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Phys. Rev. B **84**, 064424 — Published 24 August 2011

DOI: [10.1103/PhysRevB.84.064424](https://doi.org/10.1103/PhysRevB.84.064424)

Phonon Density of States of Fe₂O₃ across High-Pressure Structural and Electronic Transitions

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Abstract

High-pressure phonon density of states of Fe₂O₃ across structural and electronic transitions has been investigated by nuclear resonant inelastic X-ray scattering and first-principles calculations, together with synchrotron Mössbauer, X-ray diffraction, and X-ray emission spectroscopies. Drastic changes in elastic, thermodynamic, and vibrational properties of Fe₂O₃ occur across the Rh₂O₃(II)-type structural transition at 40-50 GPa, whereas the Mott insulator-metal transition occurring after the structural transition only causes nominal changes in the properties of the Fe₂O₃. The observed anomalous mode softening behavior of the elastic constants is associated with the structural transition at 40-50 GPa, leading to substantial changes in the Debye-like part of the PDOS in the terahertz acoustic phonons. Our experimental and theoretical studies provide new insights into the effects of the structural and electronic transitions in the transition metal oxide compounds.

30 **Introduction**

31 Hematite (Fe₂O₃), an antiferromagnetic insulator under ambient conditions, is
32 regarded as an archetypal Mott-insulator for the trivalent transition-metal oxides (TMOs),
33 the majority of which crystallize in the corundum-type structure [1,2]. Since iron is the
34 most abundant transition metal in the Earth, hematite as an end-member ferric iron (Fe³⁺)
35 oxide is also an important proxy for characterizing the oxidation state, geomagnetism,
36 and geochemistry of the planet's interior [3-5]. Recent multidisciplinary studies have
37 shown that hematite undergoes a number of transitions under high pressures including
38 structural, insulator-metal, magnetic collapse, and electronic spin-pairing transitions [5-
39 18]. Of particular interest is the pressure-induced Mott insulator-metal transition with the
40 breakdown of the strong electronic *d-d* correlation and closure of the Mott-Hubbard *d-d*
41 band gap, resulting in a metallic phase with zero moment at approximately 50 GPa [1,10-
42 12,18]. Although the Mott transition was first proposed to occur concurrently with the
43 Rh₂O₃(II)-type structural and electronic transition, effectively decreasing the Fe³⁺ radius
44 and the unit cell volume in the shift to the low-spin paramagnetic state, the metallic state
45 is reported to occur in the Rh₂O₃(II)-type phase [5-18]. This observation calls for further
46 understanding on the interplays and associated effects between the structural, electronic,
47 and magnetic transitions in Fe₂O₃ at high pressures.

48 Electronic and structural transitions in the TMOs have been recently found to
49 significantly affect their physical and chemical properties under high pressures [19-25].
50 Specifically, the electronic spin-pairing transition of iron in ferropericlase ((Mg,Fe)O), an
51 abundant mineral in the Earth's lower mantle, is found to result in changes in density,
52 some of the elastic constants, and transport and rheological properties, which in turn
53 affect a broad spectrum of geophysical, geochemical, and geodynamic implications of the

54 deep Earth [22-24]. Furthermore, recent studies in the TMOs under high pressures also
55 show anomalous softening in elasticity as a result of the strong phonon-magnon coupling,
56 leading to substantial changes in the Debye-like part of the phonon density of states
57 (PDOS) [19-20]. Understanding the high-pressure PDOS behavior of hematite is
58 particularly interesting because of the multiple aforementioned transitions and their
59 potential couplings. Yet, the effects of these transitions on elastic, thermodynamic, and
60 vibrational properties of Fe₂O₃ remain largely lacking. Here we have measured the partial
61 PDOS of iron in Fe₂O₃ by nuclear resonant inelastic X-ray scattering (NRIXS) up to 85
62 GPa in a diamond anvil cell (DAC). NRIXS is a relatively new synchrotron technique
63 that has been successfully applied to study lattice dynamics of ⁵⁷Fe-containing
64 compounds under extreme pressures and temperatures [25-28]. Together with first-
65 principles theoretical calculations, the PDOS are used to characterize the behavior of
66 Fe₂O₃ across the structural and electronic transitions. We observed significant softening
67 in the elastic properties as well as other major changes in thermodynamic and vibrational
68 properties of Fe₂O₃ between 40 and 50 GPa, which are attributed to the structural
69 transition in hematite. The magnetic collapse is reported to occur in the Rh₂O₃(II) phase
70 between 55 GPa and 75 GPa, causing only nominal changes in the PDOS of the Fe₂O₃.

71

72 **Experimental Details**

73 ⁵⁷Fe-enriched Fe₂O₃ (>95% enrichment) was purchased from the Cambridge Isotope
74 Laboratories, Inc., and was characterized for its crystal structure and chemical
75 composition by X-ray diffraction and electron microprobe analyses, respectively.
76 Samples measuring approximately 25 μm thick and 40 μm wide were loaded into DACs
77 with 300 μm flat culets or beveled diamonds (150 μm in the inner culets and 300 μm in

78 the outer culets) with Be gaskets of 3 mm in diameter and cubic BN gasket inserts.
79 Pressures were determined from the ruby spheres in the sample chamber using the ruby
80 fluorescence scale [29]. High-pressure NRIXS experiments were conducted at sector 3 of
81 the Advanced Photon Source (APS), Argonne National Laboratory (ANL) [25-28].
82 Energy spectra were obtained by tuning the X-ray energy (approximately ± 80 to ± 95
83 meV in steps of 0.25 meV) around the nuclear transition energy of 14.4125 keV with an
84 energy resolution of 1 meV and an X-ray beamsize of approximately 10 μm . The $K\alpha,\beta$
85 fluorescence radiation from the ⁵⁷Fe-enriched Fe₂O₃ sample, emitted with time delay
86 relative to the incident X-ray pulses due to the lattice excitations of the iron sublattice,
87 was collected by three avalanche photodiode detectors (APD), whereas the synchrotron
88 Mössbauer spectroscopy (SMS) spectra were collected by a fourth detector in the forward
89 direction. A quasi-harmonic model was used to extract the PDOS from the measured
90 energy spectra (Fig. 1) [25-27]. In this model, the atomic motions relative to the
91 temperature-dependent averaged position are assumed to be harmonic under the given
92 conditions of pressure, temperature, and other parameters. Thermal effects such as
93 change of force constants with atomic distances are allowed to change, but the vibrations
94 are still assumed to occur in a harmonic potential [25-27]. Previous studies have
95 confirmed the reliability of this model to extract the PDOS and the bulk Debye sound
96 velocity (V_D) of Fe-containing compounds under high pressures [28].

