

CHCRUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

Improved pseudopotential transferability for magnetic and electronic properties of binary manganese oxides from DFT+U+J calculations

Jin Soo Lim, Diomedes Saldana-Greco, and Andrew M. Rappe Phys. Rev. B **94**, 165151 — Published 21 October 2016 DOI: 10.1103/PhysRevB.94.165151

1	Improved Pseudopotential Transferability for Magnetic and
2	Electronic Properties of Binary Manganese Oxides from
3	$\mathrm{DFT}{+}U{+}J$ Calculations
4	Jin Soo Lim, Diomedes Saldana-Greco, and Andrew M. Rappe
5	The Makineni Theoretical Laboratories,
6	Department of Chemistry,
7	University of Pennsylvania,
8	Philadelphia, PA 19104-6323, USA
9	
10	(Dated: September 6, 2016)
11	Abstract
12	We employ the fully anisotropic $\mathrm{DFT}+U+J$ approach with the PBEsol functional to investigate
13	ground-state magnetic and electronic properties of bulk binary manganese oxides: MnO, Mn ₃ O ₄ ,
14	α -Mn ₂ O ₃ , and β -MnO ₂ , in order of increasing Mn valence. The computed crystal structures,
15	noncollinear magnetic ground states, and corresponding electronic structures are in good agreement
16	with the experimental data and hybrid functional calculations available in the literature. We take
17	into account the nonlinear core-valence interaction in our Mn pseudopotential designed by ourselves,
18	as it has been proven to be important for transition metal systems. Although the Hubbard U term
19	is capable by itself of opening a band gap, the explicitly defined exchange parameter J plays an
20	important role in improving the detailed electronic and noncollinear magnetic structure profiles.
21	Appropriate band gaps are obtained with U values smaller than those used in previously reported
22	calculations. Our results suggest that pseudopotential design together with $\mathrm{DFT}+U+J$ enables
23	the acquisition of accurate properties of complex magnetic systems using a non-hybrid density
24	functional.

25 I. INTRODUCTION

Due to the low cost, low toxicity, and high chemical stability, binary manganese oxides 26 have a wide range of applications, such as catalysis [1], batteries [2, 3], functional magnetic 27 and optical materials [4, 5], and electrocatalytic biosensors [6]. From a theoretical perspec-28 tive, manganese oxides attract great interest due to their strong electron correlations that 29 give rise to complex physical phenomena, including colossal magnetoresistance, charge and 30 orbital ordering, and noncollinear magnetism. Although manganese oxides have been stud-31 ied extensively, modeling their ground-state magnetic and electronic properties for different 32 oxidation states within density functional theory (DFT) poses fundamental challenges, due 33 to the inherent limitations in the approximations of the exchange-correlation functional. 34 Improvements to these simulations without resorting to higher-level methods and incurring 35 significant computational costs are therefore desirable for large-scale studies of systems in-36 volving strongly correlated materials. Here, we construct a pseudopotential that accounts for 37 nonlinear core-valence interactions, and we apply the fully anisotropic DFT+U+J method. 38 This approach accurately describes the magnetic and electronic properties of bulk man-39 ganese oxides with a variety of atomic and magnetic structures and different oxidation 40 states, namely MnO, Mn₃O₄, α -Mn₂O₃, and β -MnO₂. 41

MnO exists in a B1 rock salt structure with Mn^{2+} oxidation state [Fig. 1(a)]. It undergoes 42 paramagnetic to type-II antiferromagnetic (AFM-II) transition at $T_{\rm N}$ = 116 K, accompa-43 nied by cubic $(Fm\overline{3}m)$ to rhombohedral structural transition [7]. The ground-state magnetic 44 structure AFM-II consists of ferromagnetically aligned planes that are successively antipar-45 allel along [111] direction. Consequent magnetostriction causes rhombohedral contraction 46 along the [111] below $T_{\rm N}$, tilting the crystal axes 0.62° from the cubic directions [8, 9]. 47 MnO is a charge-transfer insulator with a large band gap of 3.6-4.2 eV measured experi-48 mentally [10, 11]. The electronic structure has been studied extensively via first-principles 49 methods, including Hartree-Fock [12–17], LDA [14, 18–20], GGA+U [15–18, 20–23], GW 50 method [24], and hybrid functionals [15-17, 21, 25]. 51

⁵² Mn₃O₄ exists in a spinel structure (AB_2O_4) , with Mn²⁺ (Mn_A) occupying the tetrahe-⁵³ dral sites (numbered Mn1-2) and Mn³⁺ (Mn_B) occupying the octahedral sites (numbered ⁵⁴ Mn3-6) [Fig. 1(b)]. Edge-sharing Mn_BO₆ octahedra form chains along **a** and **b**. It under-⁵⁵ goes cubic ($Fd\overline{3}m$) to tetragonal ($I4_1/amd$) structural transition at 1443 K [26] due to the

Jahn-Teller effect at Mn_B sites. Strong lattice frustration leads to a rich magnetostructural 56 phase diagram at low temperatures. At $T_{\rm N} = 42$ K, the material undergoes paramagnetic to 57 ferrimagnetic transition [27], adopting a triangular Yafet-Kittel [28] ferrimagnetic (YK-FiM) 58 state [29–31]. In the YK-FiM structure, Mn_A spins are ferromagnetically aligned along b, 59 and Mn_B spins are along **-b** canted toward $\pm c$ direction [32, 33]. Only a few optical mea-60 surements have been performed, reporting band gaps of 1.91 eV for bulk polycrystalline [34], 61 2.51 eV for thin film [35], and 2.07 eV for nanoparticles [36]. The electronic structure of the 62 bulk Mn_3O_4 spinel has not been studied as extensively as MnO. Computational methods 63 including Hartree-Fock [37], GGA+U and hybrid functionals [21, 34] have been employed 64 to simulate the electronic structure using only idealized collinear magnetic structures. 65

 α -Mn₂O₃ has multiple technological applications, such as synthesis substrate for mangan-66 ite oxide perovskite compounds, starting material for lithium ion battery cathode material 67 $LiMnO_2$ [38], and also an environmentally friendly catalyst for water purification [39] and 68 combustion [40]. The material exists in a bixbyite structure with Mn^{3+} oxidation state [Fig. 69 1(c)]. For O chains along **a** or **b**, there is one O atom missing per four sites, such that 70 each O atom forms a tetrahedral linkage to surrounding Mn atoms. It undergoes cubic 71 (Ia3) to orthorhombic (Pcab) structural transition at T = 308 K [41] due to the Jahn-Teller 72 effect at Mn^{3+} sites, causing 0.8% distortion from the cubic structure. The paramagnetic to 73 noncollinear-antiferromagnetic (NC-AFM) transition occurs at $T_{\rm N1} = 80-90$ K and another 74 AFM transition at $T_{N2} = 25$ K [42–46]. NC magnetic configuration was first proposed 75 assuming the cubic structure [45] but was later found to be incompatible with neutron 76 powder diffraction data [47]. An alternative collinear AFM structure with four magnetic 77 sublattices was proposed by Regulski et al. [47] using the cubic lattice (indicated as AFM1 78 in this study). However, Cockayne et al. [48] found that magnetic sublattice III of the 79 AFM1 structure is incompatible with the Pcab space group, and thereby proposed another 80 collinear AFM structure using the orthorhombic lattice (indicated as AFM2 in this study), 81 determined independently from both neutron powder diffraction and DFT+U study. Spin 82 canting of 12-34° was found to further improve the fitting of their diffraction data. Only 83 one study reported the optical band gap of 1.2 eV for nanostructures [49]. The electronic 84 structure of the bulk material has not been investigated extensively, other than two GGA+U85 studies [21, 48] reporting different results for the magnetic ground state. 86

