

# CHCRUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

# Structural and electronic phase transitions of ThS\_{2} from first-principles calculations

Yongliang Guo, Changying Wang, Wujie Qiu, Xuezhi Ke, Ping Huai, Cheng Cheng, Zhiyuan Zhu, and Changfeng Chen

Phys. Rev. B 94, 134104 — Published 7 October 2016

DOI: 10.1103/PhysRevB.94.134104

## Structural and electronic phase transitions of $ThS_2$ from first-principles calculations

Yongliang Guo,<sup>1,2</sup> Changying Wang,<sup>2</sup> Wujie Qiu,<sup>1</sup> Xuezhi Ke,<sup>1,\*</sup>

Ping Huai,<sup>2,†</sup> Cheng Cheng,<sup>2</sup> Zhiyuan Zhu,<sup>2</sup> and Changfeng Chen<sup>3,‡</sup>

<sup>1</sup>Department of Physics, East China Normal University, Shanghai 200241, China

<sup>2</sup>Shanghai Institute of Applied Physics, Chinese Academy of Sciences, Shanghai 201800, China

<sup>3</sup>Department of Physics and High Pressure Science and Engineering Center,

University of Nevada, Las Vegas, Nevada 89154, USA

Thorium and its compounds have received considerable attention in recent years due to the renewed interest in developing the thorium fuel cycle as an alternative nuclear energy technology. There is pressing current need to explore the physical properties essential to the fundamental understanding and practical application of these materials. Here we report on a computational study of thorium disulfide  $(ThS_2)$ , which plays an important role in the thorium fuel reprocessing cycle. We have employed the density functional theory and evolutionary structure search methods to determine the crystal structures, electronic band structures, phonon dispersions and density of states, and thermodynamic properties of  $ThS_2$  under various pressure and temperature conditions. Our calculations identify several crystalline phases of  $ThS_2$  and a series of structural phase transitions induced by pressure and temperature. The calculated results also reveal electronic phase transitions from the semiconducting state in the low-pressure phases of ThS<sub>2</sub> in the Pnma and  $Fm\bar{3}m$ symmetry to the metallic state in the high-pressure phases of  $\text{ThS}_2$  in the *Pnma* and *I4/mmm* symmetry. These results explain the experimental observation of the thermodynamic stability of the Pnma phase of ThS<sub>2</sub> at the ambient conditions and a pressure-induced structural phase transition in  $ThS_2$  around 40 GPa. Moreover, the present study reveals considerable additional information on the structural and electronic properties of  $ThS_2$  in a wide range of pressure and temperature. Such information provides key insights into the fundamental material behavior and the underlying mechanisms that lay the foundation for further exploration and application of  $ThS_2$ .

PACS numbers: 61.50.Ks, 62.50.-p, 63.20.D-, 71.20.-b

Keywords: Phase transition; Phonon dispersion; Elastic constant; Semiconductor-metal transition; Firstprinciples calculations

### I. INTRODUCTION

Thorium and uranium are the only two significantly radioactive elements that occur naturally in large quantities on Earth. While uranium has been the main fuel of choice in conventional reactors, there has been renewed and increasing interest in recent years to further develop thorium based nuclear energy technologies. Thorium has the potential to become a nuclear fuel because  $^{232}$ Th, which is not fissile itself, can convert to <sup>233</sup>U by absorbing slow neutrons<sup>1</sup>. A major driving factor for developing the thorium fuel cycle is the concern about the limited resources of uranium<sup>2</sup>, whereas thorium is thought to be three to four times more naturally  $abundant^{3-5}$ . Another advantage of the thorium fuel cycle is that it can be developed to produce negligible amounts of plutonium and fewer long-lived minor actinides than a uranium  $cycle^{5,6}$ , thus boosting thorium's potential for improving proliferation resistance and waste characteristics. Research in this field is currently at an early stage and reported results are still limited; but we expect that this field will attract more attention as understanding on relevant fundamental science advances and technological breakthroughs happen. It is therefore important to explore materials and processes relevant to the thorium-based fuel technology.

Crucial to the development of the thorium fuel cycle is a clear understanding of the structural, electronic and thermodynamic properties of thorium and its compounds that are present at various stages of the fuel cycle. The recovery of thorium from reprocessing the spent nuclear fuel is an important process, and it is achieved by separating and recycling thorium from the spent fuel using the sulfurization method  $^{7-12}$  where thorium dioxide can be sulfurized by carbon disulfide at high temperature. The sulfurized material can then be separated from the oxides or metals by the acid leaching method<sup>10,11</sup>. Several thorium oxides and sulfides appear during the spent fuel reprocessing procedures. Among them is thorium disulfide  $(ThS_2)$ , which appears as an intermediate phase that serves as an important link in the thorium fuel reprocessing cycle. It is known that  $ThS_2$  has orthorhombic cotunnite-type structure (space group Pnma, Z = $(4)^{13-16}$ , and at high pressures ThS<sub>2</sub> undergoes a structural phase transition around 40 GPa<sup>16</sup>. However, details about this phase transition, especially an accurate structural determination of the high-pressure phase, remain unclear. Furthermore, there is a lack of understanding of the electronic and thermodynamic properties of ThS<sub>2</sub> at either ambient or high pressure and temperature conditions. Knowledge about these properties is essential to the understanding of the fundamental material behavior of this compound and its role in the thorium fuel cycle.

