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2 **Unified band-theoretic description of structural, electronic, and**
3 **magnetic properties of vanadium dioxide phases**

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17 The debate about whether the insulating phases of vanadium dioxide (VO₂) can be described by
18 band theory or must invoke a theory of strong electron correlations remains unresolved even
19 after decades of research. Energy-band calculations using hybrid exchange functionals or
20 including self-energy corrections account for the insulating or metallic nature of different phases,
21 but have not yet successfully accounted for the observed magnetic orderings. Strongly-correlated
22 theories have had limited quantitative success. Here we report that, by using hard
23 pseudopotentials and an optimized hybrid exchange functional, the energy gaps and magnetic
24 orderings of both monoclinic VO₂ phases and the metallic nature of the high-temperature rutile
25 phase are consistent with available experimental data, obviating an explicit role for strong
26 correlations. We also found a potential candidate for the newly-found metallic monoclinic phase.

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29 I. INTRODUCTION

30 Vanadium dioxide (VO₂) exhibits a first-order phase transition from an insulating phase to a
31 metallic phase at 340 K [1], which is accompanied by a structural transition from the monoclinic
32 M1 phase to the tetragonal rutile (R) phase. VO₂ is intensively studied for such applications as
33 temperature-tuned memory materials [2] and smart windows [3], and for optoelectronic
34 devices [4]. It is also widely viewed as a model system for understanding insulator-to-metal
35 transitions in solids [5–8]. The M1 phase of VO₂ has a band gap of 0.6-0.7 eV [9,10] and can be
36 considered nonmagnetic (NM) [11] near room temperature, while the metallic R phase is

37 paramagnetic (PM) [9,12] above the transition temperature. In addition to these two phases, the
38 experimentally derived phase diagram of VO₂ [13,14] includes a second insulating monoclinic
39 phase designated as M2, which can be stabilized in doped or strained VO₂ single crystals [15,16],
40 thin films [17,18], and nanobeams [19]. Recently, stable metallic monoclinic (mM) phases were
41 found near room temperature under high pressure [20] and in thin films [21,22]. These phases
42 may be related to the transient metallic monoclinic state already reported in ultrafast
43 experiments [23,24].

44 The theoretical description of VO₂ phases has been controversial for half a century. The
45 debate has centered on whether the insulating phases can be described by single-quasiparticle
46 band theory or the band gap results from strong correlations in the Mott-Hubbard
47 sense [15,16,25,26]. In 1971, Goodenough suggested that the band gap in VO₂ can originate
48 from the formation of V-V pairs [27], but, in 1975, Zylbersztejn and Mott proposed that the band
49 gap in VO₂ originates largely from strong electron correlations [28]. This thesis subsequently
50 gained support from experimental data that showed behavior similar to the generic,
51 non-material-specific predictions of correlated-electron model Hamiltonians [25,29]. In 1994,
52 density-functional theory (DFT) calculations for the M1 phase, based on the local density
53 approximation (LDA) for the exchange-correlation potential, favored a Peierls-like dimerization
54 of V atoms as the root of insulating behavior [30]. However, these DFT calculations did not yield
55 a true band gap, a failure which strengthened arguments for a Mott-Hubbard description of the
56 band gap [29,31]. In 2005, Biermann *et al.* carried out dynamical mean-field theory (DMFT)
57 calculations, effectively building electron correlations into DFT-LDA calculations that give zero
58 energy gap [32]. They found a nonzero band gap for the M1 phase, but concluded that M1 *is not*
59 *a conventional Mott insulator*; instead, the finite band gap was attributed to a
60 *correlation-assisted Peierls transition*. The role of strong correlations in opening the band gap
61 was further corroborated in more recent calculations by Weber *et al.* [33].

62 In the last decade, single-particle theories have been extensively explored and tested against
63 experimental data. In 2007, Gatti *et al.* [34] calculated VO₂ energy bands using Hedin's GW
64 approximation for the one-electron Green's function [35], which replaces the bare Coulomb
65 potential in the Hartree-Fock (HF) approximation by an energy-dependent screened Coulomb
66 interaction. These calculations produced an energy gap in the M1 phase and a metallic rutile
67 phase. In 2011, Eyert [36] reported energy-band calculations using hybrid exchange-correlation
68 functionals, in which a fraction of the local exchange potential is replaced by HF exchange. He
69 obtained satisfactory energy gaps for the insulating phases, duplicating the success of Gatti *et al.*
70 *et al.* [34], and addressed the issue of magnetic ordering. While this initial success was followed by
71 more comprehensive studies [37–39], no single exchange-correlation functional has been found
72 that reproduces both the observed energy gaps and magnetic orderings of VO₂ phases, so that the
73 applicability of band theory to VO₂ remains in dispute. [Furthermore, fixed-node diffusion
74 quantum Monte Carlo calculations, which do not depend on a choice of functional, also predicted
75 the proper band gaps without reproducing the observed magnetic ordering \[40\].](#)

76 In this Letter, we introduce two novel elements in energy-band calculations for the principal

77 phases of VO₂: (1) significantly harder pseudopotentials for both oxygen and vanadium and (2)
78 an optimized mixing parameter in a hybrid functional for the exchange-correlation potential. The
79 calculated lattice constants, band gaps, and magnetic properties of the R, M1 and M2 phases of
80 VO₂ are consistent with available experimental data. Additionally, the calculated density of
81 states (DOS) for the M1 and R phase are quantitatively consistent with experimental x-ray
82 photoemission (XPS) data. The success of these hybrid DFT calculations demonstrates that band
83 theory can describe VO₂ phases without explicitly invoking strong correlations. Moreover, the
84 calculations predict a new monoclinic phase with a crystal structure intermediate between M1
85 and R, which we call the M0 state. The M0 phase is ferromagnetic and the true ground state of
86 VO₂ at absolute zero. Old data at liquid-helium temperature [41,42] suggest the existence of such
87 a phase at near-zero temperatures, but more comprehensive data are needed to confirm the
88 prediction. M0 may also be a candidate for the recently discovered [20–22] metallic monoclinic
89 (mM) phase of VO₂ at finite temperatures.

