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E. Skoropata, T. T. Su, H. Ouyang, J. W. Freeland, and J. van Lierop Phys. Rev. B **96**, 024447 — Published 28 July 2017 DOI: 10.1103/PhysRevB.96.024447

¹ Intermixing enables strong exchange coupling in nanocomposites: Magnetism through ² the interfacial ferrite in γ -Fe₂O₃/NiO

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 γ -Fe₂O₃ particles surface modified with NiO crystallites form a unique nanocomposite that points to how to tune strong interfacial exchange coupling. We find that Ni²⁺ migrated into the octahedral sites of the γ -Fe₂O₃ nanoparticle surface, and this NiFe₂O₄-like layer permits effective magnetic coupling of Ni and Fe sites that strengthened the interface exchange. A large increase in coercivity coinciding with a loss of exchange bias was achieved by this strong interfacial coupling that resulted in Ni²⁺ moment reversal in the NiO with the γ -Fe₂O₃. This work reveals the importance of intermixing in, and possibility to use, such an exchange coupling regime to alter substantially the coercivity and hence control an important property of exchange coupled nanocomposite magnets.

I. INTRODUCTION

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A key property of exchange coupled systems is an interfacial anisotropy that leads to an enhanced coercivity (H_c) 8 ⁹ and a unidirectional anisotropy that results in exchange bias (i.e. a measured field shift of a hysteresis loop, H_{ex}). The ¹⁰ interface magnetism of exchange coupled systems has been a subject of ongoing investigation since the phenomenon was first reported¹ in 1956. Much research has been focused on understanding ferromagnetic(FM)/antiferromagnetic(AF) 11 and ferrimagnetic(FiM)/AF coupled systems to develop a systematic and quantitative description of the interrelation-12 $_{13}$ ship between the microstructure, intrinsic magnetism of the layers, and exchange bias properties²⁻⁵. By comparison, less attention has been paid to H_c enhancement resulting from exchange coupling. Previous studies of thin films ¹⁵ have revealed that a large H_c enhancement may be obtained in a coupling regime wherein the AF reverses with the ¹⁶ FM(FiM)^{2,6,7}; a process which necessitates strong interfacial coupling. The relatively recent technological advancements that have enabled observation and characterization of interfacial intermixed layers⁸⁻¹⁰ now provide an excellent 17 opportunity to revisit this interesting aspect of exchange coupled magnetism to achieve deeper insight to the physical 18 origin of H_c enhancement. Further, the potential to obtain a large H_c in complex magnetic systems is important to 19 device development and in applications such as nanoparticle-based magnetic hyperthermia and permanent magnets¹¹. 20

To address this, we describe the magnetism of γ -Fe₂O₃ nanoparticles surface modified with small NiO particles. The 21 core γ -Fe₂O₃ nanoparticles have disordered surface spins and an H_{ex} due to interactions between the ordered core and 22 disordered surface spin populations^{12,13}. Surface modification with the NiO nanoparticles essentially eliminate H_{ex} and 23 the paramagnetic surface spins of the γ -Fe₂O₃, and substantially increase H_c . Using element-specific spectroscopic 24 techniques, we observe the formation of a Ni-ferrite interfacial layer. This layer reduced the disorder at the γ -25 Fe_2O_3 /NiO interface by increasing the coordination of surface atoms. This results in a larger interfacial exchange 26 constant J (vs. the surface J of γ -Fe₂O₃), and enables strong exchange coupling between γ -Fe₂O₃ and NiO. By 27 comparing the atomic Fe relaxation, magnetometry, and susceptometry of γ -Fe₂O₃ and γ -Fe₂O₃/NiO, we find that ²⁹ the H_c enhancement is not a due to a change in the magnetocrystalline anisotropy, K_1 , of γ -Fe₂O₃, or due to an $_{30}$ increase in the superparamagnetic blocking temperature (T_B) , but due to Ni-ion moment reversal in the NiO. Our $_{31}$ results demonstrate that interfacial intermixing leads to a strong interfacial exchange coupling (J_{ex}) which can be $_{32}$ used to enhance substantially H_c of a nanocomposite system.

