



CHORUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

High-pressure phase transitions of nitinol NiTi to a semiconductor with an unusual topological structure

Guangtao Liu, Hanyu Liu, Xiaolei Feng, and Simon A. T. Redfern

Phys. Rev. B **97**, 140104 — Published 26 April 2018

DOI: [10.1103/PhysRevB.97.140104](https://doi.org/10.1103/PhysRevB.97.140104)

32 I. Introduction

33 Nickel titanium (NiTi), termed nitinol, is widely used in automotive, aerospace, electronics and mechanical
34 engineering as well as in medical devices. It is well-known for its shape memory and superelastic properties. Its
35 shape memory behavior is driven by a reversible martensitic phase transition from a high-temperature austenitic
36 B2 ($Pm-3m$) phase into a low-temperature martensitic phase[1,2] at around 273 K. Experimentally, single crystal
37 x-ray diffraction[3,4] and powder neutron diffraction measurements[5] have determined the low-temperature
38 phase as B19' ($P2_1/m$) with $\gamma \sim 97.8^\circ$. Several experimental investigations have, however, reported the existence
39 of multiple coexisting phases.[6–8] A large number of computational studies have been performed to uncover the
40 underlying mechanism of phase transitions in NiTi.[9–16] *Ab initio* calculations revealed an additional
41 orthorhombic B19 ($Pmma$) phase as well as an ‘R-phase’ ($P3$), which are both low-symmetry distortions of the
42 high-symmetry B2 structure.[17] Density functional theory (DFT) calculations, in particular, have illuminated and
43 explained the controversies about the energy surface and properties of NiTi, and have suggested results which are
44 not always in agreement with the experimental observations. For example, the phonon frequencies of the
45 experimentally-observed ambient B2 phase were calculated to be imaginary in conventional *ab*-initial calculation,
46 indicating that this structure is dynamically unstable.[18] Moreover, the experimentally-observed low-temperature
47 B19' structure was predicted to be unstable with respect to a higher-symmetry base-centered orthorhombic (BCO)
48 B33 ($Cmcm$) structure at zero temperature[17], this B33 structure has not yet been confirmed in any experimental
49 study, furthermore this space group precludes this phase from being a shape memory martensite alloy derived
50 from the cubic structure. Many of these differences can be resolved if temperature-dependent entropic effects are
51 taken into account in the calculations. For example, a recent theoretical study has showed that B2 and B19'
52 structures are indeed stabilized, whereas the B33 structure is destabilized at finite temperature.[19]

53 Obviously, temperature factor plays a significant role in the martensitic structural phase transition of NiTi,
54 leading to its shape memory effect, as underlined in both experimental and theoretical studies. But in addition to
55 temperature, pressure also plays a controlling role in inducing such transitions by altering the crystal and/or
56 electronic structures and generally pressure is well-recognized for creating novel structures and properties in
57 materials, including metals and alloys. For example, a pressure-induced metal-semiconductor transition in lithium
58 has been observed by direct electrical resistance measurements.[20] Transparent sodium was predicted by
59 computational work, and then experimentally observed at high pressure, associated with pressure-induced *p-d*
60 hybridization of valence electrons and their repulsion by core electrons into interstitial sites in the structure.[21]
61 Further examples of pressure-induced structural transitions include that from disordered structure to a long-range
62 face centered cubic (fcc) topological ordered structure in $Ce_{75}Al_{25}$, which was observed at 25 GPa[22]. The

63 ground-state phase of the prototype CoCrFeMnNi high-entropy alloy was retained during a high-pressure
64 irreversible phase transition under compression.[23] Nowadays, pressure has been recognized and deployed as a
65 very powerful and useful technique in the design, exploration and synthesis of novel functional materials.
66 Following this path, the high-pressure properties and phase diagram of NiTi has been investigated using
67 self-consistent *ab initio* lattice dynamical calculations which suggested that the B19' and B19 structures are stable
68 high-pressure phases.[24]