In the present work we report on a systematic study of the structural, electronic, lattice dynamic, and thermodynamic properties of  $\text{ThS}_2$  over a wide range of pressure and temperature. We first determine the crystal structure of  $\text{Th}S_2$  at ambient and high-pressure conditions by performing an extensive structure search using first-principles total energy calculations and the particleswarm optimization (PSO) algorithm<sup>17</sup>. Our structure search identifies three viable crystal phases of  $ThS_2$  in  $Pnma, Fm\bar{3}m$ , and I4/mmm symmetry, respectively, and the calculated elastic parameters and phonon dispersions confirm their mechanical and dynamical stability. Further studies of the enthalpy and Helmholtz free energy of these structures reveal a series of pressureinduced structural phase transitions and a temperaturedriven phase transition at ambient conditions, which are in good agreement with previously reported experimental observations, and our calculations provide insights into the underlying mechanisms. The calculations also provide key information on the evolution of the structural and electronic properties of these phases. Calculated energetic results show that several structural phases may coexist at high pressures. Electronic band-structure calculations reveal that the highly compressed Pnma phase and the high-pressure I4/mmm phase are both metallic, suggesting an electronic phase transition from the semiconducting state in the low-pressure  $Fm\bar{3}m$  and Pnmaphases to the metallic state in the high-pressure Pnmaand I4/mmm phases. The present results establish key material properties of  $ThS_2$  that are important to both fundamental understanding and practical application of this interesting compound.

The rest of this paper is organized as follows. Section II discusses the details of the computational methods used in the present work. Section III contains the results and discussions of the crystal parameters, equations of state, phonon dispersions, elastic constants and structural and electronic phase transitions induced by pressure or temperature. A summary of our main findings is presented in Section IV.

## **II. COMPUTATIONAL METHODS**

The density-functional-theory (DFT) calculations reported in the present work were carried out using the Vienna Ab initio Simulation Package (VASP)<sup>18,19</sup> with the projector augmented wave scheme  $(PAW)^{20,21}$ . The exchange-correlation functional with the generalized gradient approximation (GGA) of Perdew-Burke-Ernzerhof  $(PBE)^{22}$  was used to solve the Kohn-Sham equations with a kinetic energy cutoff of 600 eV. Since none of the localized 5f-like bands is occupied in Th, the DFT-GGA level calculations provide adequate description for the structural and vibrational properties of Th and its  $compounds^{23-26}$ . We performed global structural optimization for  $ThS_2$  by combining first-principles total energy calculations and the particle-swarm optimization (PSO) algorithm<sup>17</sup> on structural prediction as implemented in the CALYPSO  $code^{27}$ , which has been proved to be effective and accurate in predicting the crystal structures of a large variety of materials  $2^{8-35}$ . The structure searches were performed using the CALYPSO code with system sizes ranging from 1 to 8 formula units (f.u.) per simulation cell at pressures of 0, 50 and 100 GPa. The structural relaxations were performed allowing the variations of the ionic position, cell volume, and cell shape. The Brillouin-zone (BZ) was sampled with a  $12 \times 12 \times 12$ k-point mesh, a  $8 \times 12 \times 6$  k-point mesh, and a  $12 \times 12 \times 8$  kpoint mesh generated via the Monkhorst-Pack scheme<sup>36</sup> for the  $Fm\bar{3}m$  phase, Pnma phase, and I4/mmm phase, respectively. The geometries were considered to be converged when the forces on each ion became less than 0.001 eV/Å. The convergence threshold was set to be  $10^{-7}$  eV for the total energy calculations in the electronic selfconsistent loop. The third-order Birch-Murnaghan equation of state<sup>37</sup> was used to fit the calculated results.

We also calculated the phonon dispersion to check for dynamical stability of the crystal structures of various phases of  $ThS_2$  and obtained the corresponding phonon density of states (DOS) that are used as input for calculations of selected thermodynamic functions. These phonon calculations were carried out using a supercell approach<sup>38</sup> as implemented in the PHONOPY code<sup>39</sup>. Supercells of sizes  $3 \times 3 \times 3$ ,  $2 \times 3 \times 2$  and  $2 \times 2 \times 2$  were created from the optimized crystallographic primitive cell for the  $Fm\bar{3}m$  phase, Pnma phase and I4/mmm phase, respectively. The BZ integration was performed with a  $2 \times 2 \times 2$  k-point mesh. The symmetry non-equivalent Th and S atoms were displaced from their equilibrium positions by an amplitude of 0.02 Å to construct the system dynamical matrix D(k). The forces induced by small displacements were calculated within VASP.

