

CHCRUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

Size effects in NiTi from density functional theory calculations

Karthik Guda Vishnu and Alejandro Strachan Phys. Rev. B **85**, 014114 — Published 27 January 2012 DOI: 10.1103/PhysRevB.85.014114

Size effects in NiTi from density functional theory calculations

KarthikGuda Vishnu¹ and Alejandro Strachan^{1,2,*}

¹School of Materials Engineering and ²Birck Nanotechnology Center,

Purdue University

West Lafayette, Indiana 47907, USA

Abstract

We use density functional theory to characterize how size affects the relative stability of thin NiTi slabs of different crystal structures and its implication on the martensitic phase transition that governs shape memory. We calculate the surface energies of B2' phase (austenite), B19 (orthorhombic), B19' (martensite) and a body centered orthorhombic phase (BCO), the theoretically-predicted ground state. We find that $(110)_{B2}$ surfaces with in-plane atomic displacements stabilize the austenite phase with respect to B19' and BCO, thus slabs with such orientations are predicted to exhibit a decrease in martensite transition temperature with decreasing thickness. Our calculations predict a critical thickness of 2 nm, below which the transition would not occur. The opposite trend is observed in slabs with atomic displacements along the surface normal: the phase transformation temperature increases with decreasing size.

^{*} Corresponding author: <u>strachan@purdue.edu</u>

I. INTRODUCTION

Shape memory materials are an important class of active materials with wide range of applications. They are used in medicine as implant devices due to their excellent biocompatibility, as damping devices and mechanical actuators.¹ Their name originates from the fact that after inelastic deformation these materials recover their original shape upon heating. The shape memory behavior is due to a solid-solid, diffusion-less phase transformation (called martensitic) between a high temperature phase (austenite) and a low temperature phase (martensite). Usually, the martensitic phase has lower symmetry than austenite and shape memory is possible when the symmetry groups of both austenite and martensite are included in a common finite symmetry group,² and when the transformation is atomistically reversible i.e. all the variants of martensite transform to a unique austenite variant upon heating.³ Scaling specimen size down to the nanoscale leads to significant changes in the thermo-mechanical response of these materials and recent theoretical predictions³ indicate that NiTi, the most widely used shape memory alloy (SMA), may loose its memory effect entirely at the nanoscale. Thus a fundamental understanding of the atomic level mechanisms that govern the response of SMA and its size effects, including the role of free surfaces, interfaces, and nano-structure⁴ is critical both from the applied and basic science points of view.

The martensitic transformation can be either thermally or mechanically induced and is characterized by a critical temperature and a critical stress. For macroscopic polycrystalline samples with grain sizes larger than approximately 100 microns^{5,6,7,8} the critical transformation temperature is a function of composition alone and relatively independent of microstructure and cooling rate. However, for nanoscale and nanostructured materials surfaces and interfaces play a large role and affect the phase transition temperature. Experimental studies on nanocrystalline NiTi revealed strong size effects for grain sizes in the range of 50 nm to 350 nm.⁹ For samples with mean grain diameter of approximately 60 nm and with 90% of the grains smaller than 100 nm, the martensite start (M_s) and martensite finish (M_f) transformation temperatures decrease from 330 K to 319 K and from 302 K to less than 197 K respectively, when compared to a coarse grained polycrystalline sample. TEM studies show that grains with diameters less than 50 nm fail to transform to martensite even after quenching to 197 K. Experimental studies by Glezer and collaborators¹⁰ on Ni₅₀Ti₂₅Cu₂₅ nano-particles embedded in an amorphous matrix show similar trends for the B2 – B19 martensite phase transformation. The authors find either partially or fully suppressed transformation at sizes less than 25 nm and indicate that for spheres of diameter less than 16 nm, the martensite transformation is completely suppressed. In case of Fe-Ni–B alloys, the authors show that the martensite transformation is not completely suppressed but the critical transformation temperature is reduced to as low as 4.2 K. Similar trends have been observed in fine Cobalt powders,¹¹ polycrystalline samples of low-alloy steels,¹² Au-Cd,¹³ and Fe-Ni systems.¹⁴ Size also affects the mechanical response of shape memory materials. Submicrometer pillars of Cu-Ni-Al shape memory alloys show extraordinarily large mechanical hysteresis,¹⁵a fact that could be exploited for applications requiring high mechanical damping. Several mechanisms have been postulated to contribute to such size effects including surface and interfacial energies, mechanical constraints, and the resulting changes in the martensite microstructure.^{4,16,17}Surfaces are also known to drive phase transformations in nanoscale FCCwires.^{18,19,20}MD simulations showedthat surface stresses cause<100> nano-wires to spontaneously change their orientation to <110> in Ni, Ag and Cu; in the case of Au

