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Structural and ferroelectric phase evolution in $[KNbO_{3}]_{1-x}[BaNi_{1/2}Nb_{1/2}O_{3-\delta}]_{x}$ (x=0,0.1)

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Structural and ferroelectric phase evolution in $[\text{KNbO}_3]_{1-x}[\text{BaNi}_{1/2}\text{Nb}_{1/2}\text{O}_{3-\delta}]_x$ (x = 0, 0.1)

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The phase transition evolution for $[\text{KNbO}_3]_{1-x}[\text{BaNi}_{1/2}\text{Nb}_{1/2}\text{O}_{3-\delta}]_x$ (x=0, 0.1) is determined via complementary dielectric permittivity and Raman scattering measurements. Raman scattering by optical phonons over the range of 100-1000 cm⁻¹ for 83 K < T < 873 K reveals six discernible zone-center optical phonon modes. Mode behaviors are observed through structural and ferroelectric phases in the solid solution x = 0.1 and compared with those for end member x = 0 and with the results of temperature-dependent dielectric permittivity. Rigorous peak fitting analyses of spectra collected from the solid solution and end member indicate structural and ferroelectric phase transition temperatures that are close to those for the KNbO₃ end member despite the inclusion of 5 atomic % of ferroelectrically inactive Ni cations. Density functional theory calculations were performed in the solid solution and end member using both cation displacement and Berry phasebased methods. Differences in the electronic and polar properties between the solid solution and the end member highlights local and non-local characteristics, which are discussed in relation to the experimental data.

I. INTRODUCTION

¹³ Ferroelectrics have long been recognized as an interesting alternative to conventional semiconductor junctions for ¹⁴ photo-excited carrier separation and photovoltaic solar energy conversion¹⁻¹⁸, exhibiting the bulk photovoltaic (PV) ¹⁵ effect^{19–22} and other ferroelectric polarization-induced or polarization-influenced carrier separation mechanisms, in-¹⁶ cluding those due to domain walls and Schottky barriers. The design and realization of visible wavelength-absorbing ¹⁷ ferroelectric oxide perovskites poses special challenges for practical photovoltaic solar energy conversion due to the inherently large band gap ($\gtrsim 2.7$ eV) associated typically with an O 2p valence and a transition metal d con-18 duction band. Inclusion of other *B*-site cations (*e.g.*, Pd, Pt, etc.), substitutionally, within $PbTiO_3^{23}$ has been proposed as a strategy for band gap lowering²⁴⁻²⁷. Introduction of Ni onto *B*-site KNbO₃ in the solid solution 19 20 $[KNbO_3]_{1-x}[BaNi_{1/2}Nb_{1/2}O_{3-\delta}]_x$ (x = 0.1) (KBNNO) was shown to result in significant decrease in band gap with-21 out loss of ferroelectric ordering¹⁰, and relatively large shift currents in KBNNO (compared with $BiFeO_3$) have been 22 predicted²⁸. Recently, room-temperature Raman scattering spectra in bulk ceramic KBNNO have been reported²⁹ 23 for several different values of x. 24

25 Incorporation of ferroelectrically inactive cations can be accomplished without complete suppression of ferroelec-26 tricity. However, concomitant decreases in ferroelectric phase transition temperature, T_c , with inactive cation concentration can be expected in accordance with well-known systems, e.g. Zr in $PbTiO_3$ (PZT)³⁰, and Sr in $BaTiO_3$ 27 forming the well-known incipient ferroelectric solid solution $Ba_{1-x}Sr_xTiO_3$ (BST)³¹. Raman scattering spectroscopy 28 is effective for probing the onset of a phase transition with local correlation lengths as short as several unit cells. The 29 structural evolution and determination of T_c in KNbO₃ and related A-site (A = Na, Li, La) solid solutions ANbO₃ 30 (KNNO and KLNO) has been reported via temperature-dependent dielectric, Raman scattering, and infrared spec-31 troscopic analyses^{32'40}. Other studies of T_c evolution in doped KNbO₃ have also been reported⁴¹⁻⁴⁵. Unlike systems 32 that involve isovalent substitutions, such as PZT, BST, KNNO, $KTa_{1-x}Nb_xO_3$, and $K_{1-x}Li_xNbO_3$, the aliovalent 33 substitution of Ni^{2+} onto the B-site of KNbO₃ is accompanied by formation of a neighboring O vacancy, producing 34 localized tensile strain near the cation-vacancy pair. Ferroelectric phase stability is a key consideration for potential 35 ³⁶ optoelectronic applications including PV solar energy conversion, making it important to consider the extent to which the substitutional incorporation of Ni cations influences the structural and ferroelectric phase stability in the solid 37 solution KBNNO. 38

