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Charge and spin order in the perovskite CaFe_{0.5}Mn_{0.5}O_{3}: Charge disproportionation behavior of randomly arranged Fe^{4+} Yoshiteru Hosaka, Noriya Ichikawa, Takashi Saito, J. Paul Attfield, and Yuichi Shimakawa Phys. Rev. B **94**, 104429 — Published 26 September 2016 DOI: 10.1103/PhysRevB.94.104429 Charge and spin order in the perovskite $CaFe_{0.5}Mn_{0.5}O_3$: Charge disproportionation behavior of randomly arranged Fe^{4+}

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Abstract

B-site-disordered perovskite CaFe_{0.5}Mn_{0.5}O₃ with unusually high valence Fe⁴⁺ was synthesized using a high-pressure technique. Fe⁴⁺ randomly distributed at the half of the *B* sites shows charge disproportionation to Fe³⁺ and Fe⁵⁺. The spins of Fe³⁺, Fe⁵⁺ and Mn⁴⁺ order below 90 K. Analysis of low-temperature neutron powder diffraction data revealed a G-type antiferromagnetic structure—where all the nearest neighboring spins of Fe³⁺, Fe⁵⁺, and Mn⁴⁺ couple antiparallel— and the small ordered moment of 0.58 μ_B reveals local charge ordering that gives rise to predominant Fe³⁺-O-Fe⁵⁺ antiferromagnetic arrangements. Despite the identical chemical compositions of CaFe_{0.5}Mn_{0.5}O₃ and Ca₂FeMnO₆, the magnetic structure of the present CaFe_{0.5}Mn_{0.5}O₃ is very different from the noncollinear one of layered *B*-site -ordered Ca₂FeMnO₆.

Introduction

At low temperature some perovskite structure oxides with the unusually high valence Fe^{4+} show characteristic charge behaviors relieving the instability of the high valence state. The simple perovskite CaFeO₃ and the *A*-site-ordered perovskite CaCu₃Fe₄O₁₂ are examples of such oxides, and they show a charge disproportionation (CD) transition in which Fe^{4+} changes to Fe^{3+} and Fe^{5+} ($2Fe^{4+} \rightarrow Fe^{3+} + Fe^{5+}$)¹⁻⁹. In the CD states at low temperature the Fe^{3+} and Fe^{5+} ions are ordered in a rock-salt manner, and magnetic moments of the Fe ions give complicated magnetic structures. In the charge disproportionated CaFeO₃, both ferromagnetic and antiferromagnetic interactions compete with each other¹⁰, and as the result, a helical magnetic structure is stabilized below 115 K, which is below the CD transition temperature^{11,12}. In the CD state of CaCu₃Fe₄O₁₂, on the other hand, a ferrimagnetic structure, where the *A*-site Cu spins couple antiferromagnetically with the *B*-site Fe spins, is stabilized¹³. Strong antiferromagnetic interactions between the *A*-site Cu²⁺ and *B*-site Fe³⁺ spins and between the Cu²⁺ and Fe⁵⁺ spins force to align the nearest neighbor Fe³⁺ and Fe⁵⁺ spins ferromagnetic.

We recently found that a similar CD occurred in the two-dimensional Fe^{4+} sublattice of the layered double perovskite Ca₂FeMnO₆, which consists of alternately stacked $Fe^{4+}O_6$ and MnO₆ octahedral layers^{14,15}. In that material the Fe⁴⁺ in the two-dimensional layers shows CD below about 200 K, and the resultant Fe³⁺ and Fe⁵⁺ are ordered in a checkerboard manner. Further cooling results in a magnetic transition at 95 K, and an unusual noncollinear ferrimagnetic spin structure is stabilized. The neighboring spins in the FeO₆ octahedral layers, *i.e.* the charge disproportionated Fe³⁺ and Fe⁵⁺ spins, couple antiferromagnetically, resulting in ferrimagnetism¹⁶.