To study the effect of temperature on the relative stability of the  $Fm\bar{3}m$  and Pnma phases of ThS<sub>2</sub> at zero pressure, we have calculated the Helmholtz free energy F(V,T) defined as:

$$F(V,T) = E(V) + F_{vib}(V,T) + F_{el}(V,T), \qquad (1)$$

where E(V) is the ground-state total energy,  $F_{vib}(V,T)$ is the vibrational energy of the lattice ions and  $F_{el}(V,T)$ is the thermal electronic contribution. Since both  $Fm\bar{3}m$ and Pnma phases are insulators, the  $F_{el}(V,T)$  term makes negligible contributions and can be neglected<sup>40,41</sup>. The vibrational term  $F_{vib}(V,T)$  dominates and is determined by

$$F_{\rm vib}(V,T) = k_B T \int_0^\infty g(\omega) \ln\left[2\sinh\left(\frac{\hbar\omega}{2k_B T}\right)\right] d\omega, \quad (2)$$

where  $g(\omega)$  is the phonon DOS. This formula strictly requires that all the phonon frequencies  $\omega$  are positive, and therefore it is not suitable for dynamically unstable phases<sup>41</sup>. This condition is satisfied in our calculations since the structural phases of ThS<sub>2</sub> studied in the present work are all dynamically stable in the parameter regions where the thermodynamic properties are examined.

#### III. RESULTS AND DISCUSSIONS

#### A. Structure identification and characterization

At ambient conditions,  $ThS_2$  adopts the PbCl<sub>2</sub>-type structure with the space group  $Pnma^{13-16}$  as shown in Fig. 1(b). It is interesting to note that this space group is also adopted by the high-pressure phase of thorium dioxide  $(ThO_2)^{42-45}$ . To explore the structural evolution of  $ThS_2$  under pressure, we first pressurize the *Pnma* phase to investigate the variation of the lattice constants. The calculated lattice constants (a, b and c) as a function of pressure are presented in Fig. 2. The results show that the Pnma phase of  $ThS_2$  is more easily compressible along the a direction than along the b or c directions, indicating that the compression process would be anisotropic. Interestingly, as the pressure reaches around 43 GPa, there is an abrupt decrease in the lattice constant a accompanied by steep increases in the lattice constants b and c. Such a behavior is characteristic of an isostructural phase transition where the space group of a crystalline structure remains unchanged while the lattice constants undergo sudden changes. This phenomenon was also observed in  $ThO_2^{45}$ ,  $UO_2^{46}$  and  $PuO_2^{40,46}$ . Results in Fig. 2 show that there are two iso-structural phase transitions in the Pnma phase of ThS<sub>2</sub> under compression, one around 43 GPa to a phase termed Pnma-I and the other around 52 GPa to another phase termed Pnma-II.

To explore possible new phases of  $ThS_2$  at high pressures, we have performed an extensive structure search up to 100 GPa using the CALYPSO code. The procedure successfully reproduced the previously known Pnma phase of  $ThS_2$ , and the structure search also identifies two new structures, one in a face-centered-cubic structure with the space group  $Fm\bar{3}m$  and the other in a body-centered-tetragonal structure with the space group I4/mmm, as illustrated in Fig. 1 (a) and (c), respectively. It is known that the ground-state structure of  $ThO_2$  adopts the  $Fm\bar{3}m$  symmetry<sup>42-45</sup>. We will show below that our energetic calculations also identify the  $Fm\bar{3}m$  phase as the ground-state structure for ThS<sub>2</sub> at low temperatures, but the crystal is thermodynamically driven into the Pnma phase around the room temperature, which explains the experimental observation of the Pnma phase of ThS<sub>2</sub> at ambient conditions. The  $Fm\bar{3}m$  phase contains four ThS<sub>2</sub> f.u. per unit-cell and each cation in this phase has a coordination number of eight. In the  $Fm\bar{3}m$  phase the Th atoms occupy the 4aWyckoff positions (0, 0, 0) and the S atoms occupy the 8c Wyckoff positions (0.25, 0.25, 0.25). Meanwhile, the I4/mmm phase of ThS<sub>2</sub> has two f.u. per unit-cell and each cation has a coordination number of ten. In the I4/mmm phase of ThS<sub>2</sub> the Th atoms occupy the 2a positions (0, 0, 0) and the S atoms occupy the 4*e* positions (0, 0, 0.344). The calculated lattice constants for these three  $ThS_2$  phases are listed in Table I and compared with available experimental data for the Pnma phase.

TABLE I: Calculated lattice constants of the three  $\text{ThS}_2$  phases compared to available experimental data.

	$a_0(\text{\AA})$	$b_0({ m \AA})$	$c_0(\text{\AA})$	
Pnma	7.280	4.300	8.651	Our work
	7.275	4.283	8.617	$Exp.^{14}$
	7.267	4.273	8.615	$Exp.^{15}$
	7.260	4.269	8.600	Exp. <sup>16</sup>
$Fm\overline{3}m$	6.676			Our work
I4/mmm	3.751		8.902	Our work

The calculated results are in good agreement with the experimental data  $^{14-16}$ .