nanowireswith diameters less than 2 nm, a transformation a body centered tetragonal structure has been observed.¹⁸Furthermore, Cu and Ni, square cross-section nanowires also exhibit shape memory and pseudoelasticity due to the above-mentioned structural relaxations and their large stacking fault energies.²¹Gold nano-films of thicknessless than 2 nm (8 atomic layers) have also been experimentally observed to transform from a(001) orientation to (111) spontaneously.²²These findings emphasize that surface properties are extremely important to understand the stability of phase transformations in nanoscale specimens. An understanding of the role of free surfaces and interfaces on the structural transformations and properties of NiTi and related shape memory alloys is lacking. This knowledge is critical to understand the size effects in shape memory and pseudoelasticity for these materials and this paper focuses on the role of free surfaces on the relative energetics of thin NiTi slabs of various crystal structures.

In this paper we use density functional theory (DFT) to predict the surface energy of the various phases that play a role in NiTi shape memory: B2', B19, B19' and BCO. In equiatomicNiTithe high-temperature, austenite phase is B2, however, B2 is known to be unstable with respect to atomic displacements along the [110] direction at zero temperature; the resulting phase, denoted B2',^{23,24} will be taken in this paper as the high temperature austenite phase. B19 is an orthorhombic phase observed in this system when Cu is present as an impurity. The monoclinic B19' is the experimentally-observedmartensite structure. However, recent DFT calculations,³ predict a different phase, body-centerded orthorhombic BCO, to be the ground state of this system. This finding is important because the B2-BCO-B2 phase transformation is not atomistically reversible and hence would not result in shape memory. These results were later confirmed in various ab initio studies and transformation paths between the above phases^{25,26,24} and free energy barriers²⁷ were studied. However, the BCO phase has still not been confirmed experimentally and the B19' phase may be stabilized by internal stresses that developed during the martensitic transformation.²⁴ Our results show that free surface energies affect the relative stability of the various phases in thin NiTi slabs in a significant way and under certain circumstances can arrest the martensite phase transition.

The remainder of the paper is organized as follows: Section II describes simulation details; section III describes how surface energies vary across different phases and different slab thicknesses. In Section IV we discuss the structures of relaxed slabs and in section V we discuss how size influences the relative energetics of NiTi slabs.Finally, in section VI we present a summary and our conclusions.

II. SIMULATION DETAILS

A. DENSITY FUNCTIONAL THEORY CALCULATIONS

Simulations have been carried out using SeqQuest,^{28,29,30} a DFT³¹ code developed at Sandia National Laboratories, within the generalized gradient approximation of Perdew, Burke and Ernzerhof (PBE).³²SeqQuest uses contracted Gaussian functions as a basis set and our calculations are performed using double zeta polarization basis sets. Norm conserving pseudopotentials of the Hamann type,³³ parameterized for the PBE functional are used. All our calculations are spin-independent and are performed at a 0.04 eV electronic temperature. SeqQuest uses the maximum change in any Hamiltonian matrix element as its convergence

criterion; this has been set to be 2.72×10^{-4} eV for all calculations. All the slab structures have been fully relaxed with respect to atoms using the Broyden method.³⁴ Convergence was assumed when the absolute value of the atomic force on every atom was less than or equal to 25×10^{-3} eV/Å. 14 and 10 k-points are used in the **a** and **b** periodic directions (described below) and 3, 2 and 1 k-points are used in the non-periodic direction for the 3, 5 and 7 unit-cell slab structures described below.

