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Electronic structure and magnetic properties in $T_2\text{AlB}_2$ ($T = \text{Fe, Mn, Cr, Co, and Ni}$) and their alloys

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Abstract

The electronic structure and intrinsic magnetic properties of Fe_2AlB_2 -related compounds and their alloys have been investigated using density functional theory. For Fe_2AlB_2 , the crystallographic a -axis is the easiest axis, which agrees with experiments. The magnetic ground state of Mn_2AlB_2 is found to be ferromagnetic in the basal ab -plane, but antiferromagnetic along the c -axis. All $3d$ dopings considered decrease the magnetization and Curie temperature in Fe_2AlB_2 . Electron doping with Co or Ni has a stronger effect on the decreasing of Curie temperature in Fe_2AlB_2 than hole doping with Mn or Cr. However, a larger amount of Mn doping on Fe_2AlB_2 promotes the FM \rightarrow AFM transition. A very anisotropic magneto-elastic effect is found in Fe_2AlB_2 : the magnetization has a much stronger dependence on the lattice parameter c than on a or b , which is explained by electronic-structure features near the Fermi level. Dopings of other elements on B and Al sites are also discussed.

I. INTRODUCTION

Magnetic cooling, which is based on the magnetocaloric effect (MCE) and discovered one century ago, has long been used in scientific laboratories to attain extremely low temperatures. A major breakthrough came in the late 1990s when Pecharsky and Gschneidner discovered giant MCE around room temperature (RT) in a new class of magnetic materials¹. This discovery has rekindled research interest in utilizing MCE for much broader applications, such as domestic appliances, which usually operate around RT. If successful, this more energy-efficient and environment-friendly magnetic cooling technique may replace conventional compressor-based refrigeration and revolutionize the cooling industry. This new era may arrive in the near future—only if one can find or engineer a proper MCE material which has large MCE under a magnetic field that can be generated by permanent magnets, and is also abundant, affordable, and has a good lifespan.

Fe_2AlB_2 is one of the promising candidates for this purpose and has attracted great attention since the recent discovery of its substantial MCE around RT². The reported entropy change has a value of $\Delta S_m = 4.1\text{--}7.7 \text{ J kg}^{-1} \text{ K}^{-1}$ in the presence of an external field $B = 2\text{--}5 \text{ T}$. Although Fe_2AlB_2 does not have the largest MCE of all materials, it is easy to synthesize and does not contain any rare, expensive, or toxic elements³. Moreover, its volume barely changes during the magnetic transition, which may ensure Fe_2AlB_2 a good life span for refrigerator applications operating at high cycle frequencies^{4,5}.

The MCE often peaks at the Curie temperature T_C of the material, however real applications require materials with a large MCE over a certain operating temperature range. This likely needs to be achieved by using composite materials with multiple compositions, so the system can have MCE over the whole operating temperature range for specific applications. It seems the first logical selection of doping would be the substitution of Fe with other $3d$ transition metal elements

T . Pure Mn_2AlB_2 ⁶ and Cr_2AlB_2 ⁷ can be formed and share the same structure of Fe_2AlB_2 . Combining theory with experiments, Kadás *et al.* studied the phase stability in $T_2\text{AlB}_2$ with $T = \text{Cr, Mn, Fe, Co, and Ni}$. They found that although compounds are metastable with $T = \text{Co and Ni}$, $\text{Fe}_{2-x}\text{Co}_x\text{AlB}_2$, $\text{Fe}_{2-x}\text{Ni}_x\text{AlB}_2$, or even $(\text{Fe}_{2-x-y}\text{Co}_x\text{Ni}_y)\text{AlB}_2$ could be stable⁸. However, the magnetic properties of those alloys or even their parent compounds are not well understood. For example, the magnetic ground state of Mn_2AlB_2 had been reported to be ferromagnetic (FM)⁶, but recent experiments concluded that it should be either nonmagnetic (NM) or antiferromagnetic (AFM)⁹. To provide a guidance on tuning Fe_2AlB_2 , a better understanding of the magnetic properties of pure $T_2\text{AlB}_2$ and their alloys is desired.

The high melting temperature of FeB makes it a difficult impurity to remove from Fe_2AlB_2 samples. The rapid cooling by melt spinning had been used to greatly improve the Fe_2AlB_2 purity by suppressing the growth of FeB⁴. On the other hand, extra Al is often added during the synthesis to decrease the formation of FeB impurities^{4,10}. With a higher Al content, $\text{Al}_{13}\text{Fe}_4$ becomes the main impurity phase. Reported magnetization values measured at low temperature vary between $1.0\text{--}1.32 \mu_B/\text{Fe}$ and the T_C values vary between $282\text{--}320 \text{ K}$ ^{2,4,9,11}. The variation of experimental values may due to the existence of an impurity phase in the Fe_2AlB_2 sample.

Experimentally, substitution of Mn^{4,5,9} or up to 15 % of Co¹⁰ on Fe sites had been reported; a larger amount of Mn- or Co-doping makes the structures unstable⁴ or phase inhomogeneous⁹. Both dopings decrease the magnetization and T_C in Fe_2AlB_2 . A spin-glass state had been found in $\text{Fe}_{1.5}\text{Mn}_{0.5}\text{AlB}_2$ at low temperature⁵. With Co-doping, T_C and the Co content are linearly correlated, which makes Co content a convenient parameter to tune the material to have MCE between RT and 200 K ¹⁰. A few studies have been reported to investigate the electronic structures and structural, electronic, and magnetic properties of those compounds^{9,11–13}.

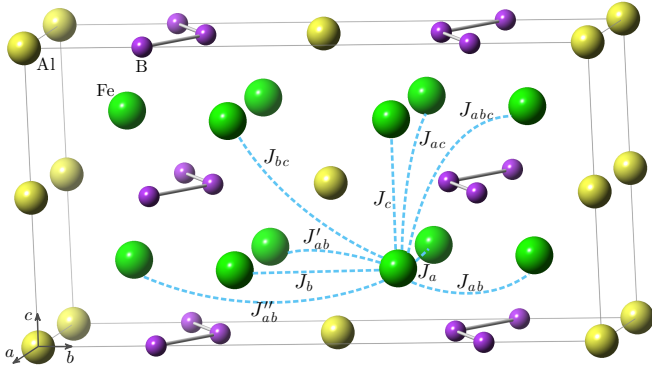


FIG. 1: Schematic representation of the crystal structure of Fe_2AlB_2 . The conventional cell is doubled along the c -axis to depict pair exchange parameters for the first few neighbors of Fe atoms (large green colored spheres). Al atoms are indicated with yellow colored spheres. B atoms, indicated by small purple spheres, form zigzag chains along the a -axis. Exchange parameters are labeled according to the axis or plane of the connecting vector.

In this work, using density functional theory (DFT), we investigate the intrinsic magnetic properties including magnetization, exchange parameters, Curie temperature, and magnetocrystalline anisotropy in Fe_2AlB_2 and their alloys. The magneto-elastic effect in Fe_2AlB_2 and the dopings of various elements on B and Al sites are also discussed. Electronic structures are studied to understand the magnetic properties.