90 II. COMPUTATIONAL DETAILS

91 Hybrid DFT calculations for each VO₂ phase were performed using a plane-wave basis and
92 the projector-augmented-wave method [43] as implemented in the Vienna Ab initio Simulation
93 Package (VASP) [44]. Several magnetic configurations were calculated to determine the
94 magnetic ordering for each VO₂ phase. The exchange and correlation were described by a tuned
95 PBE0 hybrid functional [45,46] that contains 7% HF exchange, which yields an energy gap for
96 M1 in agreement with experiment. These calculations provide a more accurate description of the
97 vanadium and oxygen atoms for two reasons. Firstly, thirteen electrons ($3s^23p^63d^44s^1$) were
98 treated as valence electrons for vanadium instead of the typical eleven electrons [36,38]. For the
99 oxygen atoms, six electrons ($2s^22p^4$) were treated as valence electrons as usual. Second, the
100 oxygen pseudopotential in these calculations was harder than typically used (i.e., the core radius
101 is smaller). The AFM-M1 phase is metastable using typical oxygen pseudopotentials but is
102 unstable using a hard potential, which reflects a delicate balance between competing effects, as
103 manifest by a complex phase diagram with multiple competing phase transitions. [The hardness
104 of pseudopotential has an effect on the magnetic order because it affects bond lengths \(and/or
105 bond angles\) and this indirectly affects whether a certain magnetic order can be stabilized or not,
106 which is known as the Goodenough-Kanamori rule \[47–49\].](#)

107 Such materials may also require a description using hard pseudopotentials. As required by
108 the harder oxygen pseudopotential, the plane-wave cutoff energy was set at 700 eV; a cutoff
109 energy of 800 eV caused no appreciable changes. All Brillouin-zone sampling was based on
110 Γ -centered k-point grids. We used $3\times 3\times 3$ grids for the M1 and M0 unit cells that each contain 12
111 atoms, a $4\times 4\times 6$ grid for the R unit cell with 6 atoms, and a $1\times 2\times 2$ grid for the M2 unit cell with
112 24 atoms. The self-consistent electronic calculations were converged to 10^{-4} eV between
113 successive iterations and the structural relaxations were converged so that the total-energy
114 difference between two successive ionic steps is 10^{-3} eV. [The initial magnetic configuration was
115 set by assigning a moment of 0, +1, or -1 Bohr magneton on each vanadium atom, resulting in
116 three possible initial configurations: NM \(all moments set at 0\), FM \(all moments set at +1\), and
117 AFM \(moments alternating between +1 and -1 along V-chains\). During self-consistency](#)

118 calculations of the electronic structure, the magnetic moments on all atoms were allowed to vary.

119 III. RESULTS

120 The optimized crystal structures in Figure 1 have all expected features of the
121 experimentally-derived structures: all V-V chains of M1 and M0 are both canted and dimerized,
122 R has only undimerized straight V-V chains, and the monoclinic M2 phase has both straight
123 dimerized V-V chains and undimerized but canted antiferromagnetic V-V chains [16,22,50–52].
124 In addition to that qualitative agreement, the calculated lattice constants and angles as well as
125 vanadium-vanadium (V-V) bond lengths and V-V angles are in good agreement with
126 corresponding experimental values (see Table I). Although our lattice constants and V-V bond
127 lengths are somewhat smaller than the corresponding experimental values, density functional
128 theory calculations simulate atoms at 0 K, not the finite temperatures available to experiments.

129 First, we consider the magnetic and electronic properties of the R phase. Experiments have
130 shown that the R phase is PM above the transition temperature of 340 K [9,12]. According to the
131 present calculations, the total energies of antiferromagnetic R (AFM-R) and NM-R are higher
132 than ferromagnetic R (FM-R) by 125 and 140 meV per formula unit, respectively. Although the
133 calculations predict FM-R to be the ground state of R, the temperature at which DFT calculation
134 must be performed (0 K) is well below any hypothetical Curie temperature of R-VO₂. However,
135 the crystal structure of VO₂ is monoclinic at temperatures below 340 K so we cannot directly
136 compare the calculated FM ground state to an experimentally-observed state, so we can only
137 state that our FM-R prediction is consistent with the experimental observations of PM-R [9,12].
138 As shown in Table II, FM-R is metallic, in agreement with experiment [9,12], DMFT
139 calculations [32], and a previous hybrid-functional calculation [53], but unlike other hybrid
140 calculations [38,54]. In Figure 2(a), the total DOS of FM-R is compared to the experimental XPS
141 spectra [55] and with DMFT results [32]. The overall shape of the DOS agrees with the
142 experimental data. In particular, a feature at -1.3 eV that is present in the experimental data [55],
143 in previous DMFT results (attributed to a lower Hubbard band) [32], and in GW calculations
144 (attributed to a plasmon) [34] is reproduced in the DOS computed in the present work.

145 We next consider the magnetic and electronic properties of the M1 phase. Conflicting reports
146 of paramagnetic [9,12] and diamagnetic [56] susceptibilities for M1 suggest that M1 probably
147 has a negligible magnetic susceptibility, and that experimental values are potentially affected by
148 fabrication parameters; we therefore designate it as NM as previous authors have done [38]. The
149 optimized AFM-M1 spin configuration relaxes to the more stable NM-M1 in contrast to previous
150 hybrid DFT results [36–38,53] but consistent with experiment [9,11,12]. As can be seen in Table
151 II, we obtain a band gap of 0.63 eV for NM-M1 in good agreement with the experimental
152 value [9,10,55] of 0.6-0.7 eV and the values obtained from DMFT [32,33] and GW [34]
153 calculations. In Figure 2, the total DOS of NM-M1 is compared to the experimental XPS
154 spectra [55] and the GW DOS of Ref. [34]. The shape of the DOS and the positions of peaks
155 from -10 to 0 eV agree well with the experimental results [55] and with the GW DOS. This
156 comparison confirms that the electronic structure of the insulator phase NM-M1 is correctly
157 reproduced by the present hybrid DFT calculations.

158 In addition to the NM-M1 and FM-R states, the present hybrid DFT calculations predict a
159 stable ferromagnetic state, FM-M0, with a structure intermediate between NM-M1 and FM-R.
160 Calculations starting from the FM-M1 configuration converge to FM-M0 during geometry
161 optimization. Since the total energy of FM-M0 is lower than the calculated energy of the
162 commonly accepted ground state, NM-M1, by ~ 50 meV per formula unit, we suggest that VO₂
163 may be ferromagnetic at very low temperatures. A low Curie temperature could account for the
164 discrepancy between the predicted ferromagnetism and the finite magnetic susceptibility
165 observed in experiments at moderately low temperatures [41,42]. Between 10 K and the
166 insulator-to-metal transition at ~ 340 K the magnetic susceptibility is small [42], reinforcing the
167 conventional wisdom that NM-M1 is the stable phase above 10 K.