97

98 **Experimental Results**

99 Elastic, thermodynamic and vibrational properties of the iron component in Fe₂O₃
100 have been derived from the integration of the measured PDOS in Fe₂O₃ (Figs. 1,2). The
101 V_D of the sample is derived from parabolic fitting of the low-energy slope of the PDOS in

102 the range of approximately 0.2 meV to 15 meV (Fig. 2) [26,27]. In this lattice dynamics
 103 model, the Debye velocity of a material is derived from the initial slope of the density of
 104 states versus energy squared (E^2) plot (a parabolic function in the PDOS versus energy).
 105 In the acoustic region, the translation motions of all the atoms in the system (in this case
 106 both Fe and O) are in phase. Therefore, the Debye velocity estimated from the Fe PDOS
 107 should be exactly the same as from the full consideration of the density of states. The
 108 Debye theory is valid for the acoustic modes in the harmonic solids with Debye-like low-
 109 frequency dynamics, and Fe₂O₃ has been shown to exhibit Debye-like lattice dynamic
 110 behavior in previous studies [28]. It has also been demonstrated that the Debye sound
 111 velocity (V_D) measured in NRIXS corresponds to the V_D of the whole matrix (not the Fe
 112 atoms alone) [28]. That is, the projected density of states of Fe represents the total
 113 phonon density of states of Fe₂O₃ in the low-frequency region. For this reason, the
 114 extraction of the long wavelength properties of the Fe₂O₃ from just the Fe partial DOS is
 115 thus in principle valid.

116 The characteristic temperature or Lamb-Mössbauer temperature (T_{LM}) in Fig. 2 is
 117 defined as:

$$118 \quad \frac{1}{T_{LM}} = k_B \int \frac{2E_{Recoil}}{E^2} g(E) dE$$

119 where k_B is the Boltzmann's constant, E_{recoil} is the recoil energy for ⁵⁷Fe nucleus (1.96
 120 meV), and $g(E)$ is the partial phonon density of states. The recoil free fraction, commonly
 121 known as Lamb-Mössbauer factor (f_{LM}), is related to the characteristic temperature, or
 122 Lamb-Mössbauer temperature (T_{LM}), as follows:

$$123 \quad -\ln f_{LM}(T) = \frac{T}{T_{LM}} = k_0^2 \langle x^2 \rangle$$

124 where k_0 is the momentum of the photons with 14.412 keV, which equals to 7.3 Å, and
 125 $\langle x^2 \rangle$ is the average displacement. One can also relate T_{LM} to the Debye temperature, as
 126 follows:

$$127 \quad -\ln f_{LM}(T) = k_B T \frac{6E_{Recoil}}{E_D^2(T)}$$

128 where E_D is the Debye cut-off energy, as given by the maximum in the Debye
 129 distribution of the phonon density of states. Hence, $T_{LM} = \frac{E_D^2}{6k_B E_{Recoil}}$.

130 Abnormal behavior of these properties is observed to occur between 40 GPa and 50
 131 GPa. Specifically, V_D , the mean force constant (D_{av}), and the Lamb-Mössbauer factor
 132 (f_{LM}) suddenly drop between 40 and 50 GPa, suggesting a softening in Fe₂O₃ in this
 133 pressure range. Furthermore, vibrational specific heat (C_{vib}) and vibrational entropy (S_{vib})
 134 decrease with increasing pressures, but jump significantly between 40 and 50 GPa. To
 135 understand these observations, we have employed X-ray diffraction, SMS, and X-ray
 136 emission spectroscopy (XES) to confirm the crystal structures, hyperfine parameters, and
 137 total spin momentum of Fe₂O₃ under high pressures, respectively (Fig. 3,4). X-ray
 138 diffraction spectra were consistent with a structural transition from the corundum to the
 139 Rh₂O₃(II) structure that has been reported to occur at approximately 40-50 GPa [7-10],
 140 whereas the SMS and XES results indicated that the magnetic collapse occurs in the high-
 141 pressure Rh₂O₃(II) phase and is completed at approximately 75 GPa, consistent with
 142 previous studies [5-15,18] (Fig. 3,4).

143

144 **Theoretical Calculations**

145 In order to qualitatively decipher the interplay between the structural, magnetic, and
 146 electronic transitions of Fe₂O₃ under high pressures, first-principles electronic and

147 phonon calculations using the VASP suite program were performed to investigate
148 structural and magnetic transitions in hematite (Fig. 5-8) [30-33]. Projected augmented
149 (PAW) potentials were used for both the Fe and O atoms, and the semi-core 3s and 3p
150 orbitals of Fe were treated as valence [31,32]. The rotationally-invariant LSDA+U model
151 with the Perdew-Burke-Ernzerhof (PBE) exchange correlation function (U=4.0 eV and J
152 = 0.9 eV) in the Generalized Gradient Approximation (GGA) was used to model the
153 strongly correlated Fe 3d states in the AFM corundum structure [30-32]. Based on
154 previous studies [14], we have used the Rh₂O₃(II)-type structure (space group: *Pbcn*)
155 with experimentally-reported lattice parameters for our theoretical calculations for the
156 high-pressure phase [14]. It is well known that it is necessary to include the Hubbard U to
157 describe the band gap of the insulating magnetic state. Whether it is necessary to involve
158 the Hubbard U term to describe the metallic non-magnetic state is not certain. We have
159 performed calculations with the same Hubbard U parameter as in the corundum structure
160 but the calculated PDOS does not agree with experimental results at all. The GGA
161 method, on the other hand, agrees well with experiments, leading us to rule out the use of
162 the Hubbard U in the calculations. It is found that the GGA approximation is consistent
163 with experimental results and adequate to describe the electron and phonon structures of
164 the high pressure non-magnetic Rh₂O₃ (II) phase (Fig. 5) [34]. Phonon dispersion and Fe
165 projected density of states (PDOS) were calculated with the small displacement method
166 [33] employing supercells constructed from a 2×2×2 replicate of the corundum structure
167 and a 1×2×2 replicate of the Rh₂O₃(II) crystallographic unit cell. Calculations were
168 performed at 40 GPa for the AFM corundum-type phase, at 55 GPa for both the non-
169 magnetic (NM) corundum-type phase and AFM Rh₂O₃(II)-type phase, and at 70 GPa for
170 the NM Rh₂O₃(II) (Figs. 5-8). We have also calculated the stability of the NM corundum

171 and AFM Rh₂O₃(II) structures at these pressures and have found them to be unstable
172 when compared to the AFM corundum and NM Rh₂O₃(II) structures.