69 In this work, we employ crystal structure predictions and first principles total energy calculations based on
70 DFT, to further explore the high-pressure behavior of NiTi at zero temperature. Here, we find that the known B19
71 phase is dynamically unstable and report that the ground-state B33 phase transforms into a *Pbcm* phase at ~4 GPa,
72 and finally to a B32 (*Fd-3m*) phase at ~29 GPa. Actually, we find that the predicted *Pbcm* phase is a distorted
73 structure derived from the dynamically unstable B19 phase. The newly-uncovered B32 phase has a diamond-like
74 stacking sequence, whose topological crystal structure is completely different from any previously-reported
75 structural configuration of NiTi. The more intriguing prediction that the high-pressure B32 phase is a
76 semiconductor with a narrow direct bandgap of ~0.26 eV, is a consequence of the special band structure
77 associated with its unusual structural topology, which appears to be characterized by more localized *3d* orbitals.
78 Our findings indicate that metallic NiTi alloy displays a rich family of abundant structural phase transitions and
79 will transform into an interestingly semiconducting state at high pressure, appears to show a new class of physical
80 and chemical mechanisms of transformation for a transition metal alloys.

81

82 **II. Computational details**

83 The crystal structures of NiTi were probed systematically using the Crystal structure AnaLYsis by Particle
84 Swarm Optimization (CALYPSO) code[25,26], which is based on a search for the global minimum in the free
85 energy surfaces computed by the DFT total energy calculations. This method has been applied successfully to a
86 wide range of crystalline systems ranging from elemental solids to binary and ternary compounds.[27–30] The
87 simulation cell comprised 1 to 8 formula units (2-16 atoms) of NiTi at 0, 10 and 40 GPa. The population size of
88 each generation was 50. The best 20 structures from last generation and 30 new ones generated by CALYPSO
89 algorithm composed the next generation. Generally, the search was terminated after the generation of 2000-2500
90 structures.

91 DFT calculations, including structural optimizations, enthalpies, electronic structures and phonons, were
92 performed using the Vienna Ab initio Simulation Package (VASP)[31] code employing the
93 Perdew-Burke-Ernzerhof[32] exchange-correlation functional. The $3d^84s^2$ and $3d^23s^2$ electrons were treated as

94 valence electrons for Ni and Ti, respectively. Usually, the parameters of the structure searching are a little coarser
95 than those of normal optimizations and enthalpy calculations to save CPU time, where energy cutoff is 400 eV
96 and Monkhorst-Pack grid is $2\pi \times 0.05 \text{ \AA}^{-1}$ in reciprocal space here. We selected a lot of candidates (20-50) with
97 low energy, then re-optimized all of them more precisely. To ensure that all enthalpy calculations were well
98 converged, to about 1 meV/atom, a Monkhorst-Pack grid was selected with sufficient density ($2\pi \times 0.05 \text{ \AA}^{-1}$) in
99 reciprocal space, as well as appropriate energy cutoff (450 eV).[33]

100 The phonon calculations and modulation of soft phonon mode were carried out using a finite displacement
101 approach[34] through the PHONOPY code[35], which uses the Hellmann-Feynman forces calculated from the
102 optimized supercells through VASP code. In our calculations of phonon spectra, we selected $2\pi \times 0.04 \text{ \AA}^{-1}$ and 700
103 eV as the parameters and extended the unit cells to be big enough supercells with volume larger than $\sim 1000 \text{ \AA}^3$.
104 For the B19 structure, the supercell is $3 \times 4 \times 3$, including 144 atoms; for the *Pbcm* structure, the supercell is $3 \times 4 \times 1$,
105 including 96 atoms; and for the B32 structure, the supercell is $2 \times 2 \times 2$, including 128 atoms.