To verify the robustness of our predictions, we repeated our surface energy calculations for the largest slabs (7 unit-cells) with Quantum Espresso (QE),³⁵ a plane wave DFT³¹ code. We use ultra-soft pseudo-potentials³⁶ parameterized for PBE³² within the generalized gradient approximation for both Ni and Ti. For our calculations we used a cutoff of 762 eV for the plane wave expansion of the wave functions and 7,620 eV for charge density. The first Brillouin zone is sampled using a Monkhorst-Pack grid centered at the Γ point. 14 and 10 k-points are used in the periodic direction whereas 1 k-point is used to sample the non-periodic direction. We used the Methfessel-Paxton³⁷ technique for smearing with a smearing parameter of 0.04 eV. Convergence criterion for the self-consistent field calculation is set at 1 x 10⁻⁸Ryand the mixing factor is set at 0.3. All our slab structures have been fully relaxed with respect to atom positions using the Broyden-Fletcher-Goldfarb-Shanno^{38,39,40,41} minimization method and the convergence was assumed when all the components of the force vector on every atom are less than or equal to 25 x 10⁻³ eV/Å. All our calculations are spin-independent.

B. INITIAL SLAB STRUCTURES

The initial slab structures are prepared starting from the fully relaxed DFT-GGA crystal structures of different phases from our earlier work.²⁴ All our unit cells are oriented with lattice parameter **a** parallel to $[100]_{B2}$, **b** along $[110]_{B2}$ and **c** along $[-110]_{B2}$ of the B2 crystal structure. The unit cells are replicated 3, 5 and 7 times in the direction normal to the surface of interest and two free surfaces are created using a vacuum of 12.70 Å; periodic boundary conditions are imposed in all three directions. This type of boundary conditions are commonly used in surface energy calculations, see for instance Ref [42]. This corresponds to an infinite number of slabs periodic in 2D and separated by a vacuum region sufficiently thick for the interactions between slab replicas to be negligible. To verify this, we performed simulations with 2D periodic boundary conditions and open boundaries along c and found free surface energy differences of about 1%. All our calculations focus on low-energy $(110)_{B2}$ surfaces. Since the B2', B19, B19' and BCO crystal structures exhibit atomic displacements with respect to the high-symmetry B2 configuration, two $(110)_{B2}$ surfaces are possible in each case; one where atomic displacements are normal to the free surface (denoted hereafter out-of-plane) and one where bulk displacements are parallel to the free surface (in-plane surfaces). In the case of out-of-plane surfaces we studied both Ti terminated (Ti-out) and Ni terminated (Ni-out) surfaces, see Fig 1. To obtain relaxed surface structures and energies we start from the bulk structure and minimize the total energy with respect to atomic positions keeping the transverse lattice parameters fixed. No symmetry is imposed in any of the calculations.

For the QE calculations we use the lattice parameters of Huang *et al*,³ for all phases corresponding to the GGA-USPP flavor of DFT; after atomic relaxation the stress components in the structures are no larger than -0.87 GPa indicating the appropriateness of the lattice

parameters chosen. The initial slab structures are built the same way as the SeqQuest structures and the surface energies and relaxed structures are found by relaxing these structures with respect to atomic positions keeping the lattice parameters fixed.

III. SURFACE ENERGY OF THIN NITI SLABS

Surface energy (γ) is calculated from the total energy of the relaxed slab and that of the perfect crystal structure as:

$$\gamma^{\alpha} = \frac{E_{Slab}^{\alpha}(N) - N \cdot n_{uc}^{\alpha} \cdot E_{Bulk}^{\alpha}}{2 \times A^{\alpha}}$$
(1)

Where the superscript α denotes the crystal structure, $E_{Slab}^{\alpha}(N)$ is the total energy of a slab consisting of N unit cells, E_{Bulk}^{α} is the energy of the corresponding bulk per formulaunit, n_{uc}^{α} is the number of formula units per unit cells and A^{α} is the corresponding cross sectional area. For completeness, Table I summarizes the bulk lattice parameters and cohesive energies per formula unit for each phase predicted from our DFT-GGA calculations²⁴ using SeqQuest, previous ab initio calculations³ using plane waves and ultra-soft pseudopotentials (whose lattice parameters we use for the QE calculations) and experimental values that show the accuracy of the ab initio predictions.

