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Crystal structure and partial Ising-like magnetic ordering of orthorhombic Dy₂TiO₅

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1. Abstract

The structure and magnetic properties of orthorhombic Dy_2TiO_5 have been investigated using Xray diffraction, neutron diffraction, and ac/dc magnetic susceptibility measurements. We report a continuous structural distortion below 100 K characterized by negative thermal expansion in the [0 1 0] direction. Neutron diffraction and magnetic susceptibility measurements revealed 2D magnetic ordering begins at 3.1 K which is followed by a 3D magnetic transition at 1.7 K. The magnetic structure has been solved through a representational analysis approach and can be indexed with the propagation vector $k = [0 \frac{1}{2} 0]$. The spin structure corresponds to a coplanar model of interwoven two-dimensional "sheets" extending in the [0 1 0] direction. The local crystal field is different for each Dy^{3+} ion (Dy1 and Dy2), one of which possesses strong uniaxial symmetry indicative of Ising-like magnetic ordering. Consequently, two succeeding transitions under magnetic field are observed in the ac susceptibility which are associated with flipping each Dy3+ spin independently.

2. Introduction

Insulators of general formula A₂TiO₅ have attracted significant attention in recent years due to their structural and chemical diversity. Depending on the A-site cation size and/or sample synthesis method, these complex oxides can form cubic, orthorhombic, hexagonal, and monoclinic polymorphs without altering the stoichiometry [1-5]. As a result, these materials are suitable for a wide array of technological applications including potential actinide hosts for long-term storage in a geological repository [6-8], ion conductors for fuel cells and oxygen sensors [2], and nanoparticles in oxide dispersion strengthened (ODS) steels [9]. These materials can readily incorporate rare-earth elements into their A-site resulting in many complex magnetic

interactions. They are frequently end members in solid solution series' of general formula $Ln_2(Ti_{2-x}Ln_x)O_{7-x/2}$ (often referred to as "stuffed pyrochlores") in which magnetic lanthanide elements are incrementally "stuffed" into the Ln₂Ti₂O₇ pyrochlore matrix increasing the relative number of spins involved in magnetic interactions [10]. Ho₂Ti₂O₇, for example, forms the wellstudied spin-ice state at low temperatures with locally ordered magnetic moments analogous with protons in water ice [11,12]. Stuffing additional magnetic Ho³⁺ atoms into the pyrochlore causes interesting, and seemingly counterintuitive, behavior [13]. A fully stuffed Ho₂(Ti_{2-x}Ho_x)O_{7-x/2} corresponds to $Ho_{2,67}Ti_{1,33}O_{6,67}$ (x = 0.67) or Ho_2TiO_5 . Depending on the sample synthesis method, this disorders the pyrochlore (at least partially) into the isometric defect-fluorite average structure with cation mixing between the Ho³⁺ and Ti⁴⁺ crystallographic sites. The structure is effectively converted from a network of corner sharing tetrahedra, essential for the spin-ice state, to a network of side-sharing tetrahedra with intrinsic disorder. Despite the increased concentration of magnetic moments and partially disordered structure, the zero-point entropy per spin characteristic of frustration remains more or less unchanged from that of the original pyrochlore spin-ice [13]. Other stuffed pyrochlores with cubic Ln_2TiO_5 endmembers have also been studied in detail for both structural and magnetic properties [10]. Interestingly, none of these displayed evidence of long range magnetic order above 2 K. Magnetic interactions were shown to be predominately antiferromagnetic while Ln^{3+} spins were shown to be strongly anisotropic similar to spin-ice.

Magnetic properties of the orthorhombic polymorphs, however, have not been fully characterized. $Dy_2Ti_2O_7$, for example, is another prototypical spin-ice [14] that can be readily transformed into an orthorhombic polymorph through the stuffing procedure [3]. The fully stuffed Dy_2TiO_5 endmember (P*nam* space group) is an important material in the nuclear power

industry where, due to dysprosium's large thermal neutron absorption cross section ($\sigma_a = 997$ b), it is used as a neutron absorber in control rods in Russian VVER type reactors [15,16]. This, however, makes Dy-based compounds difficult to characterize with neutron diffraction which has perhaps deterred detailed studies into any magnetic structure. Dy₂TiO₅ is isostructural with orthorhombic Y_2TiO_5 first reported by Mumme [1] in which Y^{3+} (or Dy^{3+} in this case) and Ti^{4+} are 7- and 5-coordinated with oxygen respectively (Fig. 1a). This mixture of 7- and 5-fold coordination is rather unique as compared with other rare earth titanates of the same ternary system (such as pyrochlore or layered perovskites) in which Ln³⁺ and Ti⁴⁺ usually form distorted cubes and octahedra respectively. Here the 7-coordinated Ln^{3+} ions are in a monocapped octrahedral configuration while 5-coordinated Ti⁴⁺ form square pyramidal polyhedra (Fig. 1b and 1c). All atoms are located in distinct 4c Wyckoff positions each of which requiring an x- and zcoordinate to describe the atomic positions within the unit cell. This creates structural flexibility allowing for significant distortions of local polyhedra. While both Dy³⁺ ions form similar monocapped octahedra locally, they are coordinated differently at longer length scales. The first Dy3+ monocapped octahedron, from now on referred to as Dy1, is edge-sharing with five additional monocapped octahedra and two square pyramids and is corner sharing with the apex of two square pyramids and the basal corner of one square pyramid. The second Dy³⁺ octahedron, Dy2, is edge-sharing with seven monocapped octahedra and two square pyramids and is corner sharing with the basal corner of one square pyramid. The magnetic properties of this orthorhombic polymorph of Dy₂TiO₅ remains unexplored to this point. One consequence of the differing connectivity for each Dy³⁺ ion creates is the distinct possibility of magnetic moments that can order independently for both Dy1 and Dy2. In this paper we investigate the