In the examples described above, CD of the unusual high valence Fe^{4+} ions produces large $Fe^{3+}O_6$ and small $Fe^{5+}O_6$ octahedra alternately three-dimensionally (CaFeO₃ and CaCu₃Fe₄O₁₂) or two-dimensionally (Ca₂FeMnO₆). Thus, freezing of a Fe⁴⁺O₆ breathing phonon mode appears to be responsible for causing such unusual charge behaviors. Additionally, in the CD states at low temperature, the nearest neighbor Fe³⁺ and Fe⁵⁺ spin pairs play important role in the magnetism, although the observed magnetic structures of the compounds are rather complicated due to the competing magnetic interactions. Considering the long-range charge and spin orders in the CD states, rigidly-arranged neighboring Fe⁴⁺ network seems to be important to cause the charge disproportionation behavior. Our primary concern is thus "to break the rigid Fe⁴⁺ network". For this purpose we have synthesized CaFe_{0.5}Mn_{0.5}O₃, which has a chemical composition identical to that of the layered double perovskite Ca₂FeMnO₆ but a disordered arrangement of *B*-site Fe and Mn. The compound contains Fe⁴⁺ randomly distributed at the half of the *B* sites, and thus stabilization of the long-range charge and spin orders of Fe³⁺ and Fe⁵⁺ is expected to be challenging. In the present study, we found, surprisingly, that this *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃ also

shows a CD of Fe⁴⁺ at a low temperature. We have used neutron powder diffraction (NPD) analysis to investigate the magnetic structure of this newly synthesized compound, and here we compare the crystal and magnetic structures of the *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃ with those of the layered *B*-site-ordered Ca₂FeMnO₆. We also discuss the CD and magnetic transition behaviors of this new phase.

Experiments

A polycrystalline sample of $CaFe_{0.5}Mn_{0.5}O_3$ was synthesized in a solid-state reaction under a high-pressure and high-temperature condition. A mixture of stoichiometric amounts of $CaCO_3$, Fe_2O_3 , and $MnCO_3$ was calcined at 1250°C in air. The calcined powder was then sealed in a Pt capsule with the oxidizing agent KClO₄ and held at 5.3 GPa and 1100°C for 30 minutes before quenching to room temperature. The pressure was then reduced slowly to ambient and the reacted sample was removed from the capsule, washed several times with distilled water to remove residual KCl and KClO₄, and was then dried.

The crystal and magnetic structures were analyzed by measuring NPD patterns. Time-of-flight NPD experiments were performed at the WISH beamline of ISIS at RAL. The data were analyzed with the FullProf Rietveld program¹⁷. The valence state of Fe was estimated with Mössbauer spectroscopy measurements. The spectra were measured in transmission geometry with a constant-acceleration spectrometer using a ⁵⁷Co/Rh radiation source and were fitted with Lorentzian functions. Magnetic properties were measured with a commercial Quantum Design MPMS SQUID magnetometer.

Results and Discussion

Figure 1(a) shows the NPD pattern at room temperature and the result of crystal structure refinements for the obtained CaFe_{0.5}Mn_{0.5}O₃. The corresponding pattern and results for the layered double perovskite are shown in Fig. 1(c). In the pattern for CaFe_{0.5}Mn_{0.5}O₃ neither superstructure peaks related to the Fe/Mn ordering at the *B*-site (e.g. those at d = 3.5 and 7.5 Å seen in the pattern for layered Ca₂FeMnO₆) nor diffuse intensities due to the short range orderings of the *B*-site cations are evident. The observed diffraction pattern of CaFe_{0.5}Mn_{0.5}O₃ is well reproduced with a GdFeO₃-type perovskite structure model with a space group *Pnma* and a $\sqrt{2a} \times 2a \times \sqrt{2a}$ unit cell (*a* represents the lattice constant of the simple perovskite structure, ~3.8 Å), and the refinement results are listed in Table 1. The Fe and Mn ions occupy the 4*b* site, and the refined occupancies are 51.2(2)/48.8 % for Fe/Mn. Note also that each refined lattice parameter is, in keeping with Vegard's law, between the parameters corresponding to the two end-composition compounds of CaFeO₃ and CaMnO₃^{8,18-20}. The above results confirm the random distribution of the Fe and Mn ions at the *B* site in CaFe_{0.5}Mn_{0.5}O₃.