Table 1: Lattice parameters (Å) and cohesive energies (per formula unit) of B2, B2', B19, B19' and BCO from our calculations as well as from previous theoretical and experimental works. DFT - GGA refers to SeqQuest calculations and USPP - GGA refers to QE calculations using lattice parameters from Ref. 3

Phase	Method	a (Å)	b(Å)	c(Å)	Monoclinic cell angle (°)	$E - E_{B2} (eV)$
B2	DFT - GGA	3.011	4.258	4.258	90.0	0.000
	USPP – GGA	3.009	4.255	4.255	90.0	0.000
	Exp ⁴³	3.014	4.262	4.262	90.0	0.000
B2'	DFT – GGA	3.011	4.258	4.258	90.0	-0.009
	USPP – GGA	3.009	4.255	4.255	90.0	-0.005
B19	DFT – GGA	2.850	4.597	4.167	90.0	-0.051
	USPP – GGA	2.776	4.631	4.221	90.0	-0.046
B19'	DFT - GGA	2.933	4.678	4.065	98.26	-0.081

	USPP – GGA	2.929	4.686	4.048	97.78	-0.075
	Exp ⁴⁴	2.898	4.646	4.108	97.8	
всо	DFT - GGA	2.926	4.925	4.012	106.5	-0.097
	USPP – GGA	2.940	4.936	4.012	107.0	-0.081
	$USPP - GGA^3$	2.940	4.936	3.997	107.0	-0.1

Figure 2 shows surface energies for all crystals and surface types as a function slab thickness. Very weak size dependence is observed, the change in surface energy going from slabs thickness of 3 to 7 unit cells is less than 2% for any of the cases studied. Out-of-plane energies in all cases correspond to the Ti terminated structures as these are found to be the low-energy configurations. For both surface types, the B2' crystal has the lowest surface energy among all phases, followed by B19, B19', and BCO. Our calculations using QE (with a different basis set and pseudopotentials as compared with SeqQuest) lead to similar energetics; the difference in surface energies for the two approaches is between 5% and 11% of each other. For both kinds of surfaces, the relative order of some of the surface energies change but the values are similar. As will be discussed in detail below this difference in surface energy causes the relative stability of the NiTi slabs with different phases to be size dependent.

We are unaware of experimental characterization of the surface energy of NiTi to validate our predictions, but the measurements for liquid NiAl^{45,46} give 1.4 J/m² at a temperature just above the melting temperature. DFT calculations in B2 NiTi by Nolan *et al*⁴⁷ predict a surface energy of 1.05 J/m² for the (110) surface for a slab of thickness 19.5 Å. Our calculations predict larger surface energies: 1.80 J/m², for SeqQuest a slab of similar thickness (21 Å) and 1.65 J/m² for QE. The origin of this discrepancy is unclear at this point; even if we use B2 as the reference structure we obtain larger surface energies than those in Ref. [47]. The systematic agreement between SeqQuest and QE simulations across sizes and structures support the accuracy of our results.

Surface Surface Neighbors and Average Distance (Å) energy Energy SeqQuest Phase **Bulk/Surface** QE J/m² and J/m^2 and $(eV/Å^2)$ $(eV/Å^2)$ 2 Ni/Ti @ 4 Ni/Ti @ Ni/Ti 2 B2' 2.53 2.62 (a) 2.69Bulk In-plane 2 broken 1.81(0.113) 1.65(0.103)1.7(0.106) 1.86(0.116) Ni-out 2 broken Ti – out 1.79(0.112)2 broken 1.65(0.103)2 -/Ni 2 Ni/Ti @ 4 Ni/Ti @ Ni/Ti 2 Ti/Ni 2 (a)B19 2.556 2.564 a 2.839 2.702 Bulk @ 2.853 In-plane 2 broken broken 2.08(0.130)1.96(0.122) 2 Ni-out 2 broken broken 2.21(0.138) 2.05(0.128)Ti – out 1.76(0.110) 2 broken 1.84(0.115)2 Ni/Ti @ 4 Ni/Ti @ Ni/Ti 2 -/Ni @ 1 B19' 2.568 2.559 Bulk (a) 2.62 2.60 In-plane 2 broken 1 broken 2.27(0.142)2.10(0.131) Ni-out 2 broken 2 broken 2.4(0.150) 2.12(0.132) Ti – out 1 broken 1.86(0.116) 1.71(0.107)2 Ni/Ti @ 4 Ni/Ti @ Ni/Ti 1 2 -/Ni @ 2.544 BCO Bulk 2.605 (a) 2.562.575 In-plane 2 broken 1 broken 2.09(0.130) 2.3(0.144) Ni-out 2 broken 2 broken 2.44(0.152)2.23(0.139) Ti-out 1 broken 1.96(0.122) 1.81(0.113)

