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Electric field cancellation on quartz by Rb adsorbate-induced negative electron affinity

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We investigate the (0001) surface of single crystal quartz with a submonolayer of Rb adsorbates. Using Rydberg atom electromagnetically induced transparency, we investigate the electric fields resulting from Rb adsorbed on the quartz surface, and measure the activation energy of the Rb adsorbates. We show that the adsorbed Rb induces a negative electron affinity (NEA) on the quartz surface. The NEA surface allows low energy electrons to bind to the surface and cancel the electric field from the Rb adsorbates. Our results are important for integrating Rydberg atoms into hybrid quantum systems and the fundamental study of atom-surface interactions, as well as applications for electrons bound to a 2D surface.

Due to recent technological advances in fabrication and trapping, hybrid quantum systems (HQS) consisting of atoms and surfaces, as well as electrons and surfaces, are fast emerging as ideal platforms for a diverse range of studies in quantum control, quantum simulation and computing, strongly correlated systems and microscopic probes of surfaces [1–5]. Miniaturization of chip surfaces is necessary to achieve large platform scalability, but decoherence and noise emerge as serious challenges as feature sizes shrink [6–8]. Mitigating noise is a fundamental step in realizing the full potential of HQS.

Combining ultracold Rydberg atoms with surfaces for HQS is attractive because Rydberg atoms can have significant transition dipole moments and strong interactions. There have been a host of theoretical proposals recently for utilizing Rydberg atoms near surfaces [2, 9–12]. Experimental progress has been hampered by uncertainties in characterizing interactions of atoms with surfaces, although some recent work in this regards are noteworthy [13–15].

To take full advantage of Rydberg atom HQS, a more complete understanding of surfaces is needed. One problem is that Rydberg atoms incident upon metal surfaces can be ionized [16, 17]. A second major hurdle is the background electric fields (E-fields) caused by adsorbates [18–23]. Rydberg states are sensitive to adsorbate E-fields because they are highly polarizable [24]. Adsorbate E-fields have caused problems for other experiments as well, including Casimir-Polder measurements [25], and sur-

face ion traps [26]. A possible solution is to minimize the E-fields by canceling them out.

A convenient surface for applications in HQS is quartz because of its extensive use in the semiconductor and optics industries. Despite numerous theoretical and experimental studies of bulk SiO₂ [27–29], the surface properties are not well understood. The (0001) surface has been the subject of recent theoretical interest, partially due to its stability and low surface energy [30–34].

In this work, we show that a quartz (0001) surface with Rb adsorbates, contrary to prevailing assumption, can have very small E-fields near the surface, Fig. 1c. We demonstrate, by appealing to theoretical arguments and *ab initio* calculations, that the reduction in E-field is caused by a transformation of the quartz into a negative electron affinity (NEA) surface via adsorption of Rb atoms on the surface. A NEA surface can bind electrons, similar to the image potential states on liquid helium (LHe) [35–37]. While the surface repulsion for electrons on LHe is provided by Pauli blocking, the repulsion on quartz occurs because the surface vacuum level dips below the conduction band minimum. We find that the binding of electrons to the surface substantially reduces the E-field above the surface.

In experiments on atom-adsorbate interactions, using different surfaces, adsorbate E-fields with magnitudes ranging from $\sim 0.1 - 10$ V cm⁻¹ have been measured at distances of $\sim 10 - 100$ μ m [13, 19, 20, 22, 25]. We measure radically different E-fields dependent upon the number of slow electrons produced