168 It is noteworthy that initial configurations of AFM-M0 and NM-M0 both converge to
169 NM-M1 when the initial magnetic moments are allowed to change during the calculation. Along
170 with the fact that FM-M1 converges to FM-M0, these calculations hint at the complex interplay
171 of magnetic and structural degrees of freedom, and highlight the necessity of more magnetic
172 measurements at low temperatures to confirm previous experimental results [41,42] and our
173 theoretical predictions. In other words, the input magnetic ordering of (FM or NM) is a stronger
174 determinant of the output crystallographic structure (M0 or M1, respectively) than the input
175 crystallographic structure. It is also interesting that our results show that both ferromagnetic
176 phases of VO₂ (M0 and R) are half metals, as is CrO₂ [57,58], suggesting that half metallicity
177 and ferromagnetism are correlated in transition-metal oxides.

178 Similar to NM-M1, the FM-M0 configuration has a simple monoclinic lattice with space
179 group P21/c (C_{2h}^5 , No. 14) and dimerized zigzag V-V chains. However, the crystal structures of
180 NM-M1 and FM-M0 exhibit subtle differences, as shown in Figures 1(a) and 1(b). The short V-V
181 bond of FM-M0 is longer and the long bond is shorter than the corresponding bonds in NM-M1.
182 Therefore, the FM-M0 crystal structure can be viewed as an intermediate state between the
183 crystal structures of NM-M1 and FM-R. In fact, both the short and long V-V bonds of FM-M0
184 are closer to the bond length found in FM-R than their NM-M1 counterparts, indicating a
185 FM-M0 intermediate state would be structurally closer to FM-R than to NM-M1. Furthermore,
186 the 175° bond angle of FM-M0 is also closer to the 180° angle found in FM-R than the 166°
187 angle of NM-M1. Diffraction measurements and optical or electrical measurements below the
188 Curie temperature are needed to verify the structure and metallic character of the FM-M0 state.

189 Recently, a stable metallic monoclinic VO₂ phase (mM) has been observed near room
190 temperature in thin films [22] and single crystals under high pressure [20]. We found that the
191 crystal structures and metallic character of the predicted FM-M0 and the experimental mM states
192 are very similar, which suggest that FM-M0 may be related to this mM phase. In the thin
193 films [22], X-ray absorption fine-structure spectroscopy (XAFS) demonstrated that the short V-V
194 bond elongates, the long V-V bond shortens, and zigzag V-V chains straighten when VO₂
195 metallizes [22], leading to an intermediate crystal structure with lattice constants and bond
196 lengths nearly identical with those for FM-M0 shown in Table I. Pressure-dependent Raman
197 spectroscopy, mid-infrared reflectivity, and optical conductivity measurements confirmed an

198 insulator-to-metal transition without an accompanying structural transition from monoclinic to
199 the rutile phase [20]. However, although a subtle change in structure was attributed to the
200 appearance of the M2 phase, that assignment explains neither the metallization nor the fact that
201 intermediate Raman spectra are unlike those of either M2 or M1 [20]. Instead, a monoclinic
202 metallic phase, such as M0, with slightly different crystal structure than either M1 or M2, would
203 explain both the mM phase in thin film samples [22] and the metallic monoclinic VO₂ phase that
204 appears under high pressure [20]. The similar crystal structures and metallic character of the
205 predicted FM-M0 and the experimental mM states suggest that FM-M0 may be related to this
206 mM phase.

207 Although most work on VO₂ over the past fifty years has focused exclusively on the
208 transition between the insulating M1 and metallic R phases, multiple authors [14,16,29,36,59]
209 have suggested that the M2 insulating phase may hold the key to a complete understanding of the
210 VO₂ phase transition. Three possible AFM configurations [60] designated as A-AFM, G-AFM,
211 and C-AFM are shown in Figure 3(a), 3(b), and 3(c), respectively. Each configuration represents
212 a unique magnetic ordering of the zigzag chains, while the straight chains have no moments. The
213 A-type and G-type exhibit antiparallel moments along the canted zigzag V-V chains [16]. For
214 A-AFM, moments on V-atoms in a canted zigzag chain are parallel to moments of its nearest
215 V-atom neighbors on the next canted chain, while they are antiparallel for G-AFM and C-AFM.
216 However, the moments of all vanadium atoms on a single chain are aligned in C-AFM.

217 Our calculations show that the A-AFM is the lowest-energy configuration of M2 and the
218 G-AFM, C-AFM, FM, and NM configurations of M2 are higher in energy than A-AFM by 4
219 meV, 27 meV, 16 meV, and 32 meV per formula unit, respectively. Although numerically
220 accurate, the small energy difference (4 meV) between A-AFM and G-AFM may not be captured
221 accurately by the approximate functionals. Nevertheless, both A-type and G-type AFM-M2 agree
222 with the experimentally derived model in which M2 is antiferromagnetic and local magnetic
223 moments are present only on the canted zigzag V-V chains [16]. Similarly, the present
224 calculations show that the local magnetic moments of AFM configurations are on the canted V-V
225 chains while the straight, dimerized chains have negligible moments. The band gap of 0.56 eV
226 calculated for A-AFM-M2 is in agreement with photoelectron spectroscopy (PES) of M2 quoting
227 a band gap greater than 0.1 eV [61]. Furthermore, our value of 0.56 eV is consistent with the
228 band model proposed by Goodenough [62] in which the band gap for M2 is comparable to, but
229 smaller than, the band gap of M1 (0.6-0.7 eV).

