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## Formation of Isolated Zn Vacancies in ZnO Single Crystals by Absorption of Ultraviolet Radiation: A Combined Study Using Positron Annihilation, Photoluminescence, and Mass Spectroscopy

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28 ZnO is an important transparent semiconducting material that has generated significant  
29 research interest [1-3]. A recent review of defects in ZnO by McCluskey and Jokela [2] has  
30 highlighted the need to understand the role of various defects on the electronic properties of ZnO.  
31 First-principles calculations place the 0/-1 acceptor level of the zinc vacancy ( $V_{\text{Zn}}$ ) 100 meV  
32 above the valence band maximum [4,5]. Therefore,  $V_{\text{Zn}}$  creation by equilibrium or non-  
33 equilibrium processes may become an initial step toward obtaining p-type ZnO. Since positrons  
34 are sensitive to neutral or negatively charged open volume  $V_{\text{Zn}}$  defects, positron annihilation  
35 spectroscopy (PAS) has been used extensively to study  $V_{\text{Zn}}$  in ZnO. Early PAS work on  
36 energetic particle irradiated ZnO demonstrated  $V_{\text{Zn}}$  formation in isolated or complex states  
37 depending on irradiation parameters like dose, energy, and type of incident particle [6-10].

38 Energetic particle irradiation generally produces vacancies and interstitials on both sub-  
39 lattices [11], and often forms undesired complexes, thus limiting its ability to create isolated  $V_{\text{Zn}}$   
40 [10]. In principle, annealing in oxygen could generate  $V_{\text{Zn}}$  defects, but this process is controlled  
41 by thermodynamics and may not produce a sufficiently high concentration. Mechanical  
42 treatments, such as polishing or indenting, have also created sub-surface defects like  $V_{\text{Zn}}$  in  
43 single crystal ZnO [12-14]. Defect creation in such non-thermal mechanical treatments is  
44 uncontrollable. On the other hand, optical excitation is an efficient and controlled non-thermal  
45 defect creation process for many materials. For example, optical excitations produce color  
46 centers in alkali halides due to their large self-trapping potentials for excitons [15].

47 In oxide systems, there is a potential barrier between the exciton self-trapping and defect  
48 pair creation [15]. The well-documented radiation hardness of ZnO also suggests that it is  
49 resilient to defect formation. However, a recent study of  $\text{Zn}^+$  ion emission predicts  $V_{\text{Zn}}$  creation  
50 upon irradiation with 193 nm excimer laser light [16]. If the predicted  $V_{\text{Zn}}$  creation does occur,  
51 then it must be due to a novel photoelectronic process. This process could open up the possibility  
52 of controlled incorporation of point defects. Therefore, studies of 193 nm photonic interactions  
53 with single crystal ZnO encompass fundamental as well as applied research.

54 This letter presents evidence for isolated  $V_{\text{Zn}}$  creation following 193 nm excimer laser  
55 interactions with single-crystal ZnO. PAS shows isolated  $V_{\text{Zn}}$  defects creation in the irradiated  
56 material. Diffusion of zinc interstitials to the surface could leave behind isolated  $V_{\text{Zn}}$  defects and  
57 promote surface sensitive  $\text{Zn}^+$  ion desorption, presumably due to zinc interstitial-vacancy Frenkel  
58 pair creation by selective excitation of Zn–O bond. Appearance of a photoluminescence  
59 excitation (PLE) peak at 3.18 eV in irradiated material suggests that the isolated  $V_{\text{Zn}}$  defects  
60 have a shallow 100 meV acceptor state.

61 The ZnO single crystals, grown by a chemical vapor transport method, were nominally n-  
62 type with a free-electron sheet concentration of  $\sim 10^{13} \text{ cm}^{-2}$ . As-grown platelet type samples, 0.2-  
63 1 mm thick, were used in this study. The sample excitation was performed under ultra high  
64 vacuum (UHV) conditions ( $\sim 1 \times 10^{-8} \text{ Pa}$ ) at 100 or 400  $\text{mJ/cm}^2$  fluence of 193 nm excimer laser  
65 pulses, each 20 ns wide, at a 1 Hz repetition rate.  $\text{Zn}^+$  ion time-of-flight (TOF) signals were  
66 acquired with a UTI-100C quadrupole mass spectrometer (QMS) that passes only  $\text{Zn}^+$  ions  
67 when tuned to mass resonance of 64 amu/e [17]. A 32-cm long free flight distance was set for the  
68  $\text{Zn}^+$  ion TOF.

