



# CHORUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

## Lattice expansion and local lattice distortion in Nb- and La-doped SrTiO<sub>3</sub> single crystals investigated by x-ray diffraction and first-principles calculations

Shunsuke Kobayashi, Yuichi Ikuhara, and Teruyasu Mizoguchi

Phys. Rev. B **98**, 134114 — Published 31 October 2018

DOI: [10.1103/PhysRevB.98.134114](https://doi.org/10.1103/PhysRevB.98.134114)

1 **Lattice expansion and local lattice distortion in Nb- and La-doped SrTiO<sub>3</sub>**  
2 **single crystals investigated by X-ray diffraction and first-principles**  
3 **calculations**

4  
5 Shunsuke Kobayashi<sup>1\*</sup>, Yuichi Ikuhara<sup>1,2</sup> and Teruyasu Mizoguchi<sup>3\*</sup>

6  
7 <sup>1</sup>Nanostructures Research Laboratory, Japan Fine Ceramics Center, Atsuta, Nagoya 456-85  
8 87, Japan

9 <sup>2</sup>Institute of Engineering Innovation, The University of Tokyo, Bunkyo, Tokyo 113-8656, Japan

10 <sup>3</sup>Institute of Industrial Science, The University of Tokyo, Meguro, Tokyo 153-8505, Japan

11  
12 \*Corresponding author

13 E-mail: s\_kobayashi@jfcc.or.jp (S. K.)

14 E-mail: teru@iis.u-tokyo.ac.jp (T. M.)

15  
16 **ABSTRACT**

17 Electron-doped SrTiO<sub>3</sub> (where the dopant can be Nb or La) has been widely investigated for  
18 both its fundamental interest in condensed matter physics and for industrial applications. Its  
19 electronic properties are closely related to the Ti–O bonding states in the SrTiO<sub>3</sub> crystal. To  
20 further develop and control these properties, it is crucial to understand the factors controlling the  
21 change in lattice parameters and local lattice distortion upon doping with various atoms. Herein,  
22 we report the changes in lattice parameters and local lattice distortion in Nb- and La-doped  
23 SrTiO<sub>3</sub> single crystals, investigated by in-plane X-ray diffraction and first-principles calculations.  
24 The lattice parameter of Nb- and La-doped SrTiO<sub>3</sub> single crystals increased with dopant  
25 concentration. The broad intensities around the Bragg peak observed in the in-plane X-ray  
26 diffraction experiments indicated that the local lattice expansion and contraction, or local lattice  
27 distortions, in the crystal were caused by the dopant atoms. First-principles calculations similarly

28 showed that the lattice expansion and local lattice distortions in the SrTiO<sub>3</sub> crystals were caused  
29 by doped Nb and La atoms. Atoms surrounding Schottky pairs of O and Sr vacancies were  
30 displaced both away from and towards the vacancies, resulting in a reduction in the lattice  
31 expansion of donor-doped SrTiO<sub>3</sub>. Donor atoms and Schottky pairs thus play an important role in  
32 determining the lattice parameters of SrTiO<sub>3</sub> crystals. These fundamental structural analyses  
33 provide a useful basis to further investigate the electronic conductivity of electron-doped SrTiO<sub>3</sub>.

34

35

## 36 I. INTRODUCTION

37 Electron-doped strontium titanate (SrTiO<sub>3</sub>) is an n-type oxide semiconductor that has been  
38 widely investigated in condensed matter physics owing to its high electron mobility [1-3],  
39 quantum phenomena [4,5], and superconductivity [6,7], as well as for its use in thermoelectric  
40 applications [8,9] and as an anode material in solid oxide fuel cells [10]. Stoichiometric SrTiO<sub>3</sub> is  
41 a typical ABO<sub>3</sub> perovskite oxide and an insulator. Electronic conductivity can be introduced by  
42 creating O vacancies through partial reduction [11], or by doping the A sites with rare earth  
43 (donor) atoms or the B sites with transition metal atoms. For example, electron-doped SrTiO<sub>3</sub> is  
44 typically obtained by partially substituting Nb for Ti and La for Sr. The lattice parameter of  
45 SrTiO<sub>3</sub> is known to increase with the concentration of the Nb [12,13] and La [14-16] dopants.

46 Changes in oxidation states upon substitution of aliovalent dopant atoms changes the effective  
47 ionic radii of the transition metals in the system. In Nb-doped SrTiO<sub>3</sub>, Nb<sup>5+</sup> replaces Ti<sup>4+</sup>, and the  
48 resulting composition can be written as Sr<sup>2+</sup>Ti<sup>4+</sup><sub>1-2x</sub>Ti<sup>3+</sup><sub>x</sub>Nb<sup>5+</sup><sub>x</sub>O<sup>2-</sup><sub>3</sub> (0 ≤ x < 0.5). Similarly, when  
49 La<sup>3+</sup> ions substitute for Sr<sup>2+</sup> in SrTiO<sub>3</sub>, the resulting structure can be represented as  
50 Sr<sup>2+</sup><sub>1-y</sub>La<sup>3+</sup><sub>y</sub>Ti<sup>4+</sup><sub>1-y</sub>Ti<sup>3+</sup><sub>y</sub>O<sup>2-</sup><sub>3</sub> (0 ≤ y ≤ 1). In the case of Nb-doped SrTiO<sub>3</sub>, lattice expansion can be

51 straightforwardly explained based on the increase in the total effective ionic radius when  $\text{Ti}^{4+}$   
52 (60.5 pm) is replaced with  $\text{Nb}^{5+}$  (64 pm), and some of the  $\text{Ti}^{4+}$  (60.5 pm) atoms are reduced to  $\text{Ti}^{3+}$   
53 (67 pm) [17]. In contrast, in La-doped  $\text{SrTiO}_3$ , although the effective ionic radius of the B cations  
54 increases because of reduction of some  $\text{Ti}^{4+}$ , the effective ionic radius of the A cations decreases  
55 upon replacing  $\text{Sr}^{2+}$  (144 pm) with  $\text{La}^{3+}$  (136 pm) [17]. The lattice expansion that occurs in the  
56 case of La-doped  $\text{SrTiO}_3$  crystals thus cannot be explained simply as a change in effective ionic  
57 radii upon doping.

58 Janotti et al. revealed that changes in the electronic states of some atoms is an important factor  
59 in the lattice expansion of La-doped  $\text{SrTiO}_3$  [14]. On the other hand, it has also been reported that  
60 the lattice parameter of La-doped  $\text{SrTiO}_3$  decreases with an increase in dopant concentration,  
61 which was attributed to the formation of non-ionized O vacancies [18,19]. As perovskite oxides  
62 contain several different types of point defects, it is possible that formation of complex defects  
63 [12,20-24] and their interactions with dopant atoms are responsible for the observed changes in  
64 lattice parameters.

65 Dopant atoms in the crystal are also expected to alter not only the overall lattice parameter but  
66 also the local crystal lattice surrounding the dopant atoms [25]. Changes in interatomic spacings  
67 and local lattice distortions as a result of changes in the ionic radii and electronic states upon  
68 introduction of donor atoms in  $\text{SrTiO}_3$  are ultimately related to changes in Ti–O bonding states,  
69 which determine the material's band structure [26-31]. Changes in Ti–O bonding states, and  
70 distortions in the  $\text{SrTiO}_3$  crystal structure, are known to decrease the effective mass of electrons,  
71 because of changes in the shape and degeneracy of the conduction band [28-30]. In particular, the  
72 electron mobility in the crystal at room temperature is known to be influenced by longitudinal  
73 optical (LO) phonon scattering [32]. Lattice distortions can also lower the LO phonon frequency,

74 reduce scattering, and possibly improve the overall conductivity [31]. Elucidating the mechanism  
75 of the lattice changes upon doping Nb and La atoms into the SrTiO<sub>3</sub> crystal is thus an important  
76 step in understanding the electronic properties of the doped material.

77 Herein, we report the lattice expansion and local lattice distortions in Nb- and La-doped SrTiO<sub>3</sub>  
78 single crystals investigated both experimentally and theoretically. Lattice parameters of Nb- and  
79 La-doped SrTiO<sub>3</sub> single crystals increased with the dopant concentration. During in-plane X-ray  
80 diffraction (XRD) analysis, broad intensities around the Bragg peak for the Nb- and La-doped  
81 SrTiO<sub>3</sub> single crystals were observed, indicating that both lattice expansion and contraction, i.e.,  
82 local lattice distortions, occurred in the crystals as a result of doping. This finding was confirmed  
83 by first-principles calculations. Furthermore, Schottky-like pairs of O and Sr vacancies were  
84 found to be capable of reducing the lattice expansion in both systems, because some atoms in the  
85 vicinity of the vacancies were displaced away from them and others toward them. These results  
86 provide useful insights into the influence of local structural changes on the electronic  
87 conductivity of electron-doped SrTiO<sub>3</sub>.

88

89

## 90 **II. METHODOLOGY**

### 91 **A. Experimental procedure**

92 Commercially available Nb-, La- and non-doped SrTiO<sub>3</sub> single crystals (Furuuchi Co.), which  
93 were grown by the Verneuil method, were used in this study. The surfaces of all the crystals were  
94 treated by chemical-mechanical polishing. The Nb concentrations in SrTi<sub>1-x</sub>Nb<sub>x</sub>O<sub>3</sub> were 0.02 ( $x =$   
95 0.0002), 0.1 ( $x = 0.001$ ), 0.2 ( $x = 0.002$ ), 1.0 ( $x = 0.01$ ), and 2.0 at.% ( $x = 0.02$ ). The La

96 concentrations in  $\text{Sr}_{1-y}\text{La}_y\text{TiO}_3$  were 0.1 ( $y = 0.001$ ), 0.5 ( $y = 0.005$ ), 1.0 ( $y = 0.01$ ), and 5.0 at.%  
97 ( $y = 0.05$ ). The size of the samples was about  $5 \times 5 \times 0.5 \text{ mm}^3$ .

98 Out-of-plane and in-plane XRD measurements were performed using a laboratory XRD system  
99 (SuperLab, Rigaku Co. [33]) at room temperature (24 °C). An incident X-ray source with a  
100 combined microfocus X-ray generator and multilayer confocal mirror optics was used, operated  
101 at 40 kV and 30 mA with  $\text{Cu K}\alpha_1$  ( $\lambda = 0.154059 \text{ nm}$ ) radiation. A monochromator and a  
102 diffracted-beam analyzer, both two-bounce Ge (220) channel-cut crystals, were used. The widths  
103 of the vertical and horizontal divergence slits were set to 1.2 and 1.8 mm, respectively. The  
104 widths of the vertical and horizontal receiving slits were set to 2 and 4 mm, respectively. These  
105 slit widths were set to the substantial maximum values of the incident X-ray source width and the  
106 acceptance angle of the diffracted-beam analyzer crystal. A schematic drawing of the  
107 experimental setup used for the in-plane XRD measurements is shown in Fig. S1.

108 The irradiation width in the vertical direction along the incident X-ray is approximately 1.8 mm,  
109 depending on the horizontal divergence slit, and that in the parallel direction is approximately 20  
110 to 30 mm, depending on the incidence angle. The incidence angles for the in-plane measurements  
111 were set to approximately  $0.35^\circ$  to  $0.40^\circ$ , slightly larger than the critical angle  $\alpha_c$  ( $\sim 0.31^\circ$ ) of  
112  $\text{SrTiO}_3$ . For these incidence angles, the X-ray attenuation length or penetration depth was  
113 approximately 80 nm [34-36]. Under these optical conditions, the full width at half maximum  
114 (FWHM) of the rocking curve of the 020 reflection in the non-doped  $\text{SrTiO}_3$  single crystal was  
115  $0.0030^\circ$  (10.8 arcsec), in good agreement with that from a study on  $\text{SrTiO}_3$  single crystals [37].

