. Phys. 44, 184008 (2011).
- [13] P. Schauß, J. Zeiher, T. Fukuhara, S. Hild, M. Cheneau, T. Macrì, T. Pohl, I. Bloch, and C. Gross, Science 347, 1455 (2015).
- [14] M. D. Lukin, M. Fleischhauer, R. Cote, L. M. Duan, D. Jaksch, J. I. Cirac, and P. Zoller, Phys. Rev. Lett. 87, 037901 (2001).
- [15] E. Urban, T. A. Johnson, T. Henage, L. Isenhower, D. D. Yavuz, T. G. Walker, and M. Saffman, Nat Phys 5, 110 (2009).
- [16] T. M. Weber, M. Höning, T. Niederprüm, T. Manthey, O. Thomas, V. Guarrera, M. Fleischhauer, G. Barontini, and H. Ott, Nat Phys 11, 157 (2015).
- [17] E. W. Müller and K. Bahadur, Phys. Rev. 102, 624 (1956).
- [18] E. W. Müller, Science 149, 591 (1965).
- [19] J. Ullrich, R. Moshammer, A. Dorn, R. Dörner, L. P. H. Schmidt, and H. Schmidt-Böcking, Rep. Prog. Phys. 66, 1463 (2003).
- [20] J. R. Rubbmark, M. M. Kash, M. G. Littman, and D. Kleppner, Phys. Rev. A 23, 3107 (1981).
- [21] L. Wang, H. Zhang, L. Zhang, C. Li, Y. Yang, J. Zhao, G. Raithel, and S. Jia, New J. Phys. 17, 063011 (2015).
- [22] T. F. Gallagher, Rydberg Atoms, Cambridge Monographs on Atomic, Molecular and Chemical Physics (Cambridge University Press, 2005).
- [23] A. Schwarzkopf, R. E. Sapiro, and G. Raithel, Phys. Rev. Lett. 107, 103001 (2011).
- [24] A. Schwarzkopf, D. A. Anderson, N. Thaicharoen, and G. Raithel, Phys. Rev. A 88, 061406 (2013).
- [25] See Supplemental Material at [URL will be inserted by publisher], which includes Refs. [7, 21, 24, 26–28].
- [26] H. A. Bethe and E. E. Salpeter, Quantum Mechanics of One- and Two-Electron Atoms (Courier Corporation, 2013).
- [27] A. D. Schwarzkopf, Ph.D. thesis, University of Michigan (2013).
- [28] A. Reinhard, T. C. Liebisch, B. Knuffman, and G. Raithel, Phys. Rev. A 75, 032712 (2007).
- [29] S. Möbius, M. Genkin, S. Wüster, A. Eisfeld, and J. M. Rost, Phys. Rev. A 88, 012716 (2013).
- [30] T. C. Killian, T. Pattard, T. Pohl, and J. M. Rost, Physics Reports 449, 77 (2007).
- [31] F. Robicheaux, J. Phys. B: At. Mol. Opt. Phys. 38, S333 (2005).
- [32] J. N. Tan, J. J. Bollinger, B. Jelenkovic, and D. J. Wineland, Phys. Rev. Lett. 75, 4198 (1995)