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X-ray-induced water vaporization

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Abstract

We present quantitative evidence for X-ray-induced water vaporization: water is vaporized at a rate of 5.5 pL/s with the 1-Å-wavelength X-ray irradiation of ~ 0.1 photons per Å²; moreover, water vapor is reversibly condensed during pauses in irradiation. This result fundamentally suggests that photoionization induces vaporization. This phenomenon is attributed to surface tension reduction by ionization and would be universally important in radiological and electrohydrodynamic situations.

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X-ray photonics is undergoing a revolution in imaging capabilities with the use of ultrabright X-ray sources. X-ray imaging at the nano- and microscale is of great interest for applications in physical and life sciences, including X-ray physics, materials science, biological imaging, environmental analysis, archaeology, paleontology, and heritage restoration [1], because it facilitates the nondestructive, direct visualization of internal structures or elements. Meanwhile, radiation damage from the energy deposited into the sample by the X-ray photons used for imaging is inevitable [2,3]. In particular, radiation-induced ionization ultimately disintegrates the sample by charge accumulation beyond a limit and is otherwise known as Coulomb explosion [2,3] and fission [4,5]. The conventional damage barrier for Coulomb explosion is known to be about 200 photons per \AA^2 for X-rays with a wavelength of 1 \AA [2,3]. Radiation damage can be mitigated by cryofixation or by ultrabright, ultrashort X-ray pulses [2]. However, radiation damage to living tissues by radiation ionization is not yet fully understood [6], although it is a limiting factor in achieving high-resolution data [2].

In previous works, we demonstrated that high-brilliance X-ray photons affect the surface tension of water by inducing ionization, using hard X-ray imaging at the 7B2 beamline (dose rate $\sim 1 \text{ kGy s}^{-1}$) of the Pohang Light Source [7–9]. A charge density over $\sim 10^{-4}$ charges per \AA^2 was required for a significant reduction ($> 30\%$) in the surface tension [9]. This condition corresponds to X-ray irradiation of ~ 0.1 photons per \AA^2 , following a simulation using the ionization rate of ~ 1 charge per \AA^2 by 2000 photons per \AA^2 with X-rays of 1 \AA wavelength [2]. We put forward a possibility of water vaporization by surface tension reduction [7,8] based on a monotonic relationship of the surface tension to the vaporization enthalpy [10,11], but quantitative evidence was lacking.

To corroborate the possibility, we conducted an elaborate experiment in a different synchrotron source while applying high-speed X-ray imaging [12,13]. Using a hydrophilic capillary tube containing a small amount of water, we first made a confined water lens with concave menisci, where the left region was *closed* and the right one was *open* in air [Fig. 1]. The changes of the water menisci were simultaneously imaged using a high-speed camera, synchronized and gated to the X-ray pulses. The image size was 768×640 pixels ($0.98 \mu\text{m}$ per pixel, 16-bit). The field of view was larger than the capillary inner diameter ($2R = 500 \mu\text{m}$). Two repeatable shutter modes were used by (i) applying a short exposure for 100 ms (to see the radiation effect) and by (ii) taking a long recess for 9.9 s (to see any possible recovery) [Fig. 1]. The undulator source at the XSD 32-ID beamline of the Advanced Photon Source of the Argonne National Laboratory provided the high X-ray brilliance necessary for the investigation of the X-ray-induced effects. The undulator gap was set to 31 mm, and most of the beam intensity was confined within the first harmonic at $\sim 13 \text{ keV}$ (1 \AA wavelength), with a peak irradiance of $\sim 10^{14} \text{ photons mm}^{-2} \text{ s}^{-1}/0.1\% \text{bw}$ (equally $\sim 1 \text{ ph \AA}^{-2} \text{ s}^{-1}/0.1\% \text{bw}$), as clearly seen by the energy spectrum from the undulator. The X-ray photons were irradiated into the confined water by this high-brilliance white X-ray beam. At the sample location ($\sim 37 \text{ m}$ from the source), the beam profile (or the 2D total intensity distribution for an on-axis aperture) shows that the capillary aperture was quasi-uniformly illuminated since $2R$ was much smaller than the beam size. Here, the irradiation for the short exposure of 100 ms was $\sim 0.1 \text{ ph \AA}^{-2}$. The exposure time for each frame was 1 ms, and 100 frames were taken in every short exposure of 100 ms, resulting in a single movie of 8000 frames during the total running time of 790.1 s.

We used a bi-concave water plug within a hydrophilic capillary tube [Fig. 2], which mimics liquid plugs in pulmonary airways [14], for simplicity in the analysis of radiation effects.