#### B. Mechanical and dynamical stability

To determine the mechanical and dynamical stability of the three structural phases of  $ThS_2$ , we have calculated their phonon dispersion curves and elastic constants. The phonon dispersion of a crystal plays a fundamental role in determining the stability of the structure. A dynamical instability of the crystal is associated with soft phonon modes with imaginary frequencies 47. To check the dynamical stability of the three  $ThS_2$  phases identified by our structure search method, we have calculated their phonon dispersion curves at ambient and high pressures. The results at 0 GPa are shown in Fig. 3, and no imaginary modes were found in the entire Brillouin zone for all three phases of ThS<sub>2</sub>, thus confirming their dynamical stability. Moreover, the calculated phonon dispersions at selected high-pressure points corresponding to the structural phase transitions (see below) shown in Fig. 4 also contain no imaginary modes. These results indicate that the  $ThS_2$  structures are dynamically stable in the entire pressure range studied here, and the absence of any phonon softening at high pressures show that the pressure-induced phase transitions studied below are all driven by the energetics (i.e., the relative enthalpy change) of the system.

The elastic constants measure the response of the crystal structure to external strain, and they also impose constraints on the structural stability. To investigate the mechanical stabilities of the three phases of  $ThS_2$ , we have derived the second-order elastic constants  $(C_{ij})$  from the stress-strain relationship<sup>48</sup>. The calculated results at zero pressure and the selected high-pressure points corresponding to the structural phase transitions are shown in Table II. The mechanical stability of a crystal requires the strain energy to be positive, which implies that the whole set of elastic constants  $C_{ij}$  should satisfy the Born-Huang criteria<sup>49</sup>. The  $Fm\bar{3}m$  phase is a cubic structure, in which the independent elastic stiffness tensor reduces to three components of  $C_{11}$ ,  $C_{44}$  and  $C_{12}$ , and the corresponding mechanical stability criteria are given by<sup>50</sup>  $C_{11} > 0, C_{44} > 0, C_{11} > |C_{12}|$  and  $(C_{11} + 2C_{12}) > 0$ . The Pnma phase is an orthorhombic structure, in which the



FIG. 1: (Color online) Crystal structures of (a)  $Fm\bar{3}m$  phase, (b) Pnma phase and (c) I4/mmm phase of ThS<sub>2</sub>.

TABLE II: Calculated elastic constants  $C_{ij}$  (in GPa) for the three ThS<sub>2</sub> phases at zero and the phase transition pressures.

		$C_{11}$	$C_{22}$	$C_{33}$	$C_{44}$	$C_{55}$	$C_{66}$	$C_{12}$	$C_{13}$	$C_{23}$
$Fm\bar{3}m$	0 GPa	169.5			18.3			48.5		
	2.3  GPa	178.2			18.8			54.2		
Pnma	0  GPa	148.6	121.1	150.1	46.2	19.3	35.5	64.7	65.7	68.5
	$2.3~\mathrm{GPa}$	156.8	136.0	165.9	51.4	25.6	40.3	73.4	71.3	80.0
	37  GPa	190.7	255.6	288.0	106.3	76.4	77.0	177.6	135.8	156.8
I4/mmm	0  GPa	203.0		188.7	81.1		60.0	65.8	54.2	
	$37 \mathrm{GPa}$	306.3		406.8	205.4		129.9	160.6	115.4	



FIG. 2: (Color online) Variation of lattice constants with pressure for the Pnma phase of ThS<sub>2</sub>.

independent elastic stiffness tensor reduces to nine components of  $C_{11}$ ,  $C_{22}$ ,  $C_{33}$ ,  $C_{44}$ ,  $C_{55}$ ,  $C_{66}$ ,  $C_{12}$ ,  $C_{13}$  and  $C_{23}$ , and in this case the mechanical stability criteria are given by<sup>50</sup>  $C_{11} > 0$ ,  $C_{22} > 0$ ,  $C_{33} > 0$ ,  $C_{44} > 0$ ,  $C_{55} > 0$ ,  $C_{66}>0,\ [C_{11}+C_{22}+C_{33}+2(C_{12}+C_{13}+C_{23})]>0, \\ (C_{11}+C_{22}-2C_{12})>0,\ (C_{11}+C_{33}-2C_{13})>0 \text{ and } \\ (C_{22}+C_{33}-2C_{23})>0. \ \text{The }I4/mmm \text{ phase is a tetrag-}$ onal structure, which has six independent elastic constants  $C_{11}$ ,  $C_{33}$ ,  $C_{44}$ ,  $C_{66}$ ,  $C_{12}$  and  $C_{13}$ , and the mechanical stability criteria are given by<sup>50</sup>  $C_{11} > 0, C_{33} > 0,$  $C_{44}>0, C_{66}>0, (C_{11}-C_{12})>0, (C_{11}+C_{33}-2C_{13})>0,$  $[2(C_{11} + C_{12}) + C_{33} + 4C_{13}] > 0$ . From the results listed in Table II, the elastic constants  $C_{ij}$  for the three ThS<sub>2</sub> phases satisfy their respective mechanical stability criteria at zero and phase transition pressures, thus confirming their mechanical stability. These results further support the conclusion that the phase transitions in  $ThS_2$ are driven by the energetics of the system. It is noted that there is a crossover between the  $C_{11}$  and  $C_{22}$  in Table II. This is due to the anisotropy effect of the Pnmaphase under pressure, i.e., it is easier to compress the structure along the y and z direction than along the xdirection.