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II. EXPERIMENTAL METHODS

The γ -Fe₂O₃/NiO nanoparticles were made using a two part seed-mediated synthesis to form γ -Fe₂O₃ cores onto which NiO was deposited. The γ -Fe₂O₃ nanoparticles were synthesized using a thermal decomposition of a Fecupferronate precursor, as described in Ref.¹⁴. To add the NiO, a precursor solution containing 1.8 mmol of Nicupferronate in octylamine was heated to 373 K in an argon atmosphere after which 4 mL of the precursor was rapidly injected into 7 mL of γ -Fe₂O₃ nanoparticle solution that had been heated to 523 K in an argon atmosphere. The entire mixture was stirred vigorously at 498 K for 30 minutes, and then stopped by cooling to room temperature. Powder samples used for x-ray diffraction (XRD), Mössbauer spectroscopy, and polarized x-ray experiments were obtained by mixing the nanoparticle stock solution with alcohols to remove excess surfactant, and air drying. Magnetometry and susceptometry experiments were done using samples prepared from 20 μ L of nanoparticle stock solution dispersed in 50 mg of paraffin wax to ensure the same particle separation. A transmission electron microscopy (TEM) sample of the nanoparticles was prepared by dropping a mixture of nanoparticle solution diluted in hexanes onto a copper socated carbon grid. TEM images and elemental mapping were collected using a JEOL 2100F.

XRD patterns were collected using a Bruker D8 DaVinci with CuK_{α} radiation. The structures and lattice param-46 47 eters were determined using a Rietveld refinement using FullProf¹⁵. Zero-field cooled (ZFC) and field cooled (FC) dc-susceptibilities were measured from 5 K to 300 K using a 0.1 mT applied field with a Quantum Design MPMS 48 XL-5. The ac-susceptibility was measured from 5 K to 300 K using a 0.25 mT applied field oscillating at 10 Hz to 49 1000 Hz. Transmission Mössbauer spectra were collected using a Janis SHI-850 closed cycle refrigeration system and 50 a WissEl constant acceleration spectrometer with a 10-GBq 57 Co**Rh** source. The drive velocity was calibrated using α -Fe at room temperature. X-ray absorption spectroscopy (XAS) and x-ray magnetic circular dichroism (XMCD) 52 measurements were done at beamline 4-ID-C of the Advanced Photon Source in a liquid helium cryostat with powder 53 samples mounted on carbon tape onto a cold finger. Spectra were collected over the L_3 and L_2 edges of Fe and Ni. ⁵⁵ All spectra were collected in total electron yield mode and the XMCD was normalized to the maximum XAS.



FIG. 1. (a) Transmission electron microscopy (TEM) image of γ -Fe₂O₃/NiO nanoparticles and (b) the elemental map of Fe (red) and Ni (green). Size distribution for (c) γ -Fe₂O₃ and (d) γ -Fe₂O₃/NiO nanoparticles.



FIG. 2. Powder x-ray diffraction pattern of γ -Fe₂O₃ and γ -Fe₂O₃/NiO nanoparticles, with the results of the refinement (black line) and Bragg markers for the NiO (Fm3m) (upper red) structure and γ -Fe₂O₃ (Fd3m) (lower black) structures. The residuals of the refinements are indicated by the solid blue lines.

III. RESULTS AND DISCUSSION

A. Structure and morphology

Transmission electron microscopy images of γ -Fe₂O₃/NiO nanoparticles are shown in Fig. 1a. The size distribution (Fig. 1c-d) obtained from ImageJ¹⁶ analysis of TEM images indicated an average size of 6.61±0.04 nm and distribution width $\ln(\sigma_{D_{TEM}})=0.05\pm0.01$ for γ -Fe₂O₃ seed particles¹⁷. For γ -Fe₂O₃/NiO we observe γ -Fe₂O₃ cores with average size 6.52±0.04 nm and $\ln(\sigma_{D_{TEM}})=0.03\pm0.01$ and additional particles with average size of 2.34±0.03 nm and $\ln(\sigma_{D_{TEM}})=0.07\pm0.01$. Elemental mapping using electron energy loss spectroscopy (EELS) shown in Fig. 1b identified clearly that small NiO crystallites that formed an incomplete shell on the γ -Fe₂O₃ seeds.