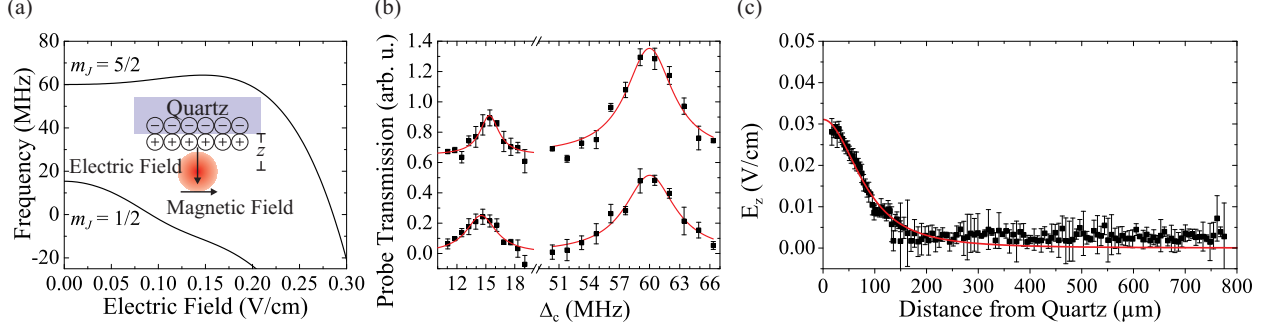


FIG. 1. (Color online) (a) Stark shift for $81D_{5/2}$, $m_J = 5/2$ and $1/2$ states in a 14.3 G magnetic field oriented perpendicular to the E-field. The inset shows the orientation of the electric and magnetic fields with respect to the quartz surface. (b) EIT spectra taken at 2 different positions $z = 150 \mu\text{m}$ (upper) and $z = 50 \mu\text{m}$ (lower) for $81D_{5/2}$ $m_J = 1/2$ (left) and $m_J = 5/2$ (right). The black points are pixel values of 3 averaged images, and the error bars are the standard deviation of the pixel values. The red lines are Lorentzian fits to the data. At $z = 50 \mu\text{m}$ the $m_J = 1/2$ state is broadened and shifted corresponding to an E-field of 0.02 V cm^{-1} . (c) In the limit of high Rydberg atom ($\text{Rb}(81D)$) population the E-field is measured at distances of $\sim 20 - 800 \mu\text{m}$ from the quartz surface at $T_{\text{sub}} = 79^\circ\text{C}$. Black points are taken from different pixels on a CCD camera. The error bars are the standard deviation of the measurement. The red line is a fit to equation (1), showing the inhomogeneity of the E-field. Our calculations indicate that the E-field at $z < 200 \mu\text{m}$ is caused by the large spacing between the electrons.

by Rydberg atoms near the surface. We demonstrate that E-fields as small as 30 mV cm^{-1} can be obtained $20 \mu\text{m}$ from the surface.

A microscopic picture of E-field noise is obtained by considering thermal fluctuations of adsorbate dipole moments [38]. An adsorbed atom develops a dipole moment as a result of the polarization of the adatom in interaction with the surface. As the density of adsorbates increases, the E-field from neighboring dipoles reduces the dipole moment of each adatom (see Supplementary). We estimate the dipole moment for a Rb adsorbate in the limit of small coverage to be $d_0 = 12 \text{ D}$ (see Supplementary).

Adsorption of a large number of Rb atoms on the quartz surface produces macroscopic E-fields. At distances far from the surface, the E-field can be modeled as two square sheets of charge, with edge length L , separated by a small distance [21, 22]. Near the center of the sheets, the E-field is largely perpendicular to the surface,

$$E_z(z) = \frac{2\sqrt{2}\sigma d(\sigma)L^2}{\pi\epsilon_0\sqrt{L^2 + 2z^2}(L^2 + 4z^2)}, \quad (1)$$

where ϵ_0 is the permittivity of free space, σ is the adsorbate density, and $d(\sigma)$ is the coverage dependent dipole moment. The temperature dependence of σ is [39],

$$\frac{\sigma/\sigma_0}{1 - \sigma/\sigma_0} = C e^{\frac{E_a}{kT_{\text{sub}}}}, \quad (2)$$