Table 2: Near neighbors, average bond distances (Å) (correspond only to structures predicted by SeqQuest) and relaxed surface energies (J/m^2) for all phases (values in parentheses correspond to surface energies in $eV/Å^2$)

Interestingly, the relative surface energies of the various phases can be explained by a simple analysis of broken bonds. Table 2 lists: i) the first nearest neighbors and their distance for all the phases in bulk form, ii) the bonds that are broken while creating each of the surfaces, and iii) their surface energies corresponding to the SeqQuest and Quantum Espresso results for the 7 unit-cell-thick slabs. As described earlier, at zero temperature the B2 phase is unstable with respect to atomic displacements in the [110] direction, thus the 8 first nearest neighbors in B2 separate into three sub-shells in B2' with two pairs in the first sub-shell, 4 in the second and the remaining two in the third. As can be seen in Table 2, for each crystal structure the lowest energy surface is the one that requires breaking the longest bonds. Table 2 also shows that the increase

in surface energy from B2', to B19, followed by B19' and BCO also correlates with the number of broken bonds. For example, creating an in-plane surface in B2' involves breaking 2 NiTi bonds of intermediate length, but in B19, B19' and BCO the lattice distortions and atomic displacements bring additional bonds within the range of the first nearest neighbors and breaking those leads to higher surface energies. In the case of out-of-plane surfaces, the bond counting analysis is inconclusive. While this bond counting analysis ignores surface relaxation it provides insight into the trends observed even after full relaxation for in-plane surfaces.

IV. STRUCTURES OF RELAXED SLABS

The structural aspects of surface relaxation can be divided into the rigid translation of atomic planes normal to the surface and atomic displacements with respect to their corresponding plane (both in the surface plane and normal to it). Figure 3 shows the percent change in interplanarseparation across the slabs (results for the longest 7-unit cell slabs are shown). Interplanar relaxation is defined as:

$$\frac{\Delta_{i-i+1}^{slab} - \Delta^{bulk}}{\Delta^{bulk}}(2)$$

 $\Delta^{slab}_{i-i+1} = COM_i - COM_{i+1}$

Where *i* indexes each of the interplanar spacings, Δ indicates the difference in center of mass position of neighboring planes in the direction normal to the surface both for the slab and bulk. The interplanar separation distance in the bulk structures Δ^{bulk} is constantwhile surface relaxation in the slabs leads to variations in this quantity as the free surface is approached. Negative values indicate contraction and positive values indicate expansion. In all cases we observe a contraction of the outermost interplanar layer followed by an expansion in the following layer. After a few layers bulk interplanar distances are recovered explaining the insensitivity of the surface energy to slab thickness. In both surfaces the B2' phase exhibits significantly more relaxation that the other structures. We find between 8% and 5% contraction in the outermost layer. The relaxation is in agreement with prior DFT calculations⁴⁷ that report a relaxation of 6% for the outermost layer in B2.

We now turn our attention to the atomic relaxations with respect to their corresponding planes in the relaxed structured. Atomic displacements along the a, b and c crystal axes are computed throughout the slab in the same way they are defined in the bulk;²⁴ they are measured from the symmetric atomic positions obtained from a rigid deformation of the B2 structure. For the non-periodic direction, we evaluate atomic displacements based on an instantaneous lattice parameter for each unit cell obtained from the relaxed atomic plane positions.

