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Magnetism and metal-insulator transition in oxygen deficient SrTiO₃

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First-principles calculations to study the electronic and magnetic properties of bulk, oxygen-deficient SrTiO₃ (STO) under different doping conditions and densities have been conducted. The appearance of magnetism in oxygen-deficient STO is not determined solely by the presence of a single oxygen vacancy but by the density of free carriers and the relative proximity of the vacant sites. We find that while an isolated vacancy behaves as a non-magnetic double donor, manipulation of the doping conditions allows the stability of a single donor state, with emergent local moments coupled ferromagnetically by carriers in the conduction band. Strong local lattice distortions enhance the binding of this state. The energy of the in-gap local moment can be further tuned by orthorhombic strain. Consequently we find that the free-carrier density and strain are fundamental components to obtaining trapped spin-polarized electrons in oxygen-deficient STO, which may have important implications in the design of optical devices.

Bulk SrTiO₃ (STO) at stoichiometry is a semiconductor with a substantial band gap and an empty d-shell. It has long been known that doping, particularly via Oxygen vacancies (Ov), leads to metallic behavior¹ and even superconductivity in the bulk². More recently, at controlled interfaces between insulating STO and LaAlO₃, a two-dimensional electron gas can be created³⁴. Also, confined in a thick layer at an open vacuum-cleaved surface of free-standing STO experimental evidence of charge carriers has been reported⁵. Quite surprising in view of the nearly empty d-shell has been the observation of robust ferromagnetism, coexisting with superconductivity. This has been seen in interfacial structures^{6,7}, but also in bulk and thin film, both in systems that are doped with other transition metals, but also those that are not. 1,8-11 Recent advances in the synthesis of oxygendeficient structures have allowed for the detection, measurement and control of optically induced long-lived magnetic moments in STO, further implicating Ov^{12} . Very recently is was shown than electrical gating at STO-based interfaces can be controlled to induce the appearance of magnetic moments dependent on the mobile carrier density¹³. In consequence, it is important to understand the potential roles of Ov both in doping, and in the independent generation of local moments associated with the Ov, and upon external doping.

In this study we perform first-principles calculations to show that the magnetic properties of single Ov in SrTiO₃ are richer than previously reported, exhibiting a large tunability with respect to the defect concentration and external doping rate. By modifying carrier-doping concentrations typical localized moments of high-density defects may vanish as the doping rate per unit of volume decreases, suggesting that external doping may restore the magnetism of the monovacancies and tune the magnetic moments associated with diluted Ov. The occurrence of magnetic localized states in Ov in STO is intimately connected to the overlap of their electronic wavefunctions. Altering this overlap by the application

of strain or modifying the density of free carriers may have profound effects on both the magnetism and the metallic/semiconducting character of the defected material. First-principles studies of magnetism for different unit cell parameters and doping rates in cubic oxygendeficient STO provide a unique opportunity to tune, hence help decipher, complex exchange interaction between electron-donor defects in transition metal oxides of current technological and fundamental interest. The application of a wide range of tensile strains shifts the energy position of the Ov impurity bands and modulates the optical properties of the defected material. Total density of states (DoS) and real-space projections of the net charge density will illustrate the tuning capability of oxygen-deficient STO with external physical stimuli.

The density functional theory based calculations were conducted using the projector augmented-wave method¹⁴ and the PBE-GGA exchange-correlation functional 15. To improve the description of the electrons occupying the d-orbitals of the Ti atoms at the vacant site, a Hubbard-U correction (GGA+U) as implemented in the VASP code^{16–18} was included. This approach is known to improve the description of the ground state of correlated systems by accounting for the localized nature of delectrons¹⁹. The rotationally invariant method by Dudarev et al.²⁰ with an effective U_{eff} =U-J=4.0 eV was applied to capture the strong correlations. Although higher values of U were reported to describe accurately the splitting of the t_{2g} - e_g Ti orbitals, this value of the on-site repulsion has demostrated to effectively describe the highy correlated electrons at O deficient sites of STO^{21} . The electronic wavefunctions were described using a planewave basis set with an energy cutoff of 400 eV. Atomic positions were fully relaxed in Γ -centered $3\times3\times3$ supercells for k-point sampling until residual forces were lower than 0.01 eV/Å. The number of k-points was decreased proportionally as the number of cells increased in the unit cell. Extra electrons were introduced or removed and compensated with an equally uniform background

charge of opposite sign. The external doping of the defected configuration was realised at the volume of bulk STO.