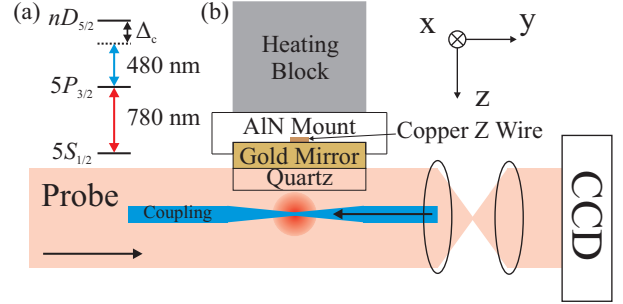


FIG. 2. (Color online) (a) Level scheme for Rydberg EIT used in our experiments. Δ_C is the coupling laser detuning. (b) A schematic of the experimental setup. Rb atoms are trapped in a mirror MOT, transferred into a magnetic trap and transported to the surface. The probe and coupling beams for Rydberg EIT are overlapped and counterpropagate. The Rydberg EIT signal is observed by analyzing the absorption of the probe beam on a CCD camera. Heaters are placed outside of vacuum to control the quartz temperature. The gold mirror is used to form the MOT. The heating block controls the temperature of the substrate. The Z-wire generates the magnetic trap. The aluminum nitride (AlN) mount insulates the Z-wire from the heating block.

where σ_0 is the density of adsorbate sites, E_a is the desorption activation energy, k is the Boltzmann constant, and T_{sub} is the substrate temperature. Equations (1) and (2) relate E_z to σ at T_{sub} .

The E-fields are determined experimentally by

measuring the frequency shift of a Rydberg state, and comparing it to a Stark shift calculation. Stark shifts of two magnetic states for $81D_{5/2}(m_J = 5/2 \text{ and } m_J = 1/2)$ are shown in Fig. 1a. An example of experimental traces at different z is shown in Fig. 1b. These types of traces were used to obtain the E-fields. The Rydberg state energy is determined using Rb Rydberg atom electromagnetically induced transparency (EIT) [40]. The energy level scheme is shown in Fig. 2a. The Rydberg EIT is detected by the absorption of the probe laser by the atomic cloud on a CCD camera [20–22], Fig. 2b. The E-field and its spatial dependence are obtained by analyzing absorption images as a function of coupling laser detuning with a spatial resolution of $5.5 \mu\text{m}$.

For the experiments, a mirror magneto-optical trap (MOT) is used to load a Rb magnetic trap $\sim 2\text{mm}$ from the quartz surface, Fig. 2b. After loading the magnetic trap, bias magnetic fields are used to move the atoms close to the surface. The atoms are released from the magnetic trap and imaged. The atomic cloud is a cigar shaped Gaussian cloud with $1/e^2$ radii of $0.5 \times 0.5 \times 1\text{mm}$.

At low Rydberg atom number, the E-field is homogeneous over the magnetic trap because the EIT signal is not detectably broadened across the extent of the atom sample, $\sim 2\text{mm}$. The variation of the E-field over $z = 200\text{--}1000 \mu\text{m}$ is $< 0.1 \text{ V cm}^{-1}$, Fig. 1c. The sensitivity is limited by the polarizability of the Rydberg state. L is estimated to be 10mm and is similar in size to other observations [21].

The main source of the Rb adsorbate E-field is the MOT atoms. Disabling the magnetic trap for ~ 10 minutes did not change the E-field. Disabling the MOT for the same period changes the E-field. In the presence of the MOT, the adsorbate coverage can be controlled by changing the surface temperature.

The adsorbate E-field points away from the surface as confirmed by an external compensating E-field. The adsorbate E-field is estimated to be normal to the surface, within 15° , based on the differential shifts of different m_J states. This further justifies the model in equation (1).

We measured the E-field as a function of T_{sub} . The results are shown in Fig. 3 at $z = 500 \mu\text{m}$. At 28°C , the E-field is $1.7 \pm 0.1 \text{ V cm}^{-1}$. Using equation (1), for a slab of length $L = 10\text{mm}$ and $d_0 \sim 12\text{D}$, we estimate $\sigma = 4 \times 10^5 \text{ atoms } \mu\text{m}^{-2}$, yielding an average Rb spacing of $\sim 1.5\text{nm}$, and an adatom coverage of 11% . Fitting all values of σ to equation (2)