116

## 117 **B. Calculation procedure**

118 The lattice expansion and local lattice distortions of Nb- and La-doped SrTiO<sub>3</sub> were investigated  
119 using first-principles calculations within the framework of the projector augmented wave (PAW)  
120 method based on density functional theory (DFT). To determine the donor-doped electronic  
121 structure correctly, spin-polarized calculations were performed. The exchange-correlation  
122 potential was treated using the generalized gradient approximation (GGA), and the on-site  
123 Coulomb potential was set at the Ti 3d orbital with  $U = 4.36$  eV [38], Nb 4d orbital with  $U = 5.00$   
124 eV [39], and La 4f orbital with  $U = 10.32$  eV [40], as implemented in the VASP code [41,42]. The  
125 cutoff energy for the planewave basis sets was 500 eV. Supercells containing 40, 135, 320, 625,  
126 and 1,080 atoms were generated by repeating the unit cell of SrTiO<sub>3</sub>  $2 \times 2 \times 2$ ,  $3 \times 3 \times 3$ ,  $4 \times 4 \times$   
127  $4$ ,  $5 \times 5 \times 5$ , and  $6 \times 6 \times 6$  times, respectively, along each of the principal directions. The k-point  
128 sampling in the Brillouin zone was performed using  $4 \times 4 \times 4$  (40 atoms),  $3 \times 3 \times 3$  (135 atoms),  
129 and  $2 \times 2 \times 2$  (320 atoms) meshes, including the  $\Gamma$  point, using a Monkhorst–Pack scheme, and  
130 single  $\Gamma$ -point calculations were performed for the 625-atom and 1,080-atom supercells. Each  
131 supercell contained either one dopant atom, one vacancy, one or two electronic defects, a single  
132 dopant/vacancy pair, or a combination of these in the case of cluster calculations. The dopant and  
133 vacancy concentrations of the 40-, 135-, 320-, 625-, and 1,080-atom supercells were  
134 approximately 12.5, 3.7, 1.6, 0.8, and 0.46 at.%, respectively. The lattice and atomic positions  
135 were fully optimized until the residual forces on all the atoms were below 0.03 eV/Å. In the  
136 calculations of charged vacancies, the total charge of the supercells was neutralized using a  
137 jellium background charge.

138 For dopant concentrations below 5 at.%, Nb- and La-doped SrTiO<sub>3</sub> were cubic at room  
139 temperature, without TiO<sub>6</sub> octahedra rotations [15,43]. To minimize the effect of octahedral  
140 rotation, simulations were performed under symmetry constraints; in most cases (i.e., single  
141 defect models), O<sub>h</sub> symmetry was applied.

142 It is known that the Pulay stress is an important factor affecting cell-size relaxation. The  
143 estimated Pulay stresses in the primitive cell calculations using cut-off energies of 400, 500, and  
144 600 eV were 1.29, -0.26, and  $0.14 \times 10^3$  bar, respectively, indicating that the 400 eV calculation  
145 provides inaccurate results. In this study, to perform systematic calculations for large models  
146 containing more than 1,000 atoms, we selected 500 eV as the cut-off energy despite its lower  
147 accuracy compared with 600 eV. For low concentrations (320-, 625-, and 1,080-atom supercells)  
148 of Nb-doped SrTiO<sub>3</sub>, the calculated Pulay stresses were  $0.2\text{--}0.4 \times 10^3$  bar, which are not greatly  
149 different from those in other calculations. Based on these preliminary results, we adjudged the  
150 present conditions to be sufficiently accurate for performing cell-size relaxation.

151

### 152 **III. RESULTS AND DISCUSSION**

#### 153 **A. Out-of-plane and in-plane XRD patterns of non-doped SrTiO<sub>3</sub> single crystal**

154 Figure 1a shows the out-of-plane and in-plane XRD patterns obtained from a non-doped SrTiO<sub>3</sub>  
155 single crystal. The out-of-plane and in-plane XRD patterns were measured under the same optical  
156 conditions, using 002 and 020 reflections of SrTiO<sub>3</sub>, respectively. The FWHM of the peaks of the  
157 out-of-plane and in-plane XRD patterns are  $0.0066^\circ$  and  $0.0052^\circ$ , respectively, indicating that  
158 both XRD patterns have a similar FWHM value. In contrast, the full width at tenth maximum  
159 (FWTM) of the out-of-plane and in-plane peaks are  $0.0128^\circ$  and  $0.0092^\circ$ , respectively, indicating  
160 that the peak shape of the out-of-plane pattern becomes much wider than that of the in-plane  
161 pattern with an increase in the distance from the Bragg peak. This is because of the effect of  
162 crystal truncation rod (CTR) scattering, which originates from the interface between the crystal  
163 surface and vacuum in the out-of-plane XRD pattern [44,45]. Conversely, when measuring  
164 in-plane diffraction, the intensity can be regarded as corresponding to diffraction from an

165 infinitely continuous crystal without any interface between the crystal surface and vacuum.  
166 Hence, the in-plane pattern can be obtained directly from the intensity of the diffracted X-rays,  
167 without taking the intensity of the CTR scattering into account.

168 The influence of CTR scattering on the out-of-plane XRD pattern can be more clearly observed  
169 on a log-scale intensity axis, as shown in Fig. 1b. For the out-of-plane method, the small peak  
170 intensities caused by the dopant atoms in the crystal are buried under the intensity of the CTR  
171 scattering. To detect the slight lattice distortions originating from dopant atoms in the crystal,  
172 in-plane analysis is a more powerful method because the peak shape of the in-plane XRD pattern  
173 is sharp in the absence of CTR scattering, as shown in the inset of Fig. 1b. In this study, we  
174 therefore investigated the lattice parameter and local lattice distortions using the in-plane XRD  
175 technique.

176

## 177 **B. Lattice expansion in Nb- and La-doped SrTiO<sub>3</sub> single crystals**

178 Figure 2 compares the in-plane XRD patterns obtained from the non-, Nb-, and La-doped  
179 SrTiO<sub>3</sub> single crystals. Each Bragg peak shifted to a lower angle, indicating that the overall lattice  
180 parameter increased with dopant concentration, as shown in Figs. 2b and 2d. The relationships  
181 between lattice expansion and concentration of the two dopants is in good agreement with earlier  
182 powder diffraction results [12-16].

183 Figure 3 shows the lattice parameters of Nb- and La-doped SrTiO<sub>3</sub> as a function of dopant  
184 concentration. The lattice parameter of the non-doped SrTiO<sub>3</sub> single crystal was  $a = 0.39051(9)$   
185 nm; the structure was estimated from the 020 diffraction to be that of a cubic perovskite, in good  
186 agreement with an earlier study of SrTiO<sub>3</sub> single crystals ( $a = 0.3905268$  nm) [37]. In the case of  
187 Nb- and La-doped SrTiO<sub>3</sub>, lattice parameters increased with dopant concentration. Results of

188 lattice parameter and rocking curve measurements for the Nb- and La-doped SrTiO<sub>3</sub> single  
189 crystals are summarized in Table 1, with plots of the rocking curves provided in Supplemental  
190 Material as Fig. S6.

191 Defect formation in complex perovskite oxides is closely related to lattice expansion [12,20-24].  
192 In this study, the non-, Nb-, and La-doped SrTiO<sub>3</sub> single crystals were grown by the Verneuil  
193 method [46]. In this method, the atmosphere for crystal growth is reducing because of the  
194 presence of hydrogen gas and because the flame temperature for melting the starting materials is  
195 approximately 2,000°C. Under a reducing atmosphere and at high growth temperatures, the Ti  
196 vacancies can be ignored because of the high formation energy for the SrTiO<sub>3</sub> crystal [47]. The  
197 mass-action relationship between Sr and O vacancies is  $[V_{Sr}''][V_O^{\bullet\bullet}] = K_s(T)$ , where  $K_s(T)$  is the  
198 temperature-dependent formation constant [48], and the substitution and defect states are  
199 described using Kröger–Vink notation [49]. The mass-action relationship indicates that the  
200 concentration of  $V_{Sr}''$  can be decreased by increasing the concentration of  $V_O^{\bullet\bullet}$  at high  
201 temperatures and under a reducing atmosphere [48,50]. For donor ( $D^{\bullet}$ )-doped SrTiO<sub>3</sub>, the  
202 mass-action relationship between O vacancies and donors in the as-grown SrTiO<sub>3</sub> crystal under a  
203 reducing atmosphere becomes  $[V_O^{\bullet\bullet}] > [D^{\bullet}] > [V_{Sr}'']$  [12,48,50]. After annealing treatments to  
204 remove only the O vacancies from donor-doped SrTiO<sub>3</sub> crystals [46,51], the mass-action  
205 relationship can be assumed to be  $[D^{\bullet}] > [V_O^{\bullet\bullet}] \approx [V_{Sr}'']$ . This leads to the favorable situation of  
206 having only residual O vacancies, owing to the lower energy of formation of the  $V_O^{\bullet\bullet} + V_{Sr}''$   
207 partial Schottky defect compared with that of a single Sr vacancy [47]. These considerations  
208 suggest that doping of SrTiO<sub>3</sub> single crystals should influence the lattice expansion by altering  
209 the balance between different defect concentrations, even though both Sr and O vacancies will  
210 still be present in the crystal to a certain degree.

211 In SrTiO<sub>3</sub> crystals grown by the Verneuil process, several other types of defect are known to be

212 present in addition to point defects and donor atoms [52-54]. It is possible that such extended  
213 defects and inhomogeneities may also affect the Bragg peak in in-plane XRD measurements. In  
214 this study, using XRD topography, transmitted light microscopy, low- and high-magnification  
215 scanning transmission electron microscopy, as well as analyzing the measurement location  
216 dependence using in-plane XRD, we detected a number of extended defects, primarily  
217 misorientations and domain boundaries; details are given in Section 3 of Supplemental Material.  
218 Although use of samples of higher crystallinity (fewer extended defects) is to be preferred, the  
219 concentrations of the extended defects were deemed to be sufficiently low that the experimentally  
220 observed structures and properties of our Nb- and La-doped SrTiO<sub>3</sub> single crystals could be  
221 interpreted primarily in terms of local lattice expansion and distortion around dopants.

222

### 223 **C. Local lattice distortions in Nb- and La-doped SrTiO<sub>3</sub> single crystals**

224 Broad intensities around the Bragg peak, indicated by the red and black arrows in Figs. 2a and  
225 2c, were observed for Nb- and La-doped SrTiO<sub>3</sub> single crystals. These broad peaks indicate that  
226 there are regions of local lattice expansion (black arrows) and contraction (red arrows) within the  
227 crystal. In neither case could the broad peaks be fitted with a single Gaussian curve. For example,  
228 Fig. 4 shows the region around the Bragg peak in the case of 0.02 at.% Nb-doped SrTiO<sub>3</sub>. The  
229 magnitudes of local lattice expansion and contraction ranged from 0.05 to 0.19 pm and from 0.04  
230 to 0.22 pm (Table 2), respectively, as calculated by the difference between the peak centers of the  
231 Gaussian curves and the Bragg peak. This suggests that there are different regions in the crystal  
232 with different magnitudes and extents of local lattice distortion, whether in compression or  
233 tension.