173 The theoretical Fe PDOS of the AFM corundum structure at 40 GPa and the NM
174 Rh₂O₃ (II) structure at 70 GPa are compared with corresponding experimental results in
175 Fig. 5a and 5c, respectively. The experimentally observed decrease in V_D of 14% from
176 the magnetic corundum to metallic Rh₂O₃(II) phase is correctly reproduced by the
177 calculated value of 16% (Figs. 2,5). The agreements are very good and all the salient
178 features of the experimental PDOS are correctly reproduced. The most distinctive
179 differences in the calculated vibrational PDOS features are the drastic shift of the first
180 PDOS peak from approximately 21 meV in the AFM corundum-type phase to
181 approximately 37 meV in the AFM Rh₂O₃(II)-type structure at 55 GPa, and as a
182 consequence, a significant drop in the gradient of the acoustic phonon dispersion. These
183 predictions are qualitatively consistent with experimental observations on the softening of
184 the Debye sound velocity across the structural transition (Figs. 1,5). There are also shifts
185 in the high frequency vibrations, dominated by the oxygen motions, to higher energy.
186 Across the structural transition, the Debye sound velocity suddenly becomes much slower
187 in the Rh₂O₃(II)-type phase than in the corundum-type phase; though, it should be noted
188 that these phases are all in the AFM state. In absence of the structural transition, the low-
189 frequency Debye-like vibrations in the corundum-type and the Rh₂O₃(II)-type structures,
190 respectively, behave rather normally with a weak pressure effect; this weak pressure
191 effect persists across the AFM to the NM transition in the Rh₂O₃(II)-type phase,
192 suggesting that electronic transitions do not significantly affect elastic and vibrational
193 properties of Fe₂O₃ at high pressures.

194

195 Discussion

196 Based on our experimental results and first-principles calculations, here we address
197 how the properties of Fe₂O₃ are affected by structural and electronic transitions under
198 high pressures. Together with the previous studies, it is now well documented that the
199 transition from the AFM hematite to the AFM Rh₂O₃(II) phase occurs at approximately
200 40-50 GPa with a volume reduction of approximately 10% and the shortening of the bond
201 lengths [5-15]. Our observed dramatic changes in the elastic, vibrational, and
202 thermodynamic properties correspond to the structural transition to the AFM Rh₂O₃(II)-
203 type occurring at 40-50 GPa, without the involvement of the spin-pairing and magnetic
204 collapse transitions. This structural transition is thus associated with the softening in the
205 lattice vibrational phonons in Fe₂O₃ and significant reductions in the V_D , Lamb-
206 Mössbauer factor, and the mean force constant, among others. The drop in the V_D across
207 the transition suggests that compressional and shear velocities of Fe₂O₃ would also
208 decrease significantly as well. On the other hand, the mean force constant represents the
209 short-range repulsive, interatomic forces arising from the charge distributions and Hund's
210 rules of atoms, and its decrease thus corresponds to a softening in the lattice
211 incompressibility and likely the lattice strength within the transition. The Lamb-
212 Mössbauer factor is related to the mean-square displacement of the iron atoms in the
213 lattice, $\langle x^2 \rangle$, through $f_{LM} = \exp(-k^2 \langle x^2 \rangle)$, where k is the wave number of the resonant X-
214 ray. That is, the structural transition reduces the lattice displacement of the Fe³⁺ atoms
215 and is manifested in the reduction of the Lamb-Mössbauer factor by approximately 5%.
216 This transition also results in an increase in the thermodynamic vibrational heat capacity
217 and entropy.

218 Previous studies showed that the electronic transition and the magnetic collapse, the
219 Mott transition, occur at higher pressures only in the Rh₂O₃(II) phase [13]. A Mott
220 transition, as defined by Mott himself, is an isostructural paramagnetic metal to
221 paramagnetic insulator transition [1,21]. This isostructural transition implies that the
222 crystal symmetry of the Rh₂O₃(II) phase is preserved across the magnetic collapse. Our
223 studies, together with previous results [7-14], confirm this behavior, in which the AFM
224 character with magnetic moments persists up to approximately 75 GPa before the
225 magnetic collapse is fully completed. Based on our NRIXS results and first-principles
226 calculations, the Mott transition does not significantly affect the elastic, thermodynamic,
227 and vibrational properties of the Rh₂O₃(II)-type Fe₂O₃; these properties change rather
228 monotonically with increasing pressures (Fig. 2). The high-spin AFM Rh₂O₃(II)-type
229 phase is found to be dynamically unstable at 55 GPa in our calculations, and a soft
230 phonon mode is found at the M symmetry point.

231

232 **Conclusions**

233 In summary, significant changes in the PDOS of Fe₂O₃ were observed with the
234 softening on the sound velocities, interatomic force constant, and Lamb-Mössbauer factor
235 associated with the corundum-type to the Rh₂O₃(II)-type structural transition at 40-50
236 GPa, but the isostructural Mott transition only occurs in the Rh₂O₃(II) phase and does not
237 have a strong effect on the properties of Fe₂O₃. The structural transition under pressure
238 corresponds to anomalous mode softening behavior of the elastic constants preceding the
239 Mott transition, leading to substantial changes in the Debye-like part of the PDOS in the
240 terahertz acoustic phonons. Our studies on the high-pressure behavior of the archetypal

241 Fe₂O₃ here provide new insights into the effects of the structural and electronic
242 transitions in the TMO compounds.