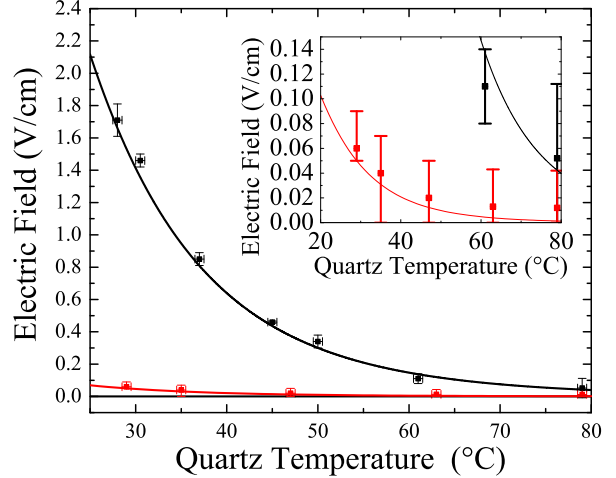


FIG. 3. (Color online) The measured E-fields due to Rb adsorbates on the (0001) surface of quartz as a function temperature, T_{sub} , at a distance of $500 \mu\text{m}$ from the surface. The E-fields are calculated by analyzing the frequency shifts of the EIT spectra. The black points are in the limit of low Rydberg atom production. The black line is a fit to the Langmuir isobar of equation (2), and yields a desorption activation energy of $E_a = 0.66 \pm 0.02 \text{ eV}$. The red data points were taken with high Rydberg atom production. The red line is explained in the text. The horizontal error bars are due to the uncertainty in the temperature, T_{sub} . The vertical error bars are the standard deviation of the experimental data. In the case of high Rydberg atom production, the Rabi frequencies of the probe and coupling lasers are $\Omega_p = 2\pi \times 3.5 \text{ MHz}$ and $\Omega_c = 2\pi \times 4 \text{ MHz}$. Approximately 200 electrons are produced during each experimental sequence, 10Hz average rate. For low Rydberg atom production, the Rabi frequencies of the probe and coupling lasers are $\Omega_p = 2\pi \times 0.5 \text{ MHz}$ and $\Omega_c = 2\pi \times 4 \text{ MHz}$. Approximately 10 electrons are created during each experimental sequence, 0.5Hz average rate. The horizontal error bars are due to the uncertainty in T_{sub} and are $\pm 0.5^\circ\text{C}$.

with a coverage dependent dipole moment, yields $E_a = 0.66 \pm 0.02 \text{ eV}$. E_a is similar to the activation energy measured for alkali atoms on similar surfaces [41–44].

Increasing the Rydberg atom number in either trap dramatically reduces the E-field, by increasing the flux of slow blackbody ionized electrons that can bind to the surface. The Rydberg atom number can be made larger by increasing the probe laser Rabi frequency. The temperature dependence of the reduced E-field is shown in Fig. 3. For typical data in Fig. 3, 300 atoms are ionized per experimental

sequence. $\text{Rb}(nS)$ and $\text{Rb}(nD)$ states were investigated for $40 \leq n \leq 100$, yielding similar results.

The Rydberg atoms are predominately ionized due to blackbody radiation; direct blackbody ionization accounts for 99% of all electrons [45] at high n . For $\text{Rb}(81D_{5/2})$, the electrons have an average kinetic energy of 10 meV. For $n \sim 40 - 100$, the electrons average kinetic energy is 8 – 15 meV [46].

If blackbody ionized electrons can bind to the surface, they can neutralize the E-field produced by the Rb-adatoms. Electrons can bind to a conducting or dielectric surface through their image potential [47]. These states are usually ultra-short lived, and rapidly collapse into the bulk. In LHe, however, the Pauli repulsion provides the necessary barrier of ~ 1 eV, to prevent decay, leading to the formation of stable bound states on the surface. In LHe, the electrons can remain in these states for tens of hours at cryogenic temperatures [37]. For adsorption on ordinary surfaces, if the vacuum energy dips below the bottom of the conduction band, a NEA surface is produced, repelling electrons from the surface.