 β -MnO₂ is widely used in Li-ion batteries [3, 50–60], Li-O₂ batteries [61, 62], super-

capacitors [63–68], adsorbents [69], and catalysts [70, 71]. β -MnO₂ exists in a tetragonal 88 $(P4_2/mnm)$ rutile structure with Mn⁴⁺ oxidation state [Fig. 1(d)]. It undergoes param-89 agnetic to screw-type spiral magnetic transition at $T_{\rm N} = 92$ K [72], where the spins lie on 90 the *ab*-plane and rotate by 129° in the next adjacent layer along the *c*-axis for a period 91 of 7 unit cells. Transport measurements suggested a very small band gap at low tempera-92 tures [72], with one study reporting a value of 0.26 eV for epitaxially grown thin films [73]. 93 Computational studies of the bulk material based on idealized MnF_2 -type collinear AFM 94 structure were performed using Hartree-Fock [74], GGA+U [21, 75, 76], and hybrid func-95 tionals [21, 76]. The spiral noncollinear magnetic structure [77] has only been simulated 96 using a tight-binding method [78] and dynamical mean-field theory [73]. 97

These manganese oxides exhibit complex magnetic and electronic properties, making 98 them a challenging set to study. Extensive reports on these systems have shown that rig-99 orous theoretical methods are required to describe their properties adequately [21]. Due 100 to the exchange-correlation functional limitations, advanced methods are needed to cor-101 rectly describe the electronic structure of these strongly correlated magnetic materials. The 102 strong electronic correlation experienced by the localized d electrons leads to unphysical 103 self-interaction of an electron with the potential it generates. Self-interaction artificially 104 raises the energy of the on-site single-particle energies in the Kohn-Sham equations, thereby 105 delocalizing the localized electronic states and leading to inaccuracies in the electronic band 106 structure. The theoretical methods developed to overcome these inherent limitations in-107 clude DFT+ U_{eff} [79], DFT+U+J [80–82], and hybrid functionals [83]. In DFT+ U_{eff} , where 108 $U_{\rm eff} = U - J$, an isotropic screened on-site Coulomb interaction is added: 109

$$E_{\text{Hub}} = \sum_{I,\sigma} \frac{U_{\text{eff}}^{I}}{2} \text{Tr}[\boldsymbol{n}^{I\sigma} (\boldsymbol{1} - \boldsymbol{n}^{I\sigma})].$$
(1)

Here, E_{Hub} is the Hubbard correction to the standard approximate DFT energy functional, I is the atomic site index, σ is the spin index, and n is the occupation matrix. In contrast, Hubbard U and J are defined distinctly in DFT+U+J, leading to a fully anisotropic treatment of the Coulomb and exchange matrices accounting for the full orbital dependence:

$$E_{\text{Hub}} = \sum_{I,\sigma} \frac{U^{I} - J^{I}}{2} \text{Tr}[\boldsymbol{n}^{I\sigma}(\boldsymbol{1} - \boldsymbol{n}^{I\sigma})] + \sum_{I,\sigma} \frac{J^{I}}{2} (\text{Tr}[\boldsymbol{n}^{I\sigma}\boldsymbol{n}^{I-\sigma}] - 2\delta_{\sigma\sigma_{\min}} \text{Tr}[\boldsymbol{n}^{I\sigma}]), \qquad (2)$$

where σ_{\min} denotes the minority spin. Compared with Eq. (1), the extra positive J term 114 in Eq. (2) discourages interactions between electrons of antialigned spins on the same site, 115 thereby encouraging magnetic ordering [82]. This fully anisotropic method has proven to 116 describe strongly correlated magnetic systems more accurately [76, 84]. The appropriate U117 value can enhance or even open up a band gap, and J can determine the noncollinear mag-118 netic ground state, thus refining the electronic structure profile of the system. Additionally, 119 hybrid functionals have been shown to overcome the deficiencies in describing these mate-120 rials (at higher computational cost) by incorporating a fraction of the exact Hartree-Fock 121 exchange into the exchange-correlation functional. Despite the improvements in results ob-122 tained from hybrid functionals, the DFT+U+J approach is nonetheless a computationally 123 much cheaper alternative that is desirable in electronic structure studies involving large-scale 124 systems, such as surfaces, supercells, interfaces, and defects. 125

In this work, we investigate the noncollinear magnetic ground states and the corresponding electronic structures of MnO, Mn_3O_4 , α - Mn_2O_3 , and β - MnO_2 using the DFT+U+Jmethod. We show that the ground-state lattice, magnetism, and electronic structure profile can be obtained with accuracy nearing that of literature hybrid functional calculations, through careful pseudopotential design and selection of fully anisotropic U and J values.

131 II. METHOD AND COMPUTATIONAL DETAILS

The magnetic and electronic structures of the manganese oxides are calculated with first-132 principles DFT using the PBEsol [85] parametrization of the generalized gradient approx-133 imation with on-site Coloumb repulsion and exchange parameters U and J, treated sep-134 arately and explicitly defined within the rotationally invariant, fully anisotropic scheme 135 (DFT+U+J) [81, 82], using the atomic orbital projection scheme [86] as implemented in 136 the QUANTUM ESPRESSO [87] package. It has been demonstrated that the fully anisotropic 137 J parameter plays an important role in describing strongly correlated noncollinear antiferro-138 magnetic systems [76]. We determine the optimal Hubbard U and J values by first testing 139 a range of values reported in the literature and changing the values as necessary, each time 140 observing the effect on the ground-state magnetic and electronic structures and properties. 141 We also employed the linear response method by Cococcioni *et al.* [88] to determine a range 142 of U values; however, the values obtained by this method were too high (> 7 eV) for ac-143

curate electronic structure profiles. The calculations account for spin-polarized electronic 144 densities by treating the Mn magnetic moments as noncollinear for all systems. All atoms 145 are represented by norm-conserving, optimized [89], designed nonlocal [90] pseudopotentials 146 generated with the OPIUM package [91], treating the 2s and 2p of O and 3s, 3p, 3d, 4s, and 147 4p of Mn as valence states. In addition to the treatment of semicore states as valence by 148 this Mn pseudopotential, nonlinear core-valence interaction via the partial core correction 149 scheme [92–94] is incorporated to account for the non-negligible overlap between the core 150 and the valence states. All calculations are run with a 70 Ry plane-wave energy cutoff to 151 ensure accuracy for small relative energies among different magnetic configurations. The 152 Brillouin zone is sampled using Monkhorst-Pack [95] k-point meshes of dimensions $6 \times 6 \times 6$, 153 $8 \times 8 \times 8$, $4 \times 4 \times 4$, and $6 \times 6 \times 6$ for MnO, Mn₃O₄, α -Mn₂O₃, and β -MnO₂, respectively. 154 A $12 \times 12 \times 12$ k-point grid is used for post-processing the electronic structure calculations 155 for all four systems. All relaxations starting from the experimental crystal structures are 156 performed without U and J, as relaxation with U leads to overestimated lattices and bond 157 lengths [21]. The magnetic and electronic structures of the optimized crystal structures are 158 then refined with U and J. 159

160 III. RESULTS AND DISCUSSION

Throughout this section, our results on the ground-state structural, magnetic, and electronic properties of the four manganese oxide systems are discussed in detail with respect to the computational parameters employed and in comparison with previously published data. Table I provides an overview of the results for each manganese oxide system from both experimental and computational studies, including DFT+U(+J) and hybrid functional studies from the literature, in comparison to our PBEsol+U+J study.