## C. Equation of state and structural phase transitions

We have calculated the energy versus volume relations for the three phases of ThS<sub>2</sub>, and the obtained results fitted by the third-order Birch-Murnaghan equation are shown in Fig. 5. The calculations reveal that the  $Fm\bar{3}m$ phase is energetically the most favorable among the three phases, and an inspection of the common tangent of these



FIG. 3: (Color online) Phonon dispersion curves for (a)  $Fm\bar{3}m$ , (b) Pnma and (c) I4/mmm phase of ThS<sub>2</sub>.



FIG. 4: (Color online) Phonon dispersion curves for (a)  $Fm\bar{3}m$ -ThS<sub>2</sub> at 2.3 GPa, (b) Pnma-ThS<sub>2</sub> at 37 GPa, (c) Pnma-I-ThS<sub>2</sub> at 43 GPa and (d) I4/mmm-ThS<sub>2</sub> at 37 GPa.

energy versus volume curves suggests a pressure-induced structural phase transition from the  $Fm\bar{3}m$  phase to the *Pnma* phase at a relatively low pressure, followed by a phase transition from the Pnma phase to the I4/mmmphase at a higher pressure. This is indeed confirmed by the calculated enthalpy versus pressure results shown in the inset of Fig. 5, which predict a transition from the  $Fm\bar{3}m$  phase to the *Pnma* phase at 2.3 GPa and a subsequent transition from the Pnma phase to the I4/mmmphase at 37 GPa. We also performed calculations to construct the volume-pressure relation for the three phases of  $ThS_2$ , and the obtained results are shown in Fig. 6. It is seen that the structural transitions between the various  $ThS_2$  phases are accompanied by significant volume collapses. We have extracted the bulk modulus for the  $ThS_2$ phases by fitting the calculated results to the third-order Birch-Murnaghan equation of state, and the obtained results are 89 GPa, 86 GPa, and 107 GPa for the  $Fm\bar{3}m$ , Pnma, and I4/mmm phases of ThS<sub>2</sub>, respectively. We also determined the bulk modulus of these phases from the calculated elastic constants, and the obtained results are practically identical to those from fitting the equation

#### of state.

We now connect our calculated results to the reported experimental observations. We first examine the structural stability of  $ThS_2$ . According to the calculated totalenergy results shown in Fig. 5, the  $Fm\bar{3}m$  phase is energetically the most stable among the three phases of  $ThS_2$ . Experimentally, however, only the Pnma phase was  $observed^{13-16}$  at ambient conditions. To elucidate the underlying mechanism, we have calculated the Helmholtz free energy F(V,T) of the  $Fm\bar{3}m$  and the Pnma phase over the relevant temperature range. The obtained results shown in Fig. 7 indicate that the  $Fm\bar{3}m$  phase is the ground-state structure below 280 K, but above this transition temperature the Pnma phase becomes thermodynamically more stable. Our calculations therefore reveal a temperature-driven structural phase transition in  $ThS_2$ , explaining the experimental observation of the *Pnma* phase at the ambient temperature; the predicted stability of the  $Fm\bar{3}m$  phase as the ground-state phase at temperatures below 280 K expands the structural phase diagram of  $ThS_2$ , and this prediction calls for further experimental verification and exploration.



FIG. 5: (Color online) Total energy versus volume results for the  $Fm\bar{3}m$  phase, Pnma phase and I4/mmm phase of ThS<sub>2</sub>. The symbols represent our calculated data points and the solid lines are the fitting curves of the third-order Birch-Murnaghan equation. The inset shows the enthalpy versus pressure curves with the enthalpy of the Pnma phase set to zero as the reference. It is noted that the E-V equation of state and the enthalpy of Pnma phase are derived from the E-V data of the Pnma phase (53 ~ 76 Å<sup>3</sup>), without using any points of the Pnma-I and Pnma-II phases.



FIG. 6: (Color online) The calculated volume versus pressure relation for the three phases of ThS<sub>2</sub>. The relative volume change at each pressure where a volume collapse occurs is measured against the initial volume  $V_0$  right before each volume drop.



FIG. 7: (Color online) Temperature dependence of the Helmholtz free energy of the Pnma phase of ThS<sub>2</sub> measured relative to that of the the  $Fm\bar{3}m$  phase.