II. COMPUTATIONAL DETAILS

A. Crystal structure

Fe_2AlB_2 crystallizes in the orthorhombic Mn_2AlB_2 -type ($Cmmm$, space group no. 65) structure. The primitive cell contains one formula unit (f.u.) while the conventional cell contains two. The crystal structure is shown in Fig. 1. We double the conventional cell along the c -axis to denote the first few exchange parameters. B atoms occupy the $4i$ ($m2m$) site, forming a zigzag chain in the ab -plane and along the a -axis. Fe atoms occupy the $4j$ ($m2m$) sites and form Fe_6B prisms with neighboring B atoms. The structure can be derived from the FeB structure by inserting one layer of Al atoms perpendicular to the b -axis, between each pair of planes containing the Fe_6B prisms¹⁴. Each Al atom, which occupies the $2a$ (mmm) site, is surrounded by eight Fe atoms. Together they form a body-center-tetragonal cell elongated along the b -axis. For this Fe-Al cage, the Fe-Al bondlength is 2.61 \AA and the Fe-Fe distances are 2.87 , 2.92 and 3.22 \AA , along the c , a and b directions, respectively. Lying in the ab -plane, the nearest Fe-Fe bond has a length of 2.72 \AA . The structure can also be derived by stacking the pure Fe plane and the Al-B plane alter-

natively along the c -axis. This view is probably more convenient to understand the magnetic properties, such as exchange coupling and magneto-elastic effect, both of which are very anisotropic along the c -axis, as we will discuss later.

B. Computational methods

Electronic structure and most magnetic properties are calculated using a standard linear muffin-tin orbital (LMTO) basis set¹⁵ generalized to full potentials (FP)¹⁶. This scheme employs generalized Hankel functions as the envelope functions. Calculations are carried out within the generalized gradient approximation (GGA) to DFT with the exchange-correlation parametrization of Perdew, Burke, and Ernzerhof (PBE)¹⁷, unless LDA¹⁸ (local density approximation, with the exchange-correlation parametrization of von Barth and Hedin) is specified.

The magnetocrystalline anisotropy energy (MAE) is calculated using the force theorem¹⁹. Starting from the self-consistent scalar-relativistic potential, the spin-orbit coupling (SOC) is included in a subsequent one-step calculation with spin being along direction \hat{n} . The MAE is characterized below as $K_{\hat{n}} = E_{\hat{n}} - E_{001}$, where E_{001} and $E_{\hat{n}}$ are the summation of occupied band energies for the magnetization being oriented along the $[001]$ and \hat{n} directions, respectively.

Exchange coupling parameters J_{ij} are calculated using a static linear-response approach implemented in a Green's function (GF) LMTO method, simplified using the atomic sphere approximation (ASA) to the potential and density^{20,21}. The scalar-relativistic Hamiltonian is used so SOC is not included, although it is a small perturbation on J_{ij} 's. In the basis set, s, p, d orbitals are included for T and Al atoms, and s, p orbitals are included for B atom. A dense k -point mesh is used to calculate exchange parameters $J(\mathbf{q})$, e.g., a 32^3 k -point mesh for the 5-atom cell. The real space $J(\mathbf{R})$ are obtained by a subsequent Fourier-transform. Curie temperatures are estimated in the mean-field approximation (MFA) with $k_B T_C = 2/3 \sum_i J_{0i}$. The coherent potential approximation (CPA) implemented within the LMTO-ASA-GF code is used to address the chemical effects of doping on magnetization and T_C . Without using supercell calculations, the CPA provides an elegant and efficient approach to investigate substitutional effects with an arbitrary composition. The details of the methods and applications can be found elsewhere^{21,22}.

Both experimental and theoretically optimized crystal structures are used to investigate the magnetic properties. We fully relax internal atomic positions and lattice constants with the PBE functional using a fast plane-wave method, as implemented within the Vienna *ab initio* simulation package (VASP)^{23,24}. The nuclei and core electrons are described by the projector augmented-wave potential²⁵ and the wave functions of valence elec-

TABLE I: Calculated atomic spin magnetic m_i , spin magnetization M , and Curie temperature T_C in Fe_2AlB_2 .

| Fe_2AlB_2 | m_i (μ_B/atom) | | | M | T_C |
|---------------------------|-------------------------------|-------|-------|-------------------------|-------|
| Method | Fe | Al | B | ($\mu_B/\text{f.u.}$) | (K) |
| FP-GGA | 1.43 | -0.04 | -0.01 | 2.73 | |
| ASA-GGA | 1.38 | -0.08 | -0.04 | 2.62 | 329 |
| FP-LDA | 1.31 | -0.01 | -0.03 | 2.54 | |
| ASA-LDA | 1.20 | -0.06 | -0.03 | 2.29 | 232 |

trons are expanded in a plane-wave basis set with a cutoff energy of up to 520 eV.

III. RESULTS AND DISCUSSION

A. Pure compounds: Exchange coupling, magnetic anisotropy, and spin configurations

Table I shows the atomic spin moments m_i at each sublattice and magnetization M in Fe_2AlB_2 , which are calculated using the experimental lattice constants and atomic position parameters from Ref. [26]. Al and B have small moments antiparallel to the Fe sublattice. Within the GGA, a magnetization of $M = 1.36$ is obtained using FP. For the sake of comparison, we carry out similar calculation for the parent compound—FeB, and obtain a magnetization of $M = 1.20$. The smaller Fe moment in FeB is likely due to its smaller Fe-Fe bondlength (2.62 Å) than in Fe_2AlB_2 . ASA gives a slightly smaller (by 4%) magnetization than FP in Fe_2AlB_2 , suggesting that ASA is suitable for this material. The calculated $T_C = 329$ K is slightly above the upper bound of the reported experimental T_C values. The agreement is fair considering the MFA generally overestimates T_C . LDA gives smaller magnetization, especially with ASA, resulting in a smaller T_C .

Starting from the FM configuration and using experimental crystal structures, we calculate the exchange coupling J_{ij} in Fe_2AlB_2 and Mn_2AlB_2 . Figure 2 shows the J_{ij} as a function of the distance R_{ij} . In both compounds J_{ij} becomes negligible after $R_{ij} > 6$ Å. The exchange parameters between first few nearest neighbors are also listed in Table II. Here, J_{ij} can be treated as stability parameters and a negative J_{ij} indicates that the given spin configuration is not favorable for that particular pair of sites²¹.

For Fe_2AlB_2 , all of the first four nearest exchange parameters are positive. The J_{ij} value increases with distance, reaching maximum at J_b , and then decrease, which generally agrees with the previous study¹³. LDA gives a similar trend but a smaller amplitude of J_{ij} than GGA, which reflects the smaller magnetic moments obtained within LDA.