Amorphous quartz has a positive electron affinity of 0.9 eV [48]. However, adsorption of atoms can change the surface properties. The dipole layer created by the adsorbates changes the electric potential at the vacuum-surface interface. By calculating the electrostatic change in energy of an electron across the surface dipole layer, an estimate of the change in electron affinity, $\Delta\chi$, can be made [49] (see Supplementary). Using $d_0 = 12$ D and $\sigma = 4.2 \times 10^5 \text{ atoms } \mu\text{m}^{-2}$ at $T_{\text{sub}} = 28^\circ\text{C}$, the change in surface electron affinity is $\Delta\chi = -1.9$ eV. This approximation suggests that Rb at our densities can shift the vacuum level ~ 1 eV below the conduction band, inducing a NEA surface on quartz. The model shows NEA up to $T_{\text{sub}} \sim 40^\circ\text{C}$.

To investigate the adatom-surface on a microscopic level, we performed total-energy calculations for the quartz (0001) surface with various Rb coverage using spin-polarized density functional theory (DFT) [50] (see Supplementary). On the surface of quartz, the Rb atom is bound to two oxygen atoms. The lowest bound state for one monolayer (ML) has an energy of $E_b = 0.35$ eV. For the lower experimentally investigated coverages, our DFT calculations show an increase of E_b by ~ 1.4 . The calculated E_b is comparable in magnitude with the measured E_a , and is consistent with the expectation $E_b \leq E_a$ [51].

We calculated the electronic density of states for bulk α -quartz and the shift of the vacuum energy

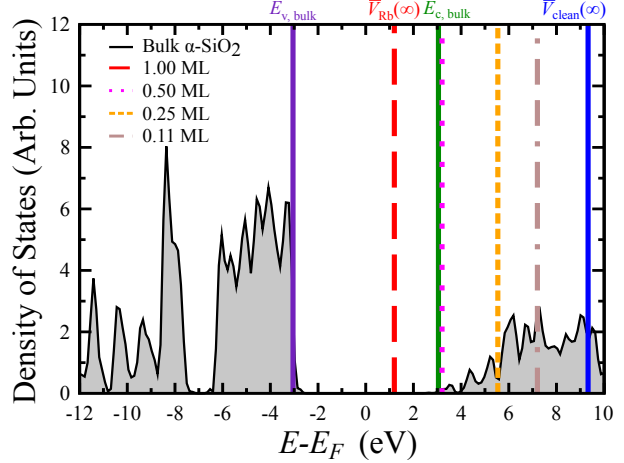


FIG. 4. (Color online) Density of states of bulk α -quartz. The Fermi level, E_F is at $E = 0$. The valence band maximum, $E_{v,\text{bulk}}$, conduction band minimum $E_{c,\text{bulk}}$ of bulk α -quartz, and the vacuum levels of the SiO_2 (0001) surface without and with Rb adsorbates, respectively, $\bar{V}_{\text{clean}}(\infty)$ and $\bar{V}_{\text{Rb}}(\infty)$ are labeled. $\bar{V}_{\text{Rb}}(\infty)$ is shown as a function of coverage in fractions of a monolayer (ML). Increasing the amount of Rb coverage shifts the vacuum level down in energy. With one ML of Rb on the surface (red line), the vacuum energy is below the bottom of the conduction band (green line), indicating the formation of a NEA surface.

with varying amounts of Rb coverage using DFT, Fig. 4. The Fermi level, E_F is set equal to zero, and lies in the middle of the band gap, between the top of the valence band, $E_{v,\text{bulk}} = -3.05$ eV, and the bottom of the conduction band, $E_{c,\text{bulk}} = 3.05$ eV. As shown in Fig. 4, the vacuum level for the clean surface, $\bar{V}_{\text{clean}}(\infty)$, has a positive electron affinity, consistent with experiment [48]. However, adsorbing Rb on the surface shifts the vacuum level downward. NEA is induced around 0.5 ML. The DFT and the straightforward electrostatic calculations, both show that the vacuum level shifts by several electron volts with *only* a modest amount of Rb coverage. The remaining discrepancy may be resolved with further improvements in DFT [52, 53]. More knowledge of the experimental surface including the Rb adsorbate structure will also help to guide the calculations.

