

CHCRUS

This is the accepted manuscript made available via CHORUS. The article has been published as:

New Verbeekite-type polymorphic phase and rich phase diagram in the math

xmlns="http://www.w3.org/1998/Math/MathML">mrow>ms ub>mi>PdSe/mi>mrow>mn>2/mn>mo>-/mo>mi>x/mi> /mrow>/msub>msub>mi>Te/mi>mi>x/mi>/msub>/mrow >/math> system

 Wenhao Liu, Mehrdad Rostami Osanloo, Xiqu Wang, Sheng Li, Nikhil Dhale, Hanlin Wu, Maarten L. Van de Put, Sabyasachi Tiwari, William G. Vandenberghe, and Bing Lv Phys. Rev. B **104**, 024507 — Published 22 July 2021 DOI: 10.1103/PhysRevB.104.024507

New Verbeekite-type polymorphic phase and rich phase diagram in the $PdSe_{2-x}Te_x$ system

Wenhao Liu¹, Mehrdad Rostami Osanloo¹, Xiqu Wang³, Sheng Li¹, Nikhil Dhale¹, Hanlin Wu¹,

Maarten L. Van de Put², Sabyasachi Tiwari², William G. Vandenberghe², and Bing Lv^{1,2*}

¹Department of Physics, University of Texas at Dallas, Richardson, Texas 75080, USA

²Department of Materials Science and Engineering,

University of Texas at Dallas, Richardson, Texas 75080, USA

³Department of Chemistry, University of Houston, Houston Texas 77004, USA

(Dated:)

We report a combined experimental and theoretical study of the $PdSe_{2-x}Te_x$ system. With increasing Te fraction, structural evolutions, first from an orthorhombic phase (space group Pbca) to a monoclinic phase (space group C2/c) and then to a trigonal phase (space group P-3m1), are observed accompanied with clearly distinct electrical transport behavior. The new monoclinic phase (C2/c) belongs to the very rare Verbeekite polymorphism and is discovered within a narrow range of Te composition ($0.3 \le x \le 0.8$). Electronic calculations and detailed transport analysis of the Verbeekite polymorphic $PdSe_{1.3}Te_{0.7}$ phase are presented. In the trigonal phase region, superconductivity with enhanced critical temperature is also observed within a narrow range of Te content ($1.0 \le x \le 1.2$). The rich phase diagram, new polymorphic structure as well as abnormally enhanced superconductivity could further stimulate more interest to explore new types of polymorphs and investigate their transport and electronic properties in the transition metal dichalcogenides family that are of significant interest.

I. INTRODUCTION

The study of a variety of polymorphic structures of transition metal dichalcogenides (TMDs), and the discoveries of their unique properties through atomic-scale structure control, have emerged as a research frontier of science for new exciting physics as well as miniaturizing of electronic devices, energy conversion and storage [1– 9]. TMD polymorphs are chemically rather simple and structured with fundamental MX_2 layers where M is a transition metal (such as V, Nb, Ta, Ti, Zr, Hf, Mo, W, Pd, and Pt) and X is a chalcogen atom (such as S, Se, or Te)[10]. With strong in-plane bonding and weak out-of-plane van der Waals interactions, these materials can be easily exfoliated down to atomic thickness, and thus enable bottom-up atomic-scale structure control and unique stacking/twisting to reveal new physics or novel functionalities [11–13].

The research of various structures of TMDs have a long and fruitful history. Various polymorphic structures appear starting from either the trigonal prismatic or the octahedral coordination of the metal atoms. Depending on the different stacking order of atomic planes and possible distortions, the most commonly encountered arrangements are 1T (T: trigonal), 2H (H: hexagonal), 3R (R: rhombohedral), 1T' (distorted 1T into monoclinic phase), and T_d (distorted 1T into orthorhombic phase)[14–18]. The digits (1-3) indicate the number of layers in the stacking sequence in the primitive unit cell.