The magnetic ground state of Mn_2AlB_2 is not well understood⁹. For simplicity, we start from the FM config-

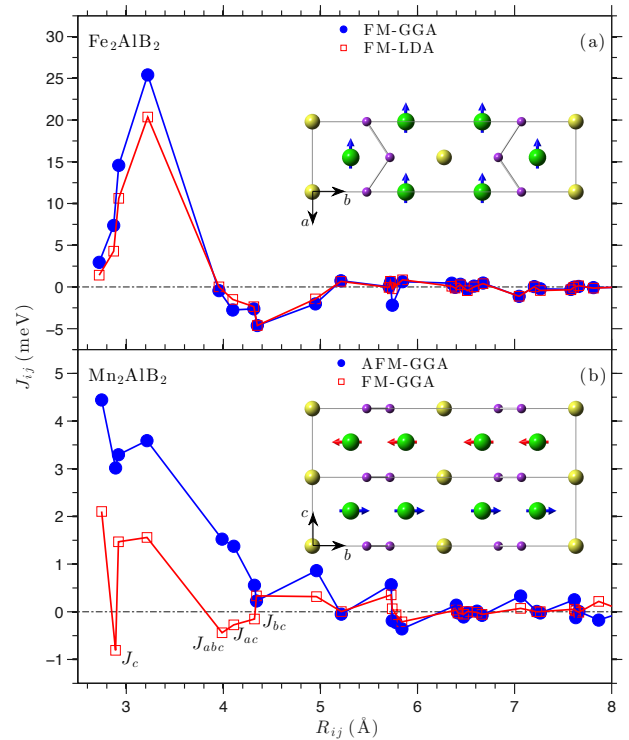


FIG. 2: Real-space magnetic exchange parameters J_{ij} in Fe_2AlB_2 (a), and Mn_2AlB_2 (b) as functions of distance. For Fe_2AlB_2 , both GGA and LDA results are shown. For Mn_2AlB_2 , both FM and AFM spin configurations are calculated within GGA. The spin configuration of their magnetic ground states are shown in insets.

uration. The calculated magnetization is $0.42 \mu_B/\text{Mn}$, which agrees well with a previous FM calculation⁹. Exchange parameters calculated in the FM configuration show a very interesting feature: all dominant J_{ij} are positive for neighbors within the Mn ab -plane but negative for neighbors between neighboring Mn layers, namely J_c , J_{abc} , J_{ac} , and J_{bc} . This suggests that the FM coupling of Mn atoms is stable within the ab -layer but not between neighboring layers. To confirm it, we calculate J_{ij} for the AFM configuration, in which FM Mn ab -layers couple antiferromagnetically along the c -axis. This AFM configuration gives lower energy than the FM configuration. Moreover, as shown in Fig. 2(b), the dominant exchange parameters become all positive and larger. To better quantify the relative stability of those two spin configurations, we fully relaxed the structure and found that the AFM configuration increases the on-site Mn moment to $0.75 \mu_B/\text{Mn}$ and lowers the total energy by 42 meV/f.u.. Unlike Fe_2AlB_2 , the largest exchange interaction in Mn_2AlB_2 is J_{ab} , which is between the nearest Mn neighbors. The Néel temperature is estimated to be $T_N = 310$ K within MFA for the AFM configuration with the experimental crystal structure.

Next we consider the SOC effect in Fe_2AlB_2 . Figure 3 shows the energy and the average orbital magnetic mo-

TABLE II: Pairwise exchange parameters J_{ij} for the Heisenberg Hamiltonian $H = -\sum J_{ij} \hat{\mathbf{e}}_i \cdot \hat{\mathbf{e}}_j$, and $\hat{\mathbf{e}}_j$ is the unit vector pointing along the direction of the local spin moment at site i . The experimental lattice parameters and atomic positions are used. For Fe_2AlB_2 , both GGA and LDA results are shown. For Mn_2AlB_2 , the PBE functional is used, and both FM and AFM spin configurations are considered.

| Fe_2AlB_2 | | R_{ij} | | $\hat{\mathbf{R}}_{ij}$ | | | J_{ij} (meV) | |
|---------------------------|-----|----------|-------|-------------------------|--------|--------|----------------|-------|
| Lbl. | No. | (Å) | (a) | x | y | z | GGA | LDA |
| J_{ab} | 2 | 2.721 | 0.931 | 0.5 | 0.785 | 0 | 2.95 | 1.41 |
| J_c | 2 | 2.870 | 0.982 | 0 | 0 | 0.982 | 7.36 | 4.29 |
| J_a | 2 | 2.923 | 1 | 1 | 0 | 0 | 14.58 | 10.63 |
| J_b | 1 | 3.222 | 1.102 | 0 | 1.102 | 0 | 25.41 | 20.37 |
| J_{abc} | 4 | 3.955 | 1.353 | 0.5 | -0.785 | 0.982 | -0.43 | 0.01 |
| J_{ac} | 4 | 4.097 | 1.401 | 1 | 0 | 0.982 | -2.76 | -1.49 |
| J_{bc} | 2 | 4.315 | 1.476 | 0 | 1.102 | 0.982 | -2.63 | -2.37 |
| J'_{ab} | 2 | 4.350 | 1.488 | 1 | 1.102 | 0 | -4.62 | -4.62 |
| J''_{ab} | 2 | 4.949 | 1.693 | 1.5 | -0.785 | 0 | -2.03 | -1.43 |
| Mn_2AlB_2 | | R_{ij} | | $\hat{\mathbf{R}}_{ij}$ | | | J_{ij} (meV) | |
| Lbl. | No. | (Å) | (a) | x | y | z | FM | AFM |
| J_{ab} | 2 | 2.747 | 0.941 | 0.5 | -0.797 | 0 | 2.10 | 4.44 |
| J_c | 2 | 2.890 | 0.990 | 0 | 0 | -0.990 | -0.81 | 3.02 |
| J_a | 2 | 2.920 | 1 | 1 | 0 | 0 | 1.47 | 3.29 |
| J_b | 1 | 3.213 | 1.100 | 0 | 1.100 | 0 | 1.56 | 3.59 |
| J_{abc} | 4 | 3.987 | 1.365 | -0.5 | -0.797 | 0.990 | -0.44 | 1.52 |
| J_{ac} | 4 | 4.108 | 1.407 | -1 | 0 | 0.990 | -0.27 | 1.37 |
| J_{bc} | 2 | 4.322 | 1.480 | 0 | 1.100 | 0.990 | -0.15 | 0.56 |
| J'_{ab} | 2 | 4.342 | 1.487 | -1 | 1.100 | 0 | 0.33 | 0.23 |
| J''_{ab} | 2 | 4.960 | 1.698 | -1.5 | -0.797 | 0 | 0.32 | 0.86 |

ments of each sublattice as functions of spin quantization axis orientation, which rotates from the $c \rightarrow a \rightarrow b \rightarrow c$ axis. The relaxed structure was used for the calculations. The a -axis is the easiest axis, which agrees with recent neutron scattering experiments¹³. The c -axis is the hardest axis, while the anisotropy within the ab plane is very small. Energy changes by $K_{100} = -0.38 \text{ meV/f.u.}$ (-1.34 MJ/m^3) when the spin quantization axis rotates from the c -axis to the a -axis. The anisotropy calculated using experimental structure²⁶ is larger by $\sim 10\%$, reaching $K_{100} = -1.47 \text{ MJ/m}^3$. B and Al atoms have negligible orbital magnetic moments, as expected for light $2p$ and $3p$ elements. The orbital moment of Fe is also rather small, and interestingly has the largest value of $0.02 \mu_B/\text{Fe}$ when spin is along the hardest c -axis instead of in the ab -plane. Similar behavior has also been found in materials such as FePt ²⁷. To elucidate the origin of MAE, we evaluate the anisotropy of the on-site SOC energy with respect to the c -axis and the ab -plane, $K_{110}^{\text{so}} = \frac{1}{2} \langle V_{\text{so}} \rangle_{110} - \frac{1}{2} \langle V_{\text{so}} \rangle_{001}$, and resolve it into four spin channels^{28,29}. It is well known that only when the MAE is dominated by the $\downarrow\downarrow$ term, one may expect an obvious