low temperature crystal structure and magnetic order of orthorhombic Dy_2TiO_5 using a combination of X-ray/neutron diffraction and ac/dc magnetic susceptibility measurements.

3. Experiment

3.1 Sample Synthesis

Stoichiometric mixtures of Dy_2O_3 and TiO_2 were combined in the following solid-state reaction:

$$Dy_2O_3 + TiO_2 \rightarrow Dy_2TiO_5$$
.

Dy₂O₃ was prefired at 1000°C for 8 hours to remove any adsorbed water. Powders were ground, mixed using an acetone slurry in a mortar and pestle, and subsequently cold pressed into a pellet using a hydraulic press upon drying. The pellet was then loaded into an alumina crucible and fired at 1200°C for 12 hours. The sample was allowed to cool to room temperature and was then reground, pressed and fired at 1500°C for an additional 12 hours. The heating and cooling rates were kept below 5°/minute. The final pellet was ground into a fine powder and checked for purity with X-ray diffraction which revealed no evidence of impurities.

3.2 Room Temperature Neutron Diffraction

Structural characterization at 300 K was carried out using neutron diffraction at the NOMAD beamline [17] at the Spallation Neutron Source at Oak Ridge National Laboratory in Oak Ridge, Tennessee (United States). Despite the large thermal neutron absorption cross-section there is a window of relatively high transmission extending from about 0.25 Å to 0.7 Å in wavelength and Dy2TiO5 was successfully measured using a small sample size. The sample was first loaded into a 2mm diameter quartz capillary filled to a height of 1 cm and measured for a

total of 140 minutes. An identical, empty quartz capillary was also measured for 140 minutes to serve as a background. Rietveld refinement was performed on diffraction patterns from detector bank 5 with an average scattering angle 2 θ of 154 degree (Q-range from 4 – 49.9 Å-1) using the FullProf code [18] to characterize the crystal structure and determine the unit cell parameters and atomic positions within the unit cell. Neutron absorption was accounted for in FullProf through the use of a refinable absorption correction parameter for time-of-flight data with cylindrical geometry.

3.3 Low Temperature X-ray and neutron Diffraction

The low temperature X-ray diffraction (XRD) patterns were measured with a HUBER X-ray powder diffractometer. Unit cell parameters were determined by Rietveld refinement using FullProf. All neutron diffraction measurements at 20K and below were performed at the Neutron Powder Diffractometer beamline (HB-2A) at the High Flux Isotope Reactor (HFIR) at Oak Ridge National Laboratory in Oak Ridge, Tennessee (United States). Custom flat-plate holders with a thickness of 0.15 mm were machined from Aluminum stock to minimize absorption from Dy atoms. A wavelength of 2.4136 Å was selected using a germanium wafer-stack monochromator to provide higher resolution and access to magnetic Bragg peaks at low scattering angles. Data was collected for 5 hours at 0.3 K and 20 K and 4 hours at intermediate temperatures. Rietveld refinement was performed at 20 K (above the magnetic transition temperature) to determine atomic positions and the unit cell parameters of the crystal structure. The magnetic structure was characterized using representational analysis. The magnetic propagation k-vector was determined using the magnetic peaks at 0.3 K with the k-search function in Fullprof. Irreducible representations and basis vectors were obtained using the SARAh representational analysis code [18].

3.4 Magnetic susceptibility measurements

The dc susceptibility measurements were performed using a Quantum Design superconducting quantum interference device (SQUID) magnetometer. The ac susceptibility was measured at the National High Magnetic Field Laboratory with the conventional mutual inductance technique at frequencies between 80 Hz and 700 Hz.

4. Results & Discussion

4.1 Structural Characterization

4.1.1 Neutron Diffraction

The previously reported orthorhombic polymorph characteristic of lanthanide titanates of the Ln₂TiO₅ composition agrees very well with the measured neutron diffraction data of Dy₂TiO₅ at room temperature (Fig. 2). The unit cell parameters were determined to be a = 10.3722(2), b = 3.71985(7), and c = 11.2379(2) Å. In general the atom positions agree well with those reported in [3] determined by synchrotron XRD, however the uncertainty is reduced by nearly an order of magnitude for the oxygens likely due to the use of neutrons in the present study (Table 1). The mean Dy-O bond length, <Dy-O>, differs for each Dy site (2.328(1) Å and 2.345(1) Å respectively). Nearest neighbor Dy atoms (Dy1 – Dy2 and Dy2 – Dy2) form two-dimensional "sheets" extending in the [0 1 0] direction (Fig. 3a).