It has been reported<sup>16</sup> that ThS<sub>2</sub> undergoes a pressureinduced, sluggish structural phase transition around 40 GPa, but the details about the nature of the phase transition and the characterization of the high-pressure phase need further examination and clarification. Our calculated enthalpy versus pressure results shown in the inset of Fig. 5 indicate a transition from the Pnma phase to the I4/mmm phase at 37 GPa, which is close to the experimentally reported transition pressure of about 40 GPa. Furthermore, during this phase transition, the volume of the *Pnma* phase is compressed to 53 Å<sup>3</sup>/f.u., and in a large range of volume under further compression, the total energy of the Pnma phase remains very close to that of the I4/mmm phase as seen in Fig. 5. Eventually, the E-V curve of the Pnma phase nearly coincides with that of the I4/mmm phase after the kink at around  $47 \text{ Å}^3$ . This kink corresponds to the iso-structural phase transition from the Pnma-I to the Pnma-II phase (see Fig. 5 and Fig. 2). This close energetic relation suggests that these two phases may coexist during the compression process, thus making the phase transition sluggish as observed in the experiment. It is interesting to note (see Fig. 2) that the Pnma phase undergoes a transformation to an isostructural Pnma-I phase at about 43 GPa. It is therefore expected that there might be multiple phases coexisting around 40 GPa where several structural and electronic (see below) phase transitions occur. The simultaneous presence of the highly compressed Pnma and I4/mmm phases together with the transformed (or partially transformed) *Pnma*-I phase could complicate the determination of the structural details near the phase transition pressure, and the theoretical results obtained here may offer a useful guidance to sort out the structural details.



FIG. 8: (Color online) Electronic band structures and total density of states (TDOS) for (a)  $Fm\bar{3}m$  phase, (b) Pnma phase, (c) Pnma-I phase (44 GPa) and (d) I4/mmm phase of ThS<sub>2</sub>.



FIG. 9: (Color online) The electronic bandgap versus pressure for the  $Fm\bar{3}m$  and Pnma phases of ThS<sub>2</sub>.

# D. Pressure-induced semiconductor-metal transitions

The electronic properties of  $ThS_2$  are also expected to be sensitive to the applied pressure. We have calculated the electronic band structures of the three phases of ThS<sub>2</sub> studied here and investigated the evolution of the electronic bandgap with increasing pressure. We first determined the band structures at zero pressure. The results in Fig. 8 show that the  $Fm\bar{3}m$  phase is a semiconductor with an indirect band gap of 2.622 eV, and the top of the valence band and the bottom of the conduction band are located at the X point and the  $\Gamma$  point, respectively. Meanwhile, the Pnma phase is also a semiconductor but with a direct band gap of 1.037 eV, and both the top of the valence band and the bottom of the conduction band are located at the  $\Gamma$  point. On the other hand, the I4/mmm phase is in a metallic state. It is noted that our calculated bandgap values for the  $Fm\bar{3}m$  and Pnmaphase may have been underestimated by the use of the PBE/GGA exchange-correlation functional and, therefore, the actual bandgaps for these phase are likely to be larger than those reported here. However, the metallic nature of the I4/mmm phase is unlikely to be affected since the steep band crossing near the Fermi energy, which is mainly contributed by the S-p electrons, is not sensitive to the choice of the potential used, and calculations using more accurate hybrid functionals are not known to open a bandgap in systems with such steep band crossings at the Fermi level. We have performed further systematic band-structure calculations and determined the bandgap variation with increasing pressure. The obtained bandgap values for the  $Fm\bar{3}m$  and Pnmaphases of  $\text{ThS}_2$  as a function of pressure are presented in Fig. 9. The results show that the bandgap of the  $Fm\bar{3}m$ phase decreases slightly with increasing pressure. Then at the transition from the Fm3m to the Pnma phase, which should be observable at low temperatures (below 280 K), there is a significant drop in the bandgap from about 2.6 eV to about 0.9 eV. At further increased pressure, the bandgap of the Pnma phase decreases gradually until 43 GPa where the bandgap collapses when the system transitions from the semiconducting Pnmaphase to the metallic *Pnma*-I phase (its electronic band structure is shown in Fig. 8(c)). It is noted, however, that this semiconductor-metal transition accompanying the isostructural phase transition in the Pnma symmetry may be preceded by another semiconductor-metal transition at 37 GPa following the structural transition from the Pnma phase to the I4/mmm phase, since the latter phase remains metallic in the entire pressure range studied here.

#### IV. SUMMARY

In this work we have conducted a systematic study of  $\text{ThS}_2$  using a host of current computational techniques to examine the structural, electronic, lattice dynamics, and thermodynamic properties that are important to the fundamental understanding and practical applications. Our structure search has successfully identified the experimentally observed *Pnma* phase of ThS<sub>2</sub> and pre-