243 We acknowledge XOR-3 and GSECARS, APS, ANL for the use of the synchrotron
244 and laser facilities, and TACC for the computational facilities. Use of the Advanced
245 Photon Source was supported by U.S. Department of Energy (DOE), Office of Science,
246 Basic Energy Sciences (BES), under contract No. DE-AC02-06CH11357. J.F.L.
247 acknowledges financial support from Energy Frontier Research in Extreme Environments
248 (EFree) Center, NSF Earth Sciences (EAR-0838221), and Carnegie/DOE Alliance Center
249 (CDAC).

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- 309

310 **Figure Captions:**

311 **FIG 1.** Representative PDOS of Fe from ⁵⁷Fe₂O₃ at high pressures. The PDOS spectra are
312 derived from the NRIXS experiments in a high-pressure diamond anvil cell.
313 Experimental error bars are shown as vertical gray bars.

314 **FIG 2.** Elastic, thermodynamic, and vibrational parameters of Fe₂O₃ as a function of
315 pressure obtained from integration of the PDOS. **(a)** Debye sound velocity; **(b)** mean
316 force constant, D_{av} ; **(c)** Lamb-Mössbauer factor, f_{LM} ; **(d)** vibrational specific heat, C_{vib}
317 (k_B , Boltzmann constant); **(e)** vibrational entropy, S_{vib} ; **(f)** characteristic temperature or
318 Lamb-Mössbauer temperature, T_{LM} . We note that these values only represent the
319 contribution of the Fe sublattice in Fe₂O₃. Using the theoretical densities we computed
320 the sound velocities of Fe₂O₃ in the corundum structure at 40 GPa and Rh₂O₃(II) structure
321 at 70 GPa to be 4.7 km/s and 3.8 km/s, respectively.

322 **FIG 3.** (Color Online) Representative X-ray diffraction **(a)** and Fe-K β X-ray emission **(b)**
323 patterns of Fe₂O₃ at high pressures. X-ray diffraction spectra confirmed the structural
324 transition from the corundum-type to the Rh₂O₃(II)-type phase at high pressures. The
325 intensity of the satellite K β ' peak can be used to understand the electronic spin-pairing
326 and metallization in Fe₂O₃. Its intensity remains similar across the structural transition,
327 but reduces significantly at higher pressures.

328 **FIG 4.** (Color Online) **(a)** Representative synchrotron Mössbauer spectra (SMS) of Fe₂O₃
329 at high pressures. Corresponding energy spectra calculated from the fits are shown in the
330 right panels. Red dots: experimental SMS spectra; black lines: modeled spectra. **(b).**
331 Analyzed hyperfine parameters and percentage of the magnetic phase in high-pressure
332 Fe₂O₃. Up to 55 GPa, the only spectral component is that of the low-pressure magnetic
333 phase characterized by the hyperfine field of 51 T, a typical value of the hyperfine field

334 for ionic ferric oxide bonding. A nonmagnetic quadrupole-split component emerges at
335 approximately 55 GPa, which we assigned as the high-pressure component, coexisting
336 with the low-pressure magnetic-split component. The relative abundance of the low-
337 pressure component continues to decrease until the magnetic collapse is completed at
338 around 75 GPa. We note that the hyperfine field of the magnetic component is slightly
339 reduced by about 10% within the transition.

340 **FIG 5.** (Color Online) Projected Fe vibrational phonon density of states (PDOS) of Fe₂O₃
341 at high pressure from first-principles calculations. **(a)** the AFM corundum-type phase at
342 40 GPa; **(b)** the theoretical AFM corundum-type phase (black line) and the NM
343 Rh₂O₃(II)-type phase (red line) at 55 GPa; Black lines, corundum-type phase; red lines:
344 Rh₂O₃(II)-type phase; **(c)** the NM Rh₂O₃(II)-type phase at 75 GPa. The black and red
345 circles in (a) and (c) represent the experimental and theoretical Fe PDOS, respectively.

346 **FIG 6.** Calculated GGA phonon dispersive curves (top) and Fe projected density of
347 states (bottom) of the non-magnetic Rh₂O₃ (II) structure of Fe₂O₃ at 55 GPa.

348 **FIG 7.** Calculated GGA phonon dispersive curves (top) and Fe projected density of
349 states (bottom) of the non-magnetic Rh₂O₃ (II) structure of Fe₂O₃ at 70 GPa.

350 **FIG 8.** (Color Online) Calculated Fe projected density of states and total density of
351 states of Fe₂O₃ in the corundum structure (top) and Rh₂O₃ (II) structure (bottom).