There is no evidence of a structural transformation or change of *Pnma* space group down to 0.3 K (although additional diffraction peaks emerge at ~1.75 K due to a magnetic transition, discussed below). The unit cell volume contracts by 0.90(5) % and <Dy-O> bond length are

reduced to 2.316(2) and 2.320(2) Å for Dy1 and Dy2 respectively (Table 1). Interestingly, Dy2 – Dy2 pairs split and are no longer nearest neighbors. Nearest neighbor Dy atoms (Dy1 – Dy2) now form two-atom "ladders" in the [0 1 0] direction (Fig. 3b). The axial positions for the two monocapped octahedra display different temperature dependence. The O1-Dy2-O3 bond angle becomes increasingly distorted at low temperatures while the O1-Dy1-O2 is much more ridged with only minor temperature dependence (Fig. 4). At 300 K both the O1-Dy1-O2 and O1-Dy2-O3 bond angles are close 180° (177.4(2) and 174.6(2) respectively). The bond angle significantly decreases for O1-Dy2-O3 as the temperature is lowered while it slightly increases for O1-Dy2-O2.

4.1.2 X-ray Diffraction

X-ray diffraction measurements also show no change in *Pnma* space group down to 10 K. The structure does, however, become continuously distorted at low temperatures (Fig. 5). The is particularly evident in *b*, which shows negative thermal expansion below 100 K (Fig. 5a). This has little effect on the unit cell volume as *b* is only $\approx 1/3^{rd}$ as large as *a* and *c* which do not show as significant of a distortion (Fig. 5b). It does, however, indicate that the Dy "ladders" described in Fig. 3b become increasingly stretched along [0 1 0] at low temperatures, which could suppress any spin canting in that direction.

4.2 Magnetic Characterization

4.2.1 dc Susceptibility

Magnetic susceptibility measurements indicate an antiferromagnetic transition with a transition temperature of 3.5 K (Fig. 6a). There is no divergence in zero-field-cooled (ZFC) and field-cooled (FC) measurements below this temperature (not shown) suggesting the absence of any irreversibility. The susceptibility follows the Curie-Weiss law above 3.5 K indicating

paramagnetic behavior. The effective magnetic moment (μ_{eff}) for Dy³⁺ was evaluated to be 10.55 μ_{B} using the Curie constant, *C*, extracted from the fit to the Curie-Weiss law. This agrees well with the moment for free Dy³⁺ ions which has a value of 10.63 μ_{B} . The Curie-Weiss temperature (θ_{CW}) was evaluated to be -10.8 K, suggesting antiferromagnetic interactions. Magnetization measurements as a function of increasing field show a saturation far below the effective moment for each Dy³⁺ (Fig. 6b). This is indicative of strong anisotropy for Dy³⁺ spins similar to that observed for the cubic Dy₂(Ti_{2-x}Dy_x)O_{7-x/2} and Ho₂(Ti_{2-x}Ho_x)O_{7-x/2} polymorphs [10,19-21]. Interestingly, magnetization measurements in these previous studies saturate at half (or slightly below) the moment for free Dy³⁺ or Ho³⁺. The orthorhombic polymorph in the present study, however, saturates at closer to 65% of the moment for free Dy³⁺ suggesting that anisotropy is partially relieved relative to spin-ice in pyrochlore.

4.2.2 ac Susceptibility

Zero-field ac magnetic susceptibility measurements also show evidence of a paramagnetic to antiferromagnetic transition beginning at 3.1 K as noted by the sluggish downturn in the real part of magnetic susceptibility, χ' (Fig. 7a). There exists only weak frequency dependence suggesting the absence of glass/ice-like dynamics. The magnetic structure is, however, strongly field dependent. There are two peaks in the field scan performed at 0.3 K (Fig. 7b). Each of the moments on both Dy³⁺ atoms (Dy1 and Dy2 discussed earlier) are likely polarized by the magnetic field independently as the magnitude of the two peaks in the field scan is identical. Cooling in the presence of a 1 T magnetic field slightly sharpens the transition and lowers the maximum to 1.1 K (Fig. 7c). A 2 T magnetic field suppressed susceptibility and further lowers the maximum to 0.8 K. Larger fields completely dampen the magnetic transition.

There is a small kink at 1.6 K which is first apparent in the 2 T measurement. This is an artifact due to He^3 condensation as the position is invariant at stronger magnetic fields.