234 Gaussian curves corresponding to expanded and contracted regions are shown as dotted orange  
235 and blue lines in Fig. 4a, respectively, and the ratios of the summed intensities of the two sets are

236 plotted in Figs. 4b and 4c as a function of dopant concentration. The fitted curves for all the  
237 in-plane XRD patterns are shown in Fig. S2 of Supplemental Material.

238 For both Nb- and La-doped SrTiO<sub>3</sub>, contraction and expansion intensities are larger and more  
239 diverse than those of non-doped SrTiO<sub>3</sub>. Contraction intensities of the Nb- and La-doped SrTiO<sub>3</sub>  
240 crystals generally decrease with increasing dopant concentration, except in the case of 5 at.%  
241 La-doped SrTiO<sub>3</sub>, where the intensity increases markedly compared to the 1 at.% crystal. On the  
242 other hand, the expansion intensities do not depend strongly on the dopant concentration, and the  
243 contraction intensities are larger than the expansion intensities for both Nb- and La-doped  
244 SrTiO<sub>3</sub>.

245 Figure 4d shows the ratios of the contraction intensities of the Nb- and La-doped SrTiO<sub>3</sub> single  
246 crystals as a function of the FWHM of the rocking curves. The contraction intensities of the two  
247 crystals tend to increase with the FWHMs of the rocking curves. This indicates that the peak  
248 widths of the in-plane XRD patterns and rocking curves broaden with a decrease in dopant  
249 concentration. The FWHMs of the rocking curves provide information on the crystallinity or lack  
250 thereof, e.g., misorientations [53,55], dislocation densities [56,57], and other defects. In this study,  
251 we carefully examined the misorientations, dislocations, and inhomogeneities in SrTiO<sub>3</sub> single  
252 crystals, and details of these measurements with further discussion are provided in Section 3 of  
253 Supplemental Material. The broad intensities around the Bragg peak and the FWHMs of the  
254 rocking curves were found to be only weakly influenced by these extended defects, however.

255 The above analysis indicates that the overall lattice expansion and local lattice distortions in Nb-  
256 and La-doped SrTiO<sub>3</sub> single crystals depend on the nature and concentration of the dominant  
257 defects, including dopant atoms.

258

259 **D. Calculations of dopant atoms and point defects in SrTiO<sub>3</sub> crystals.**

260 To understand the influence of various kinds of defects on lattice expansion and local lattice  
261 distortions in doped SrTiO<sub>3</sub>, further first-principles calculations were performed using supercells  
262 of different sizes containing positively charged defects Nb<sub>Ti</sub><sup>•</sup>, La<sub>Sr</sub><sup>•</sup>, and V<sub>O</sub><sup>••</sup>; neutral defects Nb<sub>Ti</sub><sup>x</sup>,  
263 La<sub>Sr</sub><sup>x</sup>, and V<sub>O</sub><sup>x</sup>; or isolated charges e' and 2e'. In standard Kröger–Vink notation, Nb<sub>Ti</sub><sup>x</sup>, La<sub>Sr</sub><sup>x</sup>,  
264 and V<sub>O</sub><sup>x</sup> imply that the electrons are localized on the defect, but it is more likely that the electron  
265 is donated to the transition metal atoms, so to represent this situation we write these defects as  
266 Nb<sub>Ti</sub><sup>•</sup> + e' , La<sub>Sr</sub><sup>•</sup> + e' , and V<sub>O</sub><sup>••</sup> + 2e' from here on. This was confirmed by the unbonded  
267 electrons becoming delocalized within the conduction bands (CB) of the neutral defect and  
268 isolated electron systems.

269 Figure 5 shows schematic band structures of each electron- or donor-doped system. No extra  
270 electrons are available to enter the CB in the case of states i, ii, and iii, i.e., perfect (undoped)  
271 SrTiO<sub>3</sub> and positively charged dopant or defect systems. Relaxation of the crystal structure  
272 around these defects reveals the effects of positively charged point defects (both extrinsic and  
273 intrinsic) on the lattice parameter, absent the effects of electrons in the CB or the creation of  
274 subvalent Ti ions. In the case of systems containing net neutral defect states, viz., Nb<sub>Ti</sub><sup>•</sup> + e', La<sub>Sr</sub><sup>•</sup>  
275 + e', and V<sub>O</sub><sup>••</sup> + 2e', the electrons are located in the CB, as is also the case for the electron-doped  
276 systems. Examining the relaxed structures of these systems sheds further light on the effect of  
277 dopant atoms and point defects (including trivalent Ti ions) under the appropriate experimental  
278 conditions.

279

## 280 **E. Lattice parameter changes resulting from positively charged defect states**

281 Changes in lattice parameter,  $\Delta L$ , were calculated according to  $\Delta L = L_a - L_{\text{STO}}$ , where  $L_{\text{STO}}$  is  
282 the calculated lattice parameter of non-doped SrTiO<sub>3</sub> (0.397694 nm) and  $L_a$  is the calculated  
283 lattice parameter from supercells of SrTiO<sub>3</sub> containing dopant and/or other defect species. Figure

284 6 shows a plot of calculated changes in lattice parameter for models containing positively  
285 charged defects ( $\text{Nb}_{\text{Ti}}^{\bullet}$ ,  $\text{La}_{\text{Sr}}^{\bullet}$ , and  $\text{V}_{\text{O}}^{\bullet\bullet}$ ), as a function of dopant concentration. As summarized in  
286 the illustrations in Fig. 5, no extra electrons were present in the CBs for these defect species.

287 In the case of the  $\text{La}_{\text{Sr}}^{\bullet}$  containing system, the lattice parameter decreased because of a decrease  
288 in the effective ionic size of the A-site cations when  $\text{Sr}^{2+}$  is replaced with  $\text{La}^{3+}$ . In the case of  
289  $\text{Nb}_{\text{Ti}}^{\bullet}$ , however, the lattice parameter changed only slightly, owing to the smaller difference in  
290 ionic radii between  $\text{Nb}^{5+}$  and  $\text{Ti}^{4+}$ . For the double-charged O vacancy,  $\text{V}_{\text{O}}^{\bullet\bullet}$ , the lattice parameter  
291 decreased with increasing vacancy concentration, consistent with the removal of the large  $\text{O}^{2-}$   
292 ions.

293 Previous calculations have shown that when vacancies are present in ionic crystals such as  
294 perovskite oxides the local lattice around the vacancy often expands as a result of the electrostatic  
295 repulsion between its neighboring cations [22]. Our calculations of  $\text{V}_{\text{O}}^{\bullet\bullet}$ -containing  $\text{SrTiO}_3$   
296 suggest that contraction of the lattice beyond the neighboring cation shell is enough to  
297 compensate for this expansion, producing an overall decrease in lattice volume. This suggests  
298 that the lattice contraction reported previously for La-doped  $\text{SrTiO}_3$  may have been caused by  
299 anion vacancies (i.e., oxygen substoichiometry) in the crystals [18,19].

300

### 301 **F. Lattice parameter changes for neutral defect systems**

302 In contrast to the above defect species, the neutral defect systems  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$ ,  $\text{La}_{\text{Sr}}^{\bullet} + e'$ , and  $\text{V}_{\text{O}}^{\bullet\bullet}$   
303  $+ 2e'$ , and free electron states  $e'$  and  $2e'$ , produced local lattice expansion with increasing dopant  
304 concentration, as seen in Fig. 7. Because positively charged defects on their own result in lattice  
305 contraction (Fig. 6), the presence of electrons in the CB, corresponding to creation of trivalent Ti  
306 species, appears to be responsible for the overall lattice expansion in these cases. This is most  
307 noticeable in systems containing free charges, as the increase in lattice parameters is greatest (Fig.

308 7).

309 The SrTiO<sub>3</sub> crystal with V<sub>O</sub><sup>••</sup> + 2e' exhibits only a small lattice expansion even when the dopant  
310 concentration is above 12 at.%. This is because the decrease in lattice volume caused by the  
311 positively charged vacancy is counterbalanced by the increase due to the larger ionic size of  
312 trivalent Ti ions. The slight change in lattice parameter for V<sub>O</sub><sup>••</sup> + 2e' predicted from our  
313 calculations is consistent with the earlier experimental results obtained for O-deficient SrTiO<sub>3</sub>  
314 [11,58]. In contrast, lattices of systems containing Nb<sub>Ti</sub><sup>•</sup> and La<sub>Sr</sub><sup>•</sup> expanded when donor electrons  
315 were also included.

316 To examine the origin of the lattice expansion caused by Nb<sub>Ti</sub><sup>•</sup> + e' and La<sub>Sr</sub><sup>•</sup> + e' dopant states in  
317 more detail, we compared the calculated lattice parameters with experimental values of Nb- and  
318 La-doped SrTiO<sub>3</sub> crystals as a function dopant concentration, as shown in Fig. 8a. Both  
319 experiment and theoretical modeling indicate that the rates of expansion for La-doped SrTiO<sub>3</sub> are  
320 smaller than those for the Nb-doped SrTiO<sub>3</sub> crystals, suggesting that the theoretical models  
321 describe an important mechanism behind the lattice expansion. However, although the  
322 experimental trend is qualitatively reproduced by the DFT calculations, the absolute values of the  
323 lattice parameters are overestimated for both dopants (Fig. 8a).

324 One of the reasons for the discrepancy between the calculated and experimental values could be  
325 that in the real materials defect complexes (clusters) form with native defects, such as partial  
326 Schottky defects V<sub>O</sub><sup>••</sup> + V<sub>Sr</sub><sup>''</sup>. To examine the effect of such defect complexes on the crystal  
327 lattice, we performed DFT calculations of models containing various combinations of defects  
328 with V<sub>Sr</sub><sup>''</sup> and/or V<sub>O</sub><sup>••</sup>, and the results are summarized in Fig. 8b. Lattice parameters of crystals  
329 containing point defects V<sub>Ti</sub><sup>''</sup> + 4h<sup>•</sup>, V<sub>Sr</sub><sup>''</sup> + 2h<sup>•</sup>, V<sub>O</sub><sup>••</sup> + 2e', 2Nb<sub>Ti</sub><sup>•</sup> + V<sub>Sr</sub><sup>''</sup>, and 2La<sub>Sr</sub><sup>•</sup> + V<sub>Sr</sub><sup>''</sup>, as well  
330 as the Schottky pair V<sub>Sr</sub><sup>''</sup> + V<sub>O</sub><sup>••</sup>, all increased as a result of electrostatic repulsion between the  
331 host cations surrounding the point defects [22]. The magnitude of the lattice expansion in the case

332 of the  $V_{\text{O}}^{\bullet\bullet} + V_{\text{Sr}}''$  cluster (half-filled diamond symbol in Fig. 8b) is smaller than the sum of the  
333 lattice expansions produced by  $V_{\text{O}}^{\bullet\bullet} + 2e'$  (filled triangle symbol in Fig. 8b) and  $V_{\text{Sr}}'' + 2h^{\bullet}$   
334 (reversed triangle symbol in Fig. 8b) defect states in isolation, indicating that interaction between  
335 point defects and absence of free electrons abates much of the inter-ion repulsion, leading to  
336 reduced overall lattice expansion.