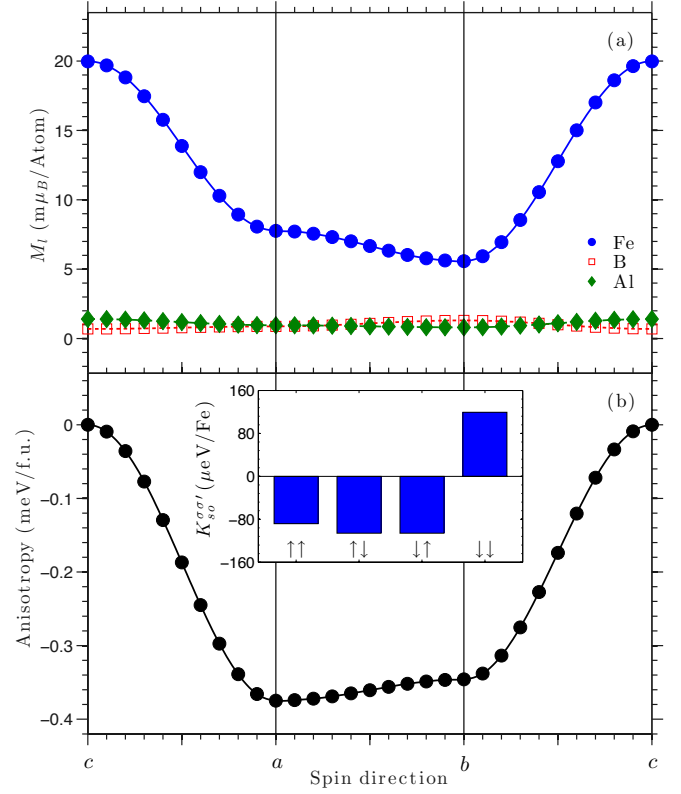


FIG. 3: Variation of (a) atomic orbital magnetic moments of Fe, Al, and B sublattices, and (b) energy as functions of spin quantization axis rotation in Fe_2AlB_2 . The inset in panel (b) shows the spin-resolved anisotropy of Fe-site spin-orbit coupling energy $K_{\text{so}}^{\sigma\sigma'} = \frac{1}{2} \langle V_{\text{so}} \rangle_{110} - \frac{1}{2} \langle V_{\text{so}} \rangle_{001}$.

correlation between the orbital moment and MAE, and a larger orbital moment along the easy axis^{28,30}. As shown in the inset of 3(b), the amplitudes of four spin components of $K_{\text{so}}^{\sigma\sigma'}$ are comparable. The $\downarrow\downarrow$ term favors the spin to be along the c -axis, while all other three terms ($\uparrow\uparrow$, $\uparrow\downarrow$, and $\downarrow\uparrow$) favor the spin to lie in the ab -plane, explaining the absence of the correlation between MAE and orbital moment in Fe_2AlB_2 . LDA gives a smaller anisotropy ($K_{100} = -0.82 \text{ MJ/m}^3$ using experimental structure) but a similar trend of angular dependence of energy. For Mn_2AlB_2 , the c -axis is also the hardest axis. Within ab -plane, the system has slightly lower energy when the spin is along the b -axis. As expected, its anisotropy is much smaller than in Fe_2AlB_2 . The schematic representation of the ground state spin configurations of Fe_2AlB_2 and Mn_2AlB_2 are shown in the insets of Fig. 2.

Figure 4 shows the scalar-relativistic partial density of states (PDOS) projected on individual elements in $T_2\text{AlB}_2$ with $T = \text{Cr, Mn, Fe, Co, and Ni}$. The total density of states (DOS), scaled by $1/2$, is also shown to compare. The total DOS of Fe_2AlB_2 compares well with previously reported calculations⁹. Al- $3s$ and B- $2s$ states are located between -12 and -7 eV below the Fermi level E_F . Al- $3p$ states hybridize with T atoms at around -4 eV

TABLE III: Lattice parameters, internal atomic positions, y_{4j} and y_{4i} , on-site atomic magnetic moment of T atoms, m_T (μ_B), relative total energy (meV/f.u.), and critical temperatures (Curie temperature in Fe_2AlB_2 or Néel temperature in Mn_2AlB_2), T_C (K) in $T_2\text{AlB}_2$ with $T = \text{Fe}, \text{Mn}, \text{Cr}, \text{Co}$, and Ni . T atom occupies the $4j$ site ($0 \leq y_{4j} \leq 1/2$) and B atom occupies the $4i$ ($0 \leq y_{4i} \leq 0$) site.

| T | a | b | c | y_{4j} | y_{4i} | m_T | ΔE | T_C |
|-----------|-------|--------|-------|----------|----------|-------|------------|-------|
| Cr | 2.921 | 11.034 | 2.929 | 0.3521 | 0.2057 | 0 | | |
| Exp. [7] | 2.937 | 11.07 | 2.971 | 0.352 | 0.220 | | | |
| Exp. [32] | 2.937 | 11.047 | 2.968 | | | | | |
| Mn-NM | 2.890 | 11.050 | 2.817 | 0.3562 | 0.2060 | 0 | 0 | |
| Mn-FM | 2.892 | 11.056 | 2.826 | 0.3551 | 0.2060 | 0.42 | -21.5 | |
| Mn-AFM | 2.887 | 11.109 | 2.830 | 0.3547 | 0.2061 | 0.75 | -63.6 | 296 |
| Exp. [26] | 2.92 | 11.08 | 2.89 | 0.355 | 0.209 | | | |
| Exp. [9] | 2.936 | 11.12 | 2.912 | | | | | |
| Fe-NM | 2.951 | 11.261 | 2.698 | 0.3531 | 0.2065 | 0 | 0 | |
| Fe-AFM | 2.941 | 11.212 | 2.739 | 0.3559 | 0.2070 | 1.06 | -97.9 | |
| Fe-FM | 2.915 | 11.017 | 2.851 | 0.3537 | 0.2063 | 1.37 | -164.1 | 298 |
| Exp. [14] | 2.923 | 11.034 | 2.870 | 0.3540 | 0.2071 | | | |
| Co | 2.962 | 11.314 | 2.689 | 0.3541 | 0.2073 | 0.21 | | |
| Ni | 2.979 | 11.041 | 2.843 | 0.3586 | 0.2101 | 0 | | |

below E_F . For Mn_2AlB_2 , the Fermi level is located at a pseudogap in the AFM configuration, and the DOS at E_F is smaller than in the FM configuration, which again suggests that the AFM configuration is more stable. The calculated hypothetical Co_2AlB_2 shows a weak magnetic moment of about $0.2 \mu_B/\text{Co}$. The two peaks at $\pm 0.1 \text{ eV}$ around the Fermi level, as shown in Fig. 4(d), will be pinched at E_F in the non-spin-polarized calculation (not shown). Thus, the small spin-polarization decreases the DOS at E_F and stabilizes the system. The calculated Cr_2AlB_2 and hypothetical Ni_2AlB_2 are non-magnetic and have small DOS at E_F . It is worth noting that $\text{Ni}_x\text{B}_{1-x}$ systems become magnetic only after $x > 0.75$ ³¹.