Reitveld refinements¹⁵ of the XRD patterns (Fig. 2) of the nanoparticle systems using the Fd $\bar{3}$ m spinel structure of Fe-oxide and the Fm $\bar{3}$ m rock-salt structure for the NiO shell indicated a lattice parameter for the spinel phase of 8.380 ± 0.002 Å, typical for γ -Fe₂O₃ or doped- γ -Fe₂O₃ nanoparticles¹⁸. The rock-salt phase lattice parameter of 4.190±0.002 Å is consistent with NiO¹⁹. By including Scherrer broadening into the refinements, an average crystallite diameter of the γ -Fe₂O₃ seeds and γ -Fe₂O₃/NiO nanoparticles of 6.5±0.5 nm indicated no change in core size, while a crystallite diameter of ~3 nm was observed for the NiO; all in agreement with the TEM.

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B. Magnetometry and Susceptometry

⁷¹ Zero-field-cooled (ZFC) and field-cooled (FC) 10 mT *dc*-susceptibility, $\chi_{dc}(T)$, and 10-1k Hz frequency dependent in-⁷² phase and out-of-phase *ac*-susceptibilities ($\chi'_{ac}(\nu, T)$ and $\chi''_{ac}(\nu, T)$, respectively) were used to measure the dynamical ⁷³ responses of the nanoparticles. This range of timescales and fields identifies the different overall responses that ⁷⁴ reflect the dynamical magnetism of the various spin populations. Shown in Fig. 3a-b, $\chi_{dc}(T)$, for γ -Fe₂O₃ and ⁷⁵ γ -Fe₂O₃/NiO is quite similar; a maximum ZFC response, and onset of ZFC/FC irreversibility indicate $T_B \sim 75 K$. ⁷⁶ $\chi'_{ac}(\nu, T)$ (Fig 3c-d) shows a frequency dependent maximum with warming that is preceded by a maximum in $\chi''_{ac}(\nu, T)$ ⁷⁷ that indicates a maximum of energy dissipation by the nanoparticles' magnetizations occurring just below T_B , and a ⁷⁸ frequency independent decrease of $\chi'_{ac}(\nu, T)$ for $T>T_B$. A comparison of $\chi_{ac}(\nu, T)$ of the same γ -Fe₂O₃ nanoparticles ⁷⁹ with a larger interparticle separation (inset of Fig. 3c), indicates some interparticle interactions, however, for the ⁸⁰ same interparticle separation, there is clearly a much broader range of temperature-dependent response of the γ -⁸¹ Fe₂O₃ cores compared to γ -Fe₂O₃/NiO indicating a change in the dynamics of one or more spin population within



FIG. 3. Zero-field cooled (ZFC) (black \bigcirc) and field-cooled (FC) (red \Box) *dc*-susceptibility of (a) γ -Fe₂O₃ and (b) γ -Fe₂O₃/NiO nanoparticles. Also shown are the in-phase (top) and out-of-phase (bottom) *ac*-susceptibilities of (c) γ -Fe₂O₃ and (d) γ -Fe₂O₃/NiO nanoparticles prepared using the same interparticle spacing. The inset of (c) shows the same measurement for γ -Fe₂O₃ nanoparticles with a larger interparticle spacing.



FIG. 4. (a) Typical hysteresis loops for γ -Fe₂O₃/NiO measured from ±5 T after cooling to 5 K in 5 T. The inset shows the temperature variation of the high-field magnetization. b) Temperature dependence of the coercivity, $H_c(T)$, for γ -Fe₂O₃ (red \Box) and γ -Fe₂O₃/NiO (black \bigcirc). The inset shows $H_c(T^{1/2})$ with the lines indicating a fit as described in the text. Temperature dependence of the saturation magnetization, $M_S(T)$, for (c) γ -Fe₂O₃ and (c) γ -Fe₂O₃/NiO. The solid lines are a fit to a modified Bloch $T^{3/2}$ law as described in the text.