106

107 III. Results and discussion

108 To validate our computational prediction method and DFT calculations in the NiTi system, using the
109 CALYPSO search, we successfully reproduced the known ambient B2 phase, low-temperature B19' phase and the
110 ground-state (0 K) B33 phase at atmospheric pressure, as well as their energy ordering: $H(\text{B2}) > H(\text{B19}') >$
111 $H(\text{B33})$, verifying the reliability of our methodology. Then a new orthorhombic *Pbcm* phase and another cubic
112 B32 phase were found during structure searching at 10 and 40 GPa, respectively. The enthalpy of each phase as a
113 function of pressure is indicated in Figure 1. We find that the enthalpy of B33 phase is just a little lower than that
114 of the B19' phase, which is consistent with previous studies.[19] The B33 phase is stable up to ~ 4 GPa, then
115 transforms to the B19 phase or the *Pbcm* phase. It can be seen that it is very difficult to distinguish them on the
116 basis of their energy, because the energy difference between these two primitive orthorhombic structures is less
117 than the convergence accuracy of the energy calculation. At higher pressure (29 GPa), the B32 phase becomes
118 enthalpically favored.

119 In addition to the enthalpy calculations for these structures, we have also calculated their phonon dispersion
120 curves. The dynamic stabilities of the B33 phase and B19' phase have been confirmed in theoretical study
121 previously.[19] For the B19 phase, we found a soft phonon branch with negative frequencies running from the
122 Γ -point (0, 0, 0) to the Z-point (0, 0, 0.5) to the T-point (-0.5, 0, 0.5) of the Brillouin Zone (BZ) indicating that this
123 structure is dynamically unstable. (Figure 2a) We therefore modulated the structure by introducing atomic

124 displacements corresponding to the wavevector of the Z -point within a supercell of the B19 structure and we
125 obtained a $Pbcm$ structure, which is the same structure as obtained in our crystal structure predictions. Actually,
126 the crystal structure of the predicted $Pbcm$ phase is a distorted supercell of the known B19 phase. Over the entire
127 BZ, we found no imaginary phonon frequencies for this $Pbcm$ phase nor for the B32 phase, confirming the
128 dynamic stability of each of these two structures. (Figure 2b and 2c) We also confirmed their mechanical
129 stabilities by computing the elastic constants.[33] Therefore, by considering both thermal and dynamical
130 stabilities, it appears safe to conclude that, at zero temperature, the $Pbcm$ phase and the B32 phase are calculated
131 to be stable above ~ 4 and ~ 29 GPa, respectively.

132 In order to understand better the nature of the phase transitions between the high-pressure polymorphs of
133 NiTi, we consider the crystal structures and details of these phases, as shown in Figure 3 and Table I. Careful
134 observation of these NiTi crystal structures reveals that all of them, except the B32 phase, share the same
135 structural topology. They can all be derived from the highly symmetric simple cubic B2 structure. In these phases,
136 each Ti atom is eight-fold coordinated by Ni atoms and then regular or irregular hexahedra form their crystal
137 structures. Both the temperature-induced martensitic phase transition ($B2 \rightarrow B19'$) and our newly-discovered
138 pressure-induced phase transition ($B33 \rightarrow Pbcm$) can be seen in terms of the varying relative displacements of Ni
139 and Ti atoms, derived from a (seriously) distorted B2 structure.

140 In contrast, the high-pressure B32 structure shows a completely different topological crystal structure, and
141 can be viewed as two interpenetrating diamond structures separated by a vector of $(0.5, 0.5, 0.5)$. Each Ti atom is
142 tetrahedrally coordinated by four Ti atoms and by four Ni atoms at identical distances, thus forming a
143 superstructure of the body centered cubic (bcc) lattice, which often occurs in certain intermetallic I-III compounds
144 (LiAl, LiGa, NaTi) and I-IIB alloys (LiZn, LiCd).[36] Thus, it can be seen that the transformation to the B32
145 phase involves a significant topological rearrangement and is a reconstructive phase transition, in contrast with the
146 others structural phase transitions in NiTi. Therefore, the energy barrier of reconstructive phase transition ($Pbcm$
147 $\rightarrow B32$) is expected to be significantly higher than that of the lower-pressure transition ($B33 \rightarrow Pbcm$).
148 Additionally, we have calculated the equation of states of all these NiTi structures.[33] We find that the B32 phase
149 is significantly denser than the other phases due to the nature change from (distorted) bcc stacking to close-packed
150 fcc one, further underlining the expectation that this phase becomes most stable at extreme pressure.