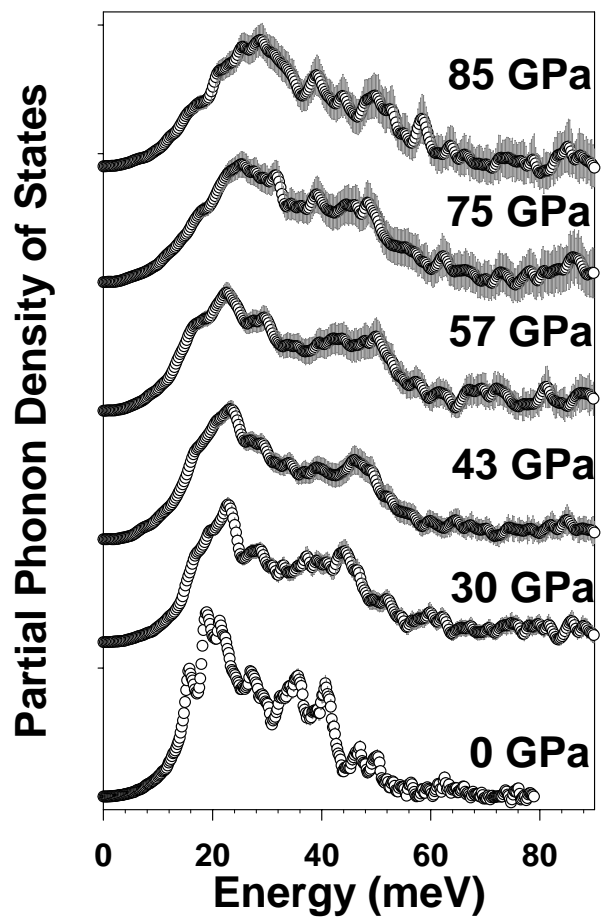


FIG. 1

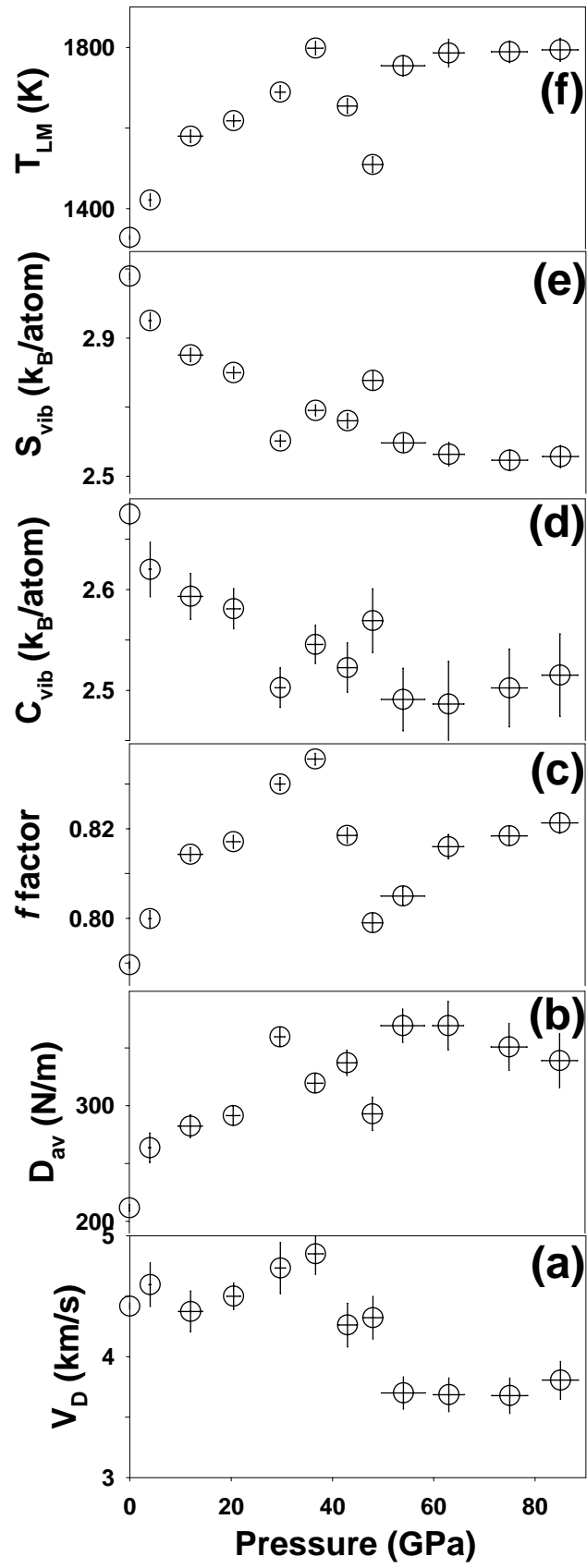


FIG. 2

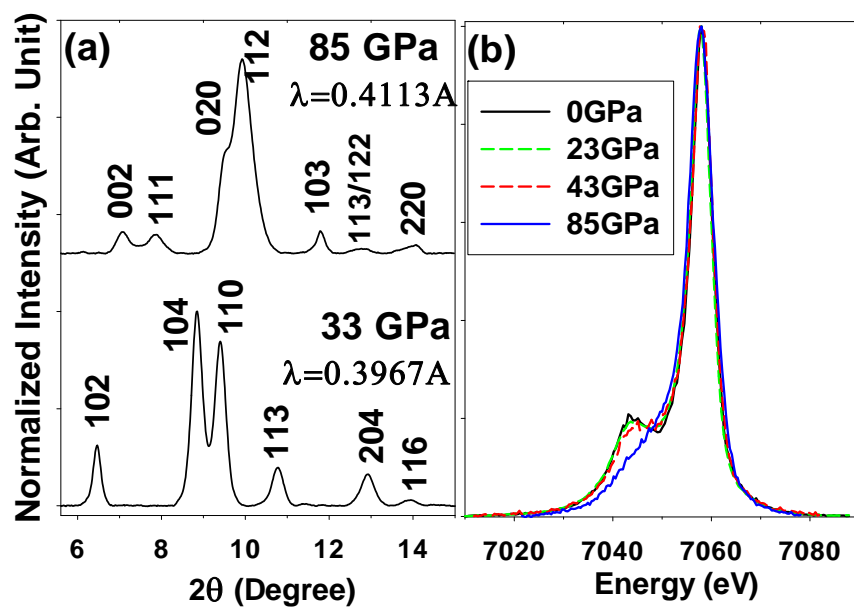