A peak in susceptibility is commonly assigned to the onset of long-range order. However, the downturn is broader than expected for a transition to long range magnetic order. There is also an observable inflection in χ ' below the maximum at 3.1 K for the zero field measurements of Dy₂TiO₅ in Fig. 7a and 7c. This is most apparent when looking at $d\chi'/dT$ as sharp peaks are evident at 1.7 K and 0.6 K for 0 and 1 T measurements respectively (Fig. 7d). The rapid increase in $d\chi'/dT$ at ~3 K corresponds to the maximum observed in χ' . It has been previously argued that the onset of long range antiferromagnetic ordering can be better predicted by a peak in the first derivative of χ' [22] suggesting that long range ordering may not begin until 1.7 K. This possibly implies that short range ordering at 3.1 K precedes the onset of long range ordering at 1.7 K. However, because the ac susceptibility shows little to no frequency dependence, this could also indicate a shift in the dimensionality of magnetic order (*i.e.* a 2D-3D transition) with $T_{N,2D} = 3.1$ K and $T_{N,3D} = 1.7$ K. Low temperature neutron diffraction provides more insight into the dimensionality of the spin structure.

4.2.3 Magnetic Neutron Diffraction

A magnetic transition is confirmed with neutron diffraction experiments (Fig. 8a). Comparing the background of the diffraction patterns taken at 20 K and 3K reveals the onset of local magnetic ordering as noted by the appearance of a broad, diffuse peak centered at $2\theta \approx$ 22.5°. Strong resolution limited Bragg peaks are apparent at 0.3K. These peaks can be indexed with a propagation vector $k = [0 \frac{1}{2} 0]$. The diffuse peak at 3 K and the (100) magnetic peak at 0.3 K are centered at the same scattering angle but with a different peak shape (Fig. 8b). To test the lower dimensionality suggested by susceptibility measurements, the diffuse peak was fit with a Warren function characteristic of 2D magnetic ordering [23-26]:

$$P(\theta) = KmF_{hk}^2 \frac{(1+\cos^2 2\theta)}{2(\sin\theta)^{\frac{3}{2}}} \left(\frac{\xi}{\lambda\sqrt{\pi}}\right)^{\frac{1}{2}} F(a)$$
(1)

where

$$F(a) = \int_0^{20} exp[-(x^2 - a)^2] dx$$
⁽²⁾

and

$$a = \frac{2\xi\sqrt{\pi}}{\lambda}(\sin\theta - \sin\theta_0) \tag{3}$$

in which *K* represents a scale factor, *m* the multiplicity of reflection, F_{hk}^2 the magnetic structure factor, ξ the spin-spin correlation length, and θ_0 the centroid of the diffuse peak. The integral in F(a) was evaluated numerically and agrees with values reported in ref. [23]. The Warren function fits the diffuse peak well with $\xi \approx 22$ Å (Fig. 9). This agrees with the susceptibility measurements and is strongly suggestive of low dimensionality ordering. A simple linear background was included in the fitting procedure consistent with ref. [26]. Although local ordering begins at 3K, a long range magnetic transition (T_N) is not apparent until 1.7 K as noted by the temperature dependence of the (100) magnetic peak intensity (Fig. 8b inset) explaining the sluggish transition observed in susceptibility (Fig. 7a) and the sharp maximum observed in the 1st derivative curve (Fig. 7d). This also suggests that Dy₂TiO₅ is only moderately frustrated, as the frustration parameter, *f*, (defined as $|\theta_{CW}|/T_N$) is equal to 6.4. Each Dy atom occupies a 4*c* Wyckoff site within the *Pnma* symmetry creating 4 equivalent positions for each Dy atom (translations are shown below).

atom 1:

$$x, y, z$$

atom 2: $x + \frac{1}{2}, -y + \frac{1}{2}, -z + \frac{1}{2}$
atom 3: $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$
atom 4: $-x + \frac{1}{2}, -y + 1, z + \frac{1}{2}$

There exist two equivalent irreducible representations (IRs) each with 12 basis vectors (ψ_n). One IR was ultimately chosen for magnetic characterization. The magnetic state is therefore described by a linear combination of 12 basis vectors (6 basis vectors each for Dy1 and Dy2, shown in Table 2). The coefficients on ψ_1 must necessarily have opposite signs for Dy1 and Dy2 or else there are prominent forbidden reflections at $2\theta = 18.7, 33.1, 42.5$ and 54.9 among other minor reflections. This also applies to the coefficients for ψ_4 . Conversely, the coefficients on ψ_2 for Dy1 and Dy2 must necessarily have the *same* sign to eliminate forbidden reflections at $2\theta =$ 52.5 and 66.2. An analogous relationship holds for ψ_5 . The coefficients on ψ_3 must also be of opposite signs for Dy1 and Dy2 to remove forbidden reflection at $2\theta = 33.1$, 42.5 and 52.5 among others which also applies to ψ_6 . There remain, however, low intensity forbidden reflections indicating that any moment canting in the [0 1 0] direction is unlikely and this component was fixed at zero. This agrees with the negative thermal expansion observed in Fig. 5a in which Dy atoms become increasingly separating along b at low temperatures. Assuming that C_n is equal for both Dy1 and Dy2 atoms despite them occupying crystallographically independent sites produces a reasonable fit to the experimental data ($R_{WP} = 18.3$, see Table 3).