230 IV. DISCUSSION

231 The kernel of the long-standing debate about VO₂ is whether the electronic properties of this
232 material are better described by band theory in which electrons are represented by
233 non-interacting quasiparticles that experience the same single-particle crystal potential, or by a
234 many-body approach in which electron-electron interactions are explicitly incorporated. In
235 principle, band theory can always describe any given material: ground-state properties are
236 describable by DFT, which is an exact theory, assuming that a satisfactory exchange-correlation
237 potential $V_{xc}(r)$ can be constructed; excitations can be described by Hedin's GW expansion of the

238 self-energy $\Sigma(r,r';E)$ followed by solving the Bethe-Salpeter equation (BSE) [63] to include
 239 electron-hole interactions. Both the DFT and Hedin equations look like Schrödinger equations:
 240 the $V_{xc}(r)$ in DFT is replaced by the nonlocal, energy-dependent $\Sigma(r,r';E)$ in order to describe
 241 excitations. Using these equations, one gets quasiparticle energy bands, single-particle
 242 excitations, excitons (via the BSE), and plasmons (from the zeros of the real part of the
 243 single-particle dielectric function [64]), but the energy dependence in $\Sigma(r,r';E)$ is often
 244 essential [33]. The standard procedure is to first solve the DFT equation with a reasonable choice
 245 of V_{xc} , and then use the solutions to construct $\Sigma(E_k)$, which are in turn used to correct the DFT
 246 energy bands. Ideally, the process should be carried to self-consistency to eliminate the effect of
 247 the initial V_{xc} choice. Gatti *et al.* [34] have already demonstrated that this process correctly
 248 predicts the band gap of insulating monoclinic VO₂, but the numerical procedures are quite
 249 cumbersome and magnetic calculations require separate, self-consistent GW calculations. Hybrid
 250 exchange-correlation functionals constitute an attempt to construct a $V_{xc}(r)$ that also serves as a
 251 local, energy-independent approximation to $\Sigma(r,r';E)$, known as the COHSEX (Coulomb hole
 252 plus screened exchange) approximation [34]. The fact that $\Sigma(r,r';E)$ is material specific justifies
 253 tuning the mixing parameter in the hybrid functional, as is done in the present paper. In this way,
 254 the tuned exchange-correlation functional models $\Sigma(r,r';E)$ for each material. Similarly, the
 255 Hubbard U, which is present in theories that incorporate explicit electron-electron interactions, is
 256 also often treated as a free parameter. Here we have demonstrated that, by tuning the mixing
 257 parameter of a hybrid functional and using harder-than-usual pseudopotentials, the single-particle
 258 approach correctly yields both the electronic and magnetic properties of VO₂ phases; however,
 259 the underlying nature of the phase transition is not addressed here.

260 DFT and GW calculations serve as rigorous *quantitative* tests of quasiparticle theories. The
 261 early conclusions that VO₂ is a strongly-correlated material were based on model many-body
 262 Hamiltonians. Experimental data in the region of the phase transition were compared with the
 263 corresponding model behavior [26,29]. The appearance of correlated behavior at the phase
 264 transition, however, does not necessarily imply that strong correlations persist at temperatures
 265 away from the phase transition. Quantitative theories based on strong correlations, such as
 266 LDA+U, GGA+U and DMFT, assume at the outset that strong electron-electron interactions,
 267 incorporated via the Hubbard-model on-site parameter U, dominate. In the case of VO₂, LDA+U
 268 yields insulating behavior for both the monoclinic and rutile phases [65,66]. The DMFT
 269 calculations by Biermann *et al.* [32] and by Weber *et al.* [33] are anchored on a zero-gap DFT
 270 calculation and found that strong correlations are needed to reproduce the observed value of a
 271 Peierls-induced energy gap. However, these methods have not yet been used to study the
 272 competing magnetic orderings. Thus, only the present calculations, based on band theory,
 273 reproduce the observed structural, electronic, *and* magnetic properties of all VO₂ phases. The
 274 present band theory, DMFT, and GW/COHSEX all give a band-gap value in accord with
 275 experiment, which raises the following challenge: If DMFT and GW/COHSEX calculations were
 276 to be anchored on the present hybrid-functional band structure, which yields a correct energy gap,
 277 instead of the zero-gap LDA band structure, would they retain this value of the energy gap? If so,
 278 the role of correlations beyond what is captured by the present hybrid functional would be

279 negligible. Clearly, such calculations would be valuable to establish the origin of the agreement
280 between seemingly incompatible theories.

281 V. CONCLUSIONS

282 In conclusion, our study underlines the power of the hybrid DFT approach to produce a
283 comprehensive theoretical picture of all the major VO₂ phases and their magnetic properties. We
284 have successfully reproduced the electronic and magnetic properties of M1, M2, and R phases of
285 VO₂ using DFT calculations with a hybrid functional and accurate pseudopotentials. The success
286 of these hybrid DFT calculations suggests that band theory can provide an adequate description
287 of VO₂ phases *despite the unusually large coupling between magnetic and structural degrees of*
288 *freedom in VO₂. The strength of that coupling is perhaps displayed more clearly in this work*
289 *than ever before given the strong influence that the initial magnetic state has on the optimized*
290 *crystal structure.* Moreover, the present calculations predict a new monoclinic ferromagnetic
291 metal state of VO₂, which accounts for the magnetic data at low temperature and is also a
292 candidate for the recently observed metallic monoclinic mM phase that appears in thin films or
293 under high pressure. In addition, the antiferromagnetic structure of M2 was predicted to be
294 A-type. Experimental verification of ferromagnetism in room-temperature VO₂ under high
295 pressure, as well as structural and electronic measurements at low temperatures in unstrained
296 VO₂, clearly set important priorities for future research to test the validity of these particular
297 findings.