337 Figure 8b shows that lattice expansion around dopant atoms is decreased in the vicinity of  $V_{\text{O}}^{\bullet\bullet}$   
338 +  $V_{\text{Sr}}''$  Schottky-type pairs. As partial Schottky defects in  $\text{SrTiO}_3$  are more energetically favorable  
339 than full Schottky defects [47], this may explain the large difference between the magnitudes of  
340 lattice expansion observed experimentally and those predicted from DFT using isolated dopant  
341 atoms and electrons only. In contrast to partial Schottky defects, defect clusters of the form  
342  $2\text{Nb}_{\text{Ti}}^{\bullet} + V_{\text{Sr}}''$  and  $2\text{La}_{\text{Sr}}^{\bullet} + V_{\text{Sr}}''$  cannot explain the difference between the experimental and  
343 simulation results, as both produce an excessive lattice expansion, despite  $\text{La}_{\text{Sr}}^{\bullet}$  dopants on their  
344 own being shown to cause lattice contraction (Fig. 6). These results suggest that charged O  
345 vacancies  $V_{\text{O}}^{\bullet\bullet}$  are important in abating lattice expansion of donor-doped  $\text{SrTiO}_3$ .

346 To better understand the relaxation behavior around vacancy pairs and dopants, we plotted local  
347 cation displacements in the vicinity of dopants/defects as vector images. Figure 9 shows the  
348 three-dimensional vector plots of Sr ion displacements around  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$ ,  $\text{Nb}_{\text{Ti}}^{\bullet} + e' + V_{\text{O}}^{\bullet\bullet} + V_{\text{Sr}}''$ ,  
349  $\text{La}_{\text{Sr}}^{\bullet} + e'$ , and  $\text{La}_{\text{Sr}}^{\bullet} + e' + V_{\text{O}}^{\bullet\bullet} + V_{\text{Sr}}''$  defect clusters in the 625-atom supercell, corresponding to  
350 the results in Fig. 8b. The vectors indicate the magnitudes and directions of the displacements  
351 from the ideal cubic perovskite lattice positions, estimated using the calculated supercell volume.  
352 Since the displacements of the Ti and O ions were much smaller than those of the Sr ions, for  
353 clarity only the displacements of the Sr ions are shown. The Sr ions adjacent to  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  (Fig.  
354 9a) and  $\text{La}_{\text{Sr}}^{\bullet} + e'$  defects (Fig. 9b) are uniformly displaced away from the defects (expansion).  
355 The maximum displacements of the Sr ions near  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  and  $\text{La}_{\text{Sr}}^{\bullet} + e'$  defects are 3.6 and 1.5

356 pm, respectively. The difference in the magnitude of these displacements depends on the type of  
357 dopant, indicating that this plays a decisive role in determining the total lattice expansion.

358 For each donor/Schottky pair cluster, the Sr ion displacements are larger than those for the  
359 single dopant atoms (maximum  $\sim 5.0$  pm for  $\text{Nb}_{\text{Ti}}^{\bullet} + e' + \text{V}_{\text{O}}^{\bullet\bullet} + \text{V}_{\text{Sr}}''$  and  $\sim 3.9$  pm for  $\text{La}_{\text{Sr}}^{\bullet} + e' +$   
360  $\text{V}_{\text{O}}^{\bullet\bullet} + \text{V}_{\text{Sr}}''$ ), as shown in Figs. 9c and 9d. In addition, the directions of displacement vectors of Sr  
361 ions around the Schottky pair are not uniform, and some relax towards the defect (local  
362 contraction). This complex arrangement of cation displacements caused by the Schottky pair  
363 reduce the strain field induced by the dopants or other point defects in the crystal, resulting in a  
364 reduction in the overall lattice expansion of donor-doped  $\text{SrTiO}_3$ .

365 When the Sr and O vacancies of the Schottky pair are far away from each other in the supercell,  
366 the lattice parameter,  $L_a$  ( $\sim 0.397851$  nm for  $\text{Nb}_{\text{Ti}}^{\bullet} + e' + \text{V}_{\text{O}}^{\bullet\bullet} + \text{V}_{\text{Sr}}''$ , or  $\Delta L = \sim 0.157$  pm, and  
367  $\sim 0.397762$  nm for  $\text{La}_{\text{Sr}}^{\bullet} + e' + \text{V}_{\text{O}}^{\bullet\bullet} + \text{V}_{\text{Sr}}''$ , or  $\Delta L = \sim 0.068$  pm), is slightly larger than when the Sr  
368 and O vacancies are close together, such as in the model shown in Fig. 9c ( $L_a = \sim 0.397829$  nm or  
369  $\Delta L = \sim 0.135$  pm) and Fig. 9d ( $L_a = \sim 0.397759$  nm or  $\Delta L = \sim 0.065$  pm), respectively, although in  
370 both cases the partial Schottky pair has an ameliorating effect on lattice expansion. This indicates  
371 that it is necessary to take into account both the arrangement and concentration of the dopants  
372 and point defects in the crystal to obtain a more accurate model using this theoretical approach.  
373 This may be challenging because the first-principles calculations are already computationally  
374 intensive owing to the size of the supercell and relatively large atom displacements during lattice  
375 relaxation. Nevertheless, the present results provide useful insights into the local changes in  
376 bonding environments and lattice strains caused by electron doping of  $\text{SrTiO}_3$ .

377

### 378 **G. Local lattice distortion induced by dopant atoms**

379 To further understand the changes caused by the dopants and vacancies, we analyzed the local

380 lattice expansion and contraction around the dopants in the calculated models in terms of  
381 individual lattice cells. This was done by calculating the distances between A-site cations in  
382 individual cells (i.e., cell lengths)) as well as B-site cations in individual cells for each relaxed  
383 supercells. It is important to consider both the A-centered cells and B-centered cells because  
384 changes in both sublattices influence the shape of the reciprocal lattice and hence Bragg peak.

385 As an example, Figures 10a and 10b show the relaxed structures of the  $6 \times 6 \times 6$  (1,080 atoms;  
386 dopant concentration approximately 0.46 at.%) supercells for  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  and  $\text{La}_{\text{Sr}}^{\bullet} + e'$  systems,  
387 respectively, sliced through the plane of the dopant atom with background colors indicating the  
388 magnitude of the area ratio  $S_{\text{Sr-site}}/S_{\text{ave}}$ , where  $S_{\text{Sr-site}}$  is the projected area bounded by Sr sites at  
389 the corners of a single cell, and  $S_{\text{ave}}$  is the projected area of the ideal cubic perovskite unit cell for  
390 the relaxed supercell.

391 In both doped systems, the Sr and Ti atoms immediately adjacent to the dopant atoms are  
392 displaced away from the dopants. The magnitudes of the displacements are greater in the case of  
393 the  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  system than the  $\text{La}_{\text{Sr}}^{\bullet} + e'$  system, but in both cases Sr atoms are displaced slightly  
394 more than Ti atoms. Compared to the nearest-neighbor cations, O ions adjacent to Nb are only  
395 slightly displaced from their ideal sites, away from the Nb atom as a result of the larger ionic  
396 radius of  $\text{Nb}^{5+}$  compared to  $\text{Ti}^{4+}$ . In contrast, nearest-neighbor O ions around La undergo a much  
397 larger displacement towards the dopant because the relative increase in effective charge is larger  
398 and is concentrated in a smaller volume,  $\text{La}^{3+}$  having a smaller ion radius than  $\text{Sr}^{2+}$ .

399 The differences in the responses of the cations can be explained in terms of the different  
400 coordination environments of the A- and B-site cations and the response of the nearest-neighbor  
401 O atoms to the introduction of a donor dopant; contraction of the O shell (12 atoms) around La on  
402 the Sr site effectively shields the nearest shell of Ti ions from the increase in positive charge,  
403 whereas the movement away from the Nb on the Ti site of the O shell (6 atoms) exposes the

404 nearest shells of Sr and Ti cations to the increased effective charge of the dopant, resulting in  
405 greater repulsion and hence expansion of the cells adjacent to the dopant-containing cell. In the  
406 case of 0.46 at.%  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  model, expansion of the cells continues to the boundary of the  
407 supercell in the three principal axis directions. In the case of B-site doping, expansion is confined  
408 to the cells surrounding the dopant atom.

409 The displacements of Ti and Sr cations, and hence magnitudes of the cell contraction, decrease  
410 with increasing distance from the dopant, with cations at the midplane between two supercells  
411 remaining on ideal perovskite sites because of the balancing forces from dopants in image  
412 supercells. Within a single supercell, the net effect of the cation repulsion and anion attraction in  
413 the nearest-neighbor shells is a local lattice expansion ( $S_{\text{Sr-site}}/S_{\text{ave}} > 1$ ) which subjects the outer  
414 shells of cations to compressive strain fields ( $S_{\text{Sr-site}}/S_{\text{ave}} < 1$ ), as seen in Figs. 10a and 10b.

415 Plots of the ratio  $S_{\text{Sr-site}}/S_{\text{ave}}$  as a function of position in the lattice allow the distribution of  
416 expanded and contracted regions, and their relative magnitudes, to be analyzed. Figure 11 shows  
417 results for the 0.46 at.%  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  system. Results for the  $\text{La}_{\text{Sr}}^{\bullet} + e'$  system are included in Fig.  
418 S10 of Supplemental Material. Figure 11a is a histogram of cell lengths (for clarity, only a narrow  
419 range either side of the average (overall) lattice parameter is shown), which reveals that more  
420 lattice cells are compressed than expanded. Expanded cells are primarily those adjacent to the  
421 dopant, as seen in Fig. 10 and the inset of Fig. S10a, whereas compressed cells are dispersed over  
422 a wider volume and further from the dopant. As the large local expansion near the dopant is  
423 repeated periodically in image supercells, the compressed regions are mirrored across the  
424 periodic boundaries between supercells. Although lattice expansion adjacent to the dopant is to a  
425 large extent cancelled by compression of outer regions, the overall effect is a slight expansion of  
426 the lattice, corresponding to an increase in the lattice parameter of the supercell,  $L_a$ . For the 0.46  
427 at.%  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$  system, the average Sr-centered and Ti-centered cell lengths were both 0.39792

428 nm, very close to that of the supercell as a whole ( $L_a = \sim 0.39790$  nm), and greater than that of the  
429 undoped system,  $L = 0.39051$  nm.

430 Figures 11b and 11c show plots of the ratios of compressed and expanded cells, respectively, as  
431 a function of dopant concentration. Because the experimental lattice parameters in Table 2 vary  
432 by approximately  $\pm 0.25$  pm, only contracted cells falling in the range  $-0.25 < \Delta L < 0$  and  
433 expanded cells within  $0 < \Delta L < 0.25$  were used when calculating these ratios. The plots show that  
434 the number of contracted cells of Nb- and La-doped SrTiO<sub>3</sub> decrease with increasing dopant  
435 concentration, except in the case of 3.7 at.% La-doped SrTiO<sub>3</sub>. In contrast, the number of  
436 expanded cells does not appear to depend on the dopant concentration to any meaningful degree.  
437 Furthermore, the number of contracted cells is higher than the number of expanded cells in both  
438 cases; in other words, the increase in crystal lattice volume is concentrated in the region  
439 immediately surrounding the dopant. Although the highest dopant concentration in the theoretical  
440 models was only 0.46 at.%, the pattern or distribution of contracted and expanded regions is  
441 consistent with the experimental results in Fig. 4.