As summarized in Tables I and II, end member KNbO₃ possesses fifteen modes, including three acoustic and twelve 39 $_{40}$ optical phonons. Of the optical phonons, by symmetry, $3 A_1 + 4 E$ are first-order Raman-active in the rhombohedral $(R3m \text{ or } C_{3v}^5)$ phase with an A_2 phonon mode as Raman silent, $4A_1 + A_2 + 4B_1 + 3B_2$ in the orthorhombic (Amm^2) 41 $_{42}$ or C_{2v}^{14}) phase, and $3 A_1 + B_1 + 4 E$ in the tetragonal (P4mm or C_{4v}^1) phase. First-order Raman scattering of the $3 F_{1u} + F_{2u}$ modes is forbidden in the centrosymmetric cubic $(Pm3m \text{ or } O_h^1)$ phase. Atomic site disorder lowers 43 ⁴⁴ translational symmetry, effectively relaxing Raman selection rules. This increases phonon scattering rates, manifesting 45 as broadened lineshapes, complicating determination of peak positions and mode assignments^{42–44,46}. Nevertheless, Raman-active mode assignments and transition temperatures in solid solutions such as $(K,A)NbO_3$ have been made 47 and can also be accomplished in B-site solid solution systems, based on comparison with end members. Here, we ⁴⁸ report temperature-dependent dielectric response in bulk ceramic KNbO₃ and KBNNO. Using rigorous lineshape and 49 peak fitting analyses, we show temperature-dependent evolution of the structural and ferroelectric phases in bulk so ceramic KBNNO from 100-1000 cm⁻¹ for 83 K < T < 873 K, comparing our results with bulk ceramic KNbO₃. Finally, we present the results of density functional theoretical analysis, calculating the structural relaxation and 51 $_{22}$ electric polarization and theoretically estimating the T_c in order to compare with our experimental data.

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II. EXPERIMENTAL METHODS

Samples of $[KNbO_3]_{1-x}[BaNi_{1/2}Nb_{1/2}O_{3-\delta}]_x$ (x = 0 and 0.1) are synthesized by standard solid-state reaction 54 ⁵⁵ followed by a sintering process. High purity powders, including K₂CO₃ (J.T. Baker, 99.9%), BaCO₃ (Cerac, 99.9%), Nb₂O₅ (Cerac, 99.95%), and NiO (Cerac, 99%) are dried at appropriate temperatures to remove absorbed moisture. 56 Stoichiometric amounts of starting powders are mixed by ball milling for ≥ 2 hours in ethanol with yttrium-stabilized 57 ZrO_2 grinding media. After evaporation of ethanol, the raw mixtures for samples with x = 0.1 are calcined at 1073 K for 6 hours in a muffle furnace, while for x = 0 samples, 873 K for 6 hours is used. The calcined powders are ball 59 60 milled for 12 hours to minimize the particle size and then uniaxially pressed into pellets of 8 mm diameter and 2-4 ⁶¹ mm thickness. In order to minimize the volatilization of K during sintering, all pellet samples are buried in calcined ⁶² powders with the same composition and encapsulated with platinum foil. Pellets are sintered at temperatures which ⁶³ depend on compositions, with a dwell time of 1 hour and 5 K/min ramp rate. KNO pellets are sintered at 1253 K and x = 0.1 is sintered at 1418 K. The bulk densities of sintered pellets are calculated through the weights and dimensions 65 of the samples; both the x = 0 and 0.1 samples show above 92% relative density.