151 To further investigate the electronic properties of NiTi, the band structures and the density of states (DOS) of
152 all phases were studied by first principles calculations. It is expected that the B2, B33, and $Pbcm$ phases show
153 metallic behavior, as in a typical alloy system, where the valence bands and conduction bands completely overlap

154 and the DOS at the Fermi level is very high.[33] However, our computational results indicate that the predicted
155 B32 structure shows a bandgap. It is well known that DFT usually underestimates the bandgap of materials, and
156 we thus employed the screened hybrid functional of Heyd, Scuseria, and Ernzerhof (HSE06)[37] to better
157 determine the electronic properties of this structure. The modified band structure (Figure 4) clearly shows that the
158 B32 phase shows semiconducting character with a direct bandgap of ~ 0.26 eV between the Γ and X points, which
159 is consistent with its DOS. From the projected DOS, we find that most of the electronic states below the Fermi
160 level are associated with Ni $3d$ electrons, but also partly by Ti $3d$ electrons and their s , p orbitals. In contrast, the
161 conduction band is composed of mainly by Ti $3d$ electrons.

162 Here, the novel electronic properties of B32 phase may directly contribute to nature of its crystal structure, a
163 Zintl phase. Usually, such structures are poor conductors or semiconductors. Normally, such phases are the
164 products of a reaction between group I/II elements and any post-transition metal or metalloid. The classic Zintl
165 phase, NaTl, displays complete electron transfer from the more electropositive Na to the more electronegative Tl,
166 just as in genuine ionic salts such as NaCl.[38] It behaves like a group IV element, and it forms a diamond
167 network structure. Thus, the band structure and density of states have characteristics related to those of the
168 tetrahedrally bonded IV-IV semiconductors. For example, a study of LiAl indicates that while the Li-Al bond is an
169 ionically polarized covalent bond, the Al-Al bonds are metallic-like and the Li-Li interactions are essentially
170 nonbonding. It was found that this system shows a semimetallic band structure.[39] Our Bader population
171 analysis[40] reveals an electron transfer of $\sim 0.88 e$ from Ti to Ni in the B32 phase. In this B32 phase Ti and Ni
172 show, to first approximation, +1 and -1 valence states, respectively, akin to LiAl, and the B32 structure of NiTi
173 shows an obvious bandgap-

174

175 **IV. Conclusions**

176 In summary, we performed systematically theoretical studies on the structural variability and associated
177 physical properties of NiTi alloy at high pressures. Through structure prediction method and first principles
178 calculations, at zero temperature, we identify that its ground-state B33 phase transforms into a predicted
179 higher-pressure *Pbcm* phase, and then to a highly symmetric B32 phase under increased compression. Further
180 DFT studies reveal that the newly-uncovered B32 phase of this NiTi alloy shows an interestingly semiconducting
181 behavior, which is associated with its special topological phase transition and the characteristic of $3d$ electrons.
182 The present study provides new insights into the exploration and design of distinctive alloy materials under
183 extreme condition.

184

185 **ACKNOWLEDGEMENT**

186 The authors acknowledge funding supports from the National Natural Science Foundation of China under
187 Grants No. 11604314. Work at Carnegie was supported by Energy Frontier Research in Extreme Environments
188 Center (EFree), an Energy Frontier Research Center funded by the Department of Energy (DOE), Office of
189 Science, Basic Energy Sciences under Award No. DE-SC-0001057. S.A.T.R. acknowledges the support of the UK
190 Natural Environment Research Council under grant NE/P012167/1.