The water meniscus was always hemispherical within the tube and was not changed by X-ray irradiation. The X-ray-induced reduction in the surface tension changed the vaporization enthalpy and therefore the meniscus position. To monitor the meniscus position, we used high-speed phase-contrast X-ray imaging [12]. The X-ray wavelength of $\sim 1 \text{ \AA}$ from the brightest peak ($\sim 0.1 \text{ ph \AA}^{-2}$) corresponded to the dose rate of $\sim 50 \text{ kGy s}^{-1}$, which is higher than that for the previous works ($\sim 1 \text{ kGy s}^{-1}$) [7].

Representative sequential images for a single cycle of X-ray on-off switches are demonstrated in Fig. 2 for 10.001 s. The most obvious findings are the rightward gradual shifts in both of the water menisci during the short exposure of 100 ms. The rightward shifts ($\sim 3.9 \text{ \mu m}$ for 100 ms) clearly indicate the expansion of the closed air volume and thus represent the generation of water vapor by the irradiation. Interestingly, the shifts recovered upon the long 9.9-s pause of irradiation. Here, we note that the recovery behaviors were different between the left (closed) and right (open) menisci: the left one fully recovered to the initial position, whereas the right one did not. The right meniscus rather showed a slight leftward shift from the original position ($\sim 1.0 \text{ \mu m}$ for 10 s). This shift clearly shows the existence of natural vaporization into the atmosphere. The leftward shift by natural vaporization was negligible during the short exposure of 100 ms, compared with the irradiation-induced rightward shift. This behavior (i.e., the recovery [leftward shift] of the left [right] meniscus after the long recess) was repeated for every cycle of X-ray on-off switches (see, for instance, a sequential movie [15] for the first 5 cycles).

Figure 3 quantitatively shows the variance in the positions of the left (marked as X_L) and right meniscus apexes (marked as X_R) for the 10 cycles of 100.1 s. The position changes during the irradiation of 100 ms and the pause of 9.9 s were very repeatable for every cycle. Here, the

rapid shifts of $3.9\ \mu\text{m}$ for 100 ms in both X_L and X_R were induced by the X-ray irradiation of $\sim 0.1\ \text{ph}\ \text{\AA}^{-2}$. After the 9.9-s pause, the left meniscus was completely recovered, as can be clearly seen by the same positions of the circles in Fig. 3(a). In contrast, the position of the right meniscus continued to decrease with time, as seen by the squares in Fig. 3(b). The gradual decreases in X_R at a rate of $1.0\ \mu\text{m}$ for 10 s were induced by the natural vaporization. This result shows that the X-ray-induced switching is reversible in both of the menisci.

Now we discuss the X-ray-induced vaporization rate. As the volume change of water for the short exposure of 100 ms is not detectable, the vaporization rate was estimated from the water vapor expansion, measured as $\sim 770\ \text{pL}$ for 100 ms from the volume variance $\Delta V = \pi R^2 \Delta X_L$, where the meniscus radius $R = 250\ \mu\text{m}$ and the shift $\Delta X_L = 3.9\ \mu\text{m}$ for every cycle. This expansion is a direct result of the X-ray-induced effects. When water turns into water vapor, the volume expansion is approximately 1400 times at 300 K, assuming the ideal gas law. From the vapor expansion rate, $7.7\ \text{nL}\ \text{s}^{-1}$, and the volume expansion, 1400 times, we estimated the X-ray-induced vaporization rate as $5.5\ \text{pL}\ \text{s}^{-1}$ by the irradiation of $\sim 0.1\ \text{ph}\ \text{\AA}^{-2}$. The large amount of water vapor generated at the closed region implies that X-ray-induced water vaporization might be a critical damaging factor of biological imaging.

The X-ray-induced vaporization rate at the closed region is quantitatively different with the natural vaporization rate at the open region. As demonstrated in Fig. 4, natural vaporization occurred only at the open (right) region, not at the closed (left) one. The natural water vaporization rate was estimated to be $20 \pm 0.6\ \text{pL}\ \text{s}^{-1}$ for the total cycles of 740 s at 20% humidity [Fig. 4], measured from the water volume changes based on the leftward shifts in X_R . Here the contribution of the X-ray-induced vaporization for each cycle was negligible, at $\sim 0.5\ \text{pL}$

($\approx 5.5 \text{ pL s}^{-1} \times 100 \text{ ms}$), compared with $\sim 200 \text{ pL}$ ($\approx 20 \text{ pL s}^{-1} \times 10 \text{ s}$) by natural vaporization.