The calculated moment is 8.05(12) μ_B which is reasonable but still less than the ordered moment of 10 μ_B for free Dy³⁺ ions suggesting that moments are not fully saturated even at 0.3 K or are dampened by crystal field effects. The fit is significantly improved by removing the constraint that moments be equal for Dy1 and Dy2 atoms ($R_{WP} = 13.8$, Fig. 10). This results in a magnetic moment of 8.89(28) μ_B and 7.76(31) μ_B for Dy1 and Dy2 respectively (Table 3). The spin directions within the magnetic structure in both scenarios follows the underlying interwoven 2D "sheets" created by the Dy ions (Fig. 11). This ordering could explain the 2D-3D magnetic transition suggested earlier. One possible mechanism is that moments order on 2D ladders created by nearest neighbor Dy1-Dy2 pairs at 3 K (Fig. 3b and grey lines in Fig. 11) but do not interact with other pairs on longer length scales. Beginning at 1.7 K locally ordered moments on these ladders interact with neighboring Dy1-Dy2 pairs forming the interwoven structure shown in Fig. 11. It should be noted that there is an additional magnetic peak at $2\theta = 23.9^{\circ}$ that cannot be indexed with the $k = \begin{bmatrix} 0 & \frac{1}{2} & 0 \end{bmatrix}$ propagation vector and either requires a larger unit cell, is suggestive of an incommensurate magnetic structure or is due to an unidentified low temperature phase within the measured sample.

The refined Dy1 moments always point along the O1-Dy1-O2 bond angle (Fig. 12), which is nearly 180° (178.5(7)°). This is indicative of a local Ising axis explaining the anisotropy observed in the magnetization measurements discussed earlier. Dy2, however, does not possess such an "easy axis". O1, O3, and O5 all reside within a (0 1 0) plane, however none make a 180° bond angle with Dy2. The O1-Dy2-O3 bond angle is 167.2(7)°, while O5 is an octahedral monocap without an equivalent oxygen in line with Dy2. The Dy2 moments order nearly along the O1-Dy2-O3 bond angle but are canted slightly toward O5.

It should be noted that the angle between O4-Dy2-O5 is close to 180° (177.9(6)°) and nearly parallel to O1-Dy1-O2 which could provide a local Ising axis, however, the Dy2-O4 spacing (3.5243(6) Å) is far beyond the ionic radius of Dy³⁺. This axis would also be forbidden by the restrictions set on the basis vectors earlier as Dy1 and Dy2 must have opposite signs for ψ_1 and same signs for ψ_2 . These cannot be simultaneously fulfilled for this direction. Thus, Dy2 moments do not order in a particular direction with strong uniaxial symmetry as required for Ising-like ordering.

This is consistent with the observed dc magnetization measurements. The anisotropic Dy1 spins lower the magnetization saturation point much like Dy spins in Dy2Ti2O7 spin ice (approximately $\frac{1}{2}$ of the free Dy³⁺ moment). The more isotropic Dy2 spins, however, raise the saturation relative to spin-ice pyrochlores (and cubic A_2TiO_5 polymorphs) resulting in a bulk saturation that is in between spin-ice and free Dy³⁺. The distinct ordering and anisotropies for Dy1 and Dy2 potentially explains the double peak in the magnetic field scan at 0.3K (Fig. 7b). In general, a sharp peak is indicative of a spin-flip transition. The double peak therefore indicates two successive spin-flip transitions. Since there are two unique Dy ions the most likely cause of this is that one peak corresponds to Dy1 and the other Dy2. The difference in the applied fields for the peaks therefore reflects the different energy scales of the magnetic exchange interactions for the Dy1 and Dy2. The origin of this in the lattice, as discussed, is the different crystal field and bond angles that lead to more Ising-like interactions for one Dy and less for the other. To assign the observed peaks we note that, in general, Ising transitions are typically sharper than non-Ising (Heisenberg) transitions. The first peak in Fig. 7b at 0.79 T, which is slightly more narrow than the second peak at 1.35 T, could therefore correspond to Ising-like Dy1 spin flips. This assignment is further supported by examining the number of next nearest neighbors

available for exchange interactions. As shown in Figs. 3 and 11, Dy atoms are ordered in "sheets" extending infinitely in the [010] direction. Dy1 atoms form the edges of the "sheet" while Dy2 atoms are on the interior (Dy1-Dy2-Dy2-Dy1). Dy2 has four next nearest neighbors that are approximately equidistant: two in the [010] direction and two that were originally nearest neighbors at 300 K (Fig. 3a). Dy1 atoms, however, only have the two next nearest neighbors in the [010] direction because they are not interior atoms. If Dy2 moments were to flip first (at 0.79 T), Dy1 spins would be isolated and rely solely on Dy1-Dy1 coupling in the [010] direction to remain ordered. If Dy1 atoms flipped first, however, Dy2 could still maintain order (with itself) within the "sheet". Unless the Dy1-Dy1 coupling strength is very strong, the peaks at 0.79 T and 1.35 T can reasonably be attributed to flipping/polarizing Dy1 and Dy2 spins respectively. We note, however, that this is a complex problem and without the availability of a single crystal, there is not sufficient data to definitively determine the origin of the double peak since there are many interactions involved (*i.e.*, nearest neighbor and next nearest neighbor exchange, spin anisotropy, long-range dipolar interactions).