⁸² the nanoparticle. By comparison, the γ -Fe₂O₃ and γ -Fe₂O₃/NiO nanoparticles have nearly identical $\chi''_{ac}(\nu, T)$ and ⁸³ $\chi_{dc}(T)$ that indicates comparable $T_B \sim 75$ K.

Hysteresis loops measured from 5 K to 300 K after cooling in 5 T present different $H_c(T)$ for γ -Fe₂O₃ and γ -84 Fe_2O_3/NiO nanoparticles (Fig. 4a-b). The similar T_Bs is reflected in the H_c onset temperature of $T_{B,H_c} \sim 75$ K. Inter-85 estingly, H_c was nearly doubled with the NiO crystallites (e.g. compare the 5 K values). To first order, $H_c \propto KV/M_s$, where K is the effective anisotropy, V the nanoparticle volume and M_s is the saturation magnetization. Since $H_c(T)$ 87 should be dominated by magnetic relaxation effects, described in the most straightforwards manner by a uniaxial single domain particle²⁰, $H_c(T) = \frac{2K}{M_s} [1 - \sqrt{T/T_B}]$. Fits to this (solid lines in the inset of Fig. 4a) provide an estimate 88 89 of $K=2.5\times10^4$ J/m³ for γ -Fe₂O₃ nanoparticles (consistent with previous measurements with an $M_s=3.65\times10^5$ A/m) ⁹¹ and $K=5.3\times10^4$ J/m³ for the γ -Fe₂O₃/NiO nanoparticles. Interestingly, whereas the γ -Fe₂O₃ nanoparticles have $_{92}$ $H_{ex}=5.0\pm0.5$ mT at 5 K²¹, H_{ex} is nearly eliminated in γ -Fe₂O₃/NiO nanoparticles ($H_{ex}=1.5\pm1$ mT at 5 K). Since the two systems have the same T_B , the changes in H_c and H_{ex} are a result of changes to the surface magnetism of 93 γ -Fe₂O₃, and due to magnetic interactions at the γ -Fe₂O₃/NiO interface. The lack of H_{ex} coinciding with a large 94 $_{95}$ H_c enhancement indicates that the unidirectional anisotropy was enhanced by strong exchange coupling between the γ -Fe₂O₃ and rotatable AF NiO nanoparticles². A lack of $T_{B,SP}$ enhancement, despite FiM/AF interfacial coupling is due to the $T_{B,SP} \leq 75$ K also for the surface NiO crystallites as shown in the supplemental materials²² (SM), and 97 reported by others for NiO nanoparticles of comparable size²³.

⁹⁹ Spin-wave excitations (that can be affected at the nanoscale) and surface disorder alter $M_s(T)$ of a nanoparticle.

100 We quantified $M_s(T)$ by fitting the high-field region of the loops and verifying the result by extrapolating from $M(\mu_0 H)$ at $1/\mu_0 H=0$. In nanoparticles, $M_s(T)$ is typically described by a Bloch $T^{3/2}$ dependence²⁴ that is modified ¹⁰² to include a term^{12,24} $A \exp^{-T/T_f}$ that describes qualitatively the "freezing out" of disordered surface spins that ¹⁰³ contribute at $\sim T < 5T_f$; $M_s(T) = M_0 \left[(1 - A)(1 - BT^{3/2}) + A \exp^{-T/T_f} \right]$ where the Bloch constant, $B \propto 1/J$ describes the average exchange strength. Fits to this function (solid lines in Figs. 4c-d) describe $M_s(T)$ well with $A=0.21\pm0.04$, $T_f = 3.3 \pm 0.4$ K and $B = 3.19 \pm 0.06 \times 10^{-5}$ K^{-3/2} for γ -Fe₂O₃ nanoparticles, and $A = 0.42 \pm 0.05$, $T_f = 3.2 \pm 0.5$ K and 105 $B = 3.31 \pm 0.05 \times 10^{-5} \text{ K}^{-3/2}$ for γ -Fe₂O₃/NiO nanoparticles. The fit results reveal γ -Fe₂O₃/NiO nanoparticles' disor- $_{107}$ dered surface spin population makes up a larger fraction of the low T M_s while T_f is unaffected. However, reconciling the much lower H_{ex} of the γ -Fe₂O₃/NiO nanoparticles with this result suggests strongly that uncompensated Ni²⁺ 108 ¹⁰⁹ spins from the NiO contribute to the low $T M_s(T)$ (e.g. the more pronounced upturn at 5 K). The larger B indicates ¹¹⁰ a weaker overall J amongst spins which contribute to M_s for $T \gg T_f$ (i.e. the "bulk" ordered spins). Stronger exchange ¹¹¹ interactions are expected between Fe spins at the γ -Fe₂O₃/NiO interface compared to those at the γ -Fe₂O₃ surface ¹¹² due to (better) filled coordination. However, a lower exchange strength compared to the ordered interior spins of the γ -Fe₂O₃ core is expected for coupling through Ni²⁺ (providing a weaker superexchange path compared to Fe³⁺-O²⁻- Fe^{3+}) or if some degree of disorder is retained. The larger B for γ -Fe₂O₃/NiO nanoparticles points to the recapture $_{115}$ of γ -Fe₂O₃ surface spins, increasing the "effective magnetic volume" via an interfacial population with $J < J_{core}$ but ¹¹⁶ with significantly larger exchange strength compared to J_{surf} of bare γ -Fe₂O₃.