Temperature-dependent Raman spectra are collected in the backscattering configuration $z(x, x + y)\overline{z}$ using a single 74 monochromator (XploRA, Horiba Jobin-Yvon, Edison NJ), 4 mW, 532-nm laser excitation, focused through a 10x 75 objective to a spot diameter of $\approx 10 \ \mu m$ at an intensity of $1.6 \times 10^3 \ W/cm^2$. Light is dispersed using 2400 gr/mm 76 gratings and collected using a thermoelectrically cooled array detector. The spectral resolution of this confocal micro-77 Raman instrument under these conditions is $\approx 1.2 \text{ cm}^{-1}$. The sample temperature is varied from 83 K to 873 K 78 (Linkham THMS 600, instrumental precision ± 0.1 K) in increments of 2.5 K (for KNbO₃) and 5 K (for KBNNO) at 79 a heating ramp rate of 5 K/min. The sample is also allowed to equilibrate for one minute between consecutive Raman 80 scans. 81

We use density functional theory (DFT) calculations to examine the local structure and polarization caused by 82 substitution of 0.1 $Ba(Ni_{1/2}Nb_{1/2})O_{2.75}$ (BNNO) into KNO. To directly and accurately model the low concentration set of Ba and Ni dopant atoms, we use a $4 \times 4 \times 2$ 159-atom supercell, with x = 0.125 BNNO content. This cell contains 2 Ni atoms and 4 Ba atoms substituting at 2 Nb and 4 K sites. To maintain charge balance, one O atom is removed, 85 in two different configurations. For the first configuration, the O vacancy is placed between two Ni atoms located at 86 nearest-neighbor B-sites along the z-direction, while for the second the vacancy is located between an Nb and an Ni, 87 and the two Ni are separated by 7 Å. For computational efficiency, we performed all calculations at the $KNbO_3$ lattice 88 constants. This choice is supported by the results of the XRD measurements that show that the lattice parameters 89 of KNbO₃ and x = 0.1 KBNNO are essentially the same. The positions of the ions were fully relaxed at the DFT+U 90 level of calculations, using Hubbard U values of 3.97 eV and 9.9 eV for Nb and Ni, respectively, as in previous KBNNO 91 calculations¹⁰. Calculations with other values of U showed only slight dependence of the relaxed structure on the U 92 value, supporting the reliability of our choice of U. 93

The relaxed structures are analyzed for cation displacements which can then be used to evaluate the effect of the Ba(Ni,Nb)O_{2.75} substitution on the local structure and to estimate the polarization by multiplying the cation displacements by their respective Born effective charge (Z^*) values. We also used Berry phase calculations to rigorously revaluate the polarization, with good agreement obtained between the approximate displacement-based and the Berry phase polarization values. The decomposition of P into the contributions of individual cation sites enables analysis of trends in overall P in terms of local structure changes.

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III. RESULTS AND DISCUSSION

The powder x-ray diffraction results show the formation of a stable perovskite for all compositions, while very small NiO impurity peaks are present in samples of x = 0.1 (Figure 1). Dielectric permittivities and losses are obtained as a function of temperature and frequency (Figure 2). Samples of composition x = 0.1 show dielectric anomalies which indicate phase transitions at 683 K and 533 K. The lack of frequency dependence indicates normal ferroelectric behavior, since frequency-dependent dielectric response is characteristic of a relaxor.