Table III summarizes the lattice parameters, atomic positions, atomic moment of T site, relative total energies, and critical temperatures of $T_2\text{AlB}_2$ with different magnetic configurations. The calculated lattice parameters and atomic positions agree well with experiments. The relaxed lattice parameters of $T_2\text{AlB}_2$ not only vary with element T but also depend on the spin configuration. For Fe_2AlB_2 , lattice parameters a and b decrease, and c increases when the spin configuration changes from NM \rightarrow AFM \rightarrow FM. The calculated lattice parameters using the FM configuration agree the best with experiments. For Mn_2AlB_2 , b and c increase when the spin configuration changes from NM \rightarrow FM \rightarrow AFM. Relative to those of Fe_2AlB_2 , the lattice parameter a of $T_2\text{AlB}_2$ varies within 0.06 \AA (2.2%) in the sequence of $\text{Mn} < \text{Fe} \approx \text{Cr} < \text{Co} < \text{Ni}$; b varies within 0.30 \AA (2.7%) in the sequence of $\text{Fe} < \text{Ni} < \text{Cr} \approx \text{Mn} < \text{Co}$; and pa-

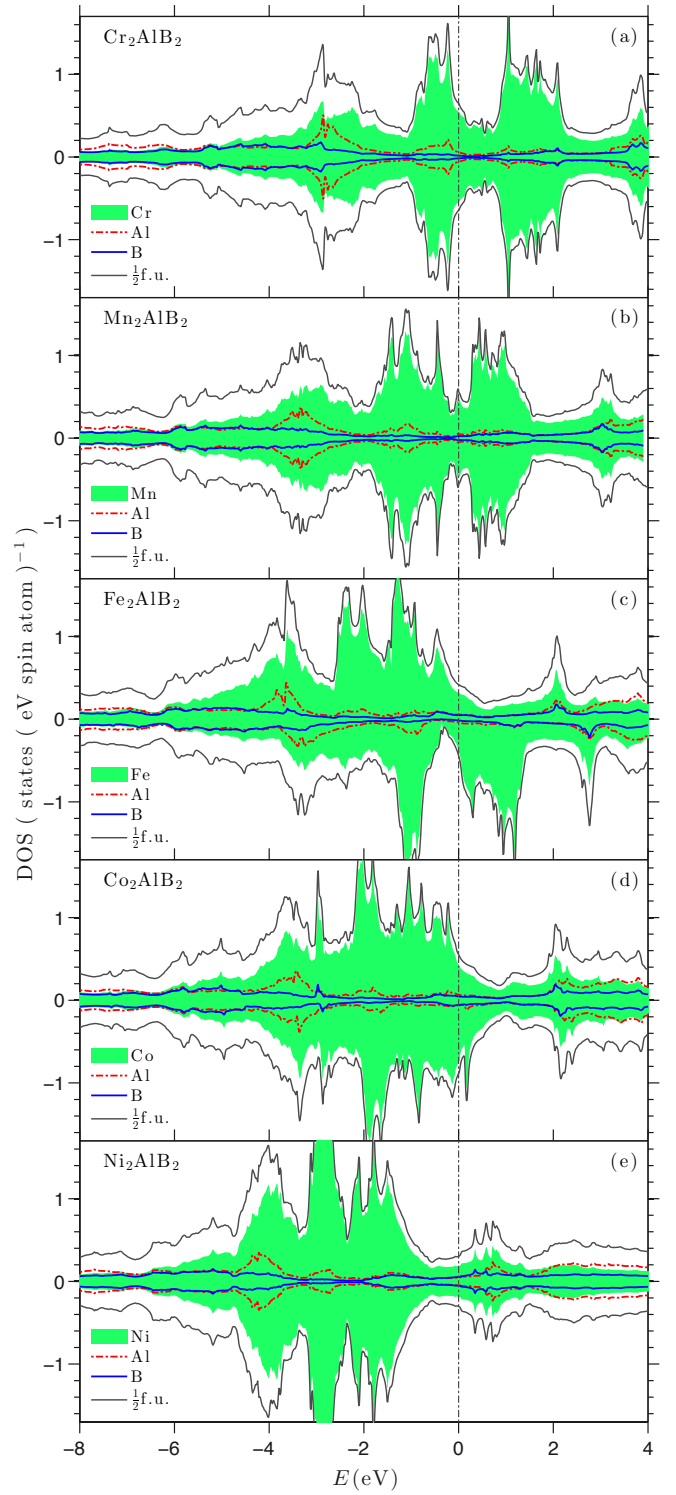


FIG. 4: Atom- and spin-projected, scalar-relativistic partial densities of states (DOS) in (a) Cr_2AlB_2 , (b) Mn_2AlB_2 , (c) Fe_2AlB_2 , (d) Co_2AlB_2 , and (e) Ni_2AlB_2 calculated within GGA and using fully relaxed structures. The total DOS of the f.u. cell are scaled by $\frac{1}{2}$ to better compare. Cr_2AlB_2 and Ni_2AlB_2 are non-magnetic. Mn_2AlB_2 is anti-ferromagnetic and Fe_2AlB_2 is ferromagnetic. Co_2AlB_2 is weakly ferromagnetic. Fermi energy E_F is at 0 eV.

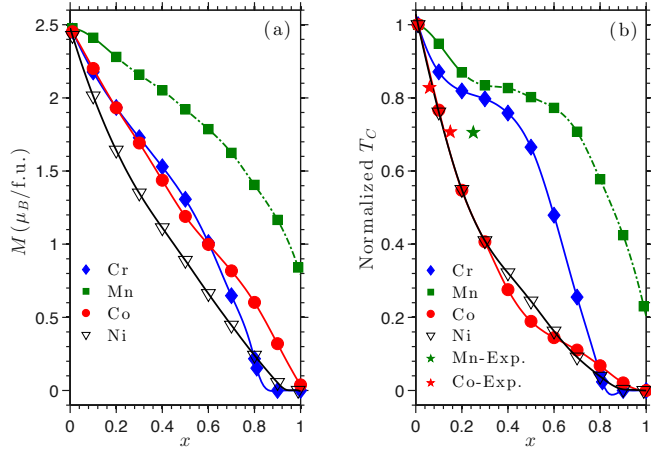


FIG. 5: Spin magnetization M (a) and normalized T_C (b) as functions of doping concentration x in $(\text{Fe}_{1-x}\text{T}_x)_2\text{AlB}_2$ with $T = \text{Cr, Mn, Co, and Ni}$. Theoretical T_C is estimated within MFA and normalized with respect to pure Fe_2AlB_2 . Experimental T_C values of Mn and Co, adopted from Ref.[5] and [10], are denoted by stars in green and red, respectively. For $(\text{Fe}_{1-x}\text{Mn}_x)_2\text{AlB}_2$, the FM configuration becomes less stable than AFM within CPA after $x > 0.2$, which are denoted by the green dashed line.

parameter c varies within 0.16 \AA (5.7%) in the sequence of $\text{Co} < \text{Mn} < \text{Ni} \approx \text{Fe} < \text{Cr}$. Percentage-wise, the largest variation occurs with lattice parameter c . As we will show later, Fe_2AlB_2 has a much stronger magneto-elastic effect along the c -axis.

B. Alloys: M and T_C in $\text{Fe}_{2-x}\text{T}_x\text{AlB}_2$

To investigate how magnetic properties change with the $3d$ substitutions on Fe sites in Fe_2AlB_2 , we first consider the chemical effect by neglecting the structure changes caused by substitution. The LMTO-ASA-CPA method is used to calculate the magnetization and the normalized effective exchange (or MFA estimation of T_C in units of pure Fe_2AlB_2) in $\text{Fe}_{2-x}\text{T}_x\text{AlB}_2$ as functions of doping concentration x , with $T = \text{Cr, Mn, Co, and Ni}$. The experimental lattice parameters and atomic positions of Fe_2AlB_2 are used and results are shown in Fig. 5. All dopings decrease the magnetization and T_C in Fe_2AlB_2 . The component-resolved atomic spin moments in those alloys are shown in Fig. 6. For Mn doping, we also consider the AFM configuration and show the absolute values of component-resolved moments in Fig. 6(b).