C. Atomic magnetism

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Clearly, a better microscopic understanding of the Fe and Ni spin composition and magnetism is necessary to 118 ¹¹⁹ identify the origin of the changes to H_c , H_{ex} , and surface magnetism from the strong exchange coupling enabled by ¹²⁰ the NiO crystallites. Mössbauer spectroscopy at 10 K ($\ll T_B$ where superparamagnetism does not alter the hyperfine parameters) provides each unique magnetic and electronic environment (site), described by a sextet characterized 121 by a Lorentzian (FWHM) linewidth Γ , hyperfine field B_{hf} , isomer shift δ , and quadrupole splitting Δ , with the 122 $_{123}$ relative abundance of each site proportional to the respective spectral areas. The majority of the spectrum of γ - $_{124}$ Fe₂O₃/NiO at 10 K is described by components (labeled A and B_I) with hyperfine parameters typical of the B-sites $(B_{hf,B_I}=53.32\pm0.06 \text{ T}, \delta_{B_I}=0.532\pm0.007 \text{ mm/s})$ and T_d A-sites $(B_{hf,A}=50.93\pm0.05 \text{ T}, \delta_A=0.393\pm0.007 \text{ mm/s})^{13}$ 125 with $\Gamma=0.26\pm0.01$ mm/s. Assuming (as usual) that the recoil-free fractions of the A and B-sites are equal at 10 K²⁶. 30% and 44% is the site abundance of the Fe-ions (versus 62% and 38% for stoichiometric γ -Fe₂O₃). An additional 127 $B_{hf,BII}=49.7\pm0.1$ T, $\delta_{BII}=0.70\pm0.03$ mm/s and $\Gamma=0.45\pm0.05$ mm/s was necessary to fully describe the spectrum, indicating a change in the environment of some of the Fe-ions occurred after adding the NiO shells, 130 comprising 22% of the Fe-sites. These hyperfine parameters are consistent with the B-sites of non-stoichiometric ¹³¹ Ni-ferrite, existing at the interface. The larger δ represents a lower Fe-valence, so that the B_{II} -site is from Fe²⁺-¹³² ions. The lower B_{hf} identifies fewer (or weakened) nearest-neighbour J's, in keeping with the $M_s(T)$ analysis. Also, ¹³³ the v=0, $B_{hf}=0$ of paramagnetic surface spins²⁵ of the γ -Fe₂O₃ nanoparticles is not present in the γ -Fe₂O₃/NiO $_{134}$ nanoparticles' spectrum, replaced with an interfacial component (observable most clearly as absorption at \sim -3 mm/s) ¹³⁵ with $B_{hf,int}=22.1\pm0.01$ T, due to a recapture of the (now) interfacial spins. $B_{hf,int}$ is lower than the ~50 T of



FIG. 5. Mössbauer spectra measured at various temperatures for γ -Fe₂O₃/NiO and the subspectrum component due to individual Fe-sites determined from the 10 K spectrum. Theoretical spectra which well describe the measured spectra for the γ -Fe₂O₃ cores²⁵ are indicated with a dashed line for comparison.