167 A. MnO

¹⁶⁸ Crystal structure relaxations with various imposed magnetic orders, antiferromagnetic ¹⁶⁹ (AFM-II, A-AFM, C-AFM) and ferromagnetic (FM), reveal the AFM-II structure as the ¹⁷⁰ magnetic ground state (Table II). The lattice constant, a = 4.40 Å, is 0.68 % smaller than the ¹⁷¹ literature value of 4.43 Å [96]. PBEsol therefore yields structural properties that are in good



FIG. 1. Optimized crystal structures of the manganese oxides in their magnetic ground states: (a) AFM-II MnO, (b) YK-FiM Mn₃O₄, (c) NC-AFM2 α -Mn₂O₃, and (d) spiral β -MnO₂. Spin-up and spin-down Mn are colored in purple and gold for MnO, respectively, with red O atoms. The magnetic ground states of Mn₃O₄, α -Mn₂O₃, and β -MnO₂ are noncollinear; these spin structures are further discussed and illustrated in the Results and Discussion section. Bonds are not shown in (c) for clearer observation of the O chains along **a** or **b**.

agreement with the experimental values. Other AFM orders result in c < a, inconsistent with experimental data, while FM order produces a severely contracted lattice structure.

Electronic structure calculation without U and J shows an underestimated band gap of 1.16 eV, consistent with previous GGA studies [15–18, 20, 23]. Applying U = 4 eV increases the band gap to 3.32 eV; this value is higher compared to other DFT+ U_{eff} . For example, Franchini *et al.* [16] obtained a band gap of 2.03 eV with U = 6 eV; they were able to increase the band gap to 3 eV only by increasing the U value up to 15 eV. Since our calculation requires much lower U value to achieve a more reasonable band gap, it suggests

TABLE I. Ground-state magnetism, lattice, magnetic moment per Mn, and band gap for each manganese oxide system reported by (1) experimental studies, (2) literature DFT+U(+J), and (3) literature hybrid functional studies, in direct comparison to (4) our PBEsol+U+J results. Our lattice constants presented here are values optimized without U and J.

System	Method	Magnetic state	Lattice constants $(Å)$	Mag. mom. $(\mu_{\rm B})$	E_{g} (eV)
MnO	(1) Experimental	AFM-II	a=4.43 [96]	4.58 [97]	3.6-4.2 [10, 11]
	(2) $PBE+U$, $U=4 eV$ [23]	AFM-II	a = 4.489	4.60	2.34
	(3) PBE0 [16]	AFM-II	a = 4.40	4.52	4.02
	(4) PBEsol+ $U+J$, U=4 eV, $J=1.2 eV$	AFM-II	a=4.40	4.56	2.81
$\mathrm{Mn_3O_4}$	(1) Experimental	YK-FiM	a=5.71, c=9.35 [33] $V_0=155.73\text{\AA}^3$ [98]	4.34, 3.64, 3.25 [33]	1.91 [34]
	(2) $PBE+U$, $U=5 eV [34]$	FiM6	N/A	4.6, 3.9	1.46
	(3) PBE0 [21]	FiM3	$V_0 = 157.42 \text{\AA}^3$	3.69 - 4.50	2.4
	(4) PBEsol+ $U+J$, U=4 eV, $J=1.2 eV$	YK-FiM	a=5.76, c=9.35 $V_0=155.50 \text{\AA}^3$	4.49, 3.74, 3.69	1.01
$\alpha\text{-}\mathrm{Mn}_2\mathrm{O}_3$	(1) Experimental	NC-AFM2 [48]	a=9.407, b=9.447, c=9.366, $V_0=834.48 \text{\AA}^3$ [41]	3.3-4.0 [47] 2.6-3.5 [48]	1.2 [49]
	(2) $PBEsol+U+J$, U=2.8 eV, J=1.2 eV [48]	AFM2	a=9.402, b=9.444, c=9.367	3.6	0.6
	(3) HSE [21]	\mathbf{FM}	$V_0 = 845.83 \text{\AA}^3$	3.81-3.84	0.1
	(4) PBEsol+ $U+J$, U=2.8 eV, J=1.2 eV	NC-AFM2	a=9.382, b=9.444, c=9.376, $V_0=830.71 \text{\AA}^3$	4.09, 2.91, 3.68, 3.83, 3.69	0.081
$\beta\text{-}\mathrm{MnO}_2$	(1) Experimental	Spiral	a=4.396, c=2.871 $V_0=55.48\text{\AA}^3$ [99]	2.35 [100]	0.26 [73]
	(2) $GGA+U+J$, U=6.7 eV, J=1.2 eV [76]	AFM	a=4.45, c=2.936	2.96	0.8
	(3) PBE0 [21]	AFM	$V_0 = 55.06 \text{\AA}^3$	2.89	1.5
	(4) PBEsol+ $U+J$, U=2.8 eV, J=1.2 eV	Spiral	a=4.402, c=2.880 $V_0=55.80 \text{\AA}^3$	2.63	0.25

¹⁸⁰ enhanced performance of our designed pseudopotential.

Despite moving the band gap closer to the experimental value, the electronic structure 181 profile is compromised by U when compared with those reported by hybrid functional cal-182 culations [15-17, 21, 25], as U shifts the energies of the valence and conduction bands 183 further apart. Previous GW studies reported that large values of U reorder the bands when 184 compared to the GW quasiparticle band structures [24, 101]. We find that applying an 185 anisotropic J = 1.2 eV enhances the profile significantly, but it reduces the band gap to 186 2.81 eV [Fig. 2(a)]. Explicitly defined Hubbard J takes into account the full symmetry of 187 d-d interactions, thereby providing a better description of orbital spin polarizations [76]. In 188 our orbital-projected density of states (DOS), the highest-energy valence band shows strong 189 mixing of O 2p and Mn e_g states enhanced by U, whereas the lowest-energy conduction band 190 primarily consists of Mn t_{2g} states. Together with the calculated magnetic moment of 4.56 191



FIG. 2. Projected density of states of the manganese oxides computed with PBEsol+U+J: (a) MnO shows a collinear AFM-II ground state with a bandgap of 2.81 eV; the valence bands are governed by the overlap between O 2p and Mn e_g orbitals, and the conduction bands by Mn t_{2g} orbitals. The rest of the systems have noncollinear magnetic ground states: (b) Mn₃O₄ shows a YK-FiM ground state with a band gap of 1.01 eV, where Mn_A and Mn_B refer to the tetrahedral and the octahedral sites, respectively; (c) α -Mn₂O₃ shows a NC-AFM2 ground state with a small band gap of 0.081 eV; (d) β -MnO₂ shows a spiral magnetic ground state with a band gap of 0.25 eV. The band gap regions are indicated with a light blue color.

 $\mu_{\rm B}$, in good agreement with the experimental value of 4.58 $\mu_{\rm B}$ [97], our electronic structure predicts MnO as a high-spin insulator of intermediate Mott-Hubbard/charge-transfer character, consistent with results from previous high-level computational studies [12, 15–17, 25].