191

192 **Figures and captions:**

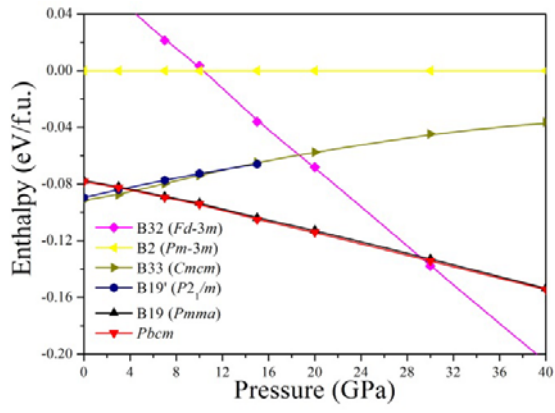
193

194 Table I. The calculated crystal structure information of NiTi at different pressures.

195

Space group	Pressure (GPa)	Lattice parameters (Å)	Atoms	<i>x</i>	<i>y</i>	<i>z</i>
<i>Pm-3m</i> (B2)	0	<i>a</i> = 3.003	Ti	0.5	0.5	0.5
			Ni	0	0	0
<i>P2₁/m</i> (B19')	0	<i>a</i> = 4.687	Ti	0.2189	0.25	0.0127
		<i>b</i> = 4.038	Ni	0.6746	0.25	0.0491
		<i>c</i> = 2.930				
		<i>β</i> = 98.97°				
<i>Cmcm</i> (B33/BCO)	0	<i>a</i> = 2.888	Ti	0	0.1440	0.75
		<i>b</i> = 9.398	Ni	0	0.5847	0.25
		<i>c</i> = 4.042				
<i>Pmma</i> (B19)	20	<i>a</i> = 4.176	Ti	0.25	0	0.2923
		<i>b</i> = 2.578	Ni	0.25	0.5	0.8172
		<i>c</i> = 4.505				
<i>Pbcm</i>	20	<i>a</i> = 2.572	Ti	0.0528	0.3975	0.25
		<i>b</i> = 9.029	Ni	0.5357	0.1598	0.25
		<i>c</i> = 4.176				
<i>Fd-3m</i> (B32)	40	<i>a</i> = 5.585	Ti	0.5	0.5	0
			Ni	0	0	0.5