Assuming the FM configuration, Mn has the slightest effect on the decrease of the magnetization and T_C . The Fe moment barely changes and even increases with a higher Mn content. The decrease of total magnetization is due to the dilution of Fe moments with smaller Mn moments. With a 25% of Mn content, the calculated T_C decreases by 20% while experiments found a larger $\Delta T_C = -30\%$. The assumption of FM configu-

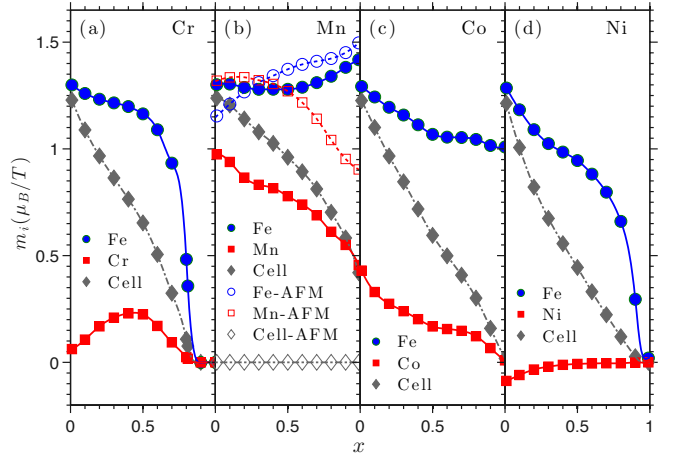


FIG. 6: Component-resolved atomic spin moments m_i as functions of doping content x in $\text{Fe}_{2-x}\text{T}_x\text{AlB}_2$ with $T = \text{Cr}$ (a), Mn (b), Co (c), and Ni (d). Calculations are carried out using the LMTO-ASA-CPA method. Structural changes due to substitution are neglected, and the experimental lattice parameters and atomic positions of Fe_2AlB_2 are used for all calculations. For $T = \text{Mn}$, the AFM configuration is also considered besides the FM configuration, and the absolute magnetic moments of Mn and Fe components are shown.

ration is logical when the Mn amount is small. However, at a higher Mn content, the AFM configuration should also be considered, given that pure Mn_2AlB_2 is more stable with the AFM configuration. Here we calculate the AFM configuration in CPA, by assuming the spin moments of $3d$ atoms (both Fe and Mn components) are parallel within the ab -plane and antiparallel between neighboring planes. As shown in Fig. 6(b), in comparison with the FM configuration, the AFM configuration gives larger Mn moments in the whole doping range and larger Fe moments at $x \geq 0.3$. Within CPA and without considering any lattice relaxation, the AFM configuration becomes more stable than the FM configuration with $x > 0.2$. Thus, the larger decrease of T_C observed in experiments is likely caused by the forming of AFM phases in the samples. By systematically investigating solid solutions $(\text{Fe}_{1-x}\text{Mn}_x)_2\text{AlB}_2$, Chai *et al.* observed both NM and FM Mössbauer spectral components in all Mn-containing samples and attributed them to the clustering of Mn-rich and Fe-rich regions in the samples. Moreover, a spin-glass state has been observed at low temperature in $(\text{Fe}_{1-x}\text{Mn}_x)_2\text{AlB}_2$ with $x = 0.25$ and this phenomenon had been interpreted as the result of geometric frustration caused by the triangular configuration of magnetic atoms⁵. Here, we argue that it could be caused by the competition between the FM and AFM configurations along the c -axis.

Given that a large DOS lies right above the Fermi level in the minority spin channel as shown in Fig. 4(c), it is not surprising that the electron doping, such as Co- or Ni-doping, decreases the magnetic moment on Fe sites. With a small amount of Co doping, the magnetization

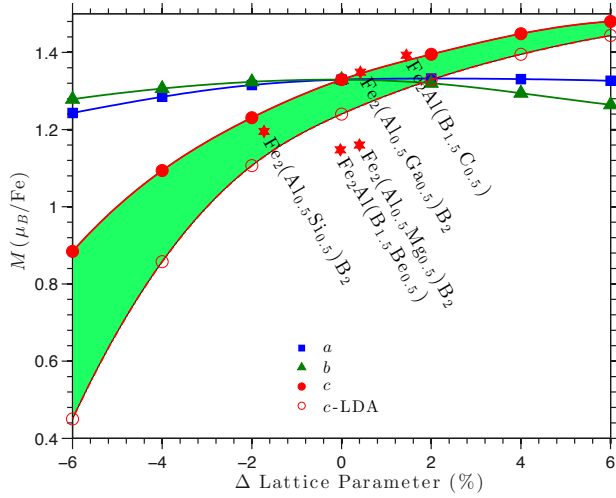


FIG. 7: Magnetization as functions of lattice parameters. Each of the three lattice parameters is varied with the other two being preserved. For the lattice distortion along the c -axis, LDA results are also shown to compare. Magnetizations calculated in fully-relaxed $\text{Fe}_2\text{Al}(\text{B}_{1.5}\text{Be}_{0.5})$, $\text{Fe}_2\text{Al}(\text{B}_{1.5}\text{Co}_{0.5})$, $\text{Fe}_2(\text{Al}_{0.5}\text{Mg}_{0.5})\text{B}_2$, $\text{Fe}_2(\text{Al}_{0.5}\text{Si}_{0.5})\text{B}_2$, and $\text{Fe}_2(\text{Al}_{0.5}\text{Ga}_{0.5})\text{B}_2$ are also shown.

and T_C decrease nearly linearly with Co content. Similar linear dependence of T_C on Co content had been observed in experiments. As shown in Fig. 5(b), calculated ΔT_C values agree very well with experiments¹⁰. With $0.4 < x < 0.9$, Co atoms in $(\text{Fe}_{1-x}\text{Co}_x)_2\text{AlB}_2$ have a small moment of $\sim 0.2 \mu_B/\text{Co}$, similar to the Co moment calculated in the fully relaxed structure of Co_2AlB_2 using FP. However, this small moment becomes unstable in ASA at $x = 1$.

In comparison to Co doping, Ni doping has a similar effect on decreasing T_C and an even stronger effect on suppressing the magnetization in Fe_2AlB_2 . As shown in Fig. 6(d), the atomic Ni moment in $(\text{Fe}_{1-x}\text{Ni}_x)_2\text{AlB}_2$ is small and coupled antiparallel with the Fe sublattice at small x , and negligible for $x > 0.3$.

In $(\text{Fe}_{1-x}\text{Cr}_x)_2\text{AlB}_2$, the Cr moment is small and parallel to the Fe sublattice. The maximum Cr moment of $0.2 \mu_B/\text{Cr}$ occurs at $x = 0.5$. Like Mn, Cr doping has a smaller effect on decreasing the T_C than Co and Ni. However, Cr doping is not likely to promote the FM \rightarrow AFM transition, which may compromise the MCE as in the case of Mn doping⁵. Thus, it is worthwhile to investigate Cr doping, which may provide a useful approach to tune the T_C and MCE in Fe_2AlB_2 .