Since phonon linewidths are temperature-dependent, vibrational mode assignment is most straightforward in low-106 temperature spectra (Figure 3, rhombohedral phase). Neither the KNO nor KBNNO samples are single-crystalline, 107 and therefore the signal measured is an average of many oblique angles not strictly associated with phonon wave vectors 108 either parallel or perpendicular to the specific crystallographic axes that would allow for the proper assignment of 109 transverse and longitudinal optical (TO, LO) phonon modes⁴⁷. Due to this formalism, the modes will be referred to 110 as $\approx 190 \text{ cm}^{-1}$, $\approx 290 \text{ cm}^{-1}$, $\approx 430 \text{ cm}^{-1}$, $\approx 529 \text{ cm}^{-1}$, $\approx 600 \text{ cm}^{-1}$, and $\approx 836 \text{ cm}^{-1}$; respectively, for both KNO and 111 112 KBNNO polycrystalline samples with the primarily contributing modes denote in Figure 3. For simplicity, a single assignment will follow each mode through softening, stiffening, and any abrupt changes due to structural changes in 113 the sample, with the understanding that there are often many modes with overlapping phonon frequency making up 114 ¹¹⁵ a spectral feature. Comparable peak energies are observed in the KBNNO sample and correspond well to reported ¹¹⁶ mode assignments in $\text{KNbO}_3^{36,37,48-51}$. The TO₁ mode, associated primarily with ferroelectricity and the polarization 117 of the material, corresponds to the central B-site Nb atoms oscillating against the oxygen octahedra and is shown ¹¹⁸ in Figure 4. This mode is particularly indicative of polarization in the material due to the parallel alignment of the ¹¹⁹ polarization vector and the Nb-O bond. The TO₂ mode corresponds to the A-site K atoms vibrating counter to the $_{120}$ greater NbO₆ octahedral structure. The TO₃ mode corresponds to the *B*-site Nb atoms and the apical oxygen atoms

¹²¹ moving parallel along the polarization vector, anti-parallel to the equatorial oxygen atoms. The A-site K atoms do ¹²² not dynamically participate in the TO_3 mode. The TO_4 mode corresponds to stationary A and B-site atoms and ¹²³ apical O atoms, while the equatorial oxygen atoms vibrate out of phase along the polarization direction, with each ¹²⁴ equatorial oxygen atom moving anti-parallel to its nearest equatorial oxygen neighbors.

Temperature-dependent Raman spectra for KNbO₃ (KNO) and KBNNO are shown in Figure 5. Peak fitting analysis is performed to quantify the variation of each peak's intensity, width, and position with temperature (Supplemental Material⁶²). Bose-Einstein correction has been performed before any data analysis of structural phase transitions; in addition, the contour plots and stacked plots are normalized for the convenience of the reader, as the low-temperature data has much greater signal and lower background than the high-temperature data. Transition temperatures are then extracted from the KNO and KBNNO samples based on reported mode assignments, as cited above.

Changes in Raman peak positions and intensities are seen at particular temperatures corresponding to first-order 131 structural phase transitions: rhombohedral (R) \rightarrow orthorhombic (O) \rightarrow tetragonal (T) \rightarrow cubic (C) (Figure 5). 132 While determination of phase transition temperatures can be done by inspection for some modes, detailed analysis 133 is required for others (Supplemental Material⁶²). For example, the transition temperatures are clear in KNO modes 134 135 of $\approx 190 \text{ cm}^{-1}$, $\approx 290 \text{ cm}^{-1}$, $\approx 529 \text{ cm}^{-1}$, and $\approx 600 \text{ cm}^{-1}$; Figure 5). By contrast, the peaks in the KBNNO solid solution sample are much broader, and specific modes (e.g. for $\approx 190 \text{ cm}^{-1}$ and $\approx 836 \text{ cm}^{-1}$) which disappear in the end member upon reaching the cubic phase do not disappear at temperatures well above the expected transition temperature for the cubic phase in the solid solution. Although the temperature dependence of the relative intensities, 138 peak widths, and position are not as distinct as in KNO, the end member's behaviors can be used as a reference for 139 mode assignment and to extract the phase transition temperatures for KBNNO. 140