dicted two new structures, the  $Fm\bar{3}m$  phase that exist as the ground-state structure at low-pressure and lowtemperature conditions and the I4/mmm phase that becomes energetically favorable at high pressures. Based on the calculated results of the Helmholtz free energy for the  $ThS_2$  phases, we find a temperature-driven phase transition at 280 K from the  $Fm\bar{3}m$  phase to the Pnmaphase, which explains the experimental observation of the Pnma phase at the ambient conditions. Our enthalpy calculations predict a structural phase transition from the Pnma phase to the I4/mmm phase at 37 GPa, which is in good agreement with the experimental observation of a phase transition around 40 GPa, and the obtained results also provide a full characterization of the nature and details about the pressure-induced structural phase transition of  $ThS_2$ . In particular, the close energetics of several structural phases, including the original Pnma phase, one of its isostructural variations, and the I4/mmm phase, around the phase transition pressure suggest their possible coexistence, which provides an explanation for the experimentally observed sluggishness of the transition and the difficulty in identifying the structural details from the previously collected experimental data alone. The calculated results presented in this work may help further resolve the pressure-induced structural evolution and phase transitions of  $ThS_2$ . Moreover, our electronic band structure calculations reveal a series of pressure-induced electronic phase transitions from the semiconducting state in the low-pressure  $Fm\bar{3}m$  and Pnma phases to the metallic state in the high-pressure Pnma and I4/mmm phases.

- \* Electronic address: xzke@phy.ecnu.edu.cn
- <sup>†</sup> Electronic address: huaiping@sinap.ac.cn
- <sup>‡</sup> Electronic address: chen@physics.unlv.edu
- <sup>1</sup> S. F. Ashley, G. T. Parks, W. J. Nuttall, C. Boxall and R. W. Grimes, Nature **429**, 31-33, (2012).
- <sup>2</sup> See http://www.world-nuclear.org/info/Nuclear-Fuel-Cycle/Uranium-Resources/Supply-of-Uranium/ (2015).
- <sup>3</sup> A. Gale, C. A. Dalton, C. H. Langmuir, Y. Su, and J.-G. Schilling, Geochem. Geophys. Geosyst. 14, 489 (2013).
- <sup>4</sup> M. B. Andersen, T. Elliott, H. Freymuth, K. W. W. Sims, Y. Niu, and K. A. Kelley, Nature **517**, 356 (2015).
- <sup>5</sup> P. Bagla, Science **350**, 726 (2015).
- <sup>6</sup> R. W. Grimes and W. J. Nuttall, Science **329**, 799 (2010).
- <sup>7</sup> S. Hirai, K. Shimakage, Y. Saitou, T. Nishimura, Y. Ue-
- mura, M. Mitomo and L. Brewer, J. Am. Ceram. Soc. 81, 145 (1998).
- <sup>8</sup> M. Ohta, H. Yuan, S. Hirai, Y. Uemura and K. Shimakage, J. Alloy. Comp. **374**, 112 (2004).
- <sup>9</sup> N. Sato, G. Shinohara, A. Kirishima and O. Tochiyama, J. Alloy. Comp. **451**, 669 (2008).
- <sup>10</sup> N. Sato and A. Kirishima, Energy Procedia 7, 444 (2011).
- <sup>11</sup> N. Sato and A. Kirishima, J. Nucl. Mater. **414**, 324 (2011).
- <sup>12</sup> M. Ohta, S. Satoh, T. Kuzuya, S. Hirai, M. Kunii and A. Yamamoto, Acta Mater. **60**, 7232 (2012).
- <sup>13</sup> E. D. Eastman, L. Brewer, L. A. Bromley, P. W. Gilles, and N. L. Lofgren, J. Am. Chem. Soc. **72**, 4019 (1950).

The results reported here provide a comprehensive description of a variety of physical properties of  $\text{ThS}_2$  over a wide range of pressure and temperature conditions and set key benchmarks for the characterization and further assessment of this compound. These results explain existing experimental observations and offer insights for understanding the mechanisms underlying the variations of the structural and electronic properties of  $\text{ThS}_2$ . Such insights are crucial to evaluating the behavior and determining the role of  $\text{ThS}_2$  in various applications, including the thorium fuel reprocessing cycle. The theoretical predictions about the new structural phases and the electronic phase transitions in the pressure-temperature phase space of  $\text{ThS}_2$  call for further experimental verification and exploration.

#### Acknowledgments

This work is partially supported by the National Natural Science Foundation of China (Grant Nos. 91326105, 11574333, and U1532124), and the Strategic Priority Research Program of the Chinese Academy of Sciences (Grant No. XDA02040100). C.F.C. was supported in part by the DOE under Cooperative Agreement No. DE-NA0001982. The TMSR supercomputer center, the Supercomputing Center, CAS, and the Shanghai Supercomputing Center are acknowledged for allocation of computing time.