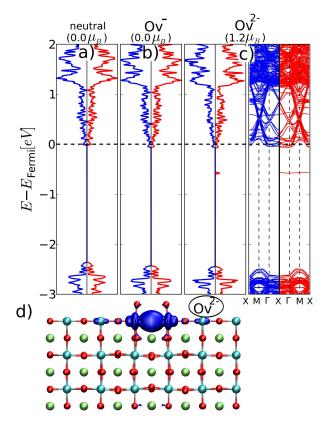


FIG. 1. In a $3\times3\times6$ supercell, we plot the density of states of a neutral (a) and singly- (b) and doubly-charged (c) vacancy. The doubly charged vacancy produces a localised moment (d) together with a strong lattice distortion. The excess of spin-up with respect to spin-down charge density in a supercell is plotted in d) for an isosurface value of $0.01~{\rm e}^{-}/{\rm Å}^{3}$.

In the cubic phase, STO is a non-magnetic widebandgap oxide material possessing the common features of perovskites, namely vertex-sharing metal-centered octahedra, square pyramids, and square planes. An oxygen vacancy is one of the intrinsic defects that may change STO's structural, magnetic and electronic properties as a result of the double-donor doping character of the defect. Indeed, an Ov is known to introduce electron carriers and cause an insulator-to-metal transition, as well as distortions in the atomic network yielding octahedra tilts²². The result of removing an oxygen atom is the enlargement of the distance between the two Ti atoms to which it was bonded, and an overall reduction of the crystal symmetry from cubic to the C_{4v} space group. The ground state of a single Ov in a 3×3×3 STO supercell is obtained upon expansion of 1% of the cubic lattice parameter. As already described, ^{21–27}, each Ti atom at the vacancy site has a dangling bond which further combine into molecular-like bonding and antibonding orbitals. These σ -orbitals are the result of the local hybridization of the Ti $3d_{z^2-r^2}$ and $4p_z$ orbitals.

In Figure 1a the DoS of a single Ov in a $3\times3\times6$ supercell is plotted, where the longest dimension is in the direction of the Ti-Ov-Ti bond. Such a large unit cell turns out to be necessary to reproduce the physics of an isolated Ov. The neutral vacancy is seen to be a donor, as conventionally assumed²⁴, and the system exhibit a metalic behavior. We now consider the effect of adding extra electrons to the system (mimicking electrostatic doping). Adding a single electron (Figure 1b) simply shifts the Fermi level and leaves the system spin symmetric and paramagnetic. Adding a second electron (Figure 1c) produces a sudden restructuring of the density of states and a localised state with a full magnetic moment appears in the gap, and the ground state has a substantial ferromagnetic moment¹³. Observing the spin density (Figure 1d), we note the strongly localized nature of this state at the vacancy position, together with substantial polaronic distortions. Adding two electrons to the Ov⁻ configuration without allowing relaxation does not bind a localized state. The separation of 4.09 Å between Ti atoms at the Ov site is reduced to 3.88 Å when the magnetic state is created, namely the lattice relaxations are critical in the description of the magnetic configuration. The electrostatic repulsion between this local charge density and the anionic near-neighbor O atom couples the localized state to a very localized atomic distortion, which is noticeable by 10° increase of the O-Ti-O angle around the Ov. This magnetic polaronic distortion is responsible for trapping the localized state by lowering its energy. The strong polaronic coupling explains that ionizing oxygendefective STO first removes electrons from the conduction band (CB) before the polaronic state is destroyed. Also to be noted is that the orbital configuration cannot be projected onto Ti d-states alone since the localised state lives in the vacancy itself. The metallic character of the oxygen-deficient STO so far described becomes semiconducting if the monovacancy is further isolated. Indeed, STO with a single Ov in a $5 \times 5 \times 5$ supercell exhibits a band gap and a complete absence of magnetic state. Janotti et al.²⁸ have recently pointed out the nearstability of polarons in the vicinity of an Ov.