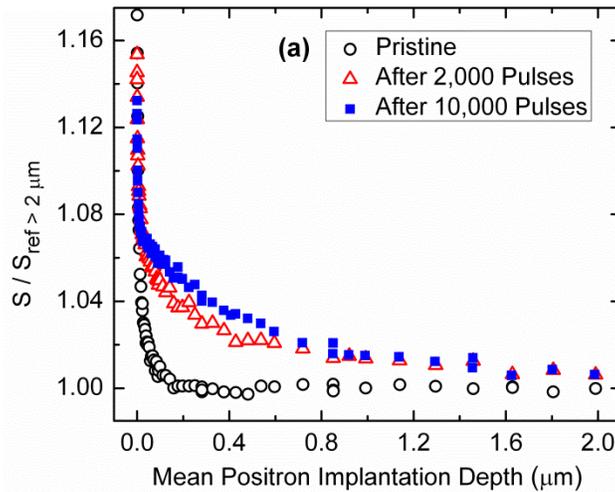
69 Positron measurements were carried out with the WSU monoenergetic variable energy  
70 positron beam. Depth resolved PAS results were extracted from the Doppler broadening of the  
71 511 keV annihilation line with an energy resolution of 1.4 keV (full width half maximum) and  
72 are expressed in terms of the S and W parameters [18]. The S parameter comprises the central  
73 fraction of the positron annihilation peak at 511 keV where lower momentum Doppler shifts  
74 dominate. It is sensitive to neutral or negatively charged vacancies with reduced concentration of  
75 high momentum electrons. The W parameter comprises the wings of the peak where higher  
76 momentum Doppler shifts dominate. It relates to the chemical environment of an annihilation  
77 site. The S and W parameters were normalized to the as-grown pristine state values of  
78  $0.4039 \pm 0.0002$  and  $0.1059 \pm 0.0001$ , respectively, at a depth in excess of 2  $\mu\text{m}$ .

79 PLE spectroscopy was performed using a JY-Horiba FluoroLog-3 spectrofluorometer  
80 equipped with double grating monochromators (1200 grooves/mm), and a R928P

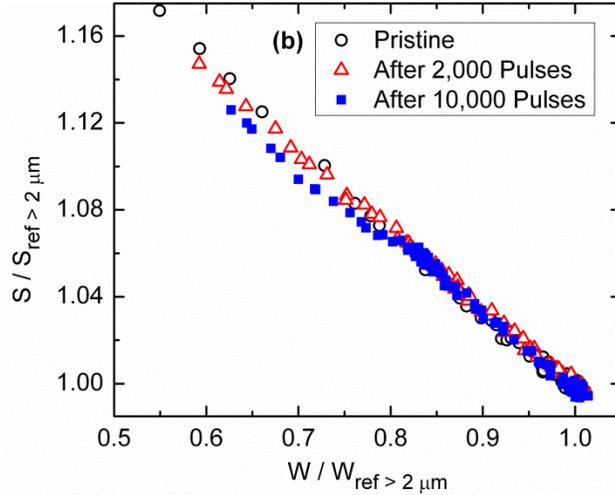
81 photomultiplier tube. A xenon 450 W lamp was used for excitation. Free carrier absorption was  
82 obtained with a Bomem DA8 vacuum Fourier transform infrared (FTIR) spectrometer equipped  
83 with a Globar light source, a KBr beamsplitter, and a liquid-nitrogen-cooled HgCdTe detector. A  
84 Janis closed-cycle helium cryostat system maintained the sample temperature.

85 Figure 1(a) shows the S parameter versus mean positron implantation depth for an as-  
86 grown sample as a function of radiation exposure performed at  $100 \text{ mJ/cm}^2$  fluence. The S  
87 parameter value at the surface, in general, reflects positronium formation with surface electrons  
88 [18]. At greater depths, the S parameter value in ZnO primarily reflects zinc vacancies [6], even  
89 though it could also reflect the concentration of di-vacancies or complexes [14,19]. In  
90 unirradiated pristine ZnO, the S values near the surface, represented by open circles, drop  
91 quickly to a plateau at increasing depths. This sharp drop to the bulk S value of  $0.4039 \pm 0.0002$   
92 reflects a reference baseline for low concentration of  $V_{\text{Zn}}$  point defects in pristine ZnO.