- <sup>14</sup> J. Graham and F. McTaggart, Aust. J. Chem. **13**, 67 (1960).
- <sup>15</sup> G. Amoretti, G. Calestani, and D. C. Giori, Z. Naturforsch. A **39**, 778 (1984).
- <sup>16</sup> L. Gerward, J. Staun Olsen, U. Benedict, H. C. Abraham, and F. Hulliger, High Press. Res. **13**, 327 (1995).
- <sup>17</sup> Y. Wang, J. Lv, L. Zhu, and Y. Ma, Phys. Rev. B 82, 094116 (2010).
- <sup>18</sup> G. Kresse and J. Furthmüller, Phys. Rev. B 54, 11169 (1996).
- <sup>19</sup> G. Kresse and J. Furthmüller, Comput. Mater. Sci. 6, 15 (1996).
- <sup>20</sup> P. E. Blöchl, Phys. Rev. B **50**, 17953 (1994).
- <sup>21</sup> G. Kresse and D. Joubert, Phys. Rev. B **59**, 1758 (1999).
- <sup>22</sup> J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. **77**, 3865 (1996).
- <sup>23</sup> L. A. Koscielski, E. Ringe, R. P. Van Duyne, D. E. Ellis, and J. A. Ibers, Inorg. chem. **51**, 8112 (2012).
- <sup>24</sup> I. R. Shein, K. I. Shein, and A. L. Ivanovskii, J. Nucl. Mater. **353**, 19 (2006).
- <sup>25</sup> D. Pérez Daroca, S. Jaroszewicz, A. M. Llois, and H. O. Mosca, J. Nucl. Mater. **437**, 135 (2013).
- <sup>26</sup> Y.-L. Guo, W.-J. Qiu, X.-Z. Ke, P. Huai, C. Cheng, H. Han, C.-L. Ren, and Z.-Y. Zhu, Phys. Lett. A **379**, 1607 (2015).
- <sup>27</sup> Y. Wang, J. Lv, L. Zhu, and Y. Ma, Comput. Phys. Com-

mun. 183, 2063 (2012).

- <sup>28</sup> J. Lv, Y. Wang, L. Zhu, and Y. Ma, Phys. Rev. Lett. **106**, 015503 (2011).
- <sup>29</sup> L. Zhu, H. Wang, Y. Wang, J. Lv, Y. Ma, Q. Cui, Y. Ma, and G. Zou, Phys. Rev. Lett. **106**, 145501 (2011).
- <sup>30</sup> H. Liu and Y. Ma, Phys. Rev. Lett. **110**, 025903 (2013).
- <sup>31</sup> D. Zhou, Q. Li, Y. Ma, Q. Cui, and C. F. Chen, J. Phys. Chem. C **117**, 5352 (2013).
- <sup>32</sup> Q. Li, D. Zhou, W. Zheng, Y. Ma, and C. F. Chen, Phys. Rev. Lett. **110**, 136403 (2013).
- <sup>33</sup> M. Zhang, H. Liu, Q. Li, B. Gao, Y. Wang, H. Li, C. F. Chen, and Y. Ma, Phys. Rev. Lett. **114**, 015502 (2015).
- <sup>34</sup> Q. Li, D. Zhou, W. Zheng, Y. Ma, and C. F. Chen, Phys. Rev. Lett. **115**, 185502 (2015).
- <sup>35</sup> L. Bai, Q. Li, S. A. Corr, M. Pravica, C. F. Chen, Y. Zhao, S. V. Sinogeikin, Y. Meng, C. Park, and G. Shen, Phys. Rev. B **92**, 134106 (2015).
- <sup>36</sup> H. J. Monkhorst and J. D. Pack, Phys. Rev. B **13**, 5188 (1976).
- <sup>37</sup> F. Birch, J. Geophys. Res. **91**, 4949 (1986).
- <sup>38</sup> K. Parlinski, Z. Q. Li, and Y. Kawazoe, Phys. Rev. Lett. 78, 4063 (1997).
- <sup>39</sup> A. Togo, F. Oba, and I. Tanaka, Phys. Rev. B **78**, 134106

(2008).

- <sup>40</sup> P. Zhang, B.-T. Wang, and X.-G. Zhao, Phys. Rev. B 82, 144110 (2010).
- <sup>41</sup> B.-T. Wang, P. Zhang, R. Lizárraga, I. Di Marco, and O. Eriksson, Phys. Rev. B 88, 104107 (2013).
- <sup>42</sup> J.-P. Dancausse, E. Gering, S. Heathman, and U. Benedict, High Press. Res. 2, 381 (1990).
- <sup>43</sup> M. Idiri, T. Le Bihan, S. Heathman, and J. Rebizant, Phys. Rev. B **70**, 014113 (2004).
- <sup>44</sup> B.-T. Wang, H. Shi, W.-D. Li, and P. Zhang, J. Nucl. Mater. **399**, 181 (2010).
- <sup>45</sup> H. X. Song, L. Liu, H. Y. Geng, and Q. Wu, Phys. Rev. B 87, 184103 (2013).
- <sup>46</sup> H. X. Song, H. Y. Geng, and Q. Wu, Phys. Rev. B 85, 064110 (2012).
- <sup>47</sup> W. Cochran, Phys. Rev. Lett. **3**, 412 (1959).
- <sup>48</sup> Y. Le Page and P. Saxe, Phys. Rev. B **65**, 104104 (2002).
   <sup>49</sup> M. Born and K. Huang, Dynamical Theory of Crystal Lat-
- tices Oxford Classic Texts in the Physical Sciences (Clarendon Press, Oxford, 1988).
- <sup>50</sup> Z.-J. Wu, E.-J. Zhao, H.-P. Xiang, X.-F. Hao, X.-J. Liu, and J. Meng, Phys. Rev. B **76**, 054115 (2007).