442 The computational results also indicate that local lattice distortions depend strongly on the  
443 dopant concentrations. The average distance between dopant atoms increases with decreasing  
444 dopant concentration in the manner shown in Fig. 12. In the region of low dopant concentrations,  
445 e.g., below 0.1 at.%, regions of lattice contraction are caused by local lattice expansion  
446 immediately surrounding the dopant atoms, as illustrated in the schematic model in Fig. 12. For  
447 lower dopant concentrations, e.g., below 0.001 at.%, the volume of unstrained crystal (regions far  
448 from dopants) is much larger than that of regions of lattice contraction and expansion, so the  
449 overall lattice parameter varies only slightly from that of the undoped crystal. For dopant  
450 concentrations greater than 1 at.%, dopant atoms are sufficiently close to each other ( $\sim 2$  nm or 5  
451 unit cells) that the regions between them undergo expansion only. In these two cases (low and

452 high dopant concentrations), the strain field distributions are more or less uniform, with less  
453 variation in local lattice distortions than is observed for intermediate dopant contents (0.1 to 1.0  
454 at.%).

455 The above computational results are consistent with the in-plane XRD measurements, which  
456 indicated that, as a volume percent, more of the lattice is in compression than tension in  
457 donor-doped crystals (Fig. 2). Although smaller in volume, the magnitude of lattice expansion in  
458 tensile regions is greater than the net contraction further away from the dopants, resulting in an  
459 overall increase in lattice parameter, as predicted from theoretical calculations. It is thus  
460 reasonable to conclude that one of the chief contributors to broadening of the Bragg peak,  
461 corresponding to expanded and contracted regions of the lattice, are the donor atoms themselves.

462

#### 463 **IV. CONCLUSIONS**

464 Lattice parameters and local lattice distortions of Nb- and La-doped SrTiO<sub>3</sub> single crystals  
465 were investigated through XRD measurements and first-principles calculations. The lattice  
466 parameters of Nb- and La-doped SrTiO<sub>3</sub> were found to increase with dopant concentration.  
467 In-plane XRD measurements allowed for detailed investigation of the sharp Bragg peak without  
468 any interference from CTR scattering. The broad intensities around the Bragg peak in Nb- and  
469 La-doped SrTiO<sub>3</sub> single crystals indicated the existence of local lattice distortions in the crystal.  
470 DFT calculations showed that the lattice expansions and local lattice distortions of Nb- and  
471 La-doped SrTiO<sub>3</sub> can be attributed to the introduction of donor dopant atoms. Lattice expansion  
472 was predominantly driven by the creation of trivalent Ti ions with electrons in the CB. The  
473 cationic displacements around the donor atoms resulted in the expansion of the local lattice, with  
474 locally contracted regions further from the defect. Formation of  $V_{Sr}'' + V_O''$  partial Schottky  
475 defects reduced the lattice expansion of Nb- and La-doped SrTiO<sub>3</sub> because some of the atoms

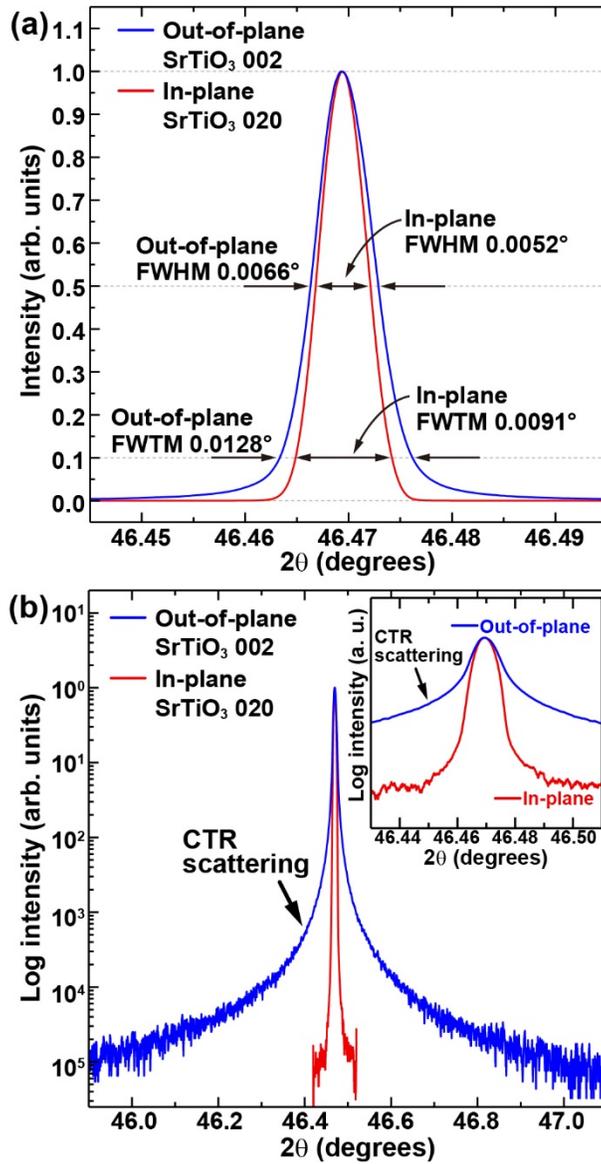
476 surrounding the vacancies and dopant atoms displace away from these point defects whereas  
477 other displace towards them. These results provide a consistent explanation of the influence of  
478 dopants on the electronic conductivity of electron-doped SrTiO<sub>3</sub> in terms of local structural  
479 distortions.

480

#### 481 **ACKNOWLEDGEMENTS**

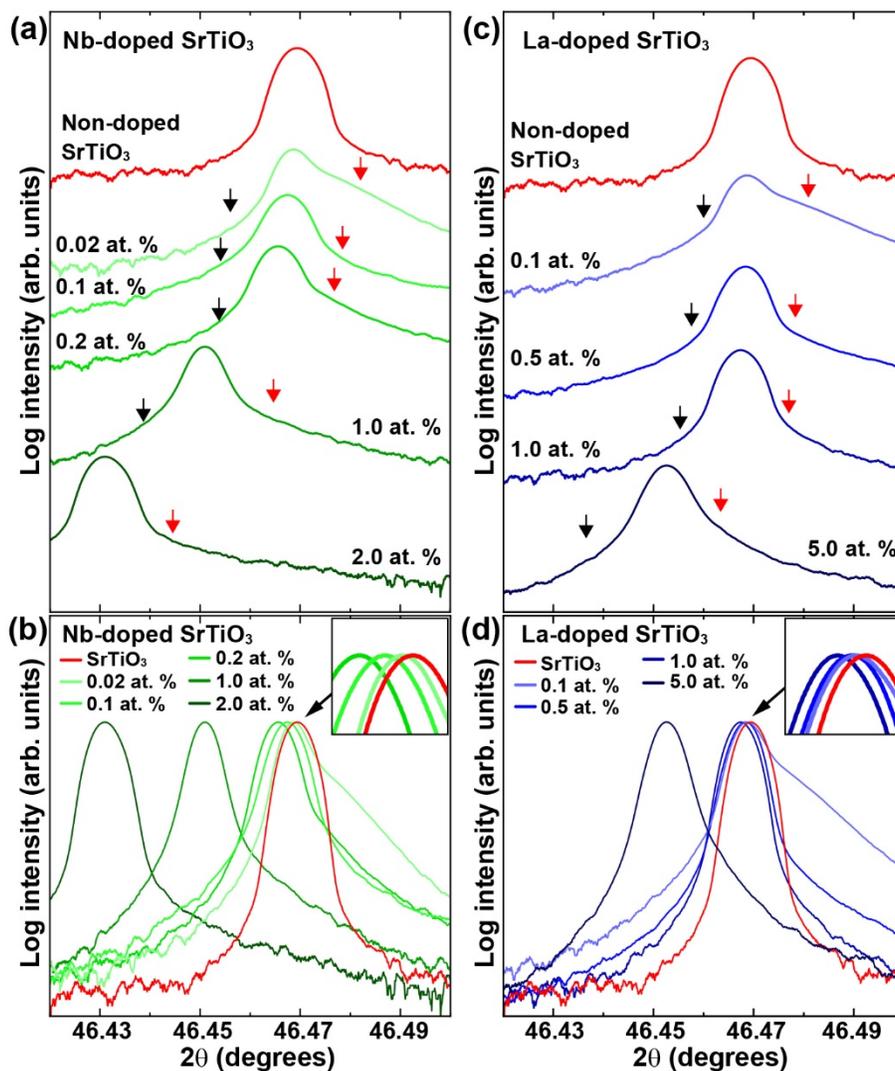
482 The authors thank C. A. J. Fisher and T. Yamamoto for the useful discussions. This work was  
483 supported by a Grant-in-Aid for Young Scientists B (Grant No. 17K14119), Scientific Research  
484 grants (Grant Nos. 25106003, 26249092, 17H06094) from the Japan Society for the Promotion of  
485 Science, JST-PRESTO, and Kazato Research Foundation of Japan. Part of this work was  
486 supported by the “Nanotechnology Platform” (project No. 12024046) sponsored by the Ministry  
487 of Education, Culture, Sports, Science and Technology of Japan.

#### 488 **Captions**



489  
 490 FIG. 1. (a) Linear and (b) log-scale intensities of out-of-plane and in-plane XRD patterns  
 491 obtained for a pure SrTiO<sub>3</sub> single crystal. Reflections of the out-of-plane and in-plane XRD  
 492 patterns are for 002 and 020 SrTiO<sub>3</sub>, respectively. Rescaled data are shown in the inset at the  
 493 upper right corner of (b).

494

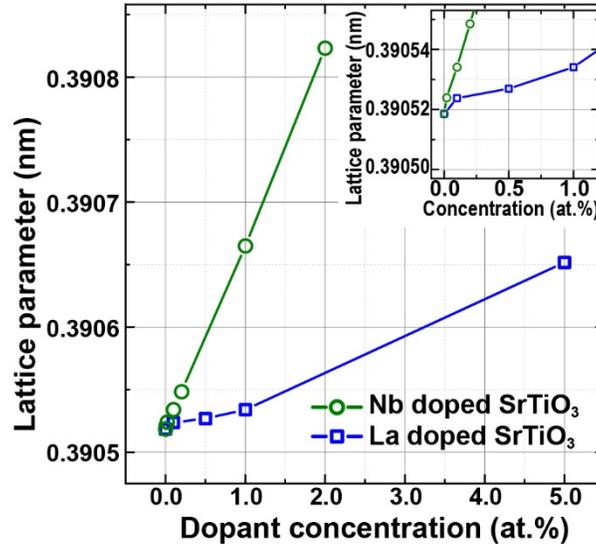


496

497

498 FIG. 2. In-plane XRD patterns obtained from 020 reflections of (a) Nb- and (b) La-doped SrTiO<sub>3</sub>  
 499 single crystals. The red and black arrows in (a) and (b) indicate the intensities for the crystal  
 500 lattice contraction and expansion, respectively. The rescaled data for Nb- and La-doped SrTiO<sub>3</sub>  
 501 are shown in (c) and (d), respectively. The red-line XRD patterns in all the images are from the  
 502 non-doped SrTiO<sub>3</sub> single crystal. The insets in the upper right corners of (c) and (d) show  
 503 magnified images of the regions around the peaks of non-doped SrTiO<sub>3</sub>.