TABLE II. Relaxed lattice constants and relative energies per formula unit of MnO with various imposed magnetic orders. Experimental lattice constant is a = 4.43 Å [96] with the AFM-II ground state.

Magnetism	Lattice constants (Å)	Relative $E \text{ (meV/f.u.)}$	
AFM-II	a = 4.40	0	
C-AFM	a = 4.41, c = 4.38	17	
A-AFM	a = 4.43, c = 4.38	80	
FM	a = 4.29	413	

195 **B.** Mn_3O_4

We determine the ground-state magnetic structure of Mn_3O_4 to be the experimentally 196 reported YK-FiM structure. This noncollinear structure is 152 meV lower in energy than the 197 lowest collinear structure. The YK-FiM structure has not been computed before; therefore, 198 we start by comparing our results with previous calculations on idealized collinear structures. 199 We compute the six idealized collinear FiM configurations (FiM1-6), in addition to the FM 200 order, as first specified in the Hartree-Fock study of Chartier et al. [37]. Six Mn atoms of the 201 unit cell are numbered as shown in Fig. 1(b), where two Mn_A are Mn1-2, two Mn_B along 202 **b** are Mn3-4, and two Mn_B along **a** are Mn5-6. Crystal structure relaxation with the six 203 imposed FiM orders shows FiM6 $(\uparrow\downarrow\uparrow\downarrow\uparrow\downarrow)$ as the lowest-energy structure when the spins are 204 held collinear. In FiM6 order, all spins are antiferromagnetic to all their neighbors, which is 205 consistent with the experimental measurements reporting the exchange interaction constants 206 to be antiferromagnetic [31, 102, 103]. However, the net magnetic moment is zero in FiM6, 207 which is inconsistent with the experimentally observed net magnetic moment of 1.84 $\mu_{\rm B}$ 208 per formula unit along b [30, 32, 33, 104]. The idealized collinear FiM configuration most 209 consistent with the experimentally observed YK-FiM structure would be FiM4 ($\uparrow\uparrow\downarrow\downarrow\uparrow\downarrow$), 210 where Mn_A spins (Mn1-2) are ferromagnetically aligned, and all Mn_B spins (Mn3-6) are 211 antiferromagnetically aligned. We find that applying U = 4 eV and J = 1.2 eV to the 212 relaxed structures lowers the energy of the FiM4 structure, making it the lowest-energy 213 collinear magnetic state (Table III). However, once the spins are allowed to be noncollinear, 214 the YK-FiM structure is the most energetically favorable. The lattice constants obtained 215



FIG. 3. Computed YK-FiM structure of Mn_3O_4 : (a) Magnetic moments are colored in green within the Mn tetrahedral (purple) and octahedral (gold) cages; (b) Side view showing the top and bottom bilayers used to illustrate the noncollinear spin pattern; (c) Top bilayer exhibiting a sinusoidal Mn_B spin pattern (along the green dashed lines), with Mn_A spin alignment along the *b*-axis; (d) the bottom bilayer shows a similar spin structure, but the pattern is related by mirror symmetry to the one shown in (c).

with FiM4 order are a = 5.76 Å and c = 9.35 Å, in good agreement with the experimental values of a = 5.71 Å and c = 9.35 Å reported by neutron diffraction study of a single-crystal sample [33].

In contrast to our results, previous computational studies reported FiM3 ($\uparrow\uparrow\downarrow\downarrow\uparrow\uparrow\uparrow$) [21] and FiM6 ($\uparrow\downarrow\uparrow\downarrow\uparrow\downarrow\uparrow\downarrow$) [34, 37] as the magnetic ground state. FiM3 order describes intrachain *B-B* interactions to be ferromagnetic, which is inconsistent with the experimental observations of the interaction to be strongly antiferromagnetic [31, 102, 103, 105]. To justify FiM4 ($\uparrow\uparrow\uparrow\downarrow\downarrow\uparrow\downarrow$) as the idealized collinear magnetic ground state, we calculated four exchange

TABLE III. Relaxed lattice constants of Mn_3O_4 with various imposed magnetic orders, in addition to relative energies per formula unit and magnetic moments per Mn obtained with U = 4 eV and J = 1.2 eV. Experimental lattice constants are a = 5.71 Å and c = 9.35 Å, and the experimental magnetic moments are 4.34, 3.64, and 3.25 μ_B for tetrahedral Mn, octahedral Mn along b, and octahedral Mn along a, respectively [33] with the YK-FiM ground state.

Magnetism	Lattice constants $(Å)$	Relative $E \text{ (meV/f.u.)}$	Magnetic moment $(\mu_{\rm B})$
YK-FiM	a = 5.76, c = 9.35	0	4.49, 3.74, 3.69
FiM4 ($\uparrow\uparrow\downarrow\downarrow\uparrow\downarrow$)	a = 5.76, c = 9.35	152	4.50, 3.85, 3.65
FiM1 ($\downarrow\downarrow\uparrow\uparrow\uparrow\uparrow)$	a = 5.74, c = 9.37	154	$4.48, \ 3.67$
FiM6 ($\downarrow\downarrow\downarrow\downarrow\downarrow\downarrow\downarrow)$	$a = 5.76 \ b = 5.78, \ c = 9.32$	158	4.48, 3.76
FiM3 ($\uparrow\uparrow\downarrow\downarrow\uparrow\uparrow)$	a = 5.75, c = 9.34	175	$4.50, \ 3.85, \ 3.65$
FiM2 ($\uparrow\downarrow\uparrow\uparrow\uparrow\uparrow)$	a = 5.77, c = 9.35	184	4.51, 3.78
FiM5 ($\uparrow\uparrow\uparrow\downarrow\uparrow\uparrow$)	a = 5.81, b = 5.78, c = 9.36	193	4.50, 3.86, 3.66
$FM (\uparrow \uparrow \uparrow \uparrow \uparrow \uparrow)$	a = 5.82, c = 9.36	208	4.52, 3.87

interaction constants: J_{AA} , J_{AB} , J_{BBsr} , and J_{BBlr} , where the last two values represent in-224 trachain (short-range) and interchain (long-range) B-B interactions, respectively. We map 225 the energies of the six FiM structures, relative to that of the FM structure, to a Heisenberg 226 Hamiltonian, as described in Ref. [37]. All interactions are antiferromagnetic, with small 227 values involving the tetrahedral Mn_A site ($J_{AA} = -0.36$ K and $J_{AB} = -2.98$ K), large and dom-228 inant intrachain B-B interaction ($J_{BBsr} = -23.9$ K), and small interchain B-B interaction 229 $(J_{BBlr} = -0.45 \text{ K})$. The J values are in reasonable agreement with the experimental values 230 obtained from a polycrystalline sample [31] ($J_{AA} = -4.9$ K, $J_{AB} = -6.8$ K, and $J_{BB} = -19.9$ 231 K). The strong antiferromagnetic intrachain B-B interaction can be understood as a result 232 of the direct exchange between overlapping neighboring $Mn_B t_{2g}$ orbitals (J_{BBsr}) dominating 233 over very weak ferromagnetic superexchange mediated by O 2p orbitals (J_{BBlr}) [103]. 234