¹³⁶ the core Fe-sites, so the interfacial spin population retained some degree of disorder (likely spin fluctuations). The ¹³⁷ interfacial Fe-sites also have $\Delta = 0.40 \pm 0.05$ mm/s due to an asymmetric local electric field that is also observed for ¹³⁸ the surface spins of γ -Fe₂O₃ (but not in the bulk). This asymmetry in crystal fields about the Fe-ions is suggestive ¹³⁹ of a larger magnetocrystalline anisotropy at the interface in the nanocomposite system trumping the ~4% decrease ¹⁴⁰ in overall J described above.

¹⁴¹ Mössbauer spectra measured at 100 K intervals (Fig. 5) aid to identify the nature of the (atomic) spin dynamics ¹⁴² in the nanocomposite. The overall temperature dependent spectral collapse that demarks $B_{hf}(T)$ for γ -Fe₂O₃/NiO ¹⁴³ nanoparticles was comparable to that of the γ -Fe₂O₃ cores²⁵ (i.e. similar overall line asymmetry and broadening, ¹⁴⁴ and B_{hf} reduction with warming). However, the temperature dependence of the spectral lineshape evolution of the ¹⁴⁵ γ -Fe₂O₃/NiO system is quite different – much slower spin dynamics at 100 and 200 K (larger spectral components ¹⁴⁶ having measurable B_{hf}). These results indicate clearly the impact on the magnetism of the Ni-ferrite interfacial layer ¹⁴⁷ from Ni-ions migrating into the surface of the γ -Fe₂O₃ nanoparticles. Bonding between interfacial Fe-ions and Ni-ions ¹⁴⁸ strengthens the J_{surf} of γ -Fe₂O₃ and recaptures the (previously) paramagnetic surface spins.

D. Element-specific magnetism

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The nature of the Fe- and Ni-sites and their magnetic couplings were further determined from x-ray absorption 150 spectra (XAS) and magnetic dichroic spectra (XMCD) measured over their $L_{2,3}$ -edges at 10 K and in ± 5 T fields 151 ($\ll T_B$ and $M=M_s$ at 5 T). XAS and XMCD provide valuable insight to the nature of interfacial layers by virtue of 152 the element- and site-specificity, and have been used extensively to study nanostructured magnets^{8,21,27-31}. XAS and 153 XMCD spectra were simulated with CTM4XAS³² using ligand field multiplet calculations of the $2p^63d^n \rightarrow 2p^53d^{n+1}$ 154 transitions for Fe^{3+} and Fe^{2+} , and Ni^{2+} , respectively, and by specifying the crystal field splitting 10Dq of O_h and T_d -sites; all sites were described using parameters typical of similar systems^{33,34}. Figure 6a-d identifies that the Fe 156 XAS and XMCD spectra were consistent with a spinel Fe-oxide, in agreement with the above Mössbauer results. The 157 XMCD spectrum shows clearly Fe^{2+} and Fe^{3+} O_h -sites whose magnetization aligns parallel to the applied magnetic 158 ¹⁵⁹ field, and T_d Fe³⁺-sites AF superexchange coupled to the O_h -sites. Keeping in mind the preferential surface sensitivity ¹⁶⁰ of total electron yield³⁵ the relative Fe-site abundances of 31% Fe²⁺ O_h , 32% Fe³⁺ T_d , and 37% Fe³⁺ O_h from a best



FIG. 6. XAS and XMCD measured over the Fe $L_{2,3}$ -edges of γ -Fe₂O₃/NiO at 10 K and 5 T compared with and ligand field multiplet (LFM) simulations of Fe²⁺ O_h , Fe³⁺ T_d , and Fe³⁺ O_h sites. Simulations of the (a) XAS and (b) XMCD of Fe-sites, and measurements (black \circ) of the (c) XAS and (d) XMCD compared to a sum of simulated sites with 31 % Fe²⁺ O_h , 32 % Fe³⁺ T_d , and 37 % Fe³⁺ O_h (grey line) with antiparallel O_h and T_d -site magnetizations. (e) XAS and (f) XMCD measured over the Ni $L_{2,3}$ -edges of γ -Fe₂O₃/NiO at 10 K and 5 T compared with and ligand field multiplet (LFM) simulations of Ni²⁺ O_h .