504

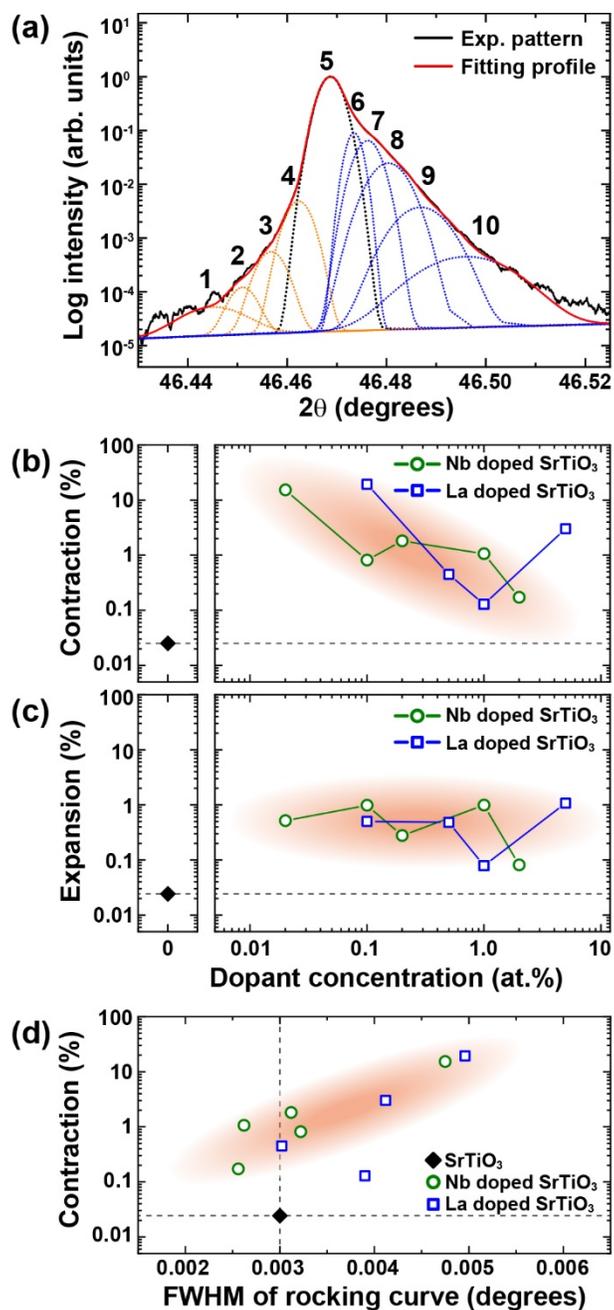


505  
 506 FIG. 3. Lattice parameters of Nb- and La-doped SrTiO<sub>3</sub> single crystals as a function of dopant  
 507 concentration. The inset shows a magnified view of the low-concentration region.

508  
 509 Table 1. XRD measurements of non-, Nb-, and La-doped SrTiO<sub>3</sub> single crystals. Lattice  
 510 parameters were calculated for a cubic perovskite structure, using Cu K $\alpha_1$  ( $\lambda = 0.154059$  nm).

511

Sample	Concentration / at.%	020 Bragg peak $2\theta / ^\circ$	Lattice parameter / nm	020 Rocking curve FWHM / $^\circ$
Non-doped SrTiO <sub>3</sub>	-	46.46941	0.39051(9)	0.0030(0)
Nb-doped SrTiO <sub>3</sub>	0.02	46.46873	0.39052(4)	0.0047(5)
	0.1	46.46745	0.39053(4)	0.0032(2)
	0.2	46.46563	0.39054(9)	0.0031(2)
	1	46.45095	0.39066(5)	0.0026(2)
	2	46.43107	0.39082(3)	0.0025(6)
La-doped SrTiO <sub>3</sub>	0.1	46.46875	0.39052(4)	0.0049(6)
	0.5	46.46835	0.39052(7)	0.0030(2)
	1	46.46745	0.39053(4)	0.0039(0)
	5	46.45265	0.39065(2)	0.0041(2)



512  
 513 FIG. 4. (a) Fitting profile obtained using Gaussian functions for the experimental XRD pattern of  
 514 0.02 at.% Nb-doped SrTiO<sub>3</sub>. The dotted black, orange, and blue lines indicate the Gaussian  
 515 functions used for the main, expansion, and contraction intensities, respectively. Gaussian curves  
 516 were numbered as labelled. (b) The contraction and (c) expansion intensity ratios calculated from  
 517 the fitting results in Fig. S2, as a function of dopant concentration. Horizontal axes in (b) and (c),  
 518 are log-scale, except for the values for non-doped SrTiO<sub>3</sub> shown as a black diamond on the left  
 519 side of each graph. (d) Ratios of the contraction intensities as a function of the FWHM of the  
 520 rocking curves. Gray dashed lines in (b), (c), and (d) are included to aid comparison between

521 Nb-doped, La-doped, and non-doped SrTiO<sub>3</sub>.

522

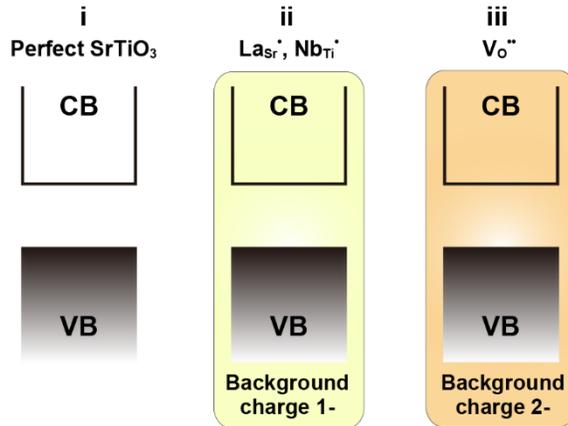
523 Table 2. Lattice parameters estimated from the center of the Gaussian curves in Fig. 4.  $d_N$  is the  
524 lattice parameter estimated from the center of each Gaussian curve for a cubic perovskite  
525 structure. The last column lists differences between the lattice parameter corresponding to the 5<sup>th</sup>  
526 peak and that corresponding to the  $N^{\text{th}}$  peak.

527

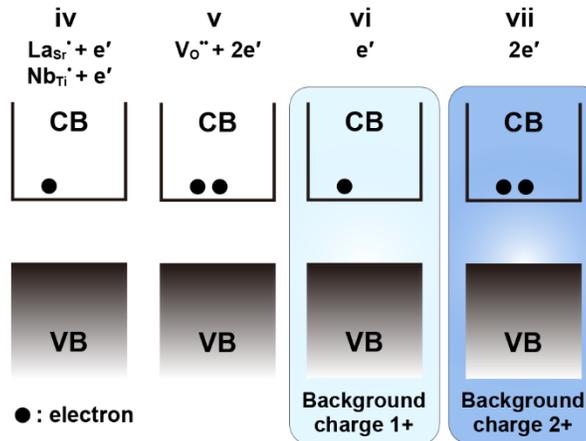
Gaussian peak	Lattice parameter $d_N / \text{nm}$	$d_N - d_5$ / pm
1	0.39071(1)	0.19
2	0.39066(4)	0.14
3	0.39061(9)	0.09
4	0.39057(5)	0.05
5	0.39052(4)	0.00
6	0.39048(6)	-0.04
7	0.39046(5)	-0.06
8	0.39043(2)	-0.09
9	0.39038(0)	-0.14
10	0.39030(7)	-0.22

528

I. No electron in conduction band



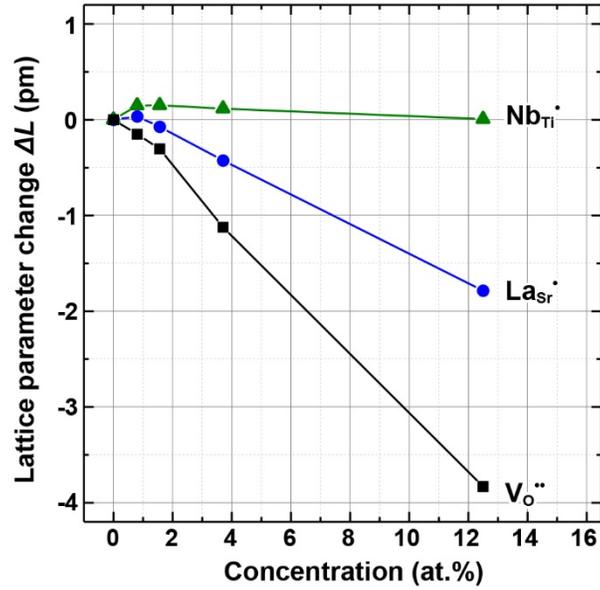
II. Electron/s in conduction band



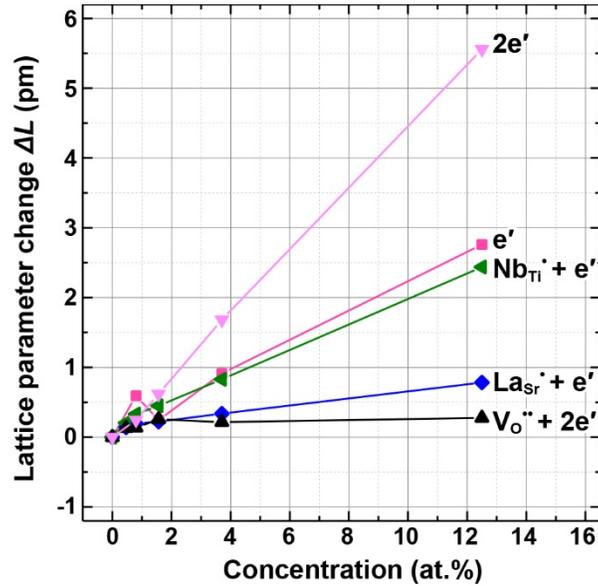
529

530 FIG. 5. Schematic diagram of electronic states in the valence band (VB) and conduction band  
 531 (CB) of  $\text{SrTiO}_3$ , as used in the calculation models. There are no electrons in the CB for **i**: initial  
 532 state, i.e., perfect  $\text{SrTiO}_3$ , **ii**: positively charged state 1+, i.e.,  $\text{La}_{\text{Sr}}^+$  and  $\text{Nb}_{\text{Ti}}^+$ , and **iii**: positively  
 533 charged state 2+, i.e.,  $\text{V}_{\text{O}}^{2+}$ . The total charges of supercells **ii** and **iii** were neutralized using jellium  
 534 background charges of 1- and 2-, respectively. **iv**: neutral state with one electron in the CB, i.e.,  $e'$ ,  
 535  $\text{Nb}_{\text{Ti}}^+ + e'$ , and  $\text{La}_{\text{Sr}}^+ + e'$ ; **v**: neutral state with two electrons in the CB, i.e.,  $\text{V}_{\text{O}}^{2+}$ ; **vi** and **vii**: neutral  
 536 states with one electron and two electrons, respectively, in the CB, i.e.,  $e'$  and  $2e'$ . The total  
 537 charge of the supercells of **vi** and **vii** was neutralized using jellium background charges of 1+ and  
 538 2+, respectively.