The R \rightarrow O transition temperature for KNO was established using the relative intensity, mode linewidth, and frequency for the $\approx 190 \text{ cm}^{-1}$, $\approx 290 \text{ cm}^{-1}$, $\approx 430 \text{ cm}^{-1}$, $\approx 529 \text{ cm}^{-1}$, and $\approx 600 \text{ cm}^{-1}$ modes (Supplemental Material⁶²), is yielding a R \rightarrow O transition temperature $T_{R\rightarrow O, \text{ KNO}}$ of 272.0 \pm 0.1 K (Table III). For the KBNNO sample the ≈ 190 is cm⁻¹, $\approx 290 \text{ cm}^{-1}$, and $\approx 430 \text{ cm}^{-1}$ modes also exhibited abrupt changes, yielding $T_{R\rightarrow O, \text{ KBNNO}} \approx 275\pm 8 \text{ K}$. A is particularly clear example is seen in Figure 6a where the degenerate $\approx 290 \text{ cm}^{-1}$ modes' frequency undergoes a sharp discontinuity which can be clearly observed in the KNO sample. This combination of modes contributing to the imperature at $\approx 290 \text{ cm}^{-1}$ softens with higher temperatures. The frequency behavior of the $\approx 290 \text{ cm}^{-1}$ modes in KBNNO is a transition and softens, although the mode is much more difficult to resolve due to the various neighboring overlapping modes at those temperatures. While the initial frequency discontinuity in the KBNNO sample is more is subtle, the softening with temperature is much more dramatic for the solid solution when compared to the end is member.

The O \rightarrow T transition temperature in KNO (512±1 K) is determined from the behavior of the \approx 190 cm⁻¹, \approx 290 cm⁻¹, \approx 430 cm⁻¹, \approx 529 cm⁻¹, and \approx 600 cm⁻¹ modes. The KBNNO transition is ascertained from the \approx 430 cm⁻¹, \approx 529 cm⁻¹ and \approx 600 cm⁻¹ modes with $T_{O \rightarrow T, \text{ KBNNO}}$ of 523±6 K. Determination of this transition temperature in the temperature in \approx KBNNO based on other modes, such as the combination of vibrational modes at \approx 290 cm⁻¹, could not be accomplished without unacceptably large error. Weighted error analysis is performed using standard expressions for the probability distribution and weighting factors (Supplemental Material⁶²). This effectively removes the contributions of the \approx 290 cm⁻¹ modes by weighting them by the inverse of their variance such that they do not significantly affect the KBNNO samples and the are significant for determining the O \rightarrow T transition (Figure 6b).

¹⁶¹ Changes in the mode frequency for $\approx 430 \text{ cm}^{-1}$ phonons signal the onset of the orthorhombic phase at ≈ 273 K. For ¹⁶² increasing temperature, the modes at $\approx 430 \text{ cm}^{-1}$ stiffens until the O \rightarrow T transition occurs, beyond which the modes ¹⁶³ soften slightly until the frequency stabilizes. A decrease in intensity to the background noise level prevents analysis ¹⁶⁴ at higher temperature. This stiffening and softening behavior around the O \rightarrow T transition in KNO is used to discern ¹⁶⁵ the KBNNO transition temperature by identifying the same stiffening and softening in the much broader behavior of ¹⁶⁶ the $\approx 430 \text{ cm}^{-1}$ modes. Without the context provided by the response of KNO, assigning the maximum frequency ¹⁶⁷ of these modes is, at best, challenging considering the various features that affect the relative intensities, linewidths, ¹⁶⁸ and frequencies of the Raman modes with temperature, including through structural phase transitions^{52–55}.