5. Conclusions

The structure and magnetic properties of orthorhombic Dy₂TiO₅ have been successfully determined using neutron diffraction, magnetic susceptibility and magnetization from 300 K to 0.3 K. The ac susceptibility shows evidence of a 2D-3D magnetic transition as noted by the sluggish antiferromagnetic transition beginning at 3 K and inflection at 1.7K. Neutron diffraction shows that local magnetic ordering (beginning at 3 K) precedes long range magnetic ordering (beginning at 1.7 K), explaining the ac susceptibility behavior. Magnetic ordering saturates at 1.2 K, however, refined magnetic moments are slightly less than the ordered moment for both Dy1

and Dy2 (8.89(28) μ_B and 7.76(31) μ_B respectively). The magnetic structure can be indexed with a propagation vector of $k = [0 \frac{1}{2} 0]$. Moments order in interwoven two-dimensional sheets extending in the [0 1 0] direction which encircle nonmagnetic Ti atoms. Dy1 shows Ising-like ordering along local O1-Dy1-O2 axes while Dy2 does not possess a direction of uniaxial symmetry.

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Figures



Fig. 1: (a) Structure of orthorhombic Dy_2TiO_5 (*Pnma* space group). (b) Dy atoms are in monocapped octahedral coordination (shown as blue and magenta polyhedra for Dy1 and Dy2 respectively). (c) Ti atoms are in square pyramidal coordination.



Fig. 2: Room temperature neutron diffraction pattern (open circles) of Dy_2TiO_5 refined with the orthorhombic (*Pnma*) structural model (solid red line). Dy_2TiO_5 can be accurately measured despite dysprosium's large absorption cross section (994 b for thermal neutrons) due to the high flux at the NOMAD beamline. Vertical blue ticks denote Bragg peak positions while the solid green line is the difference between the measured data and structural model.



Fig. 3: Dy sublattice of Dy_2TiO_5 . (a) Dy1 (dark blue spheres) has two Dy2 (cyan spheres) nearest neighbors at 300 K while Dy2 is neighbors with two Dy2 atoms. (b) At 20 K Dy2 – Dy2 pairs are split and are no longer nearest neighbors. Dy1 is nearest neighbors with two Dy2 atoms while Dy2 is nearest neighbors with Dy1 atoms.



Fig. 4: Axial bond angle temperature dependence for Dy1 and Dy2 monocapped octahedra determined by neutron diffraction. O1-Dy1-O2 becomes more symmetrical at lower temperature while O1-Dy2-O3 becomes more distorted. The dashed lines are guides to the eye. Refer to Fig. 12 for a detailed explanation of these oxygen positions.



Fig. 5: Low temperature XRD patterns. (a) The *a* unit cell parameter (black squares, left axis) indicates only a slight distortion at low temperatures while *b* (blue diamonds, right axis) shows a continual *increase* below 100 K. (b) Unlike *a* and *b*, the *c* unit cell parameter (black squares, left axis) continually decreases with temperature. The contraction of the unit cell volume (blue diamonds, right axis) saturates below 40 K.



Fig. 6: dc magnetic susceptibility and magnetization measurements of Dy_2TiO_5 . Inverse susceptibility (a) reveals an antiferromagnetic transition at 3.5 K (vertical dashed line) with paramagnetic behavior above this temperature. The Curie-Weiss law was fit to the data (dashed red line) resulting in a Curie-Weiss temperature (θ_{CW}) of -10.8 K. Magnetization measurements (b) at varying temperatures show a saturation below the moment for free Dy^{3+} ions but larger than that observed in spin-ice or cubic Dy_2TiO_5 polymorphs.



Fig. 7: ac magnetic susceptibility measurements of powder Dy_2TiO_5 . (a) Real part of zero field ac magnetic susceptibility at 700 Hz, 347 Hz, and 80 Hz. There is only weak frequency dependence suggesting the absence of local ice/glassy dynamics. A sluggish paramagnetic to antiferromagnetic transition is apparent at 3.1 K. (b) Magnetic field sweep at 0.3 K with a frequency of 347 Hz. There are two maxima of equal intensity at 0.79 T and 1.35 T. (c) Real part of magnetic susceptibility from B = 0 - 4 T. The magnetic field initially decreases the ordering temperature and sharpens the magnetic transition. The transition temperatures are 3.1K, 1.1 K, and 0.8 K for 0 T, 1 T, and 2 T respectively. Stronger fields completely dampen the transition. (d) 1st derivative of the real part of magnetic susceptibility for B = 0 T and 1 T. Sharp peaks are

apparent at 1.7 K and 0.6 K for 0 T and 1 T which correspond to the onset of long range

magnetic order.



Fig. 8: Low temperature neutron scattering patterns. (a) Neutron diffraction patterns at 0.3 K, 3 K, and 20 K. (b) Magnetic contributions only at 0.3 K and 3 K obtained by subtracting the pattern at 20 K. Only diffuse magnetic scattering is observed at 3 K while long range magnetic order begins around 1.7 K (inset). Solid squares and open circles were measured on separate days.



Fig. 9: Fit of the diffuse magnetic scattering at 3 K using the Warren function.