We model the electrons as a uniformly charged square sheet of length L , that overlays the adsorbate layers with $L = 10$ nm at $z = 0$. The resulting E-field is a sum of the E-fields from the adsorbates and electrons, $E_{\text{tot}} = E_{\text{ads}} + E_{\text{ele}}$. After requiring the $E_{\text{tot}} = 0$ at $z = 0$, $E_{\text{tot}}(z = 500 \mu\text{m})$ is plotted in Fig. 3. The near exact fit to the data is an indi-

cation that the reduction in the E-field is due to the formation of a NEA surface for Rb-SiO₂.

For high temperatures and high Rydberg population the E-field is low. The measured E-field as a function of z at $T_{\text{sub}} = 79^\circ\text{C}$ is shown in Fig. 1c. For $z > 200\ \mu\text{m}$ the E-field is negligible within error. At $z < 200\ \mu\text{m}$ the E-field increases to $\sim 30\ \text{mV cm}^{-1}$. Under these conditions, we estimate a surface electron density of $\sim 10\ \text{electrons mm}^{-2}$. For $z < 200\ \mu\text{m}$, approximating the electrons as a uniform sheet of charge breaks down since the electron spacing becomes larger than z . The spectral width of the EIT resonance for $81\ D_{5/2}$, ($m_J = 1/2$) increases from 2 MHz far from the surface to $\sim 4\ \text{MHz}$ at $z \leq 50\ \mu\text{m}$, Fig. 1b. We attribute this broadening to the inhomogeneity of E_{tot} near the surface. The data in Fig. 1c is fit to equation (1), and shows that the residual E-field can be modeled as a dipole patch, with $L \sim 200\ \mu\text{m}$, approximately equal to the estimated electron spacing of $\sim 300\ \mu\text{m}$.

We can remove electrons from the surface using 400 nm light generated by a light emitting diode (LED) array. With the surface saturated with electrons ($E_{\text{tot}}(z = 0) = 0$), the LEDs are pulsed on for a variable time while atoms are loaded into the MOT. The light intensity is small to avoid light induced desorption of Rb. The MOT fluorescence is monitored to verify this condition. The photodesorption rate constant has an Arrhenius behavior, with activation energy $0.7 \pm 0.07\ \text{eV}$ (see Supplementary). The activation energy is similar to E_a suggesting the electron detachment mechanism is dependent on Rb coverage. The Rb coverage affects the energy levels most strongly. It is unknown if the electrons are detached from or tunnel into the surface. The electron photodesorption is the subject of future investigation.

Over the temperature range investigated, $28^\circ\text{C} < T_{\text{sub}} < 80^\circ\text{C}$, the Rb-quartz system can bind electrons for several hours, when the MOT is on and the EIT lasers are off. The small E-fields have been repeatedly measured many times for over a year, yielding the same results within experimental error. The thermal wavelength of an electron at 28°C is 4.3 nm, indicating that the electron is not localized on one Rb adsorbate. We believe that the single crystal nature of the quartz and small surface roughness, $< 5\ \text{\AA}$, plays an important role in the uniformity of the Rb adsorbates and electron binding. We have done some simulations investigating whether the dipole potential from a patch of adsorbates or

the image potential is responsible for binding the electrons to the surface. Our results show that binding is due to the image potential of the electron. The dipole potential slightly shifts the image potential.

In summary, we have measured the activation energy of Rb on the quartz (0001) surface and shown the onset of a NEA surface capable of binding electrons upon Rb adsorption. Reducing E-fields on a quartz surface by making quartz a NEA surface by Rb adsorption is a promising pathway for coupling Rydberg atoms to surfaces. Further work can be directed towards measurements of other surface orientations and dielectrics, as well as investigating the behavior at cryogenic temperatures. The properties of the electrons, including binding energy, mobility, and effective mass, are the subject of future work.

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