The valence state of Fe in CaFe_{0.5}Mn_{0.5}O₃ was verified from the ⁵⁷Fe Mössbauer spectra. As shown in Fig. 2 (a), the observed spectrum at room temperature is well fitted with a single paramagnetic component with the isomer shift (*IS*) of 0.04 mms⁻¹. This small value is typical for unusual Fe⁴⁺ and is also very close to that observed in the layered Ca₂Fe⁴⁺MnO₆ (0.02 mms⁻¹). The absence of any trace of Fe³⁺ in the spectrum confirms that all the Fe ions in the CaFe_{0.5}Mn_{0.5}O₃ were Fe⁴⁺. Therefore, two samples having an identical chemical composition with the unusually high valence Fe⁴⁺ at room temperature can be made: one is the present *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃ and the other is the layered *B*-site-ordered Ca₂FeMnO₆ reported in Refs (14-15). Comparing the Mössbauer spectra, one sees that the local environments of Fe⁴⁺ in the two samples appear to be quite similar to each other in spite of the difference in the degree of long range order.

Importantly, as shown in the Mössbauer spectra at 3 K [Fig. 2(b)], Fe⁴⁺ in the *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃ shows CD at low temperatures. The observed spectrum is reproduced with two magnetically ordered sextets with almost equal intensities. Their *IS*s are -0.01 and 0.35 mms⁻¹ and the corresponding hyperfine fields (*HF*) are 28.8 and 44.3 T, which are close to the values for charge-disproportionated Fe⁵⁺ (*IS* = -0.01 mms⁻¹ and *HF* = 27.4 T) and Fe³⁺ (*IS* = 0.35 mms⁻¹ and *HF* = 43.4 T) in the layered *B*-site-ordered Ca₂FeMnO₆. It is surprising that all the Fe⁴⁺ ions randomly distributed at the half of the *B* sites also show CD. CD of Fe⁴⁺ has been reported to occur in the three-dimensional cubic (CaFeO₃ and CaCu₃Fe₄O₁₂) and two-dimensional square (Ca₂FeMnO₆) sublattices of Fe⁴⁺, producing respectively the rock-salt-type and checkerboard-type orderings of Fe³⁺ and Fe⁵⁺ (Ref. 12). The present results, however, show that a complete three or two dimensional network of the unusually high valence Fe⁴⁺ ions is not necessary for the CD transition. The result also implies that the CD transition is induced by the electronically instability, not by the instabilities of phonons.

It is difficult to determine the distribution of Fe^{3+} and Fe^{5+} in the CD state from the diffraction data. In the NPD pattern at 2K shown in Fig. 1(b), neither apparent superstructure nor distortion from the room temperature crystal structure is detected, and the nuclear diffraction intensities are essentially the same as those at 300 K. Although the CD transition is often accompanied by a crystal structure transition, this appears to be suppressed by substitution with Mn^{4+} in the present *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃. This also causes difficulty in determining the CD transition temperature form the structural change.

The magnetic properties and structure of charge-disproportionated $CaFe_{0.5}Mn_{0.5}O_3$ are of particular interest. As shown by the temperature dependence of magnetic susceptibility [Fig. 3(a)], $CaFe_{0.5}Mn_{0.5}O_3$ shows a magnetic transition around 90 K. Note, it is not clear at the current stage whether the magnetic transition occurs below the CD transition as seen in the layered double perovskite Ca_2FeMnO_6 , because the CD transition temperature in the present

compound has not been determined from the structural changes. However, the experimental results, that Mössbauer spectrum under the magnetic transition temperature consists of the two components and that the magnetic susceptibility shows no anomalies below the magnetic transition temperature, strongly suggest that at least the magnetic transition occurs in the CD state. Therefore, the magnetic transition temperature of $CaFe_{0.5}Mn_{0.5}O_3$ is the same as or lower than the CD transition temperature. The 90-K magnetic transition temperature is very close to the magnetic transition temperature of the layered double perovskite Ca₂FeMnO₆¹⁴⁻¹⁶. Field dependence of the magnetization revealed that the compound has a small spontaneous magnetization of about 0.2 μ_B per CaFe_{0.5}Mn_{0.5}O₃, which is also very close to that of the layered ordered Ca₂FeMnO₆ [Fig. 3(b)]. Local magnetic environments of the and charge-disproportionated Fe^{3+} and Fe^{5+} evaluated from the *HF* values in the Mössbauer spectrum [Fig. 2(b)] are also not so different from those in the layered B-site-ordered compound. It is thus notable that the two compounds with different B-site cation ordered structures show similar magnetic behaviors.