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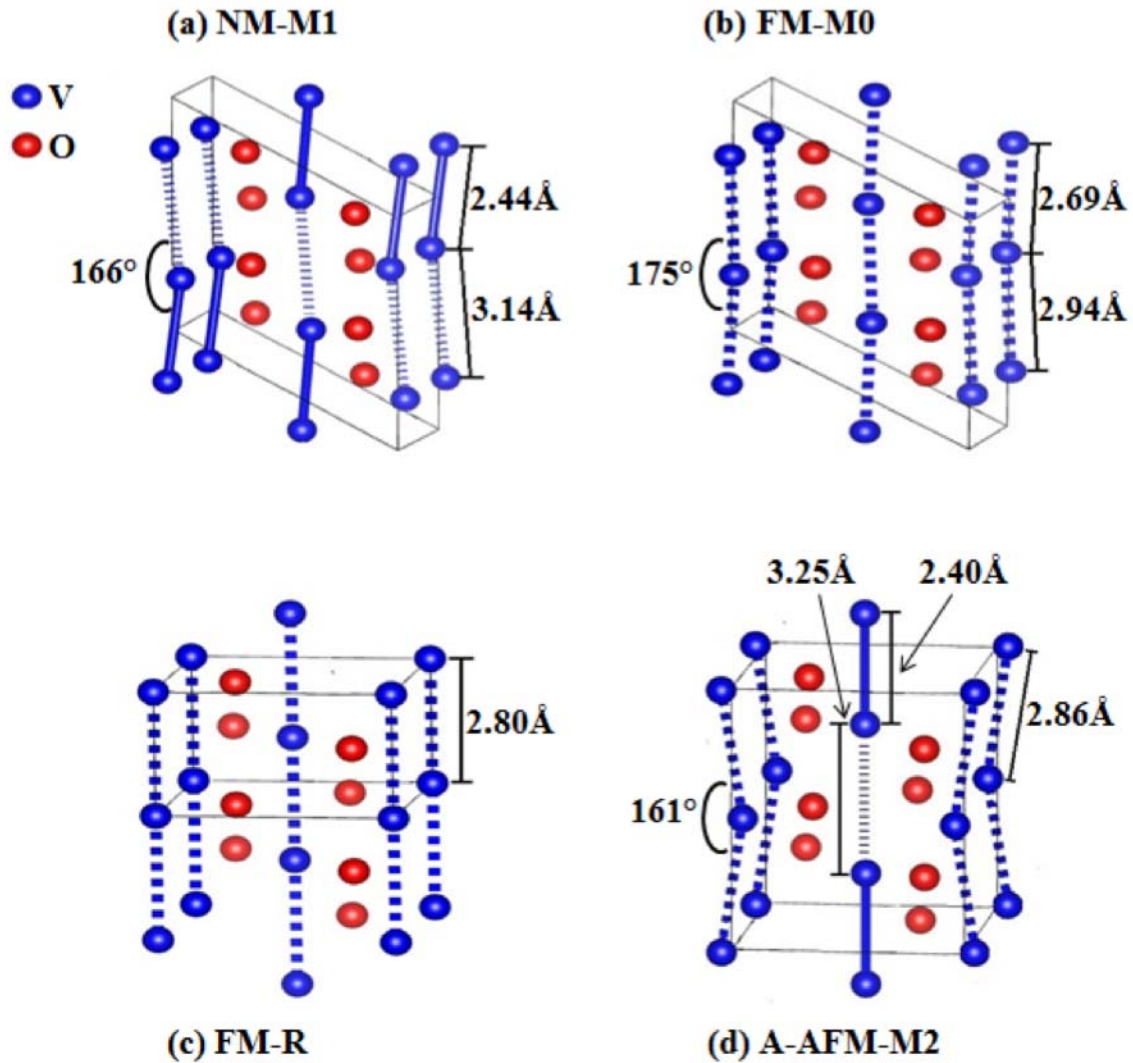
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308 [1] F. J. Morin, Phys. Rev. Lett. **3**, 34 (1959).
309 [2] T. Driscoll, H.-T. Kim, B.-G. Chae, B.-J. Kim, Y.-W. Lee, N. M. Jokerst, S. Palit, D. R.
310 Smith, M. Di Ventra, and D. N. Basov, Science (80-.). **325**, 1518 (2009).
311 [3] E. S. Lee, X. Pang, S. Hoffmann, H. Goudey, and A. Thanachareonkit, Sol. Energy Mater.
312 Sol. Cells **116**, 14 (2013).
313 [4] T. S. Kasirga, D. Sun, J. H. Park, J. M. Coy, Z. Fei, X. Xu, and D. H. Cobden, Nat.
314 Nanotech. **7**, 723 (2012).
315 [5] N. F. Brady, K. Appavoo, M. Seo, J. Nag, R. P. Prasankumar, R. F. Haglund, and D. J.
316 Hilton, J. Phys. Condens. Matter **28**, 125603 (2016).
317 [6] J. D. Budai, J. Hong, M. E. Manley, E. D. Specht, C. W. Li, J. Z. Tischler, D. L. Abernathy,
318 A. H. Said, B. M. Leu, L. A. Boatner, R. J. McQueeney, and O. Delaire, Nature **515**, 535
319 (2014).
320 [7] K. Appavoo, D. Y. Lei, Y. Sonnefraud, B. Wang, S. T. Pantelides, S. a Maier, and R. F.
321 Haglund, Nano Lett. **12**, 780 (2012).
322 [8] K. Appavoo, B. Wang, N. F. Brady, M. Seo, J. Nag, R. P. Prasankumar, D. J. Hilton, S. T.
323 Pantelides, and R. F. Haglund, Nano Lett. **14**, 1127 (2014).
324 [9] C. N. Berglund and H. J. Guggenheim, Phys. Rev. **185**, 1022 (1969).
325 [10] A. Cavalleri, M. Rini, H. H. W. Chong, S. Fourmaux, T. E. Glover, P. A. Heimann, J. C.
326 Kieffer, and R. W. Schoenlein, Phys. Rev. Lett. **95**, 067405 (2005).
327 [11] T. Kong, M. W. Masters, S. L. Bud'ko, and P. C. Canfield, APL Mater. **3**, 041502 (2015).
328 [12] K. Kosuge, J. Phys. Soc. Japan **22**, 551 (1967).
329 [13] J. Cao, Y. Gu, W. Fan, L. Q. Chen, D. F. Ogletree, K. Chen, N. Tamura, M. Kunz, C.
330 Barrett, J. Seidel, and J. Wu, Nano Lett. **10**, 2667 (2010).
331 [14] J. H. Park, J. M. Coy, T. S. Kasirga, C. Huang, Z. Fei, S. Hunter, and D. H. Cobden,
332 Nature **500**, 431 (2013).
333 [15] J. P. Pouget, H. Launois, J. P. D'Haenens, P. Merenda, and T. M. Rice, Phys. Rev. Lett. **35**,
334 873 (1975).
335 [16] J. P. Pouget, H. Launois, T. M. Rice, P. Dernier, A. Gossard, G. Villeneuve, and P.
336 Hagenmuller, Phys. Rev. B **10**, 1801 (1974).
337 [17] K. Okimura, T. Watanabe, and J. Sakai, J. Appl. Phys. **111**, 073514 (2012).
338 [18] A. Rúa, R. Cabrera, H. Coy, E. Merced, N. Sepúlveda, and F. E. Fernández, J. Appl. Phys.
339 **111**, 104502 (2012).
340 [19] E. Strelcov, A. Tselev, I. Ivanov, J. D. Budai, J. Zhang, J. Z. Tischler, I. Kravchenko, S. V
341 Kalinin, and A. Kolmakov, Nano Lett. **12**, 6198 (2012).
342 [20] E. Arcangeletti, L. Baldassarre, D. Di Castro, S. Lupi, L. Malavasi, C. Marini, A. Perucchi,
343 and P. Postorino, Phys. Rev. Lett. **98**, 196406 (2007).
344 [21] J. Laverock, S. Kittiwatanakul, A. Zakharov, Y. Niu, B. Chen, S. A. Wolf, J. W. Lu, and K.
345 E. Smith, Phys. Rev. Lett. **113**, 216401 (2014).
346 [22] T. Yao, X. Zhang, Z. Sun, S. Liu, Y. Huang, Y. Xie, C. Wu, X. Yuan, W. Zhang, Z. Wu, G.
347 Pan, F. Hu, L. Wu, Q. Liu, and S. Wei, Phys. Rev. Lett. **105**, 226405 (2010).