The highest-temperature tetragonal to cubic transition, corresponding to the ferroelectric to paraelectric transition, 170 in the KNO sample occurs at 729.2±0.2 K, as is most easily seen in the $\approx 190 \text{ cm}^{-1}$, $\approx 529 \text{ cm}^{-1}$, $\approx 600 \text{ cm}^{-1}$, and 171 $\approx 836 \text{ cm}^{-1}$ modes. At this temperature, the $\approx 290 \text{ cm}^{-1}$ modes have broadened to the point where the TO₁ and 172 TO₄+LO₄ contributions are indistinguishable. Additionally, the modes at $\approx 290 \text{ cm}^{-1}$ and $\approx 430 \text{ cm}^{-1}$ that give 173 signal above the background at lower temperatures have decreased or vanished as expected from previous studies of 174 the mode evolution in KNO^{36,37,48–51}. The $\approx 190 \text{ cm}^{-1}$, $\approx 529 \text{ cm}^{-1}$, and $\approx 600 \text{ cm}^{-1}$ modes in the Raman spectra 175 for the KBNNO sample yield $T_{T\rightarrow C, \text{ KBNNO}} \approx 736\pm7 \text{ K}$. Through this transition temperature, the $\approx 600 \text{ cm}^{-1}$ mode 176 condenses into the $\approx 529 \text{ cm}^{-1}$ mode to become a single mode at high temperatures³⁸ for both KNO and KBNNO 177 (Figure 5). The temperature dependence of the Raman mode frequency for the $\approx 529 \text{ cm}^{-1}$ mode can be traced until 178 it becomes indistinguishable with the $\approx 600 \text{ cm}^{-1}$ mode (Figure 7a). As expected with $\approx 529 \text{ cm}^{-1}$, there is a general DFT calculations are performed for cation configurations where the vacancy is located between two Ni atoms and also where the vacancy is located between Ni and Nb atoms, with another Ni surrounded by 6 O atoms. We find that the configuration with the vacancy between the two Ni is lower in energy by 1.48 eV. Such a large energy difference suggests that KBNNO will mostly contain Ni-O_{vac}-Ni configurations.

In a $4 \times 4 \times 2$ supercell, x and y are the long directions and z is the short direction. We place the two Ni along the x 189 ¹⁹⁰ direction and relax all ions starting with the cations displaced along $\langle 111 \rangle$. Berry phase polarization calculations for the relaxed structure indicate that the values of the polarization components in the x, y and z directions are 0.142191 C/m^2 , 0.192 and 0.208 C/m^2 , respectively. The polarization values in the y and z directions are slightly smaller than 192 the 0.233 C/m² obtained for the x, y and z P components of the KNbO₃ parent material in the 0 K rhombohedral 193 phase. The polarization component along the x direction is more significantly reduced to 0.142 C/m^2 . Examination 194 of the local displacements shows that the introduction of BNNO leads to a distribution of the local Nb displacements, with Nb off-centering magnitudes in a given Cartesian direction varying from 0 to 0.135 Å with the average Nb 196 ¹⁹⁷ displacement magnitude of 0.16 Å. These average displacement values are smaller than the 0.103 Å displacements along the x, y and z direction found for KNbO₃ by our DFT calculations.

Using a previously established correlation between the experimentally observed T_c and the square of the 0 K DFT polarization⁵⁶⁻⁶⁰, we estimate a decreased T_c for KBNNO of either ≈ 433 K based on the average total P, or ≈ 573 K based on the average magnitude of the Nb displacements related to the local polarization. While also predicting cubic (T-C) transition temperature ferroelectricity in KBNNO, these values underestimate both the dielectric tetragonal to Raman results. This discrepancy could be ascribed to the somewhat qualitative nature of the correlation between T_c and P^2 for materials with different local structures. The presence of Ni + O_{vac} introduces structural motifs into KBNNO that are entirely absent in KNO, diminishing the predictive accuracy of the $T_c - P^2$ correlation.