In Figure 2, we study a nearly identical system with a $3\times3\times3$ STO supercell (i.e. twice the density and the doping rate of the above neutral Ov). Here the neutral vacancy yields a single fully occupied spin-polarized bonding in-gap state at ~ 0.5 eV below the CB and a net spin-resolved charge density located in the surrounding of the vacant site (Figure2f). The oxygen-deficient STO exhibits a magnetic moment of $\sim 1.1~\mu_B$. This is in agreement with previous studies 21,22,24,29 , where the interpretation was given that correlation effects driven by the strength of the on-site Coulomb potential promote a second electron to the CB, confering the oxygen-deficient STO the metallic character observed in Figure 2c. Notice however the substantial dispersion of the in-gap state in Figure 2c which provides an explanation of the apparent discrepancy with the results of an isolated vacancy in

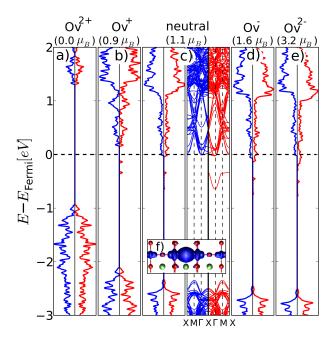


FIG. 2. Total density of states of oxygen vacancies (Ov) in STO with increasing doping rates ranging from hole to electron doping. In a), two-fold hole-doped oxygen-defected STO recovers the wide band gap of bulk STO according with double-donor character of the single Ov. In b) two Ov in a double-size supercell two-fold hole-doped are in ferromagnetic coupling. When doped with two extra ${\rm e}^-$ the vacancy nearly reaches a half-metallic state. The excess of charge density for the neutral vacancy is plotted in f) for an isosurface value of $0.01~{\rm e}^-/{\rm \mathring{A}}^3$

Figure 1. This in-gap state of Figure 2 exhibits a narrow energy spread in the k-space line joining the M and X points, related to a reduced overlap of the wave function between neighboring cells. On the contrary, a more dispersive character for the line joining the M and X points through the Γ point indicates that the impurity band is extended in the Ti-Ov-Ti bond direction more than in the perpendicular directions, allowing the wavefunction to enhance the overlap with the adjacent vacancy states and develop a dispersive band.

Hole doping upon removal of one electron from the defected system leads to the suppression of the itinerant electron and the confinement of the remaining electron in the hydrogenic orbital, rendering the system a magnetic insulator, as shown in Figure 2b. If two neighboring vacancies were allowed to interact in a larger two-fold holedoped 3×3×6 supercell with the same doping rate imposing a total magnetic moment of zero, the moments would interact ferromagnetically forming a shallow in-gap state while the extra electron would go into the conduction band rendering the system half-metalic. The latter configuration is $\sim 100 \text{ meV}$ per unit cell higher in energy than the former. Further removal of an additional electron leads to two-fold hole doping and the oxygen-deficient STO to recover the wide band gap of the pristine bulk (see Figure 2a). On the contrary, electron doping enhances the metallic character as the additional electrons populate the half-filled d-bands. For high doping rates the unbalanced occupancy of the Ti d-orbitals leads the system towards a half-metallic state.

To recapitulate: An *isolated* Ov is a double donor, as has been traditionally assumed. However, when added electrons are located in the CB and the Ov wavefunctions overlap a magnetic configuration is stabilised, with each Ov a *single* donor (and therefore a local moment) coupled by spin-polarised itinerant carriers by double exchange, rendering the system ferromagnetic. A similar ferromagnetic state can also be obtained in an undoped but higher-density system of Ov, where the overlap between the vacancies stabilises the magnetism.

These results are in excellent agreement with the reported experimental evidence of metallic conductivity in STO single crystals. This metallicity was removed after a treatment of reduction to induce a self-healing process and a decreasing of the density of initially introduced oxygen-vacancy defects and charge carriers. The consistency of our results with previous experimental observations allows us to conclude that the magnetic and electronic properties of high-dense arrays of Ov ranging from a total absence to highly localized magnetic moments are subjected to overlapping degree conditions, density of defects and doping rate.

Many of the outstanding properties of transition metal oxides stem from the interactions between lattice, charge and spin degrees of freedom. Modifications on the lattice parameter as a result of tensile or compressive strain can be used to tune the electrical and magnetic properties of these materials. Typical strains within the range from -4% to +10% have been applied on epitaxially growth thin films and remarkably different properties were obtained. For instance, pristine STO remains paraelectric at very low temperatures and the application of stress results in ferroelectricity³⁰. Epitaxial strain was used to perform a large increase of the STO transition temperature and produce room-temperature ferroelectricity³¹. demonstrating that engineering strain provides an effective way to tune STO intrinsic properties and achieve novel functionalities. We next analyze the effect of strain on the electronic and magnetic properties of high-density oxygen-deficient STO.