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Fig. 1. (Color online) (a) PAS S parameter versus mean implantation depth and (b) corresponding S-W parameter plots for single crystal ZnO before 193 nm laser irradiation, after 2,000 pulses, and after 10,000 pulses. The irradiation was performed at 100 mJ/cm<sup>2</sup> fluence per pulse under UHV conditions.

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The sample was then irradiated to 2,000 pulses. In this initial irradiated state, the S parameter remains significantly above the baseline, indicating open volume defect creation in the near surface region [Fig. 1(a)]. In addition, the corresponding S-W parameter curves in Fig. 1(b) show the same straight line trend for the pristine and initial irradiated states, indicating that irradiation creates the same kind of defects originally present in the pristine state. The observed (S,W) values in the initial irradiated state are consistent with the published (S,W) values for isolated  $V_{Zn}$  defects, which represent a range (1.04–1.05, 0.87–0.79) depending on experimental configuration [6,7,9]. The increased S values below the surface in the initial irradiated state, and the slope of the S-W lines in the pristine and initial irradiated states, are consistent with formation of isolated  $V_{Zn}$  defects [6]. Therefore, the PAS results suggest that 193 nm excitation produces isolated  $V_{Zn}$  defects in single-crystal ZnO. Moreover, the S parameter in the initial irradiated state extends more than 2 μm into the bulk material, indicating the presence of isolated  $V_{Zn}$  defects in the bulk.

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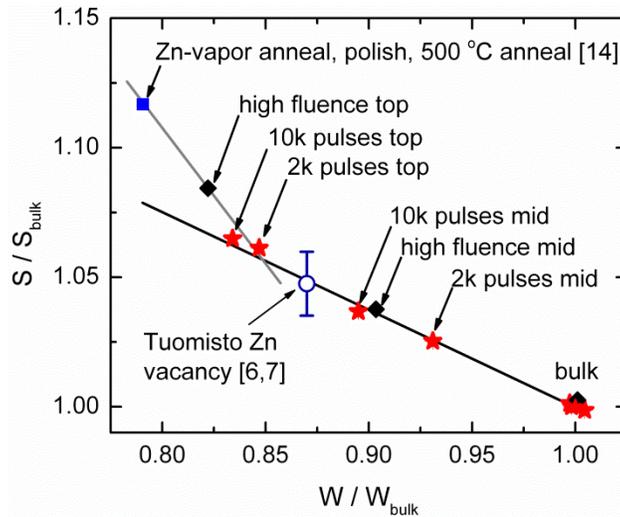
The sample was further irradiated to a total of 10,000 pulses. In this final irradiated state, the S parameter is only slightly higher than in the initial irradiated state [Fig. 1(a)]. While further increment of S values in final irradiated state might suggest more isolated  $V_{Zn}$  creation,

118 the corresponding S-W curve in Fig. 1(b) does not follow the straight line trend previously  
119 obtained on the pristine and initial irradiated states, likely due to saturation. Saturation positron  
120 trapping can occur for two reasons. First, if the isolated  $V_{\text{Zn}}$  concentration reaches a critical level  
121 (high  $10^{18} \text{ cm}^{-3}$ ), and all positrons annihilate at these defects. Second, if open volume defects  
122 with larger trapping coefficients are created, positrons may annihilate at these defects instead of  
123  $V_{\text{Zn}}$  defects.

124 To understand the nature of the observed saturation, VepFit analysis that incorporated  
125 three layers (top 100–150 nm, mid 200–700 nm, and the bulk) was performed. The results are  
126 shown in Fig. 2. The VepFit results for initial and final irradiated states, represented by stars,  
127 clearly pass through the black line, which connects Tuomisto's experimental result for the  
128 isolated  $V_{\text{Zn}}$  defects, open circle, and the referenced bulk defects. This black line represents an  
129 increasing concentration of isolated  $V_{\text{Zn}}$  defects with respect to the bulk. This analysis indicates  
130 only isolated  $V_{\text{Zn}}$  creation during 193 nm photonic excitation, and demonstrates a controlled  $V_{\text{Zn}}$   
131 fabrication near the saturation concentrations for positrons. Assuming a bulk positron lifetime of  
132 170 ps [6], a trapping coefficient of  $3 \times 10^{15} \text{ s}^{-1}$ , and an atomic density of  $8.3 \times 10^{22} \text{ cm}^{-3}$ , an  
133 estimate for the  $V_{\text{Zn}}$  concentration can be made. The 2,000-pulse initial irradiation creates  
134  $1.4 \times 10^{18}$  and  $5.5 \times 10^{16} \text{ cm}^{-3}$  zinc vacancies in the top and mid layers, respectively; the 10,000-  
135 pulse total irradiation increases these concentrations to  $2.2 \times 10^{19}$  and  $1.2 \times 10^{17} \text{ cm}^{-3}$ ,  
136 respectively.