Figure 4 shows the relaxed slab structures as well as atomic displacements as a function of position along the slabs' thickness for all the cases considered. As is the case for inter-planar separations, B2' exhibits larger atomic displacements than the other phases. Bulk B2' and B19 structures exhibit displacements along the $[110]_{B2}$ direction (Y in Figures 4) and out-of-plane surfaces affect the magnitude of these displacements near the free surface (significantly in B2' and very little in B19) but do not lead to displacements in other directions. In contrast, in-plane B2' and B19 surfaces lead to atomic displacements not just along $[110]_{B2}$ but also along the

surface normal $[-110]_{B2}$ (Z in figure 4); this new displacement is confined to the outermost unit cell in the slabs. Bulk B19' and BCO exhibit displacements along $[110]_{B2}$ and $[001]_{B2}$ and the two surfaces modify the magnitude of these displacements but do not lead to significant displacements in the third direction, see Figs. 4 (e-h). Overall, B2' and B19 surfaces lead to more pronounced structural changes that explain their lower surface energies beyond the simple bond breaking analysis discussed above.

V. RELATIVE ENERGETICS OF THIN NITI SLABS

In this section we discuss the implications of our results on the relative stability of slabs of the various phases as a function of their thickness. Based on equation (1) for surface energy, we can express the energy of a slab consisting of N unit-cells in terms of their bulk energy and surface energy:

 $E_{Slab}^{\alpha}(N) = N \cdot n_{uc}^{\alpha} \cdot E_{Bulk}^{\alpha} + 2 \times A^{\alpha} \gamma^{\alpha}(3)$

Where the superscript α denotes the crystal structure, $E^{\alpha}_{Slab}(N)$ is the total energy of a slab consisting of N unit cells, E_{Bulk}^{α} is the energy of the corresponding bulk per unit formula, n_{uc}^{α} is the number of formula units per unit cell, A^{α} is the corresponding cross sectional area and γ^{α} is the corresponding surface energy. Figure 5 shows the energy difference, per formula unit, between B19' and B2' slabs, Fig. 5(a), and BCO and B2', Fig. 5(b), as a function of their thickness both for SeqQuest (full lines) and QE (dashed lines) surface energies. The surface energy used for the functions in Fig. 5 is that of the thickest slabs and the length reported in terms of that of the B2' slabs. For out-of-plane (Ti-out) slabs, B19' and BCO become more stable than B2' as the slab thickness decreases. This result may appear surprising since B2' has the lowest surface energy of the phases, however, as Eq. 4 shows, it is the product of free surface energy and cross-sectional area that governs the slab energetics. While entropic effects should be taken into account for a definite conclusion, these results indicate that the martensite transition temperature would increase with decreasing slab thickness. For in-plane slabs, B2' becomes more stable than B19' and BCO with decreasing size; our results predict a critical size of 2 nm below which martensitic transformation would not occur. The accuarcy of our mathematical model and the convergence of surface energies can be confirmed from Figure 5, where the actual DFT calculations (points) are compared with the model (lines). Our model indicates that the size effects the phase transformation temperature at sizes beyond the actual DFT calculations; for example, the energy difference of 20 nm thick slabs of B19' and B2' phases will be 15% smaller than the energy difference in the bulk for the in-plane slabs. It is also clear that these effects will be more important in wires (1D) or in clusters (0D).

VI. SUMMARY AND CONCLUSIONS

We used DFT-GGA to study the atomic structure and energetics of thin NiTi slabs of the various crystal structures that play a role in shape memory. We focus on the low energy [110] surfaces of B2', B19, B19' and BCO crystals. We find that the high-temperature austenite phase (B2') has the lowest surface energy followed by B19, B19' and BCO. For slabs with atomic displacements parallel to the free surface we predict a decrease in martensite transition temperature with decreasing slab thickness and a critical size of 2 nm for NiTi slabs, below which transformation would not occur. Such an inverse relationship between the critical transformation temperature

 (T_c) and size (diameter) has been observed in cylindrical iron nanowires with diameters ranging from 2.5 nm to 4 nm.⁴⁸In contrast, crystals with atomic displacements normal to the free surface, the martensite phases B19' and BCO become more stable with respect to B2' with decreasing slab thickness indicating an increase in the transition temperature. These results indicate a complex role of free surfaces on size effects in martensite transition temperature and additional DFT calculations on wires and spherical particles could provide important new insight. In addition, characterizing the effect of different surface passivations on the relative energy of nanoscale specimens with different crystal structure is critical to assess the potential use of NiTi and other shape memory alloys in nanoscale applications.