Fig. 10: Magnetic neutron refinement at 0.3 K. The refinement (solid red line) agrees well with the measured data (open circles). The nuclear structure, magnetic structure, and Al holder were all refined together. Refined Bragg peaks from each phase are shown as vertical green ticks. The solid black line represents the difference between the measured data and refinement. There exists an additional magnetic peak at 23.8° that cannot be indexed with the k = $[0 \frac{1}{2} 0]$ propagation vector and requires a larger unit cell.



Fig. 11: Refined magnetic structure of Dy_2TiO_5 . Dy atoms are shown as grey spheres while magnetic moments are shown as red arrows. Solid gray lines designate nearest neighbor Dy pairs. Ordered moments encircle nonmagnetic Ti⁴⁺ cations (not shown).



Fig. 12: Relation between refined magnetic moments and oxygen coordination for Dy1 and Dy2 at 0.3 K. Magnetic moments on Dy1 atoms (cyan spheres) order along the 180° angle formed by O1-Dy1-O2 which corresponds to a local Ising axis. Dy2 (black sphere) does not possess such a crystal field and the moment is ordered between the O1-Dy2-O3 and O1-Dy2-O5 bond angles. O5-Dy2-O4 does form a 180° angle that is nearly parallel to O1-Dy1-O2, however, the Dy2-O4 distance (shown as a dashed line) is far beyond the ionic radius of Dy³⁺.

Table 1: Refined structural parameters for orthorhombic Dy₂TiO₅ at 300 K and 0.3 K determined by neutron diffraction. Data at 300 K was collected at the NOMAD beamline of the Spallation Neutron Source while data at 0.3 K was collected at the Hb-2a beamline of the High Flux Isotope Reactor.

300 K	a (Å)	b (Å)	c (Å)	V (Å ³)	-		
Pnma	10.3722(2)	3.71985(7)	11.2379(2)	433.59(1)	_		
Atom	Х	у	Z	\mathbf{B}_{iso}			
Dy1	0.11421(13)	0.25	0.22290(11)	0.0056(3)			
Dy2	0.13611(13)	0.25	0.55759(13)	0.0062(3)			
Ti1	0.1740(6)	0.25	0.8833(7)	0.0081(9)			
01	0.4948(4)	0.25	0.1032(4)	0.0070(6)			
O2	0.2255(4)	0.25	0.0433(4)	0.0098(8)			
O3	0.2598(4)	0.25	0.7294(4)	0.0069(7)			
O4	0.5097(5)	0.25	0.6537(4)	0.0146(10)			
05	0.2659(4)	0.25	0.3833(4)	0.0065(7)	_		
Bond Length	~ 1			<u>.</u>	o •	W 0	
(A)	01	02	03	04 2 201(2)	05	<x-o></x-o>	
Dv1	2314(5)	2 325(5)	2 227(3) ×2	2.391(3) ×2	2 393(5)	2328(1)	
	2.317(5)	2.525(5)	2.227(3) 2	2	2.595(5)	2.520(1)	
Dy2	2.327(3) 2 359(3) ×2	2.355(3)	2.318(5)		2.377(5)	2.345(1)	
					1.962(2)		
Ti		1.876(9)	1.945(9)	1.754(8)	×2	1.900(3)	
					_		
0.3 K	a (Å)	b (Å)	c (Å)	V (Å ³)			
Pnma	10.344(2)	3.7114(8)	11.193(2)	429.74(16)			
					-		
Atom	х	у	Z	B _{iso}			
Dy1	0.116(2)	0.25	0.225(1)				
Dy2	0.143(2)	0.25	0.559(1)				
Ti1	0.216(7)	0.25	0.854(11)				
01	0.500(6)	0.25	0.091(6)				
O2	0.244(5)	0.25	0.035(6)				
O3	0.249(7)	0.25	0.741(4)				
O4	0.485(4)	0.25	0.662(4)				

O5 0.261(6)		0.25 0.368(5)				
Bond Length	01	02	03	04	05	<y ()=""></y>
(A) Dy1	2.38(7)	2.50(7)	2.33(5) ×2	2.24(3) ×2	2.19(7)	2.316(2)
Dy2 2.24(7) 2.40(5) ×2		2.21(4) ×2	2.31(6)		2.47(6)	2.320(2)
Ti		2.06(14)	1.31(13)	2.40(10)	1.877(18) ×2	1.90(4)

Table 2: Symmetrically allowed basis vectors (BV) for the Γ_2 IR for the *Pnma* space group with a $\mathbf{k} = [0 \frac{1}{2} 0]$ propagation vector. Note: Dy1 and Dy2 need not necessarily have the same combination of basis vectors.