137 The same analysis was performed on another sample irradiated to 2,000 pulses at high  
138 fluence ( $400 \text{ mJ/cm}^2$ ). The corresponding VepFit results are shown by diamonds in Fig. 2. A  
139 clear deviation from the black line is noticed for the top layer, heading toward Selim's  
140 experimental result for larger defects in ZnO [14]. The large open volume defect creation can be  
141 understood from the perspective of significant neutral particle (atomic Zn and O) emission  
142 during the laser pulse. At high fluence, a detected particle emission rate of  $\sim 0.1$  monolayer per  
143 pulse is consistent with leaving behind large open volume defects [20]. On the other hand, during  
144 low fluence excitation, the neutral particle emission rate falls below  $10^{-9}$  monolayer per pulse

145 [20]. The lack of detectable neutral particle emission for ZnO under low fluence excitation  
 146 correlates with only isolated  $V_{Zn}$  creation.  
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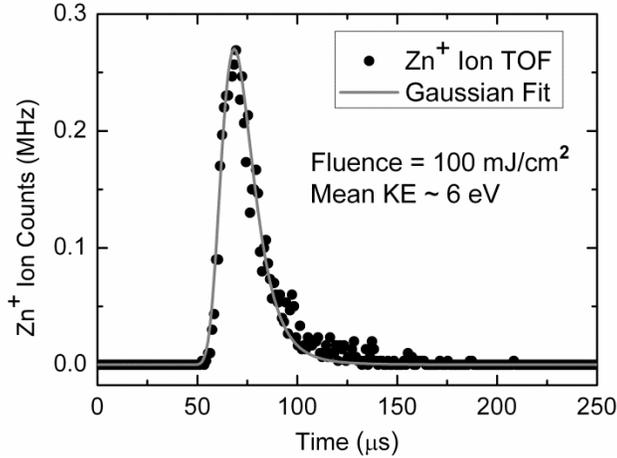
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 150 Fig. 2. (Color online) PAS results calculated using the VepFit model, incorporating two thin  
 151 layers (top, mid) and bulk. The open circle shows Tuomisto’s experimental result for the  
 152 isolated Zn vacancy [6,7]. Stars show isolated Zn vacancies created by laser exposure at  
 153  $100 \text{ mJ/cm}^2$  fluence excitation. Diamonds show large open volume defects creation in the  
 154 top layer (100–150 nm) at high fluence ( $400 \text{ mJ/cm}^2$ ) excitation.  
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156 The low fluence PAS results reveal the highest concentration of isolated  $V_{Zn}$  defects in  
 157 the top layer, which overlaps with absorption depth for 193 nm photons in ZnO ( $d_{Ab} \sim 200 \text{ nm}$ ).  
 158 The dominant  $V_{Zn}$  presence in the top layer is consistent with 193 nm photons creating these  
 159 defects in the interaction layer, defined by laser absorption depth. The increased  $V_{Zn}$  presence in  
 160 the mid layer and beyond is likely related to a secondary response induced by the laser pulse.  
 161 Transient laser heating can trigger diffusion of newly created  $V_{Zn}$  in the interaction volume. A  
 162 diffusion penetration depth can be estimated by  $d_{th} = \sqrt{\alpha\tau}$ , where  $\alpha = 0.8 \text{ cm}^2/\text{s}$  is the thermal  
 163 diffusivity of single crystal ZnO [21], and  $\tau = 20 \text{ ns}$  is the laser pulse duration. An approximate  
 164 value of  $d_{th} \sim 1 \text{ }\mu\text{m}$  allows the transient thermal field to extend about a few  $\mu\text{m}$  into the bulk,  
 165 which is consistent with the observed  $V_{Zn}$  density profile, indicating that diffusion of the newly  
 166 created defects is likely occurring.