VII. ACKNOWLEDGEMENTS

This work was supported by the US Department of Energy Basic Energy Sciences (DoE-BES) program under program number DE-FG02-07ER46399 (Program Manager: John Vetrano). Computational resources of nanoHUB.org are gratefully acknowledged.

FIGURE CAPTIONS

Figure 1: Snap-shots of various crystal structures of NiTi and an in-plane view of how the Ni terminated and the Ti terminated [110] surfaces are cleaved. a) B2', b) B19, c) B19' (martensite), d) BCO. Ti and Ni atoms are indicated by blue and red spheres respectively.

Figure 2: Surface energy as a function slab thickness for the various crystal structures and orientations. Symbols joined by lines represent SeqQuest results (dashed lines are used for out-of-plane displacements and solid lines for in-plane cases. Light symbols(red online) correspond tosurface energycalculationsusing QuantumEspresso for the largest slab size considered to assess the accuracy of the predictions.

Figure 3: Percent surface relaxation for the 7-unit-cell-thick slabs for all crystal structures. (a) Inplane slabs, (b) Out-of-plane (Ti – terminated) slabs.

Figure 4: Atomic displacements of relaxed slabs (largest size) for all surfaces under consideration and snapshots of the relaxed structures (both side and cross-sectional views). (a) B2' out-of-plane slab, (b) B2' in-plane slab, (c) B19 out-of-plane slab, (d) B19 in-plane slab, (e) B19' out-of-plane slab (f) B19' in-plane slab (g) BCO out-of-plane slab (h) BCO in-plane slab.

Figure 5: Relative phase stability of the martensite phases B19' (a) and BCO (b) with respect to B2'. Full lines and dashed lines are model predictions using the surface energies predicted by SeqQuest and QE respectively, points (open symbols correspond to SeqQuest calculations and full symbol correspond to QE calculations) represent the actual calculations.

REFERENCES

⁴Alexander Thompson and Alejandro Strachan, Physical Review B, 81, 085429 (2010)

⁵C. Lopez del Castillo, B. G. Mellor, M. L. Blazquez and C. Gomez ScriptaMetallurgica, 21, 1711 – 1716 (1987)

⁶K. Mukunthan and L. C. Brown, Metallurgical Transactions A, 19A, 2921 – 2929 (1988)

⁷P.F. Brofman and G.S. Ansell, Metallurgical Transactions A, 14A, 1929 – 1931 (1983)

⁸Ying Chen and Christopher A. Schuh, ActaMaterialia, 59, 537 – 553 (2011)

¹K. Otsuka and X. Ren, Progress in Materials science 50, 511 – 678 (2005).

²Kaushik Bhattacharya, Sergio Conti, Giovanni Zanzotto and Johannes Zimmer Nature, 428, 55 – 59 (2004)

³Xiangyang Huang, Graeme J. Ackland and Karin M. Rabe, Nature Materials, 2, 307 – 311 (2003)

⁹T. Waitz, T. Antretter, F.D. Fischer and H.P. Karnthaler, Materials science and technology 24(8), 934 – 940 (2008).

¹⁰A.M. Glezer, E.N. Blinova, V.A. Pozdnyakov and A.V. Shelyakov, Journal of Nanoparticle Research 5, 551 – 560 (2003)

¹¹Osamu Kitakami,Hisateru Sato, Yutaka Shimada, Futami Sato and Michiyoshi Tanaka Physical Review B 21, 13849 (1997).

¹²Hong Seok-Yang and H.K.D.H. Bhadeshia, ScriptaMaterialia 60, 493 – 495 (2009).

¹³K. Asaki, T. Tadaki and Y. Hirotsu, Philosophical Magazine A 82, 463 – 478 (2002).

¹⁴K. Asaki, Y. Hirotsu and T. Tadaki, Journal of electron microscopy 48(4), 387 – 391 (1999).

¹⁵Jose San Juan, Maria L. Nŏ and Christopher A. Schuh, Nature Nanotechnology 4. 415 – 419 (2009).