- 348 [23] D. Wegkamp, M. Herzog, L. Xian, M. Gatti, P. Cudazzo, C. L. McGahan, R. E. Marvel, R.
349 F. Haglund, A. Rubio, M. Wolf, and J. Stähler, *Phys. Rev. Lett.* **113**, 216401 (2014).
- 350 [24] V. R. Morrison, R. P. Chatelain, K. L. Tiwari, A. Hendaoui, A. Bruhacs, M. Chaker, and B.
351 J. Siwick, *Science* (80-.). 346, 445 (2014).
- 352 [25] H.-T. Kim, Y. W. Lee, B.-J. Kim, B.-G. Chae, S. J. Yun, K.-Y. Kang, K.-J. Han, K.-J. Yee,
353 and Y.-S. Lim, *Phys. Rev. Lett.* **97**, 266401 (2006).
- 354 [26] H.-T. Kim, B.-G. Chae, D.-H. Youn, S.-L. Maeng, G. Kim, K.-Y. Kang, and Y.-S. Lim,
355 *New J. Phys.* **6**, 52 (2004).
- 356 [27] J. B. Goodenough, *J. Solid State Chem.* **3**, 490 (1971).
- 357 [28] A. Zylbersztejn and N. F. Mott, *Phys. Rev. B* **11**, 4383 (1975).
- 358 [29] R. M. Rice, H. Launois, and J. P. Pouget, *Phys. Rev. Lett.* **73**, 3042 (1994).
- 359 [30] R. M. Wentzcovitch, *Phys. Rev. Lett.* **72**, 3389 (1994).
- 360 [31] R. M. Wentzcovitch, W. W. Schulz, and P. B. Allen, *Phys. Rev. Lett.* **73**, 3043 (1994).
- 361 [32] S. Biermann, A. Poteryaev, A. I. Lichtenstein, and A. Georges, *Phys. Rev. Lett.* **94**, 026404
362 (2005).
- 363 [33] C. Weber, D. D. O’Regan, N. D. M. Hine, M. C. Payne, G. Kotliar, and P. B.
364 Littlewood, *Phys. Rev. Lett.* **108**, 256402 (2012).
- 365 [34] M. Gatti, F. Bruneval, V. Olevano, and L. Reining, *Phys. Rev. Lett.* **99**, 266402 (2007).
- 366 [35] L. Hedin, *Phys. Rev.* **139**, A796 (1965).
- 367 [36] V. Eyert, *Phys. Rev. Lett.* **107**, 016401 (2011).
- 368 [37] X. Yuan, Y. Zhang, T. A. Abtew, P. Zhang, and W. Zhang, *Phys. Rev. B* **86**, 235103 (2012).
- 369 [38] R. Grau-Crespo, H. Wang, and U. Schwingenschlögl, *Phys. Rev. B* **86**, 081101 (2012).
- 370 [39] B. Xiao, J. Sun, A. Ruzsinszky, and J. P. Perdew, *Phys. Rev. B* **90**, 085134 (2014).
- 371 [40] H. Zheng and L. K. Wagner, *Phys. Rev. Lett.* **114**, 176401 (2015).
- 372 [41] K. Kosuge, Y. Ueda, S. Kachi, T. Shinjo, T. Takada, and M. Takano, *J. Solid State Chem.*
373 **23**, 105 (1978).
- 374 [42] P. Pouget, P. Lederer, D. Schreiber, H. Launois, D. Wohlleben, A. Casalot, and G.
375 Villeneuve, *J. Phys. Chem. Solids* **33**, 1961 (1972).
- 376 [43] G. Kresse and D. Joubert, *Phys. Rev. B* **59**, 1758 (1999).
- 377 [44] G. Kresse and J. Furthmüller, *Phys. Rev. B* **54**, 11169 (1996).
- 378 [45] J. P. Perdew, M. Ernzerhof, and K. Burke, *J. Chem. Phys.* **105**, 9982 (1996).
- 379 [46] C. Adamo and V. Barone, *J. Chem. Phys.* **110**, 6158 (1999).
- 380 [47] J. B. Goodenough, *J. Phys. Chem. Solids* **6**, 287 (1958).
- 381 [48] J. B. Goodenough, *Phys. Rev.* **100**, 564 (1955).
- 382 [49] J. Kanamori, *J. Phys. Chem. Solids* **10**, 87 (1959).
- 383 [50] J. M. Longo, P. Kierkegaard, C. J. Ballhausen, U. Ragnarsson, S. E. Rasmussen, E. Sunde,
384 and N. A. Sørensen, *Acta Chem. Scand.* **24**, 420 (1970).
- 385 [51] D. B. McWhan, M. Marezio, J. P. Remeika, and P. D. Dernier, *Phys. Rev. B* **10**, 490
386 (1974).
- 387 [52] M. Marezio, D. B. McWhan, J. P. Remeika, and P. D. Dernier, *Phys. Rev. B* **5**, 2541
388 (1972).