	Atom 1		Atom 2				Atom 3			Atom 4			
BV	m _x	$m_{\rm y}$	mz	m _x	$m_{\rm y}$	mz	BV	m _x	$m_{\rm y}$	mz	m _x	$m_{\rm y}$	mz
ψ_1	2	0	0	2	0	0	Ψ3	0	$\overline{2}$	0	0	$\overline{2}$	0
ψ_2	0	0	2	0	0	$\overline{2}$	Ψ4	$\overline{2}$	0	0	$\overline{2}$	0	0
Ψ_6	0	2	0	0	2	0	Ψ5	0	0	$\overline{2}$	0	0	2

Table 3: Refined coefficients for magnetic basis vectors. C_n refers to the coefficient on ψ_n

described in Table 2. Data was collected at 0.3 K at the Hb-2a beamline of the High Flux Isotope

Same Moments ($R_{WP} = 18.3$)									
Atom	C_1	C_2	C ₃	C_4	C_5	C_6	μ		
Dy1	2.61(6)	-3.06(6)	0	-2.61(6)	3.06(6)	0	8.05(12)		
Dy2	-2.61(6)	-3.06(6)	0	2.61(6)	3.06(6)	0	8.05(12)		
Different	t Moments ($R_{WP} =$	13.9)							
Atom	C_1	C_2	C ₃	C_4	C_5	C_6	μ		
Dy1	2.35(15)	-3.78(13)	0	-2.35(15)	3.78(13)	0	8.89(28)		
Dv2	-3.00(17)	-2.46(12)	0	3.00(17)	2.46(12)	0	7.76(31)		

Reactor. The refinement is improved if coefficients on Dy1 and Dy2 are independently refined.

References

[1] W. G. Mumme and A. D. Wadsley, Acta Crystallographica Section B-Structural Crystallography and Crystal Chemistry **B 24**, 1327 (1968).

[2] A. V. Shlyakhtina, D. A. Belov, O. K. Karyagina, and L. G. Shcherbakova, Journal of Alloys and Compounds **479**, 6 (2009).

[3] R. D. Aughterson, G. R. Lumpkin, G. J. Thorogood, Z. M. Zhang, B. Gault, and J. M. Cairney, Journal of Solid State Chemistry **227**, 60 (2015).

[4] Y. F. Shepelev and M. A. Petrova, Inorganic Materials 44, 1354 (2008).

[5] M. A. Petrova and R. G. Grebenshchikov, Glass Physics and Chemistry 34, 603 (2008).

[6] K. E. Sickafus, L. Minervini, R. W. Grimes, J. A. Valdez, M. Ishimaru, F. Li, K. J. McClellan, and T. Hartmann, Science **289**, 748 (2000).

- [7] R. C. Ewing, W. J. Weber, and J. Lian, Journal of Applied Physics 95, 5949 (2004).
- [8] W. J. Weber *et al.*, Journal of Materials Research **13**, 1434 (1998).
- [9] Y. Jiang, J. R. Smith, and G. R. Odette, Acta Materialia 58, 1536 (2010).
- [10] G. C. Lau, B. D. Muegge, T. M. McQueen, E. L. Duncan, and R. J. Cava, Journal of Solid State Chemistry **179**, 3126 (2006).

[11] M. J. Harris, S. T. Bramwell, D. F. McMorrow, T. Zeiske, and K. W. Godfrey, Physical Review Letters **79**, 2554 (1997).

[12] S. T. Bramwell and M. J. P. Gingras, Science 294, 1495 (2001).

[13] G. C. Lau, R. S. Freitas, B. G. Ueland, B. D. Muegge, E. L. Duncan, P. Schiffer, and R. J. Cava, Nature Physics **2**, 249 (2006).

[14] A. P. Ramirez, A. Hayashi, R. J. Cava, R. Siddharthan, and B. S. Shastry, Nature **399**, 333 (1999).

[15] V. D. Risovany, E. E. Varlashova, and D. N. Suslov, Journal of Nuclear Materials **281**, 84 (2000).

- [16] A. Sinha and B. P. Sharma, Journal of the American Ceramic Society 88, 1064 (2005).
- [17] J. Neuefeind, M. Feygenson, J. Carruth, R. Hoffmann, and K. K. Chipley, Nuclear Instruments & Methods in Physics Research Section B-Beam Interactions with Materials and Atoms **287**, 68 (2012).

[18] A. S. Wills, Physica B **276**, 680 (2000).

- [19] D. J. P. Morris *et al.*, Science **326**, 411 (2009).
- [20] B. G. Ueland, G. C. Lau, R. S. Freitas, J. Snyder, M. L. Dahlberg, B. D. Muegge, E. L. Duncan, R. J. Cava, and P. Schiffer, Physical Review B 77, 144412 (2008).

[21] G. C. Lau, R. S. Freitas, B. G. Ueland, M. L. Dahlberg, Q. Huang, H. W. Zandbergen, P. Schiffer, and R. J. Cava, Physical Review B **76**, 054430 (2007).

- [22] M. E. Fisher, Philosophical Magazine 7, 1731 (1962).
- [23] B. E. Warren, Physical Review **59**, 693 (1941).
- [24] J. E. Greedan, Journal of Materials Chemistry 11, 37 (2001).

[25] A. S. Wills, N. P. Raju, C. Morin, and J. E. Greedan, Chemistry of Materials 11, 1936 (1999).

[26] C. S. Knee, D. J. Price, M. R. Lees, and M. T. Weller, Physical Review B **68**, 174407 (2003).