FIG. 3

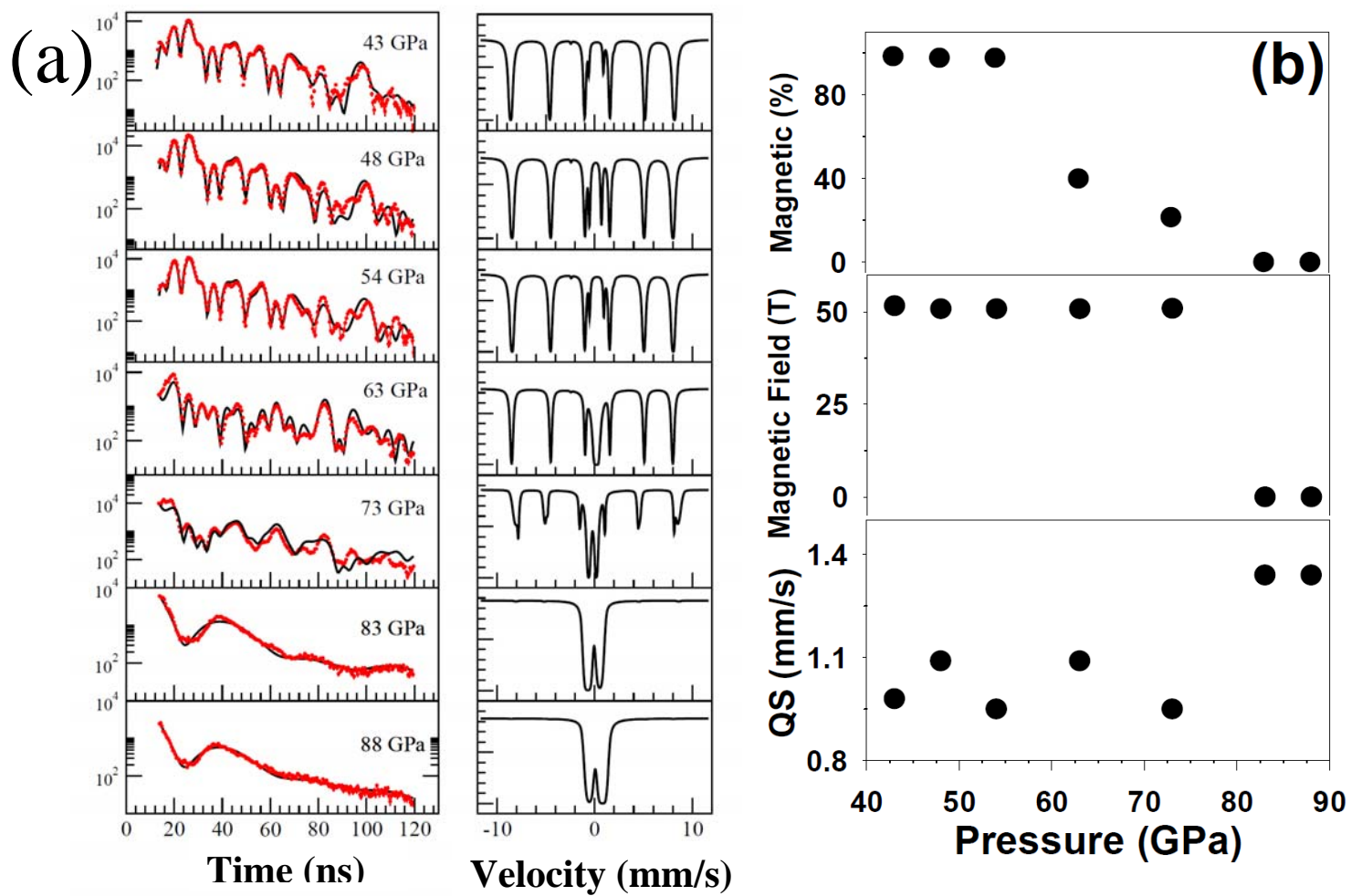


FIG4

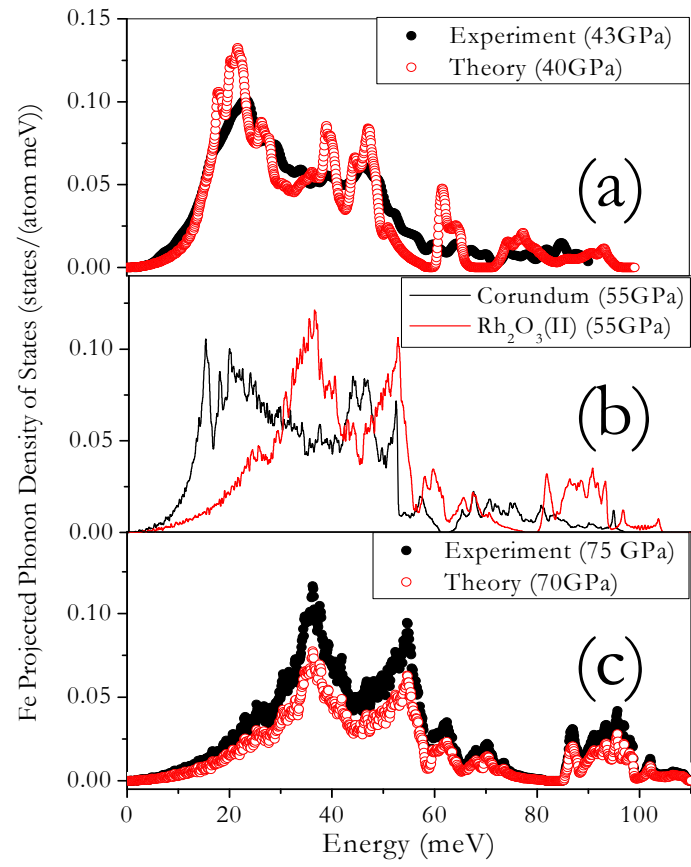


Fig. 5

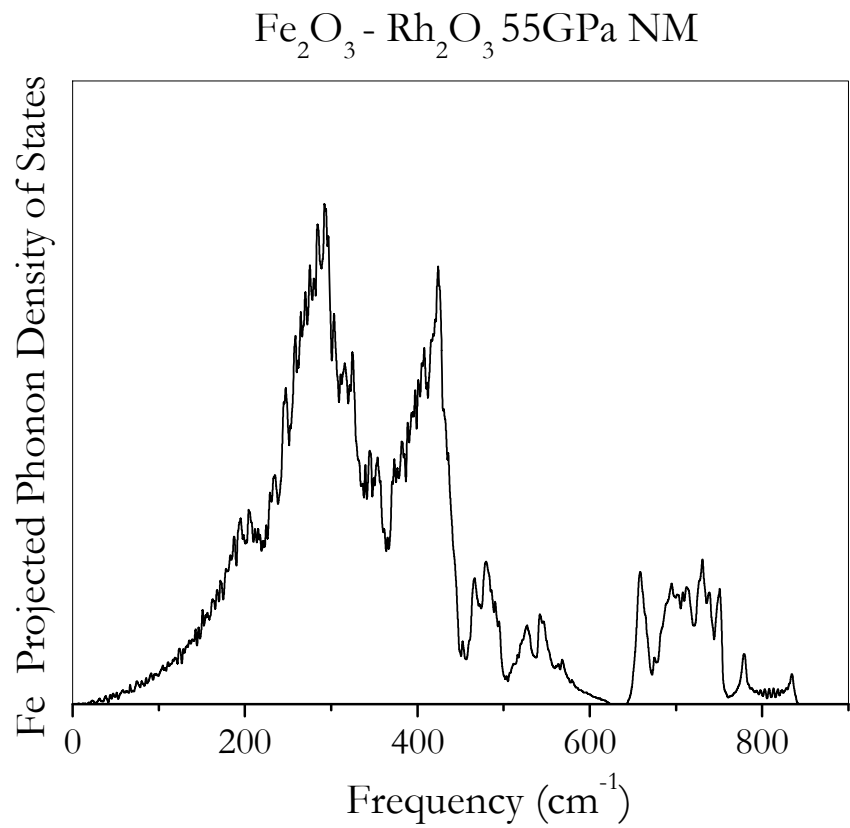