The noncollinear magnetic ground state shows an exotic spin pattern, illustrated in Fig. 3. All the spins lie on the bc plane. Mn_A spins are aligned along the *b*-axis, as if they were ferromagnetic in that direction, with small deviations from the *b*-axis. However, Mn_B spins show a sinusoidal spin pattern that is related by mirror symmetry for different bilayers of the system [Fig. 3(c) and (d)]. Calculating the electronic structure of the YK-FiM ground

state with U = 4 eV and J = 1.2 eV yields a band gap of 1.01 eV [Fig. 2(b)]. This opening 240 of a gap is remarkable when compared with the PBE+U study of Franchini *et al.* [21], 241 where only half-metallic states with gaps of 0.3-0.5 eV were obtained with U = 3-6 eV for 242 FiM3 order. Since our noncollinear magnetic ground state is the experimentally reported 243 structure, YK-FiM, rather than FiM3 [21] or FiM6 [34, 37], the electronic structure profile 244 cannot be directly compared with those reported by the previous computational studies. 245 However, several features are in agreement. The valence band consists of widely spread Mn 246 3d states with a large mixing of O 2p states. The conduction band mostly consists of Mn_B 247 3d states, with a characteristic splitting of ≈ 0.21 eV, which was also reported by Hirai et 248 al. [34] as well. The calculated magnetic moments are 4.49, 3.74, and 3.69 $\mu_{\rm B}$ for the spins of 249 Mn_A , Mn_B along b, and Mn_B along a, respectively, in good agreement with the experimental 250 values of 4.34, 3.64, and 3.25 $\mu_{\rm B}$ reported by neutron diffraction study of a single-crystal 251 sample [33]. The splitting of Mn_B magnetic moment was also observed by a ⁵⁵Mn NMR 252 study [106]. Our study predicts Mn_3O_4 as an insulator with the YK-FiM ground state, in 253 agreement with the experimental reports. 254

255 **C.** α -**Mn**₂**O**₃

We compute the magnetic and electronic structures of α -Mn₂O₃ using the AFM or-256 derings proposed by Regulski et al. [47] (AFM1) and Cockayne et al. [48] (AFM2) while 257 allowing spin noncollinearity. Cockayne et al. [48] determined the NC-AFM2 order to be 258 the magnetic ground state [Fig. 4(a)-(b)], independently from neutron powder diffraction 259 and PBEsol+U+J study in concurrence with a cluster-expansion model, suggesting that 260 the ground-state magnetic structure of α -Mn₂O₃ has largely been solved. In agreement with 261 this observation, our structural, electronic, and magnetic relaxations with each candidate 262 magnetic order confirm the NC-AFM2 structure as the magnetic ground state [Fig. 4(a)]. 263 Applying U = 2.8 eV and J = 1.2 eV further stabilizes the NC-AFM2 structure (Table 264 IV). This complex spin structure is easier to understand in terms of four magnetic sublat-265 tices [Fig. 4(c) and (d)]. These four magnetic sublattices correspond to the Mn Wyckoff 266 positions: sublattice I consists of Mn 4(a) and Mn 4(b), and sublattices II-IV consist of Mn 267 8(c) with different spin patterns. The spin deviation from the c-axis varies from 4-23°, as 268 shown in Table V, in general agreement with the experimental work, where the spins deviate 269



FIG. 4. Magnetic and electronic structures of α -Mn₂O₃: (a) Computed NC-AFM2 structure, with green and yellow spins indicating up and down directions, respectively; (b) The corresponding band structure shows an insulating state with a gap of 0.081 eV; (c) Four magnetic sublattices based on Mn Wyckoff positions; (d) The magnetic sublattices from (c) untangled for clarity, where sublattice I consists of Mn 4(a) and 4(b) in a C-type magnetic structure, sublattice II consists of Mn 8(c) in an A-type magnetic structure, sublattice III consists of Mn 8(c) in a G-type magnetic structure, and sublattice IV consists of Mn 8(c) in a unique magnetic structure, where each magnetic ion has three nearest neighbors, one with identical spin direction and two with opposite spin direction, similar to E-type [107].

²⁷⁰ in a range of 12-34°. More importantly, our computed electronic structure shows a band ²⁷¹ gap of 0.081 eV [Fig. 4(b)]. Although our results are consistent with the experimental and theoretical work presented by Cockayne *et al.* [48], they are in disagreement with the results reported by Franchini *et al.* [21], where both PBE+U and hybrid functional calculations, HSE and PBE0, yielded FM ground state. This disagreement suggests that those levels of theory incorrectly predict the ground state of this complex magnetic system and that a noncollinear description of this system is needed. As for the lattice structure, relaxation with the AFM2 order yields a = 9.382 Å, b = 9.444 Å, and c = 9.376 Å, in good agreement with the experimental values of a = 9.408 Å, b = 9.449 Å, and c = 9.374 Å [48].

TABLE IV. Relaxed lattice constants of α -Mn₂O₃ with various imposed magnetic orders, in addition to relative energies per formula unit obtained with U = 2.8 eV and J = 1.2 eV. Experimental lattice constants are a = 9.407 Å, b = 9.447 Å, and c = 9.366 Å [41] with the NC-AFM2 ground state [48].

	Magnetism	Lattice constants $(Å)$	Relative $E \text{ (meV/f.u.)}$
202	NC-AFM2	a = 9.382, b = 9.444, c = 9.376	0
283	NC-AFM1	a = 9.410, b = 9.387, c = 9.399	4.1
	FM	a = 9.438	50

TABLE V. Magnetic moments per Mn and spin angles for the NC-AFM2 structure of α -Mn₂O₃, based on the five unique Mn Wyckoff positions. The spin angles are relative to the *c*-axis. The experimental magnetic moments are in the range of 2.6-4.0 $\mu_{\rm B}$ [47, 48].

	Mn Wyckoff position	Magnetic moment $(\mu_{\rm B})$	Spin angle (°)
288	Mn 4(a)	4.09	8.7
	Mn 4(b)	2.91	17.3
	Mn 8(c)	3.68	23.4
	$Mn \ 8(c)$	3.83	21.8
	Mn 8(c)	3.69	4.5

289

²⁹⁰ Calculating the electronic structure of the NC-AFM2 order with U = 2.8 eV and J = 1.2²⁹¹ eV yields an insulating state with a gap of 0.081 eV [Fig. 2(c)], with the projected DOS pro-²⁹² file in good agreement with that reported by Cockayne *et al.* [48]. The calculated magnetic ²⁹³ moments, shown in Table V, are in general agreement with the experimental values of Cockayne *et al.* [48], which vary from 2.6-4.0 $\mu_{\rm B}$. Within the framework of DFT+*U*+*J*, our study predicts α -Mn₂O₃ as an insulator with the NC-AFM2 ground state, in agreement with the experimental observations. Achieving accurate magnetic properties with our computational setup is a significant leap forward to understanding these complex magnetic systems.

298 **D.** β -MnO₂



299

FIG. 5. Computed spiral magnetic structure of β -MnO₂: (a) Side view showing the seven unit cell period of the spin spiral, numbered for clarity, with magnetic moments colored in green; (b) Top view showing the spin rotation of 129° from each layer. The darker the spin, the closer it is to the viewer.