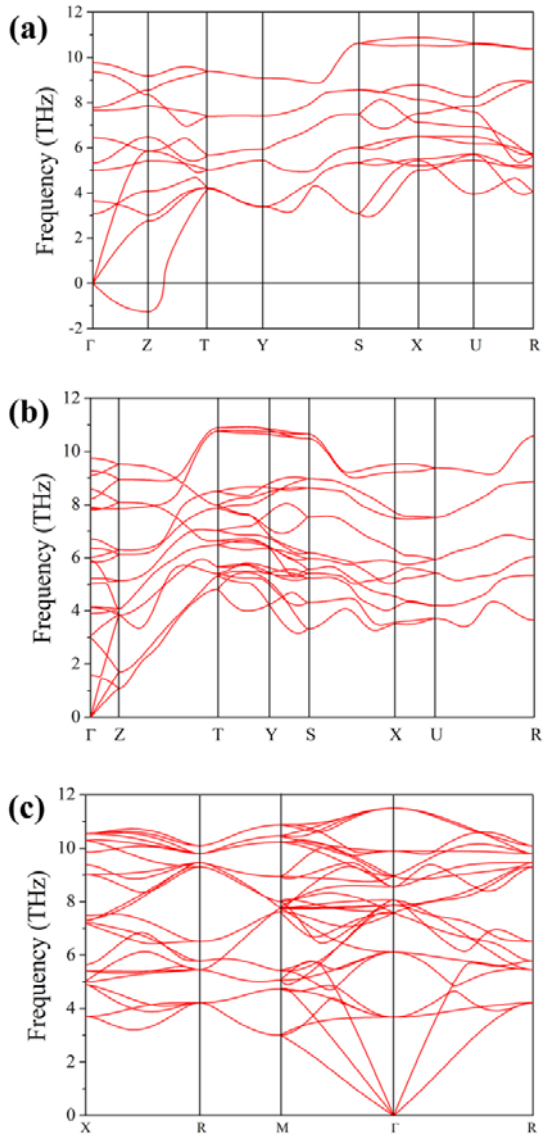
196



197

198 Figure 1: The calculated enthalpy as a function of pressure for each NiTi structure.

199

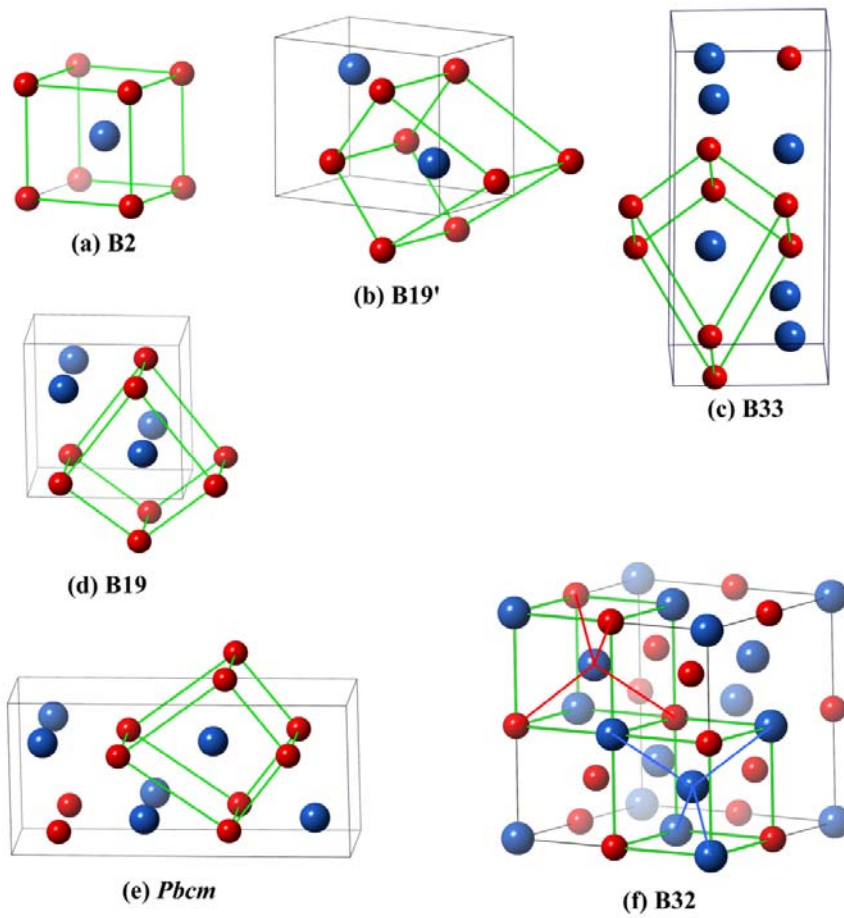


200

201 Figure 2: The phonon dispersions of NiTi in its (a) B19 structure at 20 GPa, (b) *Pbcm* structure at 20 GPa, (c) B32

202 structure at 40 GPa.