- 389 [53] H. Wang, T. A. Mellan, R. Grau-Crespo, and U. Schwingenschlögl, *Chem. Phys. Lett.* **608**,
390 126 (2014).
- 391 [54] B. Xiao, J. Sun, A. Ruzsinszky, and J. P. Perdew, *Phys. Rev. B* **90**, 085134 (2014).
- 392 [55] T. C. Koethe, Z. Hu, M. W. Haverkort, C. Schüßler-Langeheine, F. Venturini, N. B.
393 Brookes, O. Tjernberg, W. Reichelt, H. H. Hsieh, H. J. Lin, C. T. Chen, and L. H. Tjeng,
394 *Phys. Rev. Lett.* **97**, 116402 (2006).
- 395 [56] R. Molaei, R. Bayati, S. Nori, D. Kumar, J. T. Prater, and J. Narayan, *Appl. Phys. Lett.*
396 **103**, 252109 (2013).
- 397 [57] K. Schwarz, *J. Phys. F Met. Phys.* **16**, L211 (1986).
- 398 [58] I. I. Mazin, D. J. Singh, and C. Ambrosch-Draxl, *Phys. Rev. B* **59**, 411 (1999).
- 399 [59] H. Guo, K. Chen, Y. Oh, K. Wang, C. Dejoie, S. a Syed Asif, O. L. Warren, Z. W. Shan, J.
400 Wu, and a M. Minor, *Nano Lett.* **11**, 3207 (2011).
- 401 [60] E. O. Wollan and W. C. Koehler, *Phys. Rev.* **100**, 545 (1955).
- 402 [61] K. Okimura, N. Hanis Azhan, T. Hajiri, S. Kimura, M. Zaghrioui, and J. Sakai, *J. Appl.*
403 *Phys.* **115**, 153501 (2014).
- 404 [62] J. B. Goodenough and H. Y.-P. Hong, *Phys. Rev. B* **8**, 1323 (1973).
- 405 [63] M. Rohlfing and S. G. Louie, *Phys. Rev. Lett.* **81**, 2312 (1998).
- 406 [64] F. J. Nelson, J.-C. Idrobo, J. D. Fite, Z. L. Mišković, S. J. Pennycook, S. T. Pantelides, J.
407 U. Lee, and A. C. Diebold, *Nano Lett.* **14**, 3827 (2014).
- 408 [65] M.A. Korotin, N. A. Skorikov, and V. I. Anisimov, *Phys. Met. Met.* **94**, 17 (2002).
- 409 [66] A. Liebsch, H. Ishida, and G. Bihlmayer, *Phys. Rev. B* **71**, 085109 (2005).

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 431 FIG. 1 (color online). Optimized structures of different VO₂ phases: (a) NM-M1, (b) FM-M0, (c)
 432 FM-R, and (d) A-AFM-M2. Short V-V bonds (<2.50Å) are shown as solid lines (●—●)
 433 while long bonds (>3.00Å) have dotted lines (●⋯⋯⋯●). V-V bonds with lengths between 2.50
 434 and 3.00Å have dashed lines (●- - - -●).

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446 TABLE I. Comparison of lattice constants, V-V bond lengths, and V-V bond angles from this
 447 work and experiment (Exp). Note that the FM-M0 state values are compared to the monoclinic
 448 metallic state (mM) values as determined from x ray absorption fine structure measurements.

		M1	NM-M1	mM	FM-M0	R		M2	
		Exp [50]	This work	Exp [22]	This work	Exp [51]	This work	Exp [52]	This work
a (Å)		5.75	5.53	5.69	5.59	4.55	4.42	9.07	8.98
b (Å)		4.54	4.51	4.59	4.50	4.55	4.42	5.80	5.65
c (Å)		5.38	5.28	5.29	5.29	2.85	2.80	4.53	4.48
α, γ (°)		90	90	90	90	90	90	90	90
β (°)		122.65	121.93	122.61	122.05	90	90	91.88	91.88
V-V bond (Å)	short	2.62	2.44	2.72	2.69	2.85	2.80	2.54	2.40
	middle							2.93	2.86
	long	3.17	3.14	2.98	2.94			3.26	3.25
V-V angle (°)		168	166		175	90	90	162	161

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463 TABLE II. Calculated magnetic ground states and band gaps of VO₂ phases compared to
 464 experiment.

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		Experiment	Theoretical results					
			This work	HSE			GW	DMFT
				[36] ^c	[38] ^d	[37]	[34]	[32] ^g
Magnetic ground states	M0	FM/PM [41,42] ^a	FM					
	M1	NM [11,56] ^b	NM		AFM	AFM		
	M2	AFM [16]	A-AFM			FM		
Band gap (eV)	M1	0.6-0.7 [9,10]	0.63	1.10	2.23 (AFM) 0.98 (NM) ^e		0.65	0.60
	M2	>0.10 [61]	0.56	1.20				
	R	0 [9,10]	0	0	1.43 (FM) 0 (NM) ^f		0	0

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 467 ^a Divergence of the magnetic susceptibility below 30 K underlines the importance of exploring the unknown
 468 low-temperature magnetic properties.

469 ^b The disagreement of measurements of small positive [11] susceptibility and another publication [56]
 470 reporting small negative susceptibility justified our designation of M1 as NM as similar to previous
 471 authors [38].

472 ^c Band gap of each VO₂ phase was calculated by assuming the magnetic state found in experiments.