To compute the magnetic structure of β -MnO₂, we use the screw-type spiral order, in 304 addition to the AFM and FM orders that previous computational studies have employed [21, 305 74–76]. Crystal and magnetic structure relaxations yield the spiral structure as the magnetic 306 ground state (Table VI). The lattice constants, a = 4.402 Å and c = 2.880 Å, are in excellent 307 agreement with the experimental values of a = 4.404 Å and c = 2.877 Å [108, 109]. The 308 spiral structure [Fig. 5] consists of spins on the *ab*-plane rotating by 129° across each layer 309 along the *c*-axis. A total of seven unit cells (14 layers) are needed for a complete magnetic 310 spiral period (5 spin revolutions). 311

Calculating the electronic structure of the spiral order with U = 2.8 eV and J = 1.2eV yields an insulating state with a gap of 0.25 eV [Fig. 2(d)], in good agreement with the

TABLE VI. Relaxed lattice constants of β -MnO₂ with different imposed magnetic orders, in addition to relative energies per formula unit and magnetic moments per Mn obtained with U = 2.8eV and J = 1.2 eV. Experimental lattice constants are a = 4.396 Å and c = 2.871 Å [99], and the magnetic moment is 2.35 $\mu_{\rm B}$ [100] with the spiral magnetic ground state.

Magnetism	Lattice constants $(Å)$	Relative $E \text{ (meV/f.u.)}$	Magnetic moment (μ_B)
Spiral	a = 4.402, c = 2.88	0	2.63
AFM	a = 4.402, c = 2.88	47	2.65
FM	a = 4.422, c = 2.89	650	2.91

value of 0.27 eV reported by PBEsol+U+J study [75] and 0.26 eV reported by optical mea-314 surements of a thin film sample [73]. Similar to Mn_3O_4 , the opening of a gap is remarkable 315 when compared with PBE+U study of Franchini *et al.* [21], where only metallic states were 316 obtained with U values up to 6 eV. The projected DOS profile is also in excellent agreement 317 with that reported by hybrid functional calculations of Franchini et al. [21]. In accordance 318 with the Mn⁴⁺ oxidation state and the octahedral crystal-field splitting, the valence band 319 shows a single broad Mn t_{2q} band with a large mixing of O 2p states, whereas the conduc-320 tion band consists mostly of Mn e_q states with small O 2p mixing. The calculated magnetic 321 moment of 2.63 $\mu_{\rm B}$ is in good agreement with the experimental value of 2.35 $\mu_{\rm B}$ reported by 322 neutron powder diffraction study [100]. Our electronic structure accurately predicts β -MnO₂ 323 as an insulator with the spiral magnetic ground state. 324

325 IV. SUMMARY AND CONCLUSIONS

Our computational study of manganese oxides, using the fully anisotropic PBEsol+U+J326 approach, yields ground-state structural, magnetic, and electronic properties of quality and 327 accuracy that are comparable to previously reported hybrid functional and experimental 328 studies. We show that the limitations of conventional DFT regarding the magnetic and 329 electronic structures of insulating transition metal oxides can be improved by pseudopoten-330 tial design and careful selection of fully anisotropic U and J values. The resulting magnetic 331 ground states (AFM-II, YK-FiM, NC-AFM2, and spiral for MnO, Mn_3O_4 , α -Mn₂O₃, and β -332 MnO_2 , respectively) correspond to the experimentally observed configurations. All relaxed 333

lattice constants, obtained with PBEsol alone, are in good agreement with the experimental 334 values. Appropriate band gaps were obtained with U values smaller than those used by 335 previous GGA+U studies, while reproducing the electronic structure profiles in good agree-336 ment with those reported by previous hybrid functional studies. Our results overall suggest 337 the enhanced performance of our designed pseudopotential with semicore and partial core 338 correction, thereby offering a promising potential of the DFT+U+J approach for electronic 339 structure studies involving other strongly correlated, complex magnetic systems with accu-340 racy nearing that of more computationally expensive methods such as hybrid functionals. 341

342 ACKNOWLEDGEMENTS

The authors acknowledge support from the Department of Energy, Division of Basic Energy Sciences, under grant DE-FG02-07ER15920. J. S. L. wishes to thank the Vagelos Integrated Program in Energy Research (VIPER) at the University of Pennsylvania. Computational support was provided by the High-Performance Computing Modernization Office and the National Energy Research Scientific Computing Center.

- [1] Z. Yang, Y. Zhang, W. Zhang, X. Wang, Y. Qian, X. Wen, and S. Yang, J. Solid State
 Chem. 17, 679 (2006).
- [2] M. M. Thackeray, W. I. F. David, P. G. Bruce, and J. B. Goodenough, Mater. Res. Bull.
 18, 461 (1983).
- ³⁵² [3] M. M. Thackeray, Prog. Solid State Chem. 25, 1 (1997).
- [4] G. H. Lee, S. H. Huh, J. W. Jeong, B. J. Choi, S. H. Kim, and H.-C. Ri, J. Am. Chem. Soc.
 124, 12094 (2002).
- ³⁵⁵ [5] S. Thota, B. Prasad, and J. Kumar, Mater. Sci. Eng. B **167**, 153 (2010).
- ³⁵⁶ [6] R. Jothiramalingam and M. K. Wang, J. Porous Mater. **17**, 677 (2010).
- ³⁵⁷ [7] C. Kittel, Introduction to Solid State Physics (Wiley, New York, 1986).
- ³⁵⁸ [8] G. A. Slack, J. Appl. Phys. **31**, 1571 (1960).
- ³⁵⁹ [9] H. Shaked, J. Faber Jr., and R. L. Hitterman, Phys. Rev. B **38**, 11901 (1988).

- [10] R. N. Iskenderov, I. A. Drabkin, L. T. Emel'yanova, and Y. M. Ksendzov, Fiz. Tverd. Tela
 (Leningrad) 10, 2573 (1968).
- [11] J. van Elp, R. H. Potze, H. Eskes, R. Berger, and G. A. Sawatzky, Phys. Rev. B 44, 1530
 (1991).
- ³⁶⁴ [12] W. C. Mackrodt, N. M. Harrison, V. R. Saunders, N. L. Allan, M. D. Towler, E. Apra, and
 ³⁶⁵ R. Dovesi, Philos. Mag. A 68, 653 (1993).
- ³⁶⁶ [13] M. D. Towler, N. L. Allan, N. M. Harrison, V. R. Saunders, W. C. Mackrodt, and E. Apra,
 ³⁶⁷ Phys. Rev. B **50**, 5041 (1994).
- ³⁶⁸ [14] S. Massidda, M. Posternak, A. Baldereschi, and R. Resta, Phys. Rev. Lett. **82**, 430 (1999).
- ³⁶⁹ [15] X. Feng, Phys. Rev. B **69**, 155107 (2004).
- [16] C. Franchini, V. Bayer, R. Podloucky, J. Paier, and G. Kresse, Phys. Rev. B 72, 045132
 (2005).
- ³⁷² [17] A. Schron and F. Bechstedt, Phys. Rev. B 82, 165109 (2010).
- ³⁷³ [18] P. Dufek, P. Blaha, V. Sliwko, and K. Schwarz, Phys. Rev. B 49, 10170 (1994).
- ³⁷⁴ [19] J. Hugel and M. Kamal, Solid State Commun. **100**, 457 (1996).
- J. E. Pask, D. J. Singh, I. I. Mazin, C. S. Hellberg, and J. Kortus, Phys. Rev. B 64, 024403
 (2001).
- [21] C. Franchini, R. Podloucky, J. Paier, M. Marsman, and G. Kresse, Phys. Rev. B 75, 195128
 (2007).
- ³⁷⁹ [22] A. Schron, C. Rodl, and F. Bechstedt, Phys. Rev. B 86, 115134 (2012).
- ³⁸⁰ [23] A. Schron, M. Granovskij, and F. Bechstedt, J. Phys.: Condens. Matter **25**, 094006 (2013).
- ³⁸¹ [24] H. Jiang, R. I. Gomez-Abal, P. Rinke, and M. Scheffler, Phys. Rev. B 82, 045108 (2010).
- ³⁸² [25] E. Engel and R. N. Schmid, Phys. Rev. Lett. **103**, 036404 (2009).
- ³⁸³ [26] J. B. Goodenough and A. L. Loeb, Phys. Rev. Lett. **98**, 391 (1955).
- ³⁸⁴ [27] A. S. Borovik-Romanov and M. P. Orlova, Soviet Phys. JETP 5, 1023 (1957).
- ³⁸⁵ [28] Y. Yafet and C. Kittel, Phys. Rev. Lett. 87, 290 (1952).
- ³⁸⁶ [29] I. S. Jacobs, J. Phys. Chem. Solids **11**, 1 (1959).
- ³⁸⁷ [30] K. Dwight and N. Menyuk, Phys. Rev. Lett. **119**, 1470 (1960).
- ³⁸⁸ [31] G. Srinivasan and M. S. Seehra, Phys. Rev. B 28, 1 (1983).
- ³⁸⁹ [32] B. Boucher, R. Buhl, and M. Perrin, J. Appl. Phys. **42**, 1615 (1971).
- ³⁹⁰ [33] G. B. Jensen and O. V. Nielsen, J. Phys. C: Solid State Phys. 7, 409 (1974).