The different polymorphs in TMDs often display drastically different physical phenomena. For example, three different polymorphs showing distinct properties are found for the MoTe₂ phases: 2H-MoTe₂ shows semiconducting behaviors with a bandgap of about 1.1 eV, 1T'-MoTe₂ has a metallic behavior[19, 20], and T_d-MoTe₂ is topologically nontrivial and a candidate Weyl semimetal[21, 22]. Another example is found in the TaSe_{2-x}Te_x system, where various polymorphic structures have been observed. With increasing Te concentration, the material appears as an incommensurate chargedensity-wave (ICDW) 2H-TaSe₂ phase, as a 3R phase with a maximal superconducting transition temperature (T_c) of 2.4K, as a different 1T phase (also superconducting with lower T_c at 0.5-0.7 K), and finally as a normal metallic monoclinic TaTe₂ phase[23].

Interestingly, PdS₂ and PdSe₂, less studied TMDs[24– 26], have a puckered pentagonal configuration with orthorhombic space group Pbca shown in Fig. 1(a). The *Pbca* space group is rarely found in TMDs and hosts pentagons. Pentagons are often considered as topological defects or geometrical frustrations [27]. Pentagonal graphene and SnS_2 have been theoretically predicted to possess quite unique physical properties, such as unusual negative Poissons ratio or a room-temperature quantum spin Hall insulator state [28, 29]. Different from the other TMDs where the metal coordination is either trigonal prismatic or octahedral (sometimes distorted), the Pd metal configuration in PdS_2 and $PdSe_2$ are in fact rectangular nets. $PdSe_2$ is highly stable in air down to the monolayer limit[30]. The bandgap varies greatly from 0.5 eV in bulk to 1.37 eV in monolayers, and the thin PdSe₂ field-effect transistors exhibit intrinsic ambipolar characteristic and high electron mobility [31, 32]. Under high pressure, bulk PdSe₂ transforms from a pentagonal layered structure to a pyrite type structure and superconductivity up to 13.1 K emerges, which is the highest

^{*} blv@utdallas.edu

critical temperature (T_c) among all the TMD materials to date[33]. On the other hand, PdTe₂ adopts the layered 1T structure with space group P-3m1 (Fig.1(b)). PdTe₂ has an octahedral (i.e. trigonal antiprismatic) coordination, is metallic, and is characterized by an ABC Se-Pd-Se stacking order within the layer. Dirac semimetal and superconductivity with T_c of 1.65 K have been experimentally verified to coexist in PdTe₂[34–36].

The distinct difference in structure and electronic properties of PdSe₂ and PdTe₂ motivate us to explore the possible structural evolution/transformation, and their associated electrical transport behavior changes in the PdSe_{2-x}Te_x system. We have demonstrated previously that the superconductivity of PdTe₂ can be enhanced up to 2.73 K when half of the Te in PdTe₂ is replaced by Se atoms[37]. Here, we study the full range of the PdSe_{2-x}Te_x ($0 \le x \le 2$) system and discover a new C2/cpolymorphic structure analogues to the high-pressure PdSe₂ Verbeekite phase[38, 39]. We complement our experimental results with theoretical first principles calculations. The PdSe₂-PdTe₂ phase diagram and associated electrical transport results, including superconductivity, are presented.

II. EXPERIMENT DETAILS

The $PdSe_{2-x}Te_x$ crystals were synthesized using selfflux methods. Pd ingots (99.95%, Alfa Aesar), Se shots (99.999%, Alfa Aesar) and Te pieces (99.999%, Alfa Aesar) were mixed in the right stoichiometric doping ratio in an Ar glovebox with a total moisture and oxygen level less than 0.1 ppm. The source elements were loaded inside a silica tube, which was then flame sealed under vacuum and placed in a box furnace. The temperature



was slowly increased to 800 °C, held for three days and followed by furnace cooling down to room temperature.

Chemical composition of the yield crystals was verified by energy-dispersive X-ray spectroscopy (EDX) on a DM07 Zeiss Supra 40 Scanning Electron Microscope. Powder X-ray diffraction (XRD) measurements were carried out at room temperature on crushed crystals using a Rigaku SmartLab X-ray diffractometer equipped with Cu-K α radiation. Rietveld refinement was carried out using GSAS-II[40]. Resistivity was measured using the four-probe method in a Quantum Design Physical Property Measurement System (PPMS) down to 1.8 K. Four gold wires (30 μm in diameter) were pasted on the sample surface by silver epoxy as four probes.