203

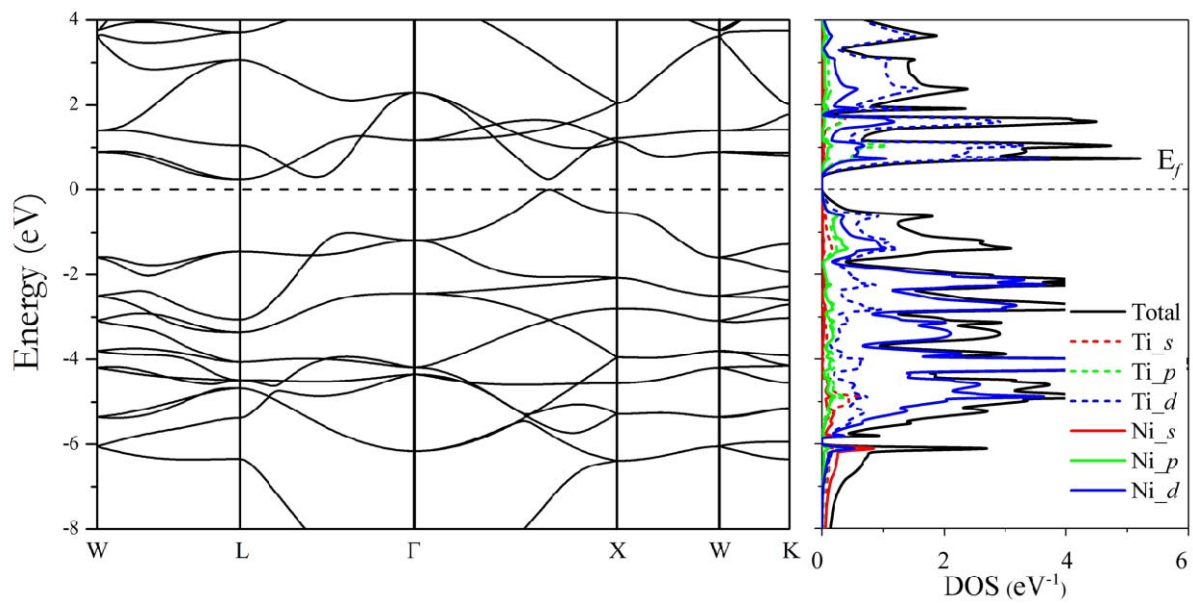


204

205 Figure 3: The schematic crystal structures of the NiTi phases (a) B2, (b) B19', (c) B33, (d) B19, (e) *Pbcm* and (f)

206 B32. Large blue and small red spheres represent Ni and Ti atoms, respectively.

207



208

209 Figure 4: Band structure along high symmetry path and the projected DOS of the B32 structure at 40 GPa. The

210 Fermi level has been set to 0 eV.

211

212 **References:**

213

- 214 [1] Shaw John A and S. Kyriakides, *J. Mech. Phys. Solids* **43**, 1243 (1995).
- 215 [2] O. Benafan, S. A. P. II, R. D. Noebe, T. A. Sisneros, and R. Vaidyanathan, *J. Appl. Phys.* **112**, 93510 (2012).
- 216 [3] Y. Kudoh, M. Tokonami, and K. Otsuka, *Acta Met.* **33**, 2049 (1985).
- 217 [4] G. M. Michal and R. Sinclair, *Acta Crystallogr. Sect. B Struct. Crystallogr. Cryst. Chem.* **18**, 1803 (1981).
- 218 [5] H. Search, C. Journals, A. Contact, M. Iopscience, and I. P. Address, *J. Phys. F Met. Phys.* **13**, L77 (1983).
- 219 [6] P. Taylor, M. J. Marcinkowski, A. S. Sastri, and D. Koskimaki, *Philos. Mag.* **18**, 945 (1968).
- 220 [7] J. M. Carpenter, *Acta Met. Mater* **38**, 1291 (1990).
- 221 [8] F. E. Wang, S. J. Pickart, and H. A. Alperin, *J. Appl. Phys* **43**, 97 (1972).
- 222 [9] J. Wang and H. Sehitoglu, *Appl. Phys. Lett.* **101**, 81907 (2012).
- 223 [10] A. Pasturel, C. Colinet, D. N. Manh, A. T. Paxton, and M. van Schilfgaarde, *Phys. Rev. B* **52**, 15176 (1995).
- 224 [11] X. Huang, C. Bungaro, V. Godlevsky, and K. M. Rabe, *Phys. Rev. B* **65**, 14108 (2001).
- 225 [12] X. Wang, *Phys. Rev. B* **78**, 092103 (2008).
- 226 [13] N. Hatcher, O. Y. Kontsevoi, and A. J. Freeman, *Phys. Rev. B* **80**, 144203 (2009).
- 227 [14] N. Hatcher, O. Y. Kontsevoi, and A. J. Freeman, *Phys. Rev. B* **79**, 20202 (2009).
- 228 [15] D. Holec, M. Friak, A. Dlouhy, and J. Neugebauer, *Phys. Rev. B* **84**, 224119 (2011).
- 229 [16] K. G. Vishnu and A. Strachan, *Phys. Rev. B* **85**, 14114 (2012).
- 230 [17] Huang Xiangyang, G. J. Ackland, and K. M. Rabe, *Nat. Mater.* **2**, 307 (2003).
- 231 [18] K. Parlinski, *Phys. Rev. B* **66**, 64307 (2002).