Biaxial strain was modelled by means of successive modifications of two of the lattice parameter of a $3\times3\times3$ supercell with a single Ov. This supercell size allows us to observe the evolution of the electronic and magnetic properties of the localized vacant state when tuning from compresive -3% to tensile +6% biaxial strain, as regularly achieved in epitaxial oxide growth experiments³². We noticed that an expansion of 1% of the defectless bulk STO lattice constant is required upon formation of the hybrid orbitals for stabilizing the formation energy of the vacant site. The magnetic moment is barely modified throughout the expansion until the transition to a new type of in-gap localized state (a Ti $3d_{xy}$ orbital) at $+6\%^{28}$. As the oxygen-deficient supercell varies from compressive to

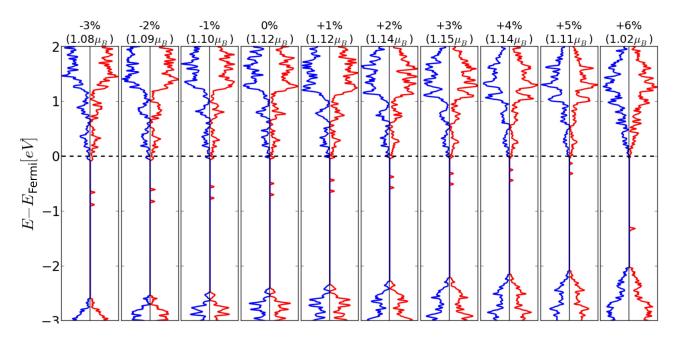


FIG. 3. Evolution of the electronic states of oxygen-deficient STO under biaxial strain.

tensile biaxial strain at fixed composition and charge carrier stoichiometry, the energy of the in-gap state progressively evolves from \sim -0.7 eV up to energy values close to the CB electrons. The trapped electron evolves from a deep in-gap to a shallow state as a result of the the different degrees of hybridization of the atomic orbitals involved in the formation of the hydrogenic-like orbital as the distance between Ti atoms changes. According to the general theory of impurity bands in n-type oxides 33 , the former state fulfills the electronic structure diagram required by 3d-transition metal oxide based materials to exhibit low Curie temperature, whereas the proximity of the impurity band 3d-orbitals to the Fermi level of the former facilitates the high Curie temperature. A band gap reduction of ~ 0.5 eV is also observed.

Light emission in irradiation-induced metallic oxygendeficient ${\rm STO}^{34}$ exhibits an extraordinary tunability depending on the recombination between excited electrons or holes with conducting or defect-level electrons. The possibility to strain-engineering the oxygen vacancy state may lead to important applications on oxide-based electronic and optic devices that depend on the doped charge carrier concentrations and trapped electron energy level position with respect to the CB to modulate the lumi-

nescence induced by photo-excited carriers.

In conclusion, we find that electronic doping due to oxygen vacancies in STO is a more complex process than has been considered up to now. Isolated vacancies are predicted to be hydrogenic double donors and are non magnetic. Adding further carriers or coupling vacancies induces an instability to the formation of a local moment trapped at the vacancy site, with strong lattice distortions, together with exchange-driven spin-polarisation of the remaining carriers in extended states. This physics is not easily captured by simplified tight-binding models of the STO bands, requiring at the minimum the inclusion of extra local orbitals at the vacancy site (as in conventional optical centers), and strong coupling between lattice and magnetic degrees of freedom.

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¹ K. Szot, W. Speier, R. Carius, U. Zastrow, and W. Beyer, Phys. Rev. Lett. 88, 075508 (2002).

² C. S. Koonce, M. L. Cohen, J. F. Schooley, W. R. Hosler, and E. R. Pfeiffer, Phys. Rev. 163, 380 (1967).

³ A. Ohtomo and H. Hwang, Nature **427**, 423 (2004).

⁴ C. Cen, S. Thiel, G. Hammerl, C. W. Schneider, K. E. Andersen, C. S. Hellberg, J. Mannhart, and J. Levy, Nature Materials 7, 298 (2008).

⁵ A. F. Santander-Syro, O. Copie, T. Kondo, F. Fortuna, S. Pailhes, R. Weht, X. G. Qiu, F. Bertran, A. Nicolaou, A. Taleb-Ibrahimi, P. Le Fevre, G. Herranz, M. Bibes,