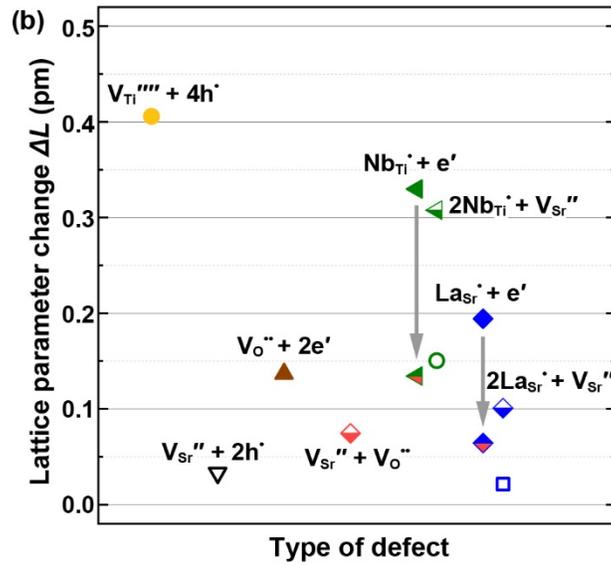
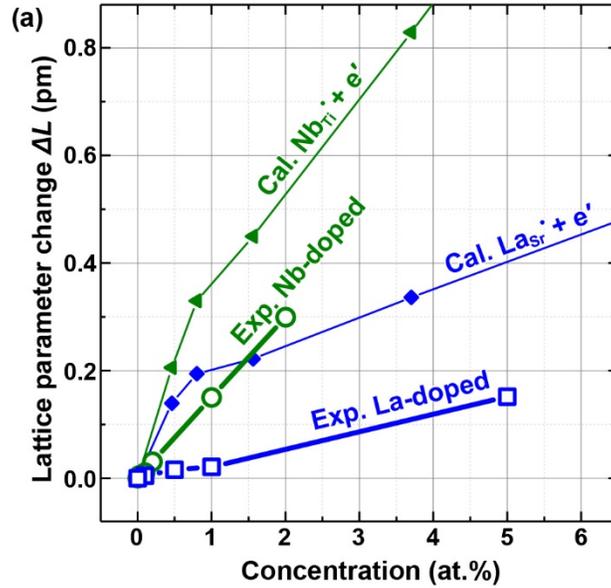
539



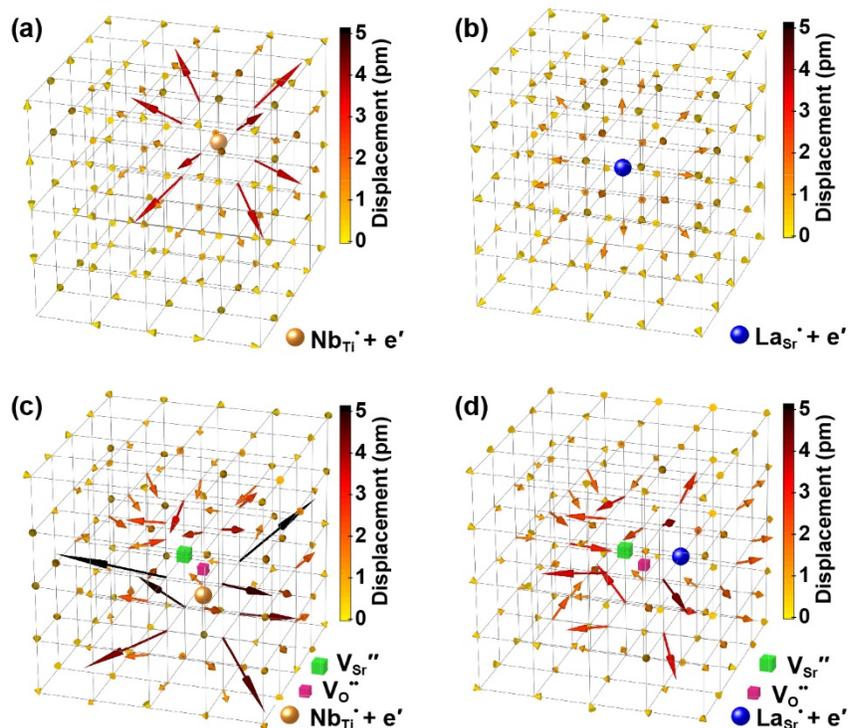
540  
 541 FIG. 6. Calculated lattice parameter changes as a function of dopant concentration for positively  
 542 charged defects  $\text{Nb}_{\text{Ti}}^{\bullet}$ ,  $\text{La}_{\text{Sr}}^{\bullet}$ , and  $\text{V}_{\text{O}}^{\bullet\bullet}$  when there are no electrons in the CB of  $\text{SrTiO}_3$ .  
 543



544  
 545 FIG. 7. Calculated lattice parameter changes as a function of dopant concentration for neutral  
 546 defects  $\text{Nb}_{\text{Ti}}^{\bullet} + e'$ ,  $\text{La}_{\text{Sr}}^{\bullet} + e'$ ,  $\text{V}_{\text{O}}^{\bullet\bullet} + 2e'$ , and free electrons  $e'$ , and  $2e'$  when electrons are present in  
 547 the CB of  $\text{SrTiO}_3$ .



548  
 549 FIG. 8. Changes in (a) experimental (Exp.) and calculated (Cal.) lattice parameters of SrTiO<sub>3</sub>  
 550 containing Nb<sub>Ti</sub><sup>•</sup> + e' and La<sub>Sr</sub><sup>•</sup> + e' defects, as a function of dopant concentration. (b) Changes in  
 551 lattice parameters for V<sub>Ti</sub><sup>''''</sup> + 4h<sup>•</sup>, V<sub>Sr</sub><sup>''</sup> + 2h<sup>•</sup>, V<sub>O</sub><sup>••</sup> + 2e', V<sub>Sr</sub><sup>''</sup> + V<sub>O</sub><sup>••</sup>, Nb<sub>Ti</sub><sup>•</sup> + e', La<sub>Sr</sub><sup>•</sup> + e', 2Nb<sub>Ti</sub><sup>•</sup> +  
 552 V<sub>Sr</sub><sup>''</sup>, and 2La<sub>Sr</sub><sup>•</sup> + V<sub>Sr</sub><sup>''</sup> defect models, as well as each dopant with a V<sub>Sr</sub><sup>''</sup> + V<sub>O</sub><sup>••</sup> partial Schottky  
 553 defect. Dopant and vacancy concentrations in the theoretical models were 0.8 at.% in all cases.  
 554 The experimental results for 1.0 at.% Nb-doped (green circle) and La-doped (blue square) SrTiO<sub>3</sub>  
 555 are included in (b) for comparison.



557

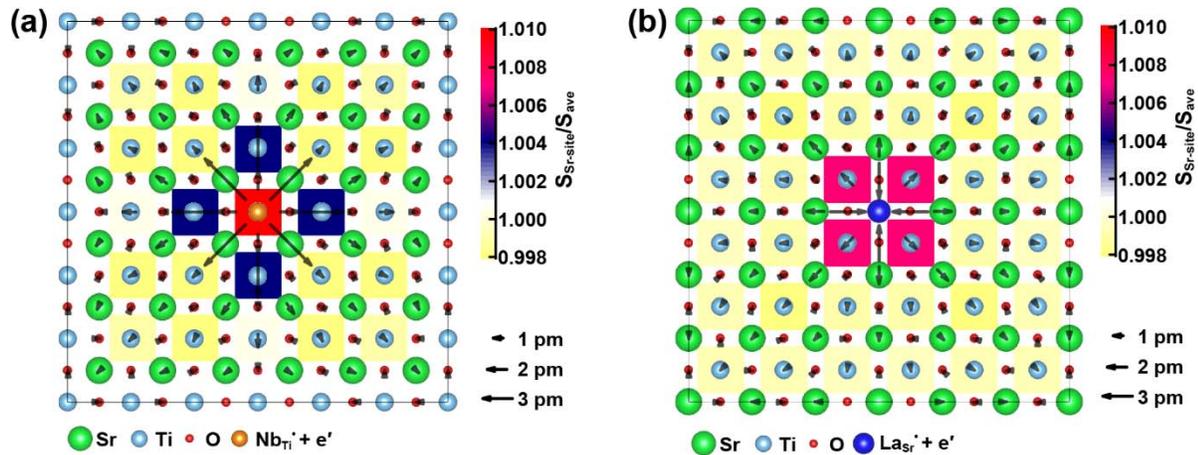
558

559 FIG. 9. Three-dimensional vector plots of Sr displacements around defects/dopants: (a)  $\text{Nb}_{\text{Ti}} + e'$ ,  
 560 (b)  $\text{La}_{\text{Sr}} + e'$ , (c)  $\text{Nb}_{\text{Ti}} + e' + \text{V}_{\text{Sr}}'' + \text{V}_{\text{O}}''$ , and (d)  $\text{La}_{\text{Sr}} + e' + \text{V}_{\text{Sr}}'' + \text{V}_{\text{O}}''$ . Sr, Ti, and O ions are  
 561 located at the corners, centers, and face centers of the lattice cells, respectively. Vectors indicate  
 562 the magnitudes and directions of Sr displacements from their ideal cubic perovskite lattice  
 563 positions and are color-coded according to the scale bar on the right of each figure.

564

565

566



567

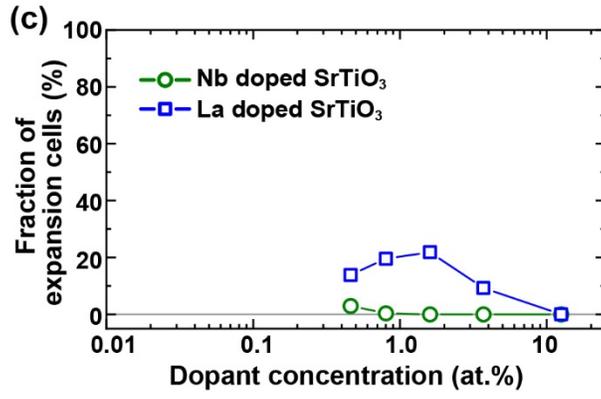
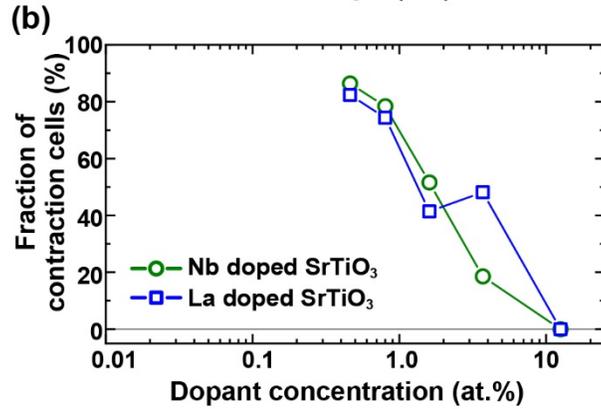
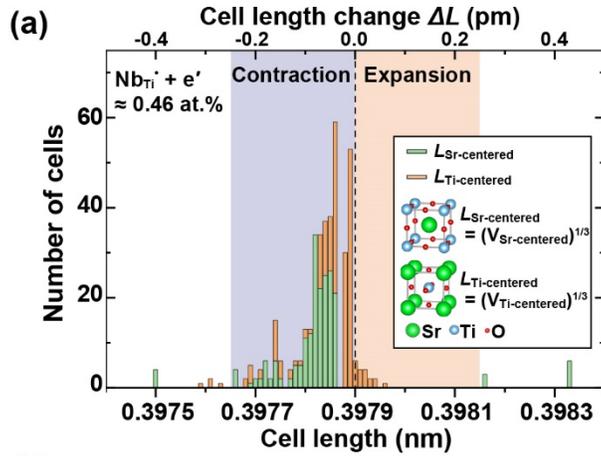
568

569 FIG. 10. Lattice models of (a) Nb- and (b) La-doped SrTiO<sub>3</sub> crystals as viewed along the  $\langle 100 \rangle$   
570 axis in the plane of the dopant atom. Background colors indicate the magnitude of  $S_{\text{Sr-site}}/S_{\text{ave}}$   
571 corresponding to the color scale on the right of each figure. The vector on each atom shows the  
572 direction and magnitude of displacement from its ideal lattice position. Vectors of O overlapping  
573 with those of Ti along the projected direction have been removed for clarity. Vectors at the  
574 bottom of (a) and (b) serve as a scale for the magnitude of displacement. Crystal structures were  
575 drawn using the VESTA program [59].