167 The transient temperature rise can be estimated by  $\Delta T = 2(1-R)(F/k)(\alpha/\pi\tau)^{1/2}$  [22].  
168 Here  $k \approx 1 \text{ Wcm}^{-1}\text{K}^{-1}$  is the thermal conductivity of single-crystal ZnO [23], and  $R \approx 0.12$  is the  
169 reflectivity of ZnO at 193 nm [24]. At  $F=100 \text{ mJ/cm}^2$  fluence, the transient temperature rises to  
170  $\sim 200 \text{ }^\circ\text{C}$ . After characterizing the final irradiation state, the sample was also annealed at  $200 \text{ }^\circ\text{C}$   
171 for one hour under UHV conditions. Subsequent PAS measurement showed a slightly more  
172 pronounced drop in the  $V_{\text{Zn}}$  density profile, indicating that the  $V_{\text{Zn}}$  defects are almost distributed  
173 by transient heating, and, therefore, are relatively immobile up to  $200 \text{ }^\circ\text{C}$  anneal. Considering  
174 extremely fast diffusion time scale ( $d_{\text{Ab}}^2/\alpha$ ), about a few ns, a  $\sim 200 \text{ }^\circ\text{C}$  transient heating seems  
175 sufficient to diffuse point defects created in ZnO [25,26].

176 Surface defects can act as a sink for the diffusing interstitial zinc ions ( $\text{Zn}_i^+$ ) created  
177 during laser pulse by 193 nm photonic excitation. Upon reaching the surface, a  $\text{Zn}_i^+$  ion could  
178 find a neutral surface oxygen vacancy site and adsorb as a weakly bound surface defect complex,  
179 which acts as an ideal source for ion emission during excimer laser interaction [16,17,27]. Ion  
180 desorption experiments performed on a ZnO sample during  $100 \text{ mJ/cm}^2$  fluence excitation under  
181 UHV conditions result in only  $\text{Zn}^+$  ion emission shown by the TOF (Fig. 3). A Gaussian fit to  
182 the TOF [16] yields a mean kinetic energy of  $\sim 6 \text{ eV}$  for the emitted  $\text{Zn}^+$  ion distribution. A  
183 simple electrostatic model can explain this hyper-thermal emission mechanism that requires  
184 photoionization of the oxygen vacancy of such a surface defect complex [16,17,27]. This  
185 photoionization step immediately switches the weakly bound complex into a repulsive center.  
186 The repulsive Coulomb force between the  $\text{Zn}^+$  ion and positively charged oxygen vacancy emits  
187 the  $\text{Zn}^+$  ion with hyper-thermal energies. In ZnO, a  $0.21 \text{ nm}$  oxygen vacancy to adsorbed  $\text{Zn}^+$   
188 ion effective bond length can deliver up to  $6.9 \text{ eV}$  of maximum kinetic energy [16], which is  
189 consistent with the observed mean kinetic energy of  $6 \text{ eV}$ .

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Fig. 3. The 193 nm interaction leads into only hyper-thermal Zn<sup>+</sup> ion emission from ZnO surface under UHV conditions at 100 mJ/cm<sup>2</sup> fluence. The Zn<sup>+</sup> ion TOF was acquired at 64 amu/e mass resonance and best QMS mass resolution.