In contrast, magnetic reflections of CaFe_{0.5}Mn_{0.5}O₃ in the NPD pattern obtained at 2 K [Fig. 1(b)] are very different from those of the layered B-site-ordered Ca₂FeMnO₆, indicating that the magnetic structures of the two compounds are not identical. The present CaFe_{0.5}Mn_{0.5}O₃ shows an apparent magnetic superstructure peak at $d \approx 4.3$ Å, while the layered B-site-ordered Ca₂FeMnO₆ shows three strong magnetic superstructure peaks indexed as 010/001, 110/011, and 210/201 [Fig. 1(d)]. The peak at $d \approx 4.3$ Å corresponds to 110 or 011 reflections of a $\sqrt{2a \times 2a \times \sqrt{2a}}$ unit cell, indicating the presence of a G-type antiferromagnetic (AFM) spin structure, where all the nearest neighboring spins couple antiparallel to each other. The fit of a G-type AFM spin structure to the NPD data at 2 K gives an average refined moment of 0.58 μ_B per cation. This ordered moment is very small compared to the ideal 3.5 μ_B average magnetic moment for 25% Fe^{3+} (S = 5/2), 25% Fe^{5+} (S = 3/2), and 50% Mn^{4+} (S = 3/2) ions randomly distributed at the B sites and demonstrates that there is large disordered component to the magnetic order. It is important to point out that the present *B*-site disordered CaFe_{0.5}Mn_{0.5}O₃ shows higher structural symmetry due to randomly distributed Fe and Mn ions than the layer-ordered Ca₂FeMnO₆. Because the collinear G-type AFM structure is much simpler than the noncollinear magnetic structure observed in the layer-ordered Ca₂FeMnO₆, the difference in the magnetic structures between the two compounds appears to be reasonable in symmetrical stability point of view.

This simple G-type AFM spin structure indicated by the NPD result is apparently incompatible with the spontaneous magnetization of 0.2 μ_B observed in the magnetization measurement. There are two plausible explanations for this. The first is that weak ferromagnetism is symmetry-allowed in *Pnma* perovskites with G-type AFM order, and is

observed in many $RFeO_3$ and $RCrO_3$ materials for various rare earth elements R^{21-23} . A second explanation is that ferrimagnetism arise from the formation of charge disproportionated Fe^{3+} and Fe⁵⁺ ordering. This is the more likely origin given the similarity of the magnetic properties to those of the cation and charge ordered Ca₂FeMnO₆. Local charge ordering below the CD transition temperature favors Fe³⁺-O-Fe⁵⁺ arrangements over the Fe³⁺-O-Fe³⁺ and Fe⁵⁺-O-Fe⁵⁺ alternatives, so in the G-type AFM spin structure below the magnetic transition temperature, which is also below the CD transition temperature, the Fe^{3+} magnetic moments are aligned in the opposite direction to the Fe⁵⁺ moments, giving net ferrimagnetic moments (Fig. 4). The Mn⁴⁺ cations are equally substituted for Fe³⁺ and Fe⁵⁺, and so their ordered magnetic moments cancel each other, and so do not give net magnetization. This model predicts an ideal magnetization of 0.5 $\mu_{\rm B}$ for CaFe_{0.5}Mn_{0.5}O₃, which is consistent with the observed magnetization of 0.2 $\mu_{\rm B}$ given further reductions due to disorder. Although this model assumes dominant AFM Fe³⁺-O-Fe⁵⁺ interactions, the observed helical magnetic structure of CaFeO₃ shows that these connections can lead to competing AFM and ferromagnetic (FM) interactions. Competition between AFM and FM Fe³⁺-O-Fe⁵⁺ and also Mn⁴⁺-O-Fe⁵⁺ interactions, plus the structural disorder present in CaFe_{0.5}Mn_{0.5}O₃ results in a large proportion of the spins being disordered or remaining dynamic below the magnetic ordering transition. This is consistent with the 5 K paramagnetic susceptibility of $CaFe_0 SMn_0 SO_3$ being greater than that of B-site cation-ordered Ca₂FeMnO₆, as shown by the high field *M*-*H* slopes in Fig. 3(b).