The DFT finding of a distribution of Nb displacement magnitudes and variation in the local structure is consistent 207 with broadening in both the dielectric and Raman data, including the Raman linewidth. In particular, the temperature 208 evolution of the Raman shift used to assign the T-C transition temperature changes from an abrupt drop at T_c observed 209 for KNO to a gradual decrease to a constant value observed for KBNNO. It is likely that the distribution of local 210 environments found in KBNNO leads to a broader distribution of Raman shifts, with the local environments displaying 211 ²¹² enhanced (relative to KNO) Nb displacements accounting for the higher temperature plateauing of the Raman shift that signifies the presence of the T-C transition. Such dependence of the Raman shift on the distribution of local 213 structures (*i.e.* on the strong displacement contributions) is perhaps the explanation for the disagreement between 214 $_{215}$ the enhanced T_c found by Raman results and the decreased T_c found by the dielectric measurements. The dielectric $_{216}$ results are given by the average structure of the material, where a decrease in the displacement and P is observed for $_{217}$ KBNNO, corresponding to a less stable P and a lower FE-PE transition temperature. On the other hand, the Raman ²¹⁸ measurements are sensitive to the variation in the local structure, where some local environments exhibit enhanced ²¹⁹ 0 K displacements that are likely to persist to higher temperature than the FE structure in the parent KNO material.

IV. CONCLUSIONS

Dielectric permittivity measurements and Raman spectroscopy are used to probe temperature-dependent structural 221 and ferroelectric phase evolution. Changes in the temperature-dependent Raman shifts are diffuse in KBNNO as com-222 pared with abrupt transitions observed in the end member KNO. Taken together with variations in DFT-calculated 223 local atom displacements and local and averaged polarizations (the extent of applicability of $T_c - P^2$ scaling notwith-224 standing), and the absence of frequency dispersion in the dielectric permittivity, these results indicate persistence of 225 ferroelectric ordering in KBNNO to temperatures that approach the T_c of the end member KNbO₃. This identifica-226 tion of structural and ferroelectric phases, combined with the ability to tune these solid-solution ceramic materials 227 ²²⁸ compositionally, highlights these oxides' promising application as semiconducting ferroelectric absorbers and carrier ²²⁹ separating layers in novel photovoltaic devices.

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TABLE I. Twelve optical phonon modes of KNO, as predicted by Postnikov *et al.*⁶¹ for the room temperature orthorhombic phase. Included are experimental values from this study.

Mode	Symm.	Calculated Frequency (cm^{-1})	Experimental Values (cm^{-1})				
	B_2	soft					
TO_1	B_1	232	250				
	A_1	257	296				
	B_2	185	192				
TO_2	B_1	146	a				
	A_1	186	275				
	B_2	467	532				
TO_3	B_1	528	Ь				
	A_1	593	596				
TO_4	B_1	297	с				
	A_1	307	с				
	A_2	224	с				
^{<i>a</i>} indistinguishable with $B_2(TO_2)$							

 b indistinguishable with $B_{2}(TO_{3})$

 c indistinguishable with $A_{1}(TO_{1})$

 TABLE II. Summary of optical modes associated with the four structural phases in KNO and KBNNO.

 Phase
 Space Group Point Group

 Optical Modes

Phase	Space Group	Point Group	Optical Modes		
Rhombohedral	R3m	C_{3v}^{5}	$3 A_1(\mathbf{T}_z, \alpha_{xx} + \alpha_{yy}, \alpha_{zz}) + A_2^* + 4 E (\mathbf{T}_x, \mathbf{T}_y, \alpha_{xx} - \alpha_{yy}, \alpha_{xy}, \alpha_{xz}, \alpha_{yz})$		
Orthorhombic	Amm2	C_{4v}^{14}	$4 A_1(T_z, \alpha_{xx}, \alpha_{yy}, \alpha_{zz}) + A_2(\alpha_{xy}) + 4 B_1(T_x, \alpha_{xz}) + 3 B_2(T_y, \alpha_{yz})$		
Tetragonal	P4mm	C^1_{4v}	$3 A_1 (T_z, \alpha_{xx} + \alpha_{yy}, \alpha_{zz}) + B_1(\alpha_{xx} - \alpha_{yy}) + 4 E(T_x, T_y, \alpha_{xz}, \alpha_{yz})$		
Cubic	Pm3m	O_h^1	$3 F_{1u}^*(\mathbf{T}_x, \mathbf{T}_y, \mathbf{T}_z) + F_{2u}^*$		
* mode is not Raman active					

TABLE III. Summary of structural phase transition temperatures as determined from Raman.