- ³⁹¹ [34] S. Hirai, Y. Goto, A. Wakatsuki, Y. Kamihara, M. Matoba, and W. L. Mao, (2014).
- ³⁹² [35] D. P. Dubal, D. S. Dhawale, R. R. Salunkhe, V. J. Fularim, and C. D. Lokhande, J. Alloys
 ³⁹³ Compd. 497, 166 (2010).
- ³⁹⁴ [36] A. Jha, R. Thapa, and K. K. Chattopadhyay, Mater. Res. Bull. 47, 813 (2012).
- ³⁹⁵ [37] A. Chartier, P. D'Arco, R. Dovesi, and V. R. Saunders, Phys. Rev. B **60**, 14042 (1999).
- ³⁹⁶ [38] Q. Liu, Y. Li, Z. Hu, D. Mao, C. Chang, and F. Huang, Electrochim. Acta **53**, 7298 (2008).
- ³⁹⁷ [39] N. N. Tusar, D. Maucec, M. Rangus, I. Arcon, M. Mozaj, M. Cotman, A. Pintar, and
- ³⁹⁸ V. Kaucic, Adv. Funct. Mater. **22**, 820 (2012).
- [40] M. Baldi, V. S. Escribano, J. M. G. Amores, F. Milella, and G. Busca, Appl. Catal. B:
 Environ. 17, L175 (1998).
- ⁴⁰¹ [41] S. Geller, Acta Cryst. **27**, 821 (1970).
- ⁴⁰² [42] E. G. King, J. Am. Chem. Soc. **76**, 3289 (1954).
- ⁴⁰³ [43] J. Cable, M. Wilkinson, E. Woolan, and W. Koehler, Phys. Prog. Rep., 43 (1957).
- ⁴⁰⁴ [44] R. R. Chevalier, G. Roult, and E. F. Bertaut, Solid State Commun. 5, 7 (1967).
- ⁴⁰⁵ [45] R. W. Grant, S. Geller, J. A. Cafe, and G. P. Espinosa, Phys. Rev. Lett. **175**, 686 (1968).
- ⁴⁰⁶ [46] S. Geller and G. P. Espinosa, Phys. Rev. B **1**, 3763 (1970).
- ⁴⁰⁷ [47] M. Regulski, R. Przenioslo, I. Sosnowska, D. Hohlwein, and R. Schneider, J. Alloys Compd.
 ⁴⁰⁸ 362, 236 (2004).
- ⁴⁰⁹ [48] E. Cockayne, I. Levin, H. Wu, and A. Llobet, Phys. Rev. B 87, 184413 (2013).
- ⁴¹⁰ [49] Q.-u.-a. Javed, W. Feng-Ping, M. Y. Rafique, A. M. Toufiq, and M. Z. Iqbal, Chin. Phys. B
 ⁴¹¹ 21, 117311 (2012).
- ⁴¹² [50] J. M. Tarascon and D. Guyomard, Electrochim. Acta **38**, 1221 (1993).
- ⁴¹³ [51] G. Pistoia and G. Wang, Solid State Ionics **66**, 135 (1993).
- ⁴¹⁴ [52] H. Huang and P. G. Bruce, J. Power Sources **54**, 52 (1995).
- ⁴¹⁵ [53] A. R. Armstrong and P. G. Bruce, Nature **381**, 499 (1996).
- ⁴¹⁶ [54] X. Wang and Y. Li, J. Am. Chem. Soc. **124**, 2880 (2002).
- ⁴¹⁷ [55] W. Tang, X. Yang, Z. Liu, and K. Ooi, J. Mater. Chem. **13**, 2989 (2003).
- ⁴¹⁸ [56] F. Cheng, J. Zhao, W. Song, C. Li, H. Ma, J. Chen, and P. Shen, Inorg. Chem. **45**, 2038 ⁴¹⁹ (2006).
- ⁴²⁰ [57] J.-Y. Luo, J.-J. Zhang, and Y.-Y. Xia, Chem. Mater. **18**, 5618 (2006).
- ⁴²¹ [58] F. Jiao and P. G. Bruce, Adv. Mater. **19**, 657 (2007).