- 232 [19] J. B. Haskins, A. E. Thompson, and J. W. Lawson, *Phys. Rev. B* **94**, 214110 (2016).
- 233 [20] T. Matsuoka and K. Shimizu, *Nature* **457**, 186 (2009).
- 234 [21] Y. Ma, M. Eremets, A. R. Oganov, Y. Xie, I. Trojan, S. Medvedev, A. O. Lyakhov, M. Valle, and V. Prakapenka, *Nature* **458**, 182 (2009).
- 235 [22] Q. S. Zeng, H. Sheng, Y. Ding, L. Wang, W. Yang, J.-Z. Jiang, W. L. Mao, and H.-K. Mao, *Science*. **332**,
237 1404 (2012).
- 238 [23] F. Zhang, Y. Wu, H. Lou, Z. Zeng, V. B. Prakapenka, E. Greenberg, Y. Ren, J. Yan, J. S. Okasinski, X. Liu, Y.
239 Liu, Q. Zeng, and Z. Lu, *Nat. Commun.* **8**, 1 (2017).
- 240 [24] C. Hu, Z. Zeng, Z. Niu, L. Cai, C. Kong, and Y. Cui, *J. Alloys Compd.* **608**, 258 (2014).
- 241 [25] Y. Wang, J. Lv, L. Zhu, and Y. Ma, *Phys. Rev. B* **82**, 94116 (2010).
- 242 [26] Y. Wang, J. Lv, L. Zhu, and Y. Ma, *Comput. Phys. Commun.* **183**, 2063 (2012).
- 243 [27] J. Lv, Y. Wang, L. Zhu, and Y. Ma, *Phys. Rev. Lett.* **106**, 15503 (2011).
- 244 [28] G. Liu, S. Besedin, A. Irodova, H. Liu, G. Gao, M. Eremets, X. Wang, and Y. Ma, *Phys. Rev. B* **95**, 104110
245 (2017).
- 246 [29] M. Zhang, H. Liu, Q. Li, B. Gao, Y. Wang, H. Li, C. Chen, and Y. Ma, *Phys. Rev. Lett.* **114**, 15502 (2015).
- 247 [30] Y. Li, J. Hao, H. Liu, S. Lu, and J. S. Tse, *Phys. Rev. Lett.* **115**, 105502 (2015).
- 248 [31] G. Kresse and J. Furthmüller, *Phys. Rev. B* **54**, 11169 (1996).
- 249 [32] E. M. PERDEW J, BURKE K, *Phys. Rev. Lett.* **77**, 3865 (1996).
- 250 [33] Supplementary Material
- 251 [34] K. Parlinski, Z. Q. Li, and Y. Kawazoe, *Phys. Rev. Lett.* **78**, 4063 (1997).
- 252 [35] A. Togo, F. Oba, and I. Tanaka, *Phys. Rev. B* **78**, 134106 (2008).
- 253 [36] M. Tadin, J. Schneider, H. Boysen, and F. Frey, *Mater. Sci. Forum* **79–82**, 635 (1991).
- 254 [37] J. Heyd, G. E. Scuseria, and M. Ernzerhof, *J. Chem. Phys.* **118**, 8207 (2003).
- 255 [38] S. C. Sevov, *Intermetallic Compounds – Principles and Practice – Volume 3: Progress* (2002).
- 256 [39] Z. Alex, *Phys. Rev. B* **17**, 2582 (1978).

257 [40] W. Tang, E. Sanville, and G. Henkelman, J. Phys. Condens. Matter **21**, 84204 (2009).
258
259