FIG. 7. XMCD of the L_3 and L_2 -edges of (a) Fe and (b) Ni for γ -Fe₂O₃/NiO at 10 K and 5 T. The integrated XMCD intensities are shown in dashed lines and p and q are the integrated XMCD of the L_3 and $(L_3 + L_2)$ -edges, respectively.



FIG. 8. Temperature (a) and field (b) dependent overall magnetism (\bigcirc) obtained from hysteresis loop measurements, and sitespecific magnetism of Fe (\Box) and Ni (\diamond) obtained from the L_3 -edge XMCD of γ -Fe₂O₃/NiO nanoparticles. Field-dependent measurements were done at 10 K, and temperature dependent measurements were done using 1 T. Also provided is the overall magnetism of D = 4 nm NiO nanoparticles including (c) the field-dependent magnetism obtained from hysteresis loop measurement at 10 K and (d) the temperature dependent magnetism in 1 T. Note the difference in scale between (b) and (d).

¹⁶¹ weighted sum of simulated Fe-sites is in good agreement with the Mössbauer spectral fits, since the larger fraction ¹⁶² of Fe²⁺ from XAS and XMCD is a result of the different "surface sensitivity". The Fe XMCD spectra also clearly ¹⁶³ do not match³⁶ a pure NiFe₂O₄ – in keeping with the γ -Fe₂O₃ core/Ni-ferrite interface/NiO nanocomposite. The Ni ¹⁶⁴ XAS and XMCD spectra (Fig. 6e-f) are of O_h Ni²⁺ with a magnetization aligned with the Fe O_h -sites^{34,37,38} from ¹⁶⁵ the formation of the Ni-ferrite intermixed layer^{39,40}. The relatively small Ni²⁺ XMCD signal (compared to NiFe₂O₄) ¹⁶⁶ is a result of an under-representation of the normalized XMCD from the XAS that speaks to the compensated Ni²⁺ ¹⁶⁷ O_h -sites within the AF NiO particles that contribute to the XAS but not the XMCD.

Sum-rules⁴¹⁻⁴³ were used to obtain the orbit-to-spin moment ratios, $m_{\ell}/m_s = 2p/(3p - 6q)$, where p and q are the integrated XMCD intensities shown in Fig. 7. For all Fe sites, $m_{\ell}/m_s = -0.02\pm0.02$, and $m_{\ell}/m_s = 0.13\pm0.02$ for Ni. While there are practical limitations in transition metal-oxide systems⁴⁴ to obtaining a precise m_{ℓ}/m_s ($L_3 - L_2$ -edge mixing, sensitivity to data normalization, etc.), the results are consistent with Fe- and Ni-sites of spinel ferrites^{34,45}.

Field and temperature dependent Fe and Ni L_3 XMCD, shown in Fig. 8, demonstrate a clear coupling of all sites 172 within the intermixed layer, and provide insight to the overall magnetism. $M_{\rm Fe}(T,\mu_0H=1 \text{ T})$ shows a similar modified 173 Bloch-like behaviour as M(T) from magnetometry. $M_{\rm Ni}(T,\mu_0H=1~{\rm T})$ from Ni²⁺ is similar to $M_{\rm Fe}(T)$, with a notable 174 difference from the expected behaviour of NiO nanoparticles which have a nearly linear M(H) behaviour and M(T)175 that varies much more strongly with temperature. This confirms that $M_{\rm Ni}$ is dominated by the sites within the 176 interfacial Ni-ferrite layer. The stronger $M_{\rm Fe}$ and $M_{\rm Ni}$ variation with temperature compared to the overall M from 177 magnetometry is likely due to the previously discussed over-representation of the interfacial layer, and a weakened 178 compared to that of the interior γ -Fe₂O₃; consistent with the larger B of γ -Fe₂O₃/NiO versus γ -Fe₂O₃, and the 179 results from Mössbauer spectroscopy. While we have identified clearly the existence of the interfacial Ni-ferrite, and 180 the exchange pathways which result in a strong magnetic coupling between the FiM γ -Fe₂O₃ and AF NiO, further 181 182 measurements using high-resolution TEM in the vicinity of the interface and in-field Mössbauer spectroscopy could ¹⁸³ shed further light on the atomic-scale structure and magnetism of the interfacial Ni-ferrite.