In comparison, the natural evaporation rate for a similar open-closed capillary system ($2R = 800 \text{ }\mu\text{m}$) [Fig. 5] was separately measured as $\sim 14 \text{ pL s}^{-1}$ (= slope $[6.4 \times 10^{-5}] \times \text{initial volume [212 nL]}$; half-closed circles) at 40% humidity. The evaporation rate for the open-open system was very high as $\sim 62 \text{ pL s}^{-1}$ (= slope $[2.8 \times 10^{-4}] \times \text{initial volume [222 nL]}$; open circles) by the convection effect [16], whereas the closed-closed system (closed circles) showed no evaporation. We find a good agreement in the natural evaporation rates for the open-closed systems [Figs. 4 and 5], supporting no influence of the X-ray irradiation on the natural evaporation rate.

Notably, X-ray-induced vaporization is quantitatively different from radiolysis-induced gas generation for water [17]. The X-ray-induced vapor expansion rate of $\sim 770 \text{ pL}$ for $\sim 0.1 \text{ ph } \text{\AA}^{-2}$ ($\approx 5 \text{ kGy}$) at 300 K , corresponding to $\sim 7 \text{ nmol MGy}^{-1}$, is much larger than the radiolysis-induced gas generation rate of $\sim 0.05 \text{ nmol MGy}^{-1}$ at 5 K or $\sim 0.01 \text{ nmol MGy}^{-1}$ at 100 K [17]. In addition, radiolysis-induced gas generation becomes negligible over 100 K and is obviously irreversible [17]. X-ray-induced vaporization is therefore irrelevant to radiolysis.

Is the closed air region warmed up *directly* by the X-ray irradiation? A $\sim 560\text{-}\mu\text{m}$ shift is expected by the thermal expansion, $\Delta V = \alpha P/C_p$, based on $\Delta V = \alpha V \Delta T$ and $\Delta T = P/(C_p V)$, where ΔT is the temperature rise, α is the thermal expansion coefficient ($3.50 \times 10^{-3} \text{ K}^{-1}$ [18]), C_p is the heat capacity ($1.17 \text{ mJ cm}^{-3} \text{ K}^{-1}$ [19]), and P is the adsorbed power passing through the closed air region. Here, $P \approx 25 \text{ }\mu\text{J}$ by multiplying the X-ray power (8 W mm^{-2}) by the irradiation time (0.1 s), the sample area ($\sim 1/4 \text{ mm}^2$), and the absorption rate ($\sim 0.012\%$ for a $500\text{-}\mu\text{m}$ thick air at 13 keV [20]). The expected thermal shift ($560 \text{ }\mu\text{m}$) is inconsistent with the actual shift ($3.9 \text{ }\mu\text{m}$ in Fig. 3). The expected heating, $\Delta T \approx 184 \text{ K}$ for 0.1 s at $250 \text{ }\mu\text{J s}^{-1}$ for $V = 115 \text{ nL}$, is unrealistic,

because >99.9% of the total attenuation of the beam is used up for non-thermal photoelectric absorption [21] and gas heating can occur only by extremely high fluxes (e.g., 10^{37} J s^{-1} [22]). Indeed, X-ray heating was negligible in the previous measurements [7,23]. X-ray-induced vaporization is therefore not due to heating effect.

X-ray-induced vaporization is thus attributed to a photoionization-induced surface tension reduction, which decreases the vaporization enthalpy, as previously hypothesized [7]. The reversibility of X-ray-induced vaporization [Fig. 3] can be explained by a reversible change of the enthalpy [10,11]. The charges accumulated from X-ray-induced ionization, which results in the surface tension reduction, are relaxed during the recess, allowing the surface tension to recover and thus reversibly inducing vapor condensation.

Our result suggests that photoionization induces vaporization by a significant surface tension reduction. The surface tension (γ [mN m⁻¹]) varies from the original value (γ_0) with the surface charge density (σ [C m⁻²]) below the Rayleigh limit ($\sigma_R \approx 6.89 \times 10^{-4} \text{ C m}^{-2}$ for water) as $\gamma = \gamma_0 [1 + 4(\sigma/\sigma_R)^2]^{-1/2}$ [9]. The latent heat of vaporization (H [J g⁻¹]) depends on the surface tension as $H \approx 581.1 \gamma^{1/3}$ (estimated for water from [10] at 0.01–374.15 °C; the coefficient of determination is $R^2 = 0.99942$) and thus has a general relation with the photoionization as:

$$H \approx 581.1 \gamma_0^{1/3} [1 + 4(\sigma/\sigma_R)^2]^{-1/6}.$$

Here the vapor flux (J) is inversely proportional to H at a fixed heat transfer ($Q = HJ$) [24], implying ~20% increase in J by ionization of $\sigma \approx 0.8\sigma_R$, which is comparable to our result that the X-ray-induced vaporization rate [5.5 pL s⁻¹] is ~28% of the natural rate [20 pL s⁻¹]. This physical linkage has importance in radiological and electrohydrodynamic situations. This linkage

may explain why bubbles within biological samples are generated and/or expanded by high-brilliance X-ray photons, which has been a critical problem in X-ray biological imaging [25]. Ionization-induced vaporization can be a significant source of vapor generation and may accelerate evaporation of charged drops in electrohydrodynamic situations such as electrification of aerosols, combustion of fuel droplets, spray painting, and inkjet printing [26,27].