- N. Reyren, Y. Apertet, P. Lecoeur, A. Barthelemy, and M. J. Rozenberg, Nature **469**, 189 (2011).
- ⁶ J. N. Eckstein, NATURE MATERIALS 6, 473 (JUL 2007), ISSN 1476-1122.
- ⁷ B. Kalisky, J. A. Bert, B. B. Klopfer, C. Bell, H. K. Sato, M. Hosoda, Y. Hikita, H. Y. Hwang, and K. A. Moler, NA-TURE COMMUNICATIONS 3 (JUN 2012), ISSN 2041-1723, doi: "bibinfo doi 10.1038/ncomms1931.
- ⁸ S. Middey, C. Meneghini, and S. Ray, Applied Physics Letters 101, 042406 (2012).
- ⁹ G. Herranz, R. Ranchal, M. Bibes, H. Jaffrès, E. Jacquet, J.-L. Maurice, K. Bouzehouane, F. Wyczisk, E. Tafra, M. Basletic, A. Hamzic, C. Colliex, J.-P. Contour, A. Barthélémy, and A. Fert, Phys. Rev. Lett. **96**, 027207 (2006).
- W. Xu, J. Yang, W. Bai, K. Tang, Y. Zhang, and X. Tang, Journal of Applied Physics 114, 154106 (2013).
- ¹¹ D. A. Crandles, B. DesRoches, and F. S. Razavi, Journal of Applied Physics **108**, 053908 (2010).
- W. D. Rice, P. Ambwani, M. Bombeck, J. D. Thompson, G. Haugstad, C. Leighton, and S. A. Crooker, Nature materials 13, 298 (2014).
- ¹³ F. Bi, M. Huang, S. Ryu, H. Lee, C.-W. Bark, C.-B. Eom, P. Irvin, and J. Levy, Nature Communications 5, 5019 (2014).
- ¹⁴ P. E. Blöchl, Phys. Rev. B **50**, 17953 (1994).
- ¹⁵ J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).
- ¹⁶ G. Kresse and J. Hafner, Phys. Rev. B **48**, 13115 (1993).
- ¹⁷ G. Kresse and J. Furthmüller, Phys. Rev. B **54**, 11169 (1996).
- ¹⁸ G. Kresse and D. Joubert, Phys. Rev. B **59**, 1758 (1999).
- ¹⁹ B. Himmetoglu, A. Floris, S. de Gironcoli, and M. Cococcioni, International Journal of Quantum Chemistry 114.

- ²⁰ S. L. Dudarev, G. A. Botton, S. Y. Savrasov, C. J. Humphreys, and A. P. Sutton, Phys. Rev. B **57**, 1505 (1998).
- ²¹ Z. Hou and K. Terakura, Journal of the Physical Society of Japan 79, 114704 (2010).
- ²² M. Choi, F. Oba, Y. Kumagai, and I. Tanaka, AD-VANCED MATERIALS 25, 86 (2013).
- ²⁴ C. Lin and A. A. Demkov, Phys. Rev. Lett. **111**, 217601 (2013)
- ²⁵ C. Mitra, C. Lin, J. Robertson, and A. A. Demkov, Phys. Rev. B **86**, 155105 (2012).
- ²⁶ C. Lin, C. Mitra, and A. A. Demkov, Phys. Rev. B 86, 161102 (2012).
- ²⁷ J. Carrasco, F. Illas, N. Lopez, E. A. Kotomin, Y. F. Zhukovskii, R. A. Evarestov, Y. A. Mastrikov, S. Piskunov, and J. Maier, Phys. Rev. B 73, 064106 (2006).
- A. Janotti, J. B. Varley, M. Choi, and C. G. Van de Walle, Phys. Rev. B 90, 085202 (2014).
- ²⁹ D. D. Cuong, B. Lee, K. M. Choi, H.-S. Ahn, S. Han, and J. Lee, Phys. Rev. Lett. **98**, 115503 (2007).
- ³⁰ H. Uwe and T. Sakudo, Phys. Rev. B **13**, 271 (1976).
- J. Haeni, P. Irvin, W. Chang, R. Uecker, P. Reiche, Y. Li, S. Choudhury, W. Tian, M. Hawley, B. Craigo, A. Tagantsev, X. Pan, S. Streiffer, L. Chen, S. Kirchoefer, J. Levy, and D. Schlom, Nature 430, 758 (2004).
- ³² D. G. Schlom, L.-Q. Chen, C. J. Fennie, V. Gopalan, D. A. Muller, X. Pan, R. Ramesh, and R. Uecker, MRS BULLETIN 39, 118 (2014).
- J. Coey, M. Venkatesan, and C. Fitzgerald, Nature Materials 4, 173 (2005).
- ³⁴ D. Kan, T. Terashima, R. Kanda, A. Masuno, K. Tanaka, S. Chu, H. Kan, A. Ishizumi, Y. Kanemitsu, Y. Shimakawa, and M. Takano, Nature Materials 4, 816 (2005).