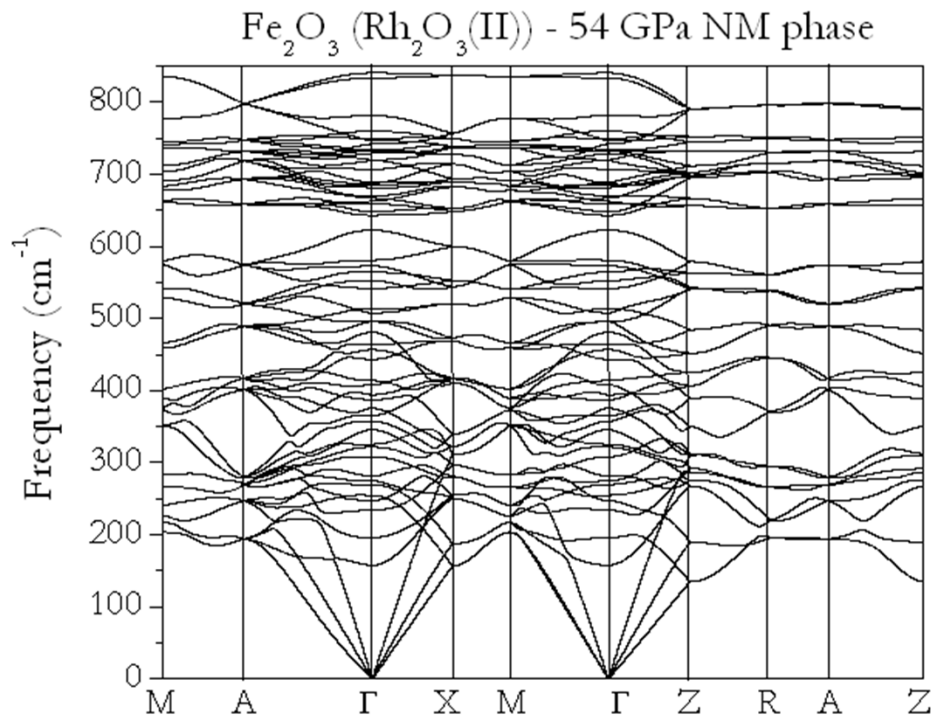


FIG6

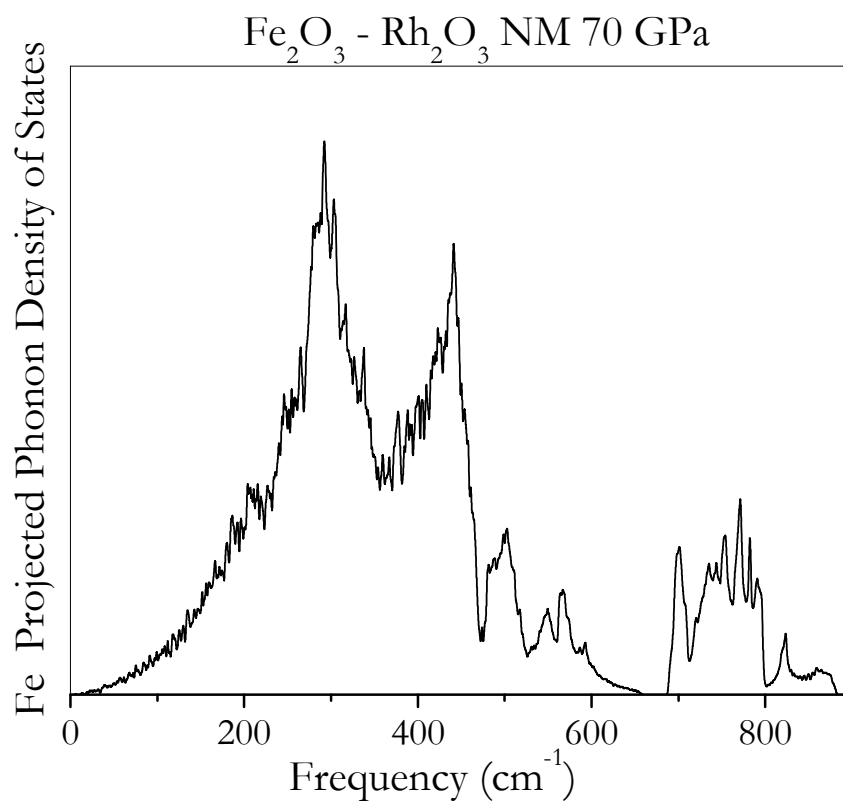
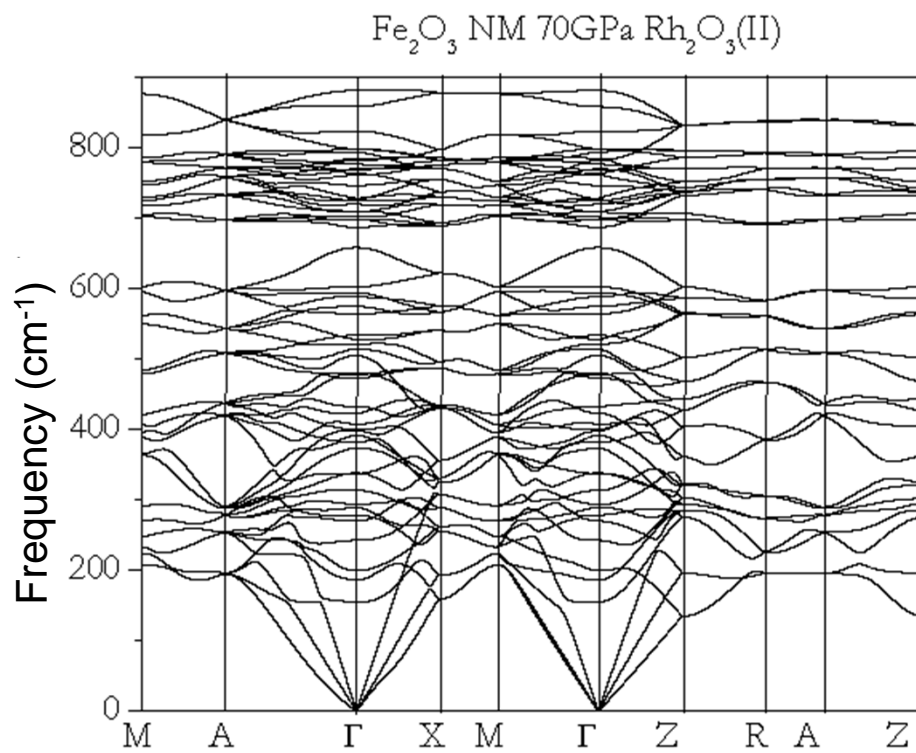