C. Effect of lattice distortion

Besides the chemical effect, the volume change caused by substitution may also affect the magnetic properties. As shown in Table III, the lattice parameters in $T_2\text{AlB}_2$ vary with the element T and spin configuration. To have a rough idea on the magneto-elastic effect in Fe_2AlB_2 ,

we calculate the magnetization dependence on the three lattice parameters, respectively. Starting from the fully relaxed structure, each of the three parameters is varied while the other two are kept constant. Interestingly, as shown in Fig. 7, the magneto-elastic effect in Fe_2AlB_2 is very anisotropic. The magnetization has a much stronger dependence on the lattice parameter c than on a or b . With $\Delta c = -6\%$, magnetization decreases by 35% within GGA and 60% within LDA.

This anisotropic magneto-elastic effect can be understood by investigating the electronic structure features near E_F and their changes caused by the lattice distortion. The bandstructures of Fe_2AlB_2 in the majority and minority channels are shown in Fig. 8(a) and (c), respectively. The PDOS projected on Fe-3d states and in two spin channels are shown in Fig. 8(b). Particularly of note is a narrow band right above E_F in the minority spin channel. This band is found to consist almost entirely of Fe- d_{xy} and $d_{x^2-y^2}$ ($m = \pm 2$) orbitals. The variations of PDOS with lattice parameter c are shown in Fig. 9. The decreasing of parameter c increases the bandwidth of d_{3z^2-1} state, which has a relatively large density right below E_F in the majority spin channel. These antibonding d_{3z^2-1} states, located between -1.3 eV and E_F , shift up toward E_F and become less occupied. Correspondingly, the aforementioned peak of $m = \pm 2$ states, located right above E_F in the minority spin channel, become more occupied. As a result, the magnetization decreases. The Fe- d_{yz} and d_{xz} states ($m = \pm 1$) have small DOS around E_F and contribute less to this magnetization change. With the further decrease of c and then magnetization, the spin splitting becomes smaller, which quickly accelerates the decrease of magnetization as Δc approaches to -6% .

D. Dopings on B and Al sites

We also consider the substitutions of B and Al atoms with their neighboring elements in the periodic table: Be and C atoms on the B site, and Mg, Si and Ga atoms on the Al site. The stabilities of those dopings are not well understood, and a careful and complete future investigation is desired. Here we focus on the possible effects of those dopings on the magnetization. Using various configurations of a 10-atom Fe_2AlB_2 unit cell, we substitute one B or Al atom with a dopant atom and fully relax the structures for $\text{Fe}_2\text{Al}(\text{B}_{0.75}\text{Z}_{0.25})_2$ with $Z = \text{Be}$ and C , and $\text{Fe}_2(\text{Al}_{0.5}\text{Z}_{0.5})\text{B}_2$ with $Z = \text{Mg}$, Si , and Ga . Their magnetizations calculated with the corresponding lowest energy configuration are denoted in Fig. 7 with respect to the change of lattice parameter c .

Only C doping on the B sites noticeably increases the lattice parameter c and magnetization, while most of other substitutions decrease the magnetization in Fe_2AlB_2 . Both chemical effect and the magneto-elastic effect contribute to the magnetization enhancement. The lattice parameter c increases by 1.6% and magnetization increases to $1.4 \mu_B/\text{Fe}$ in $\text{Fe}_2\text{Al}(\text{B}_{0.75}\text{C}_{0.25})_2$. Unlike B,

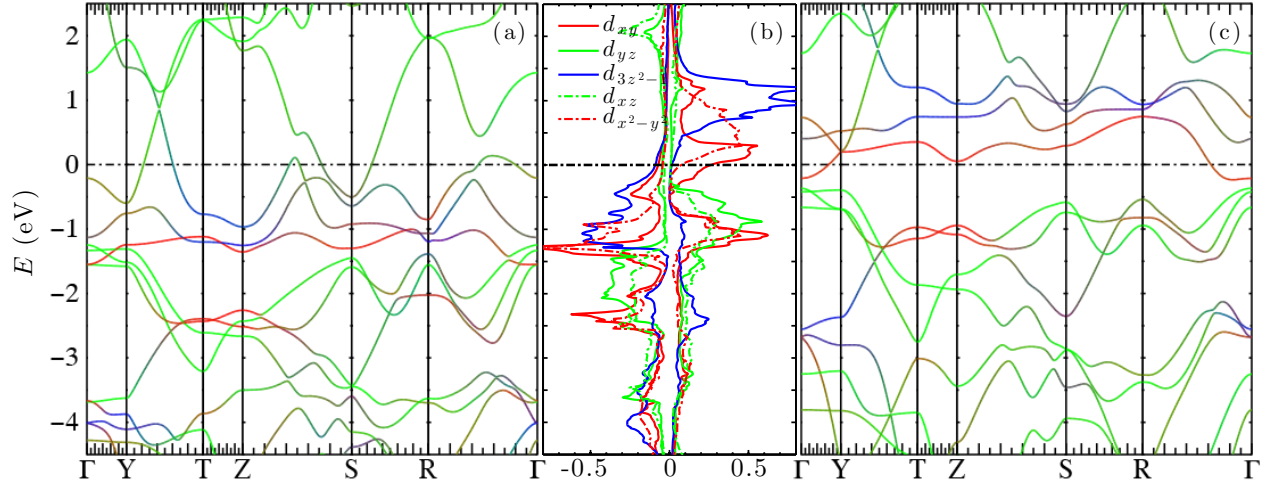


FIG. 8: Bandstructure of Fe_2AlB_2 in (a) majority and (c) minority spin channels. Bands are with color weights, with blue identifying the $\text{Fe-}d_{3z^2-1}$ states, red the $\text{Fe-}(d_{xy}, d_{x^2-y^2})$ states, and green everything else. PDOS (states $(\text{eV spin atom})^{-1}$) projected on Fe-3d states in two spin channels are shown in panel (b). The left and right portion of the panel (b) shows PDOS in the majority and minority spin channels, respectively. The horizontal dashed lines in all three panels indicate the Fermi level.

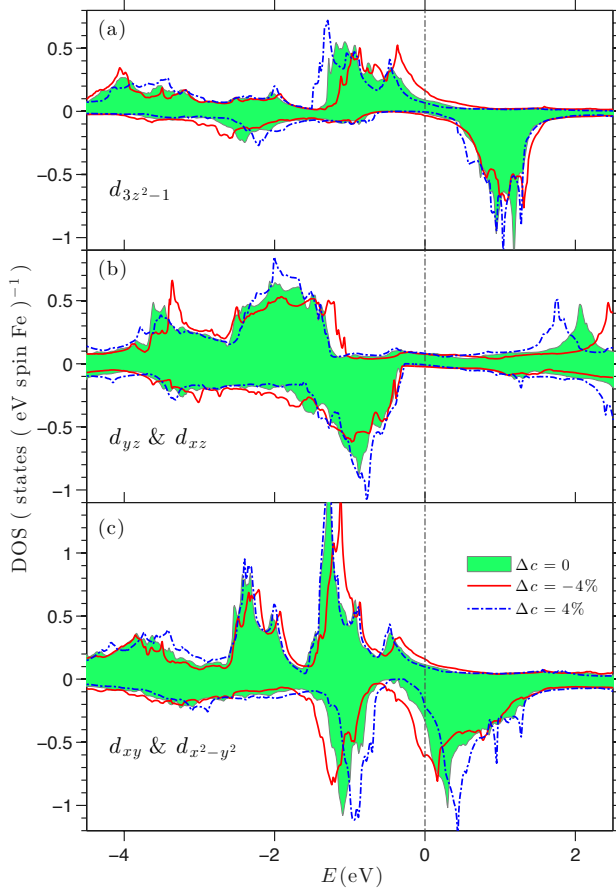


FIG. 9: The partial density of states projected on Fe 3d with different c parameters. The scalar-relativistic partial density of states projected on the 3d states of Fe sites in Fe_2AlB_2 with different lattice parameters.