¹⁶Bouville M and Ahluwalia R, ActaMaterialia 56(14), 3558 – 3567 (2008)

¹⁷ T. Waitz, K. Tsuchiya, T. Antretter and F.D. Fischer, MRS Bulletin 34, 814 – 821 (2009)

- ¹⁸ M.I. Haftel and Ken Gall, Physical Review B 74, 035420 (2006)
- ¹⁹JiankuaiDiao, Ken Gall and Martin L. Dunn Nature Materials 2, 656 660 (2003)
- ²⁰JiankuaiDiao, Ken Gall and Martin L. Dunn Physical Review B 70, 075413 (2004)

²¹ Harold S. Park, Ken Gall and Jonathan A. Zimmerman, Physical Review Letters 95, 255504 (2005) ²² Y. Kondo, Q. Ru and KunioTakayanagi, Physical Review Letters 82, 751 (1999)

²³Xiangyang Huang, Claudia Bungaro, VitaliyGodlevsky and Karin N. Rabe, Physical Review B, 65, 014108 (2001)

- ²⁴KarthikGuda Vishnu and Alejandro Strachan, ActaMaterialia 58, 745 752 (2010)
- ²⁵ N. Hatcher, O. Yu. Kontsevoi and A.J. Freeman, Physical Review B 79, 020202(R) (2009)
- ²⁶ N. Hatcher, O. Yu. Kontsevoi and A.J. Freeman, Physical Review B 80, 144203 (2009)
- ²⁷ Daniel Mutter and Peter Nielaba, Physical Review B 82, 224201 (2010)

²⁸P.A. Schultz, SeqQuest code http://dft.sandia.gov/Quest

²⁹P.J. Feibelman, Physical Review B 35, 2626 (1987).

- ³⁰P.J. Feibelman, Physical Review B 44, 3916 (1991).
- ³¹P. Hohenberg and W. Kohn, Physical Review 136, B864 (1967).

³²J.P. Perdew, K. Burke and M. Ernzerhof, Physical Review Letters 77, 3865 (1996).

³³D.R. Hamann, Physical Review B 40, 2980 (1989).

³⁴D.D. Johnson, Physical Review B 38, 12807 (1988).

³⁵P. Gianozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L. Chiarotti, M. Cococcioni, I. Dabo, A. Dal Corso, S. Fabris, G. Fratesi, S. de Gironcoli, R. Gebauer, U. Gerstmann, C. Gougoussis, A. Kokalj, M. Lazzeri, L. Martin-Samos, N. Marzari, F. Mauri, R. Mazzarello, S. Paolini, A. Pasquarello, L. Paulatto, C. Sbraccia, S. Scandolo, G. Sclauzero, A. P. Seitsonen, A. Smogunov, P. Umari and R. M. Wentzcovitch Journal of Physics Condensed matter 21, 395502 (2009)

³⁶David Vanderbilt, Physical Review B (Rapid communications) 41, 7892 (1990)

³⁷Methfessel and A.T. Paxton, Physical Review B 40, 3616 – 3621 (1989)

³⁸Broyden C G IMA Journal of Applied Mathematics 1970; 6(1): 76 – 90

³⁹Fletcher R Computer Journal 1970; 13:317 – 322

⁴⁰Goldfard D Mathematics of Computation 1970; 24: 23 – 26

⁴¹Shanno D F Mathematics of Computation 1970; 24: 647 – 656

⁴²AftabAlam, Brent Kraczek and D. D. Johnson, Physical Review B 82, 024435 (2010)

⁴³ Wang F E, Buehler W J and Pickart S, Japanese Journal of applied physics, 36(10), 3232 –

3239 (1965)

⁴⁴Kudoh Y. Tokonami M. Mivazaki S and Otsuka K. ActaMetallurgica, 33(11), 2049 – 2056

(1985)

⁴⁵G.D. Ayushina, E.S. Levin and P.V. Gel'd, Russian Journal of physics and chemistry, 43, 1548 (1969)

⁴⁶D.B. Miracle, ActaMetallurgica et Materialia, 41(3), 649 – 684 (1993)

⁴⁷Michael Nolan and Syed A.M. Tofail, Biomaterials, 31, 3439 – 3448 (2010)
⁴⁸ Luis Sandoval and Herbert M. Urbassek, Nano Letters 9(6), 2290 – 2294 (2009)











(d) BCO





Figure 3: Surface relaxation