- ⁴²² [59] V. Mathew, J. Lim, J. Kang, J. Gim, A. K. Rai, and J. Kim, Electrochem. Commun. 13,
 ⁴²³ 730 (2011).
- ⁴²⁴ [60] D. Wang, L.-M. Liu, S.-J. Zhao, B.-H. Li, H. Liu, and X.-F. Lang, Phys. Chem. Chem.
 ⁴²⁵ Phys. 15, 9075 (2013).
- ⁴²⁶ [61] A. Debart, A. J. Paterson, J. Bao, and P. G. Bruce, Angew. Chem. Int. Ed. 47, 4521 (2008).
- ⁴²⁷ [62] A. K. Thapa, Y. Hidaka, H. Hagiwara, S. Ida, and T. Ishihara, J. Electrochem. Soc. 158,
 ⁴²⁸ A1483 (2011).
- ⁴²⁹ [63] M. Toupin, T. Brousse, and D. Belanger, Chem. Mater. **14**, 3946 (2002).
- ⁴³⁰ [64] V. Subramanian, H. Zhu, R. Vajtai, P. M. Ajayan, and B. Wei, J. Phys. Chem. B 109,
 ⁴³¹ 20207 (2005).
- ⁴³² [65] S. Devaraj and N. Munichandraiah, J. Phys. Chem. C **112**, 4406 (2008).
- 433 [66] H. Zhang, G. Cao, Z. Wang, Y. Yang, Z. Shi, and Z. Gu, Nano Lett. 8, 2664 (2008).
- ⁴³⁴ [67] X. Lang, A. Hirata, T. Fujita, and M. Chen, Nat. Nanotechnol. 6, 232 (2011).
- ⁴³⁵ [68] J. Zang and X. Li, J. Mater. Chem. **21**, 10965 (2011).
- [69] L. Espinal, W. Wong-Ng, J. A. Kaduk, A. J. Allen, C. R. Snyder, C. Chiu, D. W. Siderius,
 L. Li, E. Cockayne, A. E. Espinal, and S. L. Suib, J. Am. Chem. Soc. 134, 7944 (2012).
- [70] R. Andreozzi, A. Insola, V. Caprio, R. Marotta, and V. Tufano, Appl. Catal. A: General
 138, 75 (1996).
- ⁴⁴⁰ [71] F. H. B. Lima, M. L. Calegaro, and E. A. Ticianelli, Electrochim. Acta **52**, 3732 (2007).
- ⁴⁴¹ [72] H. Sato and T. Enoki, Phys. Rev. B **61**, 3563 (2000).
- ⁴⁴² [73] X. L. Yu, S. X. Wu, Y. J. Liu, and S. W. Li, Solid State Commun. **146**, 166 (2008).
- ⁴⁴³ [74] W. C. Mackrodt and E.-A. Williamson, J. Chem. Soc., Faraday Trans. **93**, 3295 (1997).
- ⁴⁴⁴ [75] E. Cockayne and L. Li, Chem. Phys. Lett. **544**, 53 (2012).
- ⁴⁴⁵ [76] D. A. Tompsett, D. S. Middlemiss, and M. S. Islam, Phys. Rev. B 86, 205126 (2012).
- ⁴⁴⁶ [77] A. Yoshimori, J. Phys. Soc. Jpn. **14**, 807 (1959).
- ⁴⁴⁷ [78] M. Zhuang and J. W. Halley, Phys. Rev. B **64**, 024413 (2001).
- [79] S. L. Dudarev, G. A. Botton, S. Y. Savrasov, C. J. Humphreys, and A. P. Sutton, Phys.
 Rev. B 57, 1505 (1998).
- ⁴⁵⁰ [80] V. I. Anisimov, J. Zaanen, and O. K. Anderson, Phys. Rev. B 44, 943 (1991).
- ⁴⁵¹ [81] A. I. Liechtenstein, V. I. Anisimov, and J. Zaanen, Phys. Rev. B **52**, R5467 (1995).
- ⁴⁵² [82] B. Himmetoglu, R. M. Wentzcovitch, and M. Cococcioni, Phys. Rev. B 84, 115108 (2011).

- ⁴⁵³ [83] A. D. Becke, J. Chem. Phys. **98**, 1372 (1993).
- ⁴⁵⁴ [84] E. Bousquet and N. Spaldin, Phys. Rev. B 82, 220402 (2010).
- [85] J. P. Perdew, A. Ruzsinszky, G. I. Csonka, O. A. Vydrov, G. E. Scuseria, L. A. Constantin,
 X. Zhou, and K. Burke, Phys. Rev. Lett. 100, 136406 (2008).
- ⁴⁵⁷ [86] Y.-C. Wang, Z.-H. Chen, and H. Jiang, J. Chem. Phys. **144**, 144106 (2016).
- 458 [87] P. Giannozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L.
- 459 Chiarotti, M. Cococcioni, I. Dabo, A. Dal Corso, S. de Gironcoli, S. Fabris, G. Fratesi,
- 460 R. Gebauer, U. Gerstmann, C. Gougoussis, A. Kokalj, M. Lazzeri, L. Martin-Samos,
- ⁴⁶¹ N. Marzari, F. Mauri, R. Mazzarello, S. Paolini, A. Pasquarello, L. Paulatto, C. Sbraccia,
- 462 S. Scandolo, G. Sclauzero, A. P. Seitsonen, A. Smogunov, P. Umari, and R. M. Wentzcovitch,
- ⁴⁶³ J. Phys.: Condens. Matter **21**, 395502 (2009).
- ⁴⁶⁴ [88] M. Cococcioni and S. de Gironcoli, Phys. Rev. B **71**, 035105 (2005).
- [89] A. M. Rappe, K. M. Rabe, E. Kaxiras, and J. D. Joannopoulos, Phys. Rev. B 41, 1227
 (1990).
- ⁴⁶⁷ [90] N. J. Ramer and A. M. Rappe, Phys. Rev. B **59**, 12471 (1999).
- ⁴⁶⁸ [91] http://opium.sourceforge.net.
- ⁴⁶⁹ [92] S. G. Louie, S. Froyen, and M. L. Cohen, Phys. Rev. B 26, 1738 (1982).
- ⁴⁷⁰ [93] M. Fuchs and M. Scheffler, Comput. Phys. Commun. **119**, 67 (1999).
- ⁴⁷¹ [94] D. Porezag, M. R. Pederson, and A. Y. Liu, Phys. Rev. B **60**, 14132 (1999).
- ⁴⁷² [95] H. J. Monkhorst and J. D. Pack, Phys. Rev. B **13**, 5188 (1976).
- ⁴⁷³ [96] S. Sasaki, K. Fujino, Y. Takeuchi, and R. Sadanaga, Acta Cryst. A36, 904 (1980).
- ⁴⁷⁴ [97] A. K. Cheetham and D. A. O. Hope, Phys. Rev. B **27**, 6964 (1983).
- 475 [98] R. G. Wyckoff, Crystal Structures (Wiley, New York, 1963).
- ⁴⁷⁶ [99] N. Ohama and Y. Hamaguchi, J. Phys. Soc. Jpn. **30**, 1311 (1971).
- ⁴⁷⁷ [100] M. Regulski, R. Przenioslo, and I. Sosnowska, Phys. Rev. B 68, 172401 (2003).
- ⁴⁷⁸ [101] C. Rodl, F. Fuchs, J. Furthmuller, and F. Bechstedt, Phys. Rev. B **79**, 235114 (2009).
- ⁴⁷⁹ [102] B. Mehdaoui, O. Pena, M. Bahout, A. B. Antunes, and G. Martinez, Bol. Soc. Esp. Ceram.
 ⁴⁸⁰ V. 47, 143 (2008).
- [103] J.-H. Chung, J.-H. Kim, S.-H. Lee, T. J. Sato, T. Suzuki, M. Katsumura, and T. Katsufuji,
 Phys. Rev. B 77, 054412 (2008).
- ⁴⁸³ [104] B. Chardon and F. Vigneron, J. Magn. Magn. Mater. **58**, 128 (1986).

- ⁴⁸⁴ [105] A. Kuriki, Y. Moritomo, S. Xu, K. Ohoyama, K. Kato, and A. Nakamura, J. Phys. Soc.
 ⁴⁸⁵ Jpn. **72**, 458 (2003).
- ⁴⁸⁶ [106] E. Jo, K. An, J. H. Shim, C. Kim, and S. Lee, Phys. Rev. B 84, 174423 (2011).
- ⁴⁸⁷ [107] E. O. Wollan and W. C. Koehler, Phys. Rev. **100**, 545 (1955).
- 488 [108] R. G. Wyckoff, Crystal Structures, Vol. 1 (Interscience, New York, 1960).
- ⁴⁸⁹ [109] A. A. Bolzan, C. Fong, B. J. Kennedy, and C. J. Howard, Aust. J. Chem. **46**, 939 (1993).