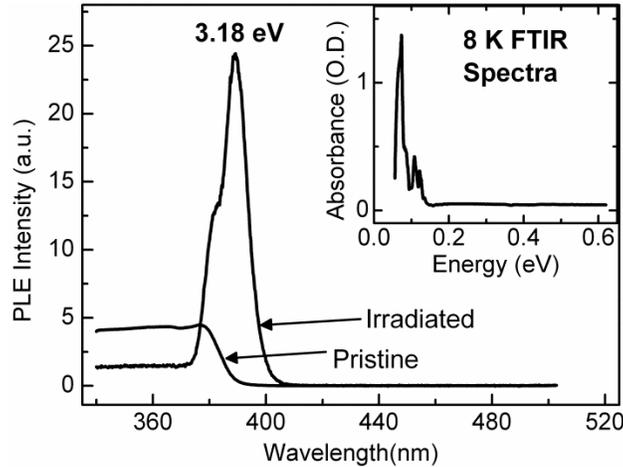
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The formation of such surface defect complexes, due to Zn<sub>i</sub><sup>+</sup> transport to the surface, could affect the ion yield. A Zn<sup>+</sup> ion emission study on ZnO at 193 nm interaction has demonstrated nearly exponential decay of Zn<sup>+</sup> ion yield as a function of exposure at fluences ≤ 40 mJ/cm<sup>2</sup> [16]. The first order exponential decay kinetics reflect only consumption of such pre-existing surface defect complexes via emission process. On the other hand, the decay kinetics change to a second order at fluences ≥ 60 mJ/cm<sup>2</sup> [16], reflecting replenishment of such defect complexes due to Zn<sub>i</sub><sup>+</sup> ion transport to the surface. The observed  $V_{Zn}$  density profile and Zn<sup>+</sup> ion emission with second order decay kinetics verify the diffusion driven distribution of these defects. Besides, this study and the previous one observe only Zn<sup>+</sup> ion emission at fluences ≤ 100 mJ/cm<sup>2</sup>, and could not verify oxygen ion emission despite a near unity ion detection sensitivity [16]. Therefore, preferential Zn<sup>+</sup> ion emission and  $V_{Zn}$  creation suggest that selective photoelectronic Zn<sub>i</sub>- $V_{Zn}$  Frenkel pair creation occurs during 193 nm irradiation.

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Transient heating at 200 °C is not sufficient to create such a Frenkel pair in ZnO. In alkali halides, Frenkel pairs are produced via large exciton self-trapping potentials [15]. ZnO, in contrast, does not exhibit exciton self-trapping. Interestingly, ZnO single crystal shows a wide 6.34 eV transition associated with hybridized O 2p and Zn 4sp atomic orbitals [28-31]. If the

213 electron wave function has nodes between the Zn and O atoms, it would be strongly antibonding.  
214 Thus, the absorption of one 6.4 eV photon could excite this wide antibonding state of the Zn–O  
215 bond parallel to the crystal c-axis. This excitation could then break the bond and create a Frenkel  
216 pair in the Zn sub-lattice.  
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220 Fig. 4. 193 nm irradiation effect on room temperature PLE spectra for 525 nm emission. The  
221 525 nm emission was collected at 1 nm resolution while the sample was excited in steps  
222 of 0.2 nm at 0.5 nm resolution. Inset shows an 8 K FTIR spectrum of the irradiated  
223 sample. Irradiation was performed at 100 mJ/cm<sup>2</sup> fluence for 2,000 pulses under UHV  
224 conditions.  
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226 The zinc vacancy is an acceptor with a theoretical predicted 0/-1 transition state at 100  
227 meV above the valence band maximum [4,5]. In order to investigate the electronic state of the  
228 isolated  $V_{Zn}$  defects, the PLE technique was employed, since PL in ZnO is a surface sensitive  
229 probe under ultraviolet excitation. Pristine and irradiated samples showed a green PL band  
230 peaked at 525 nm under steady-state ultraviolet excitation. PLE spectra for 525 nm emission are  
231 shown in Fig. 4. In the pristine state, the PLE spectrum behaves like a step function, essentially  
232 following the fundamental absorption edge. The flat response over the 340–378 nm range  
233 suggests that electron-hole pair excitation at energies higher than 3.28 eV (378 nm) efficiently  
234 produce green emission. Above 378 nm excitation, the green emission intensity falls sharply to  
235 zero, demonstrating that excitation energies less than 3.28 eV do not produce green emission in  
236 the pristine state.

237 For the irradiated sample, the PLE spectrum at energies less than 3.28 eV is dramatically  
238 different. An excitation band appears at  $\sim 3.18$  eV (389.5 nm), confirming a new defect band that  
239 acts as source for 525 nm green emission. This 3.18 eV defect absorption band is  $\sim 100$  meV  
240 below the room-temperature band gap of ZnO. This 3.18 eV absorption band is tentatively  
241 assigned to an electron transfer from the  $V_{\text{Zn}}^{0/-1}$  acceptor level to the conduction band  
242 minimum. The appearance of a  $V_{\text{Zn}}$  related band in irradiated material is consistent with creation  
243 of mostly localized defects, which is verified by the absence of any free carrier absorption in the  
244 8 K FTIR absorption spectrum shown in the inset. The fact that 193 nm excitation does not  
245 produce significant free carrier densities at 8 K suggests that it might be possible to produce  $V_{\text{Zn}}$   
246 defects that are not accompanied by compensating donor defects.