It is worth noting again that the Fe^{3+} and Fe^{5+} Mössbauer spectra and the magnetic behaviors of the present CaFe_{0.5}Mn_{0.5}O₃ are quite similar to those of the layered *B*-site-ordered Ca₂FeMnO₆ with the same chemical composition but a different magnetic structure. Although the local aliovalent Fe^{3+} -O-Fe⁵⁺ arrangements in the CD states play an important role in determining the magnetic structures in both compounds, the *B*-site ordering makes the magnetic structures different.

Conclusions

The *B*-site-disordered perovskite-structure oxide $CaFe_{0.5}Mn_{0.5}O_3$ was obtained using high-pressure synthesis. The NPD structure analysis and the Mössbauer spectroscopy results revealed that the compound contains Mn^{4+} and unusually high valence Fe^{4+} ions randomly distributed over the *B* sites, and that the chemical composition of this oxide is identical to that of the layered *B*-site-ordered Ca₂FeMnO₆. Like Ca₂FeMnO₆, CaFe_{0.5}Mn_{0.5}O₃ shows CD of Fe⁴⁺ at low temperatures and shows magnetic transition at 90 K. The CD in the *B*-site disordered CaFe_{0.5}Mn_{0.5}O₃ suggests that freezing of breathing mode phonon is not necessary for the transition. Magnetic structure analysis with the NPD data revealed that CaFe_{0.5}Mn_{0.5}O₃ has a G-type AFM spin structure although a small spontaneous magnetization is observed in magnetization measurements. This is consistent with local charge ordering that gives rise to Fe^{3+} -O-Fe⁵⁺ antiferromagnetic interactions so inexact cancellation of opposing Fe^{3+} and Fe^{5+} spins produces ferrimagnetism. The magnetic structure of $CaFe_{0.5}Mn_{0.5}O_3$ is thus very different from the noncollinear magnetic structure of Ca_2FeMnO_6 .

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Table 1. Refined crystal structure parameters of CaFe_{0.5}Mn_{0.5}O₃ at room temperature (obtained in Rietveld analysis of the NPD data.) Space group, *Pnma*, a = 5.3177(3) Å, b = 7.5035(4) Å, and c = 5.3041(3) Å. Rwp = 4.59%.

| Atom | x | Y | Ζ | $B(\text{\AA}^2)$ | occupancy |
|------|-----------|-----------|-----------|-------------------|-----------|
| Ca | 0.0282(6) | 0.25 | 0.993(1) | 0.94(6) | 1.0 |
| Fe | 0.0 | 0.0 | 0.5 | 0.11(7) | 0.510(2) |
| Mn | 0.0 | 0.0 | 0.5 | 0.11(7) | 0.490(2) |
| 01 | 0.2892(5) | 0.0350(4) | 0.7153(6) | 1.26(6) | 1.0 |
| 02 | 0.4879(6) | 0.25 | 0.0617(7) | 0.70(7) | 1.0 |

Figure captions

Figure 1. Neutron powder diffraction patterns and refinement results for *B*-site-disordered perovskite $CaFe_{0.5}Mn_{0.5}O_3$ (a) at 300 and (b) 2 K. Corresponding patterns and results for the layered double perovskite Ca_2FeMnO_6 at 300 (c) and (d) 2 K are reproduced from (Ref. 14). The dots, black lines, blue lines, and vertical marks represent observed patterns, calculated patterns, differences between observed and calculated intensities, and Bragg refraction points, respectively. Marked peaks are typical magnetic reflections.

Figure 2. Mössbauer spectra of *B*-site-disordered $CaFe_{0.5}Mn_{0.5}O_3$ measured (a) at room temperature and (b) 3 K. The circles are observed data, and the spectra are fitted with (a) a singlet component and (b) two sextets corresponding to Fe^{3+} and Fe^{5+} components.

Figure 3. Magnetic properties of *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃ (red) together with those of the layered *B*-site-ordered double perovskite Ca₂FeMnO₆ (blue). (a) Temperature dependence of magnetic susceptibility χ . (b) Magnetic field dependence of magnetic moment.

Figure 4. Idealized model for the spin and charge ordering in *B*-site-disordered CaFe_{0.5}Mn_{0.5}O₃. Red, blue, and white arrows represent magnetic spins of Fe³⁺, Fe⁵⁺, and Mn⁴⁺, respectively. G-type antiferromagnetism over local Fe³⁺-O-Fe⁵⁺ arrangements give rise to the observed ferrimagnetism.