To conclude, we provide quantitative evidence of X-ray-induced water vaporization in a closed capillary tube using high-speed phase-contrast X-ray imaging. This X-ray-induced effect can be fully recovered by a sufficient pause in the irradiation. High-brilliance X-rays can modify material properties, including surface tension and viscosity [28]. A very short exposure on the μs timescale would be useful to prevent any X-ray-induced changes [29,30]. Finally, we stress that the X-ray irradiation barrier for vaporization (~ 0.1 photons per \AA^2) is much less than the conventional damage barrier (~ 200 photons per \AA^2) by the Coulomb explosion [2]. The new barrier is important in supporting sample stability and safety from radiation damage with high-brilliance X-ray sources. This result is of significance in carrying out X-ray experiments for high-resolution analysis of small confined water or living tissues. Ionization-induced vaporization will be universally important in electrohydrodynamics of charged drops and in biological imaging when using ionizing radiation sources besides X-ray sources.

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Figure Captions

FIG. 1: (Color online) Schematic diagram of the experimental set-up and the irradiation scheme. We used a confined water plug with bi-concave menisci, where the left region was closed and the right one was open in air. The irradiation and the precise tracking of the menisci were achieved by a high-brilliance white X-ray beam. Repeatable, two different shutter modes were used by (i) applying a short exposure for 100 ms (for the radiation effect) and by (ii) taking a long pause for 9.9 s (for the possible recovery) in a single movie of 8000 frames for 790.1 s.

FIG. 2: (Color online) Sequential X-ray images for the first X-ray on-off cycle. Rightward gradual shifts appeared in both of the water menisci upon the short exposure of 100 ms. The original positions of the water menisci are marked by the solid lines. This image sequence clearly illustrates the expansion of the closed air volume because of the generation of water vapor by the irradiation at $\sim 0.1 \text{ ph } \text{\AA}^{-2}$. The left meniscus fully recovered to the initial position, whereas the right one did not, after the long 9.9-s pause of irradiation (see, in the image, 10 s 1 ms). The leftward shift at the right meniscus shows the existence of natural vaporization into the atmosphere. The same behavior was repeated at every cycle of X-ray on-off switches.

FIG. 3: (Color online) High reversibility of the X-ray-induced vaporization. The variances in the positions of the left (marked as X_L in (a)) and the right meniscus apexes (marked as X_R in (b)) were tracked for 10 cycles for 100.1 s. The position changes during the short exposure of 100 ms and the pause of 9.9 s were very repeatable for every cycle. The rapid shifts of $3.9 \text{ }\mu\text{m}$ for 100 ms in both X_L and X_R were induced by the X-ray irradiation ($\sim 0.1 \text{ ph } \text{\AA}^{-2}$). After the pause, the left

meniscus was completely recovered, as could be clearly seen by the same positions of the circles, whereas the right one continued to decrease with time, as seen by the squares. The gradual decreases in X_R at a rate of $1.0 \mu\text{m}$ for 10 s were induced by natural vaporization. This result shows that the X-ray-induced switching is highly reversible in both of the menisci.

FIG. 4: (Color online) Natural vaporization rates with X-ray irradiation. X-ray snapshots of the water menisci showed gradual water shrinkage by natural vaporization only in the open (right) region but not in the closed (left) region. The natural vaporization rate measured from the water volume changes was $20 \pm 0.6 \text{ pL s}^{-1}$ for the total cycles of 740 s .

FIG. 5: (Color online) Natural vaporization rates without X-ray irradiation. Optical micrographs for monitoring water volume changes within open-open, open-closed, and closed-closed capillary systems ($2R = 800 \mu\text{m}$) with time. The length of the open region was $\sim 10 \text{ mm}$ for all systems. The evaporation rates were measured as $\sim 14 \text{ pL s}^{-1}$ for the open-closed case (half-closed circles) and $\sim 62 \text{ pL s}^{-1}$ for the open-open case (open circles). The closed-closed case (closed circles) showed no evaporation.