FIG7

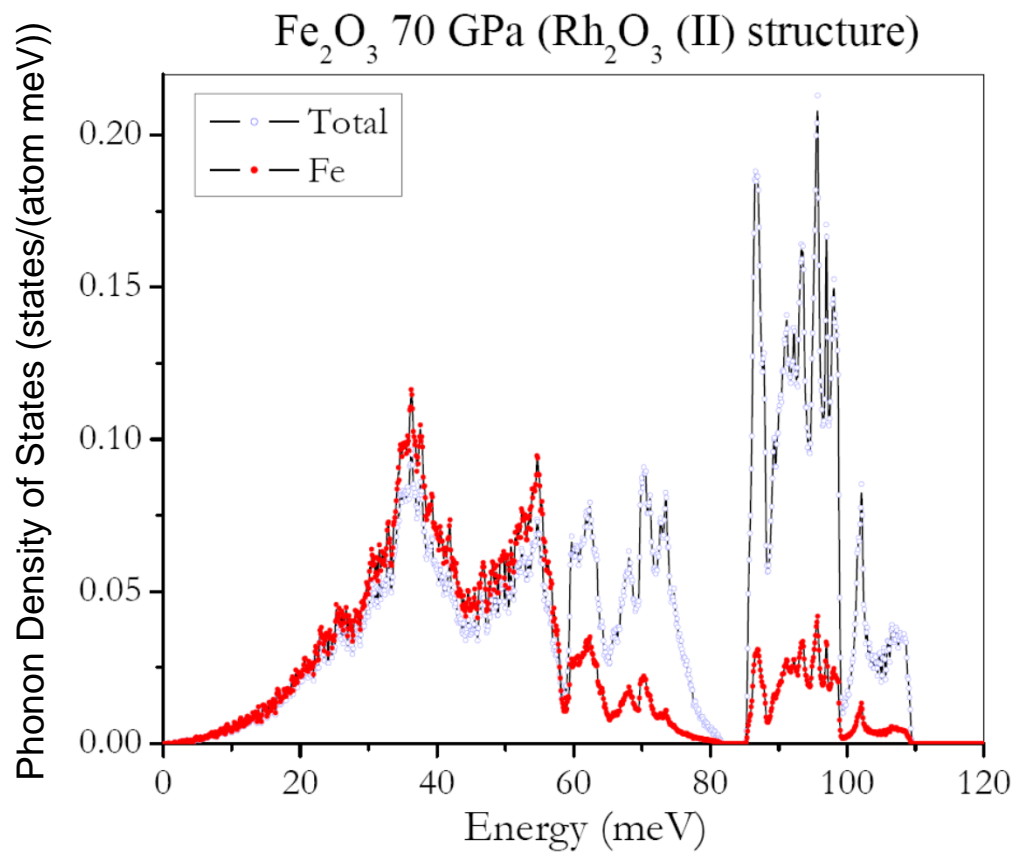
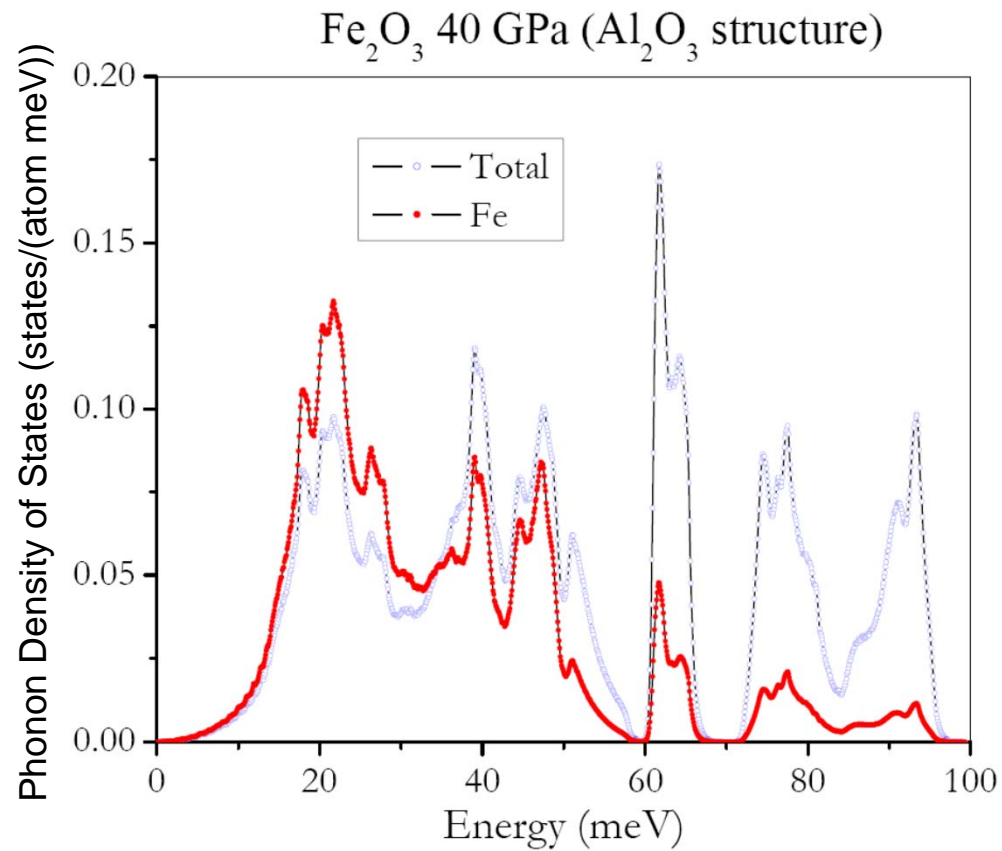


FIG8