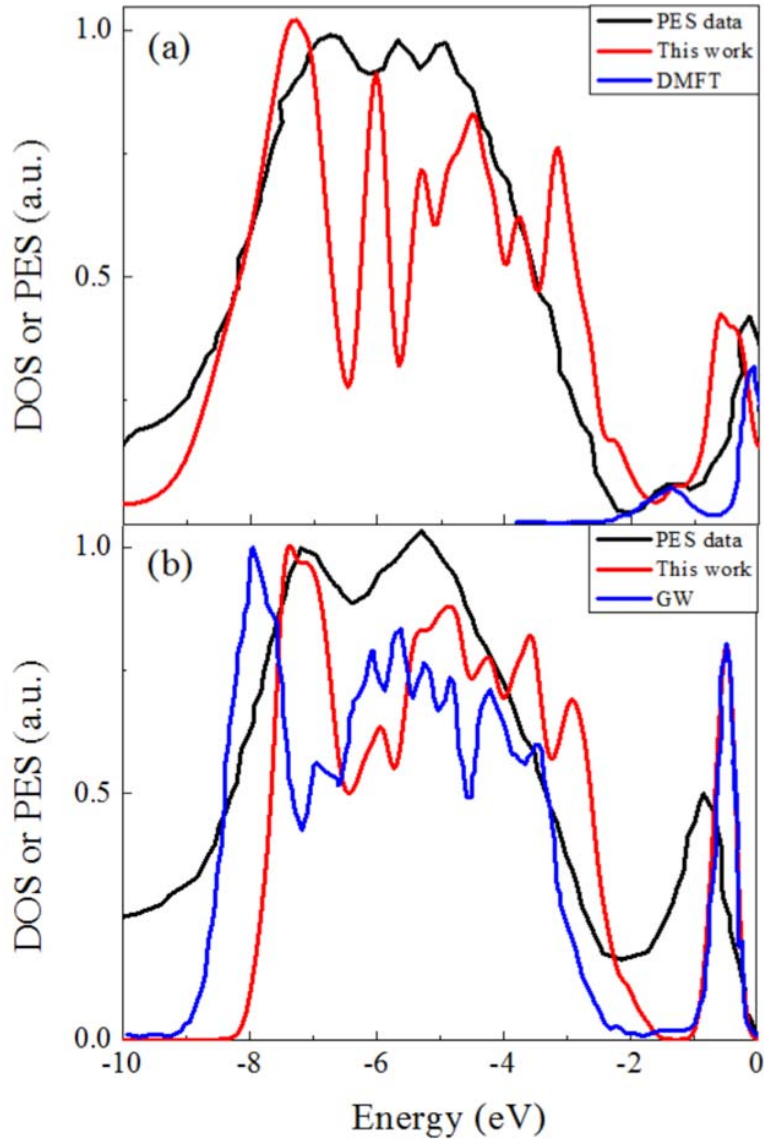
473 ^d Non-spin-polarized calculations similar to those of Eyert [36] were reproduced and then spin-polarized
 474 calculations for each potential magnetic state were performed [38].

475 ^e The correct magnetic phase, NM-M1, has a calculated band gap is close to the experimental value. However,
 476 AFM-M1 was calculated to be lower in energy, and the band gap is over thrice the expected value.

477 ^f A ferromagnetic R state with a band gap of 1.43 eV was calculated to be the ground state. However, a NM
 478 state with a correct band gap of 0 was also obtained, albeit at a higher energy.

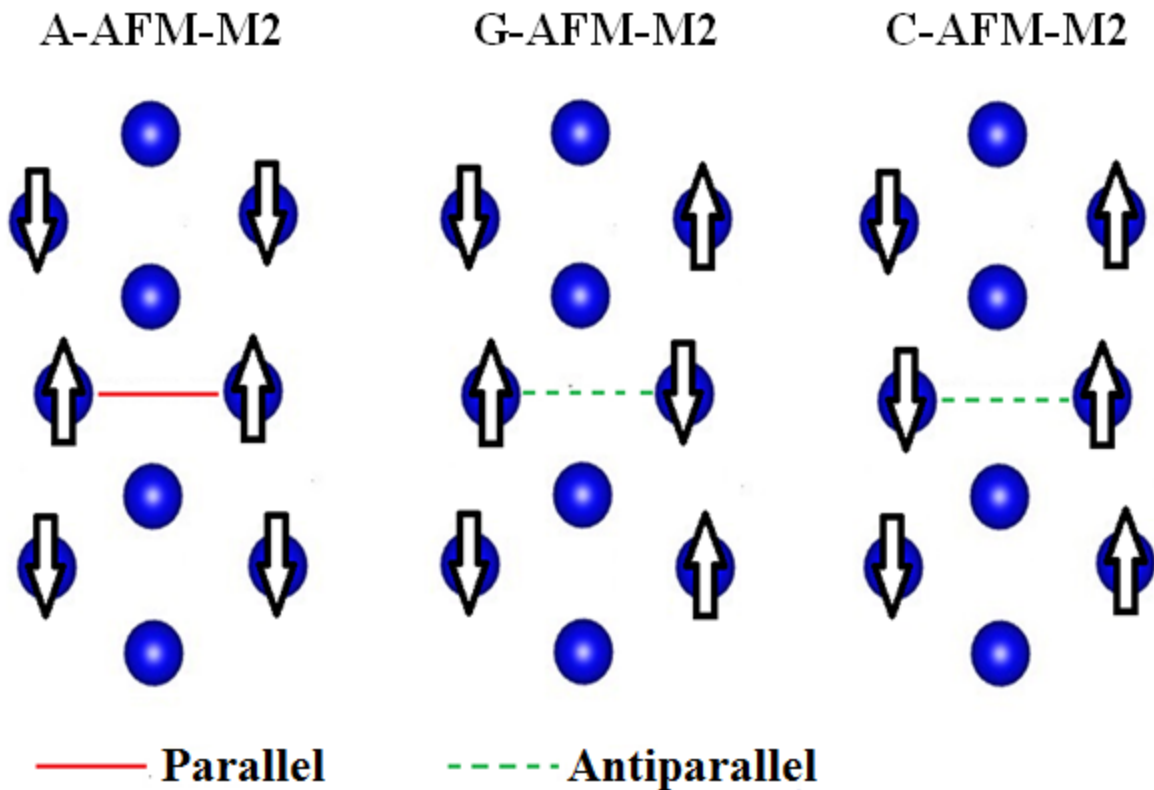
479 ^g A stable nonmagnetic structure was obtained with cluster-DMFT, but it was not compared to other magnetic
 480 states to determine the ground state.

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 487 FIG. 2 (color online). (a) The DOS of FM-R calculated in this work (red) is compared with the
 488 experimental [55] photoemission spectrum (black) and the V 3d (t_{2g}) spectral weights (blue) from
 489 LDA+DMFT calculations [32]. The 1.3 eV satellite feature is clearly found in this work. (b) The
 490 total DOS of NM-M1 calculated in this work (red) is compared with the experimental [55]
 491 photoemission spectrum (black) of the low temperature insulating M1 and the DOS (blue) from
 492 GW calculations [34]. Each DOS from this work was convoluted with a Gaussian function.

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FIG. 3 (color online). Schematic of the three possible magnetic structures of AFM-M2: A-AFM, G-AFM and C-AFM. The blue solid circles are V atoms and the white arrows represent their magnetic moments. The solid line between two adjacent canted chains represents parallel magnetic moments between the nearest vanadium atoms from each chain, while the dashed lines represent an antiparallel configuration. The A-AFM configuration has the lowest energy.