IV. SUMMARY AND CONCLUSIONS

In summary, we find an increased H_c and decreased H_{ex} of the γ -Fe₂O₃/NiO nanocomposite compared to γ -Fe₂O₃ 185 $_{156}$ nanoparticles that reveals strong coupling between the γ -Fe₂O₃ and NiO. This was enabled by an interfacial Ni-ferrite which provided stronger exchange interactions amongst interfacial Fe spins compared to bare γ -Fe₂O₃, that was 187 reflected directly in the partial recapture of the disordered surface spins into the ordered core. We observe clearly 188 that the Ni²⁺ ions are coupled to the B-sublattice of the γ -Fe₂O₃ core, and display temperature and field dependent 189 magnetism expected for a Ni-ferrite with effective Ni-O-Fe exchange pathways that enable strong J_{ex} between the 190 γ -Fe₂O₃ and NiO particles. For a typical system, the properties resulting from exchange coupling depend on K_{FiM} 191 and K_{AF} and the layer volumes V_{FiM} and V_{AF} which determine the energy barrier to the reversal of the layers' 192 ¹⁹³ magnetization, and the strength of J_{ex} . Usually $K_{FiM} \ll K_{AF} V_{AF}$, so the AF does not reverse with a field; the ¹⁹⁴ non-rotatable pinned AF spins provide the unidirectional anisotropy responsible for H_{ex} . The lack of H_{ex} despite exchange coupling to NiO is due to the low K_{AF} of NiO $(K \sim 4.3 \times 10^5 \text{ J/m}^3)^{46}$ combined with the small NiO particle ¹⁹⁶ size $-K_{AF}V_{AF} \sim 2.9 \times 10^{-21}$ J that is lower than $\sim 3.6 \times 10^{-21}$ J for γ -Fe₂O₃ core. Thus, the AF NiO does not exert ¹⁹⁷ sufficient torque on the core, so $H_{ex}=0$. However, when the interfacial coupling is strong $(J_{ex}\gg K_{AF}V_{AF})$ the AF ¹⁹⁶ spins rotate with the FiM layer which can increase H_c substantially². The exchange coupling between γ -Fe₂O₃ and ¹⁹⁹ NiO through the Ni-ferrite has a strength $J_{ex} \sim 10^{-3}$ J/m², accounting for the surface area of γ -Fe₂O₃ in contact with NiO particles. This regime has been observed in thin films^{6,7} which have shown a sharp maximum in H_c coinciding with H_{ex} onset with increasing AF layer thickness, pointing to an effective route to control H_c using 201 interface exchange coupling. We have shown that exchange interactions between γ -Fe₂O₃ and NiO which propagate 202 through an interfacial Ni-ferrite provide precisely this coupling regime, which enabled large H_c enhancement. We 203 have further demonstrated that the effective Fe-O-Ni exchange pathways in the interfacial Ni-ferrite are responsible 204 for the strong coupling between γ -Fe₂O₃ and NiO, which is essential to achieve H_c enhancement. 205

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ACKNOWLEDGMENTS

²⁰⁷ The authors thank NSERC and CFI of Canada, and NSC of Taiwan. Use of the Advanced Photon Source at ²⁰⁸ Argonne National Laboratories was supported by the US DOE under contract DE-AC02-06CH11357. The authors ²⁰⁹ also thank Dr. Shen-Chuan Lo of the Material and Chemical Research Laboratories, Industrial Technology Research ²¹⁰ Institute, Taiwan for his assistance with the TEM measurements.

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