576

577

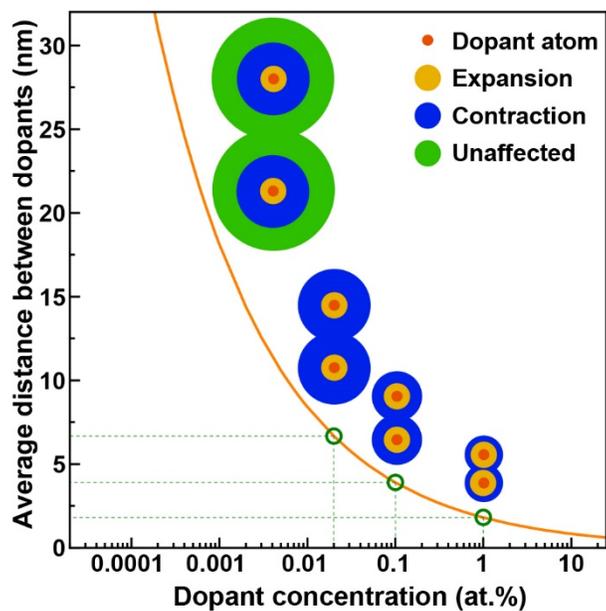
578



579  
 580 FIG. 11. (a) Histogram of cell lengths of Ti-centered and Sr-centered cells within relaxed  
 581 supercells of Nb-doped SrTiO<sub>3</sub>. For clarity, only a narrow range about the average lattice  
 582 parameter (marked with a dotted line) is shown; results  $\Delta L$  (pm) for the entire range are provided as Fig.  
 583 S10a of Supplemental Material. (b), (c): Ratios of numbers of contracted and expanded cells,  
 584 respectively, as a function of dopant concentration. The numbers of contracted and expanded  
 585 cells were taken as those  $\pm 0.25$  pm either side of the lattice parameter. As an aid to the eye,

586 lengths of contracted cells and expanded cells are highlighted with purple and orange  
587 backgrounds, respectively.

588



589

590 FIG. 12. Average distance between dopant atoms as a function of dopant concentration.

591 Schematic models consisting of unaffected, expanded, and contracted regions around dopant  
592 atoms for each concentration are also shown.

593

594 **References**

- 595 [1] O. Tufte and P. Chapman, *Phys. Rev.* **155**, 796 (1967).
- 596 [2] J. Son, P. Moetakef, B. Jalan, O. Bierwagen, N. J. Wright, R. Engel-Herbert, and S. Stemmer, *Nat. Mater.*  
597 **9**, 482 (2010).
- 598 [3] S. Kobayashi, Y. Mizumukai, T. Ohnishi, N. Shibata, Y. Ikuhara, and T. Yamamoto, *ACS Nano* **9**, 10769  
599 (2015).
- 600 [4] Y. Kozuka, M. Kim, C. Bell, B. G. Kim, Y. Hikita, and H. Y. Hwang, *Nature* **462**, 487 (2009).
- 601 [5] Y. Matsubara, K. S. Takahashi, M. S. Bahramy, Y. Kozuka, D. Maryenko, J. Falson, A. Tsukazaki, Y.  
602 Tokura, and M. Kawasaki, *Nat. Commun.* **7**, 11631 (2016).
- 603 [6] C. S. Koonce and M. L. Cohen, *Phys. Rev.* **163**, 380 (1967).
- 604 [7] H. Suzuki, H. Bando, Y. Ootuka, I. H. Inoue, T. Yamamoto, K. Takahashi, and Y. Nishihara, *J. Phys. Soc.*  
605 *Jpn.* **65**, 1529 (1996).
- 606 [8] S. Ohta, T. Nomura, H. Ohta, and K. Koumoto, *J. Appl. Phys.* **97**, 034106 (2005).
- 607 [9] H. Ohta *et al.*, *Nat. Mater.* **6**, 129 (2007).
- 608 [10] O. A. Marina, N. L. Canfield, and J. W. Stevenson, *Solid State Ionics* **149**, 21 (2002).
- 609 [11] H. Yamada and G. R. Miller, *J. Solid State Chem.* **6**, 169 (1973).
- 610 [12] P. Blennow, A. Hagen, K. Hansen, L. Wallenberg, and M. Mogensen, *Solid State Ionics* **179**, 2047 (2008).
- 611 [13] B. Zhang *et al.*, *J. Mater. Chem. C* **3**, 11406 (2015).
- 612 [14] A. Janotti, B. Jalan, S. Stemmer, and C. G. Van de Walle, *Appl. Phys. Lett.* **100**, 262104 (2012).
- 613 [15] S. A. Howard, J. K. Yau, and H. U. Anderson, *J. Appl. Phys.* **65**, 1492 (1989).
- 614 [16] I. Mahmud, M.-S. Yoon, I.-H. Kim, M.-K. Choi, and S.-C. Ur, *J. Korean Phys. Soc.* **68**, 35 (2016).
- 615 [17] R. D. Shannon, *Acta Crystallogr. A* **32**, 751 (1976).
- 616 [18] J. Liu, C. L. Wang, W. B. Su, H. C. Wang, P. Zheng, J. C. Li, J. L. Zhang, and L. M. Mei, *Appl. Phys. Lett.*  
617 **95**, 162110 (2009).
- 618 [19] P.-P. Shang, B.-P. Zhang, Y. Liu, J.-F. Li, and H.-M. Zhu, *J. Electron. Mater.* **40**, 926 (2010).
- 619 [20] Y. Tokuda, S. Kobayashi, T. Ohnishi, T. Mizoguchi, N. Shibata, Y. Ikuhara, and T. Yamamoto, *Appl. Phys.*  
620 *Lett.* **99**, 033110, 033110 (2011).
- 621 [21] S. Kobayashi, Y. Tokuda, T. Ohnishi, T. Mizoguchi, N. Shibata, Y. Sato, Y. Ikuhara, and T. Yamamoto, *J.*  
622 *Mater. Sci.* **46**, 4354 (2011).
- 623 [22] D. Freedman, D. Roundy, and T. Arias, *Phys. Rev. B* **80**, 0614108 (2009).
- 624 [23] T. Ohnishi, K. Shibuya, T. Yamamoto, and M. Lippmaa, *J. Appl. Phys.* **103**, 103703 (2008).
- 625 [24] S. Kobayashi, Y. Ikuhara, and T. Yamamoto, *Appl. Phys. Lett.* **102**, 231911 (2013).
- 626 [25] S. T. Pantelides, *Rev. Mod. Phys.* **50**, 797 (1978).
- 627 [26] W. Wunderlich, H. Ohta, and K. Koumoto, *Phys. B Condens. Matter.* **404**, 2202 (2009).
- 628 [27] J. D. Baniecki, M. Ishii, H. Aso, K. Kurihara, and D. Ricinschi, *J. Appl. Phys.* **113**, 013701 (2013).

- 629 [28] A. Janotti, D. Steiauf, and C. G. Van de Walle, Phys. Rev. B **84**, 201304 (2011).
- 630 [29] Y. J. Chang, G. Khalsa, L. Moreschini, A. L. Walter, A. Bostwick, K. Horn, A. H. MacDonald, and E.  
631 Rotenberg, Phys. Rev. B **87**, 115212 (2013).
- 632 [30] R. Berger, C. Fennie, and J. Neaton, Phys. Rev. Lett. **107**, 146804 (2011).
- 633 [31] B. Himmetoglu, A. Janotti, H. Peelaers, A. Alkauskas, and C. G. Van de Walle, Phys. Rev. B **90**, 241204  
634 (2014).
- 635 [32] H. Frederikse and W. Hosler, Phys. Rev. **161**, 822 (1967).
- 636 [33] K. Omote, J. Phys.: Condens. Matter. **22**, 474004 (2010).
- 637 [34] B. L. Henke, E. M. Gullikson, and J. C. Davis, At. Data Nucl. Data Tables **54**, 181 (1993).
- 638 [35] X-Ray attenuation length [[http://henke.lbl.gov/optical\\_constants/atten2.html](http://henke.lbl.gov/optical_constants/atten2.html)].
- 639 [36] Z. Salman *et al.*, Phys. Rev. B **83**, 224112 (2011).
- 640 [37] M. Schmidbauer, A. Kwasniewski, and J. Schwarzkopf, Acta Crystallogr. B **68**, 8 (2012).
- 641 [38] D. Cuong, B. Lee, K. Choi, H.-S. Ahn, S. Han, and J. Lee, Phys. Rev. Lett. **98**, 115503 (2007).
- 642 [39] N. Orita, Jpn. J. Appl. Phys. **49**, 055801 (2010).
- 643 [40] S. Okamoto, A. J. Millis, and N. A. Spaldin, Phys. Rev. Lett. **97**, 056802 (2006).
- 644 [41] G. Kresse and J. Furthmüller, Phys. Rev. B **54**, 11169 (1996).
- 645 [42] G. Kresse and D. Joubert, Phys. Rev. B **59**, 1758 (1999).
- 646 [43] K. Page, T. Kolodiazny, T. Proffen, A. K. Cheetham, and R. Seshadri, Phys. Rev. Lett. **101**, 205502  
647 (2008).
- 648 [44] S. R. Andrews and R. A. Cowley, J. Phys. C-Solid State Phys. **18**, 6427 (1985).
- 649 [45] I. K. Robinson, Phys. Rev. B **33**, 3830 (1986).
- 650 [46] K. Nassau and A. E. Miller, J. Cryst. Growth **91**, 373 (1988).
- 651 [47] T. Tanaka, K. Matsunaga, Y. Ikuhara, and T. Yamamoto, Phys. Rev. B **68**, 205213 (2003).
- 652 [48] R. Moos and K. H. Hardtl, J. Am. Ceram. Soc. **80**, 2549 (1997).
- 653 [49] F. A. Kröger and H. J. Vink, Solid State Phys. **3**, 307 (1956).
- 654 [50] Y. Kozuka, Y. Hikita, C. Bell, and H. Y. Hwang, Appl. Phys. Lett. **97**, 012107 (2010).
- 655 [51] A. E. Paladino, J. Am. Ceram. Soc. **48**, 476 (1965).
- 656 [52] H. J. Scheel, J. G. Bednorz, and P. Dill, Ferroelectrics **13**, 507 (1976).
- 657 [53] Y. Jun-ichi, S. Takeshi, and Y. Junji, Jap. J. Appl. Phys. **40**, 6536 (2001).
- 658 [54] C. Rodenbacher, M. Luysberg, A. Schwedt, V. Havel, F. Gunkel, J. Mayer, and R. Waser, Sci. Rep. **6**,  
659 32250 (2016).
- 660 [55] P. I. Nabokin, D. Souptel, and A. M. Balbashov, J. Cryst. Growth **250**, 397 (2003).
- 661 [56] P. Gay, P. B. Hirsch, and A. Kelly, Acta Metall. **1**, 315 (1953).
- 662 [57] J. E. Ayers, J. Cryst. Growth **135**, 71 (1994).
- 663 [58] Y. Kim, A. S. Disa, T. E. Babakol, and J. D. Brock, Appl. Phys. Lett. **96**, 251901 (2010).
- 664 [59] K. Momma and F. Izumi, J. Appl. Crystallogr. **41**, 653 (2008).