247 In conclusion, evidence is presented for the controlled formation of isolated zinc  
248 vacancies during 193 nm excimer laser irradiation at  $100 \text{ mJ/cm}^2$  fluence. The stable zinc  
249 vacancies are associated with a shallow acceptor level about 100 meV above the valence band  
250 maximum. The ability to produce relatively high densities of zinc vacancies in the sub-surface  
251 material may prove useful in tailoring the electronic properties of zinc oxide – either directly, by  
252 providing acceptor defects, or indirectly, by stabilizing implanted dopants.

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261 experimental work.

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- 265 [1] U. Ozgur *et al.*, J. Appl. Phys. **98**, ARTN 041301 (2005).  
266 [2] M.D. McCluskey and S.J. Jokela, J. Appl. Phys. **106**, ARTN 071101 (2009).  
267 [3] A. Janotti and C.G. Van de Walle, Rep. Prog. Phys. **72**, ARTN 126501 (2009).  
268 [4] A.F. Kohan *et al.*, Phys. Rev. B **61**, 15019 (2000).  
269 [5] A. Janotti and C.G. Van de Walle, Phys. Rev. B **76**, ARTN 165202 (2007).  
270 [6] F. Tuomisto *et al.*, Phys. Rev. Lett. **91**, ARTN 205502 (2003).  
271 [7] F. Tuomisto *et al.*, Phys. Rev. B **72**, ARTN 085206 (2005).  
272 [8] Z.Q. Chen *et al.*, Phys. Rev. B **75**, ARTN 245206 (2007).  
273 [9] A. Zubiaga *et al.*, Phys. Rev. B **78**, ARTN 035125 (2008).  
274 [10] Y.F. Dong *et al.*, Phys. Rev. B **81**, ARTN 081201 (2010).  
275 [11] F. Tuomisto *et al.*, Phys. Rev. B **76**, ARTN 165207 (2007).  
276 [12] Z. Takkouk *et al.*, Physica B-Condens. Matter **366**, 185 (2005).  
277 [13] Y. Ohno *et al.*, Appl. Phys. Lett. **92**, ARTN 011922 (2008).  
278 [14] F.A. Selim *et al.*, Phys. Rev. Lett. **99**, ARTN 085502 (2007).  
279 [15] N. Itoh and A. M. Stoneham, *Materials Modification by Electronic Excitation*  
280 (Cambridge University Press, United Kingdom, 2001), p. 4, 224.  
281 [16] E. Khan *et al.*, J. Appl. Phys. **111**, ARTN 063101 (2012).  
282 [17] J.T. Dickinson *et al.*, Phys. Rev. Lett. **73**, 2630 (1994).  
283 [18] P.J. Schultz and K.G. Lynn, Rev. Mod. Phys. **60**, 701 (1988).  
284 [19] M.H. Weber *et al.*, Appl. Surf. Sci. **255**, 68 (2008).  
285 [20] E. H. Khan *et al.*, Submitted to J. Appl. Phys. (2013).  
286 [21] X. Xu *et al.*, Chinese Phys. Lett. **25**, 176 (2008).  
287 [22] E. Khan *et al.*, Langmuir **25**, 1930 (2009).  
288 [23] U. Ozgur *et al.*, J. Electronic Mater. **35**, 550 (2006).  
289 [24] J. L. Freeouf, Phys. Rev. B **7**, 3810 (1973).  
290 [25] L.S. Vlasenko and G.D. Watkins, Phys. Rev. B **71**, ARTN 125210 (2005).  
291 [26] P. Erhart and K. Albe, Appl. Phys. Lett. **88**, ARTN 201918 (2006).  
292 [27] E. Khan, S. Langford, and J. Dickinson, J. Appl. Phys. **110**, ARTN 023110 (2011).  
293 [28] A. Kobayashi *et al.*, Phys. Rev. B **28**, 935 (1983).  
294 [29] P.J. Moller, S. Komolov, and E. Lazneva, J. Phys.-Condens. Matter **11**, 9581 (1999).  
295 [30] P.S. Xu *et al.*, Nucl. Instr. & Meth. Phys. Res. Sect. B **199**, 286, (2003).  
296 [31] K. Ozawa *et al.*, Phys. Rev. B **68**, ARTN 125417 (2003).

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