	<i>J</i> 1	1	
	Rhombohedral \rightarrow Orthorhombic	${\rm Orthorhombic} \to {\rm Tetragonal}$	$\mathrm{Tetragonal} \to \mathrm{Cubic}$
	$(R \rightarrow O)$	$(O \rightarrow T)$	$(T \rightarrow C)$
KNO	$272.0 \pm 0.1 \ {\rm K}$	512 ± 1 K	$729.2{\pm}0.2~\mathrm{K}$
KBNNO	275±8 K	$523 \pm 6 \text{ K}$	$736\pm7~{ m K}$



FIG. 1. X-ray diffraction spectra of end member KNO and x = 0.1 KBNNO stoichiometry. Arrows indicate the inclusion of NiO. Additional stoichiometries available in Supplemental Material⁶².



FIG. 2. Temperature dependence of dielectric permittivity measured at 1 MHz for (a) KBNNO x = 0.1; heating cycle and cooling cycle are red and black respectively. (b) Temperature-dependent dielectric permittivity and loss of KBNNO x = 0.1 for various *E*-field frequencies.



FIG. 3. Raman spectra of the KNO and KBNNO samples in the rhombohedral phase collected at 83 K with approximate mode assignments from bulk and polycrystalline litterature showing overall agreement between end member KNO and solid-solution $\rm KBNNO^{36,37,48-51}$.



FIG. 4. Ionic displacements associated with the four transverse optical phonons from lattice dynamics calculations 37 .



FIG. 5. Raman scattering spectra of KNO ceramic materials as a color map (a) and at selected temperatures (b). Raman spectra of KBNNO ceramic materials is shown as a color map (c) and at selected temperatures (d). Phase transitions more evident in KNO, but both ceramics show spectral characteristics revealing $R \rightarrow O \rightarrow T \rightarrow C$ phase transitions.



FIG. 6. (a) Temperature dependence of the overlapping phonon modes near 290 cm⁻¹. While a >4 cm⁻¹ increase in peak energy is seen for KNO, signaling the R \rightarrow O transition, a much smaller change is observed for KBNNO, as confirmed by analysis of the \approx 430 cm⁻¹ mode. (b) Temperature dependence of the \approx 430 cm⁻¹ modes shows a maximum stiffening which signals the O \rightarrow T phase transition in KNO; a broader, weaker maximum is also seen at the same temperature in KBNNO. Vertical dashed lines correspond to determined phase transition temperatures in KNO.



FIG. 7. Comparison of the fitted temperature dependences of the $\approx 529 \text{ cm}^{-1}$ and $\approx 600 \text{ cm}^{-1}$ modes (a) energy and (b) linewidth in KNO and KBNNO, enabling extraction of the O \rightarrow T transition temperature; the $\approx 529 \text{ cm}^{-1}$ modes become degenerate with the $\approx 600 \text{ cm}^{-1}$ in the cubic phase and is only shown in (c). (c) Comparison of the $\approx 600 \text{ cm}^{-1}$ mode softening in KNO and KBNNO. The behavior of the KNO and KBNNO modes here is qualitatively similar and, upon phase transition, the $\approx 600 \text{ cm}^{-1}$ becomes degenerate with the $\approx 529 \text{ cm}^{-1}$ modes to collect around 560 cm⁻¹ enabling determination of T \rightarrow C transition temperatures.