C has small moment parallel to the Fe sublattice. Moreover, C doping increases the moments of neighboring Fe atoms by about $0.1 \mu_B/\text{Fe}$.

For Si doping, without considering lattice relaxation, magnetization decreases by 6 % in $\text{Fe}_2(\text{Si}_{0.5}\text{Al}_{0.5})\text{B}$. The relaxation decreases the lattice c by 1.7 %, and further decreases the magnetization by another 6 %. Thus, both chemical effect and the magneto-elastic effect contribute to the decreasing of the magnetization. Be and Mg dopings have stronger effects on decreasing the magnetization. With Be and Mg dopings, the DOS peak right below E_F in the majority spin shifts toward E_F and becomes less occupied and the magnetization decreases to about $1.1 \mu_B/\text{Fe}$. Ga doping has very small effect on the lattice parameters and the magnetization of Fe_2AlB_2 .

IV. CONCLUSION

Using density functional theory, we investigated the intrinsic magnetic properties in $T_2\text{AlB}_2$ and their alloys. For Fe_2AlB_2 , the a -axis is the easiest axis while the c -axis is the hardest axis. For Mn_2AlB_2 , we predict that the magnetic ground state is a AFM configuration, with the neighboring Mn layers being antiferromagnetically coupled along the c -axis. Co_2AlB_2 is weakly ferromagnetic while Cr_2AlB_2 and Ni_2AlB_2 are non-magnetic. All 3d substitutions decrease the magnetization and Curie temperature of Fe_2AlB_2 in the sequence of $\text{Mn} < \text{Cr} < \text{Co} < \text{Ni}$. However, Mn promotes antiferromagnetism when its doping content is larger than 20 %. The competition between the two configurations at critical compositions may be responsible to the spin-glass states observed in experiments. Unlike Mn, Cr doping is not likely to promote

the AFM configuration, and may be useful in tuning T_C in Fe_2AlB_2 . The effect of strain and alloying on magnetic properties are also studied. A very strong anisotropic magneto-elastic effect is found. Magnetization in Fe_2AlB_2 becomes fragile and quickly decreases with the lattice parameter c , while it barely changes with a and b . This effect is explained by the displacement of anti-bonding d_{3z^2-1} states right below the Fermi level in the majority spin channel, and the filling of unoccupied d_{xy} and $d_{x^2-y^2}$ states which have a sharp peak right above the Fermi level in the minority spin channel. Doping or applying pressure to modify the inter-layer distance along the c -axis may provide an effective way to

tune the magnetic properties in Fe_2AlB_2 .

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- ¹ V. K. Pecharsky and K. A. Gschneidner, Jr., Phys. Rev. Lett. **78**, 4494 (1997).
 - ² X. Tan, P. Chai, C. M. Thompson, and M. Shatruk, Journal of the American Chemical Society **135**, 9553 (2013).
 - ³ L. Lewis, R. Barua, and B. Lejeune, Journal of Alloys and Compounds **650**, 482 (2015).
 - ⁴ Q. Du, G. Chen, W. Yang, Z. Song, M. Hua, H. Du, C. Wang, S. Liu, J. Han, Y. Zhang, et al., Japanese Journal of Applied Physics **54**, 053003 (2015).
 - ⁵ Q. Du, G. Chen, W. Yang, J. Wei, M. Hua, H. Du, C. Wang, S. Liu, J. Han, Y. Zhang, et al., Journal of Physics D: Applied Physics **48**, 335001 (2015).
 - ⁶ H. Becher, K. Krogmann, and E. Peisker, Z. Anorg. Allg. Chem. **344** (1966).
 - ⁷ N. Chaban and Y. Kuz'ma, Inorg. Mater. **9**, 1696 (1973).
 - ⁸ K. Kadas, D. Iusan, J. Hellsvik, J. Cedervall, P. Berastegui, M. Sahlberg, U. Jansson, and O. Eriksson, ArXiv e-prints (2016), 1611.07190.
 - ⁹ P. Chai, S. A. Stoian, X. Tan, P. A. Dube, and M. Shatruk, Journal of Solid State Chemistry **224**, 52 (2015), chemistry of two-dimensional materials.
 - ¹⁰ S. Hirt, F. Yuan, Y. Mozharivskyj, and H. Hillebrecht, Inorganic Chemistry **55**, 9677 (2016).
 - ¹¹ M. ElMassalami, D. da S. Oliveira, and H. Takeya, Journal of Magnetism and Magnetic Materials **323**, 2133 (2011).
 - ¹² Y. Cheng, Z. Lv, X. Chen, and L. Cai, Computational Materials Science **92**, 253 (2014).
 - ¹³ J. Cedervall, M. S. Andersson, T. Sarkar, E. K. Delczeg-Czirjak, L. Bergqvist, T. C. Hansen, P. Beran, P. Nordblad, and M. Sahlberg, Journal of Alloys and Compounds **664**, 784 (2016).
 - ¹⁴ W. Jeitschko, Acta Crystallographica Section B **25**, 163 (1969).
 - ¹⁵ O. Andersen, Phys. Rev. B **12**, 3060 (1975).
 - ¹⁶ M. Methfessel, M. van Schilfgaarde, and R. Casali, in *Lecture Notes in Physics*, edited by H. Dreyse (Springer-Verlag, Berlin, 2000), vol. 535.
 - ¹⁷ J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. **77**, 3865 (1996).
 - ¹⁸ U. von Barth and L. Hedin, Journal of Physics C: Solid State Physics **5**, 1629 (1972).
 - ¹⁹ A. Mackintosh and O. Andersen, *Electrons at the Fermi Surface* (Cambridge University Press, Cambridge, England, 1980).
 - ²⁰ L. Ke, M. van Schilfgaarde, and V. Antropov, Phys. Rev. B **86**, 020402 (2012), **Rapid communication**.
 - ²¹ L. Ke, K. Belashchenko, M. van Schilfgaarde, T. Kotani, and V. Antropov, Phys. Rev. B **88**, 024404 (2013).
 - ²² L. Ke and D. D. Johnson, Phys. Rev. B **94**, 024423 (2016).
 - ²³ G. Kresse and J. Hafner, Phys. Rev. B **47**, 558 (1993).
 - ²⁴ G. Kresse and J. Furthmüller, Phys. Rev. B **54**, 11169 (1996).
 - ²⁵ G. Kresse and D. Joubert, Phys. Rev. B **59**, 1758 (1999).
 - ²⁶ K. Cenzual, L. M. Gelato, M. Penzo, and E. Parthé, Acta Crystallographica Section B **47**, 433 (1991).
 - ²⁷ V. Antropov, L. Ke, and D. Åberg, Solid State Communications **194**, 35 (2014).
 - ²⁸ L. Ke and M. van Schilfgaarde, Phys. Rev. B **92**, 014423 (2015).
 - ²⁹ L. Ke, D. A. Kukusta, and D. D. Johnson, Phys. Rev. B **94**, 144429 (2016).
 - ³⁰ P. Bruno, Phys. Rev. B **39**, 865 (1989).
 - ³¹ A. V. D. Geest and A. Kolmogorov, Calphad **46**, 184 (2014).
 - ³² J. Lu, S. Kota, M. W. Barsoum, and L. Hultman, Materials Research Letters **0**, 1 (2016).