The crystal structure of the new PdSe_{1.2}Te_{0.8} phase was determined with single crystal X-ray data measured on a Bruker SMART diffractometer equipped with an Apex II area detector and an Oxford Cryosystems 700 Series temperature controller with a Mo $K\alpha$ source ($\lambda = 0.71073$). The collected dataset was integrated using the Bruker Apex-II program, with the intensities corrected for the Lorentz factor, polarization, air absorption, and absorption due to variation in the path length through the detector faceplate. The data was scaled, and absorption correction was applied using SADABS. The structure was solved by using the intrinsic phasing method in SHELXT and refined using SHELXL with all atoms refined anisotropically.

The theoretical calculations are based on Density Functional Theory (DFT) as implemented in Vienna ab initio simulation package (VASP) [41]. To correctly account for the van der Waals forces, we employed the nonlocal optPBE-vdW functionals proposed by Dion et al. [42] with parameters by Oyedele *et al.* and Klime *et* al.[27, 43]. Electron correlation effects in the d-orbitals of Pd were accounted for in the PBE+U approximation with U = 4 eV. To study various compositions of $PdSe_{2-x}Te_x$, we used $2 \times 2 \times 1$ supercells, providing 8 compositions of $x = \{0, \frac{1}{4}, \frac{1}{2}, \frac{3}{4}, 1, \frac{5}{4}, \frac{3}{2}, \frac{7}{4}, 2\}$. The Brillouin zone was sampled with a Γ -centered Monkhorst-Pack grid of $10 \times 10 \times 6$, $6 \times 12 \times 10$, and $15 \times 15 \times 3$ kpoints, for the *Pbca*, C2/c, and *P*-3*m*1 structures, respectively. The electronic wave functions were expanded in a plane-wave basis with a kinetic-energy cutoff of 325 eV. The atomic positions of each composition were optimized until the force on each atom was lower than $10^{-3} eV/$ and the total energy was accurate to $10^{-8} eV$. After optimization, the intra-layer bond-length and inter-layer distance was measured. The Gibbs free energy of each structure and composition was calculated using:

$$G = E_0 - TS_{vibr}$$

FIG. 1. Ball and stick models of (a) $PdSe_2$ and (b) $PdTe_2$ side view (left) and top view (right).

where E_0 is the ground state energy calculated from DFT and $E_{vibr} = TS_{vibr}$ is the vibrational Gibbs free energy associated with the phonon frequencies ω_i through

$$E_{vibr} = \sum_{i} \left\{ \frac{\hbar\omega_i}{2} + k_B T \ln\left[1 - exp\left(-\frac{\hbar\omega_i}{k_B T} \right) \right] \right\}$$

The formation energy was calculated as the difference between the Gibbs free energy of the compound and of the elemental metals, evaluated at T = 300K,

$$G_{form} = G_{PdSe_{2-x}Te_x} - G_{Pd} - (2-x)G_{Se} - xG_{Te} .$$

We obtain the phonon energies at Γ from densityfunctional perturbation (DFPT) calculations in VASP employing unit cells containing $2 \times 2 \times 2$ PdSe₂/Te₂ unit cells. All optical branches are assumed to be flat and independent of q-vector. The acoustic branches are assumed to have a linear dispersion with a sound velocity given by the first optical phonon energy divided by the smallest reciprocal lattice vector

Finally, the exfoliation energy of $PdSe_{2-x}Te_x$ was estimated as the difference between the ground state energies (E_0) of the bulk and monolayers forms per unit of surface area.

III. RESULTS

A. Structure evolution

Fig. 2(a) shows the X-ray diffraction patterns for $PdSe_{2-x}Te_x$ samples. At first glance, one can clearly observe three different types of structures with one of the corresponding characteristic peaks highlighted in the Fig. 2(a). In the Se rich region 0 < x < 0.3, the XRD peaks shifts slightly towards lower angle, as expected with increasing Te content, and the $PdSe_{2-x}Te_x$ maintains the puckered pentagonal PdSe₂ structure with orthorhombic lattice. A second phase starts to emerge at x = 0.3, although the major phase remains the PdSe₂ structure, this new phase becomes more dominant when $x \ge 0.5$ and coexists with the $PdSe_2$ -type phases until x =0.6. In the composition range of $0.7 \le x \le 0.8$, an XRD pure quality of the new phase was obtained. With further increasing of the Te content for $0.9 \le x \le 2.0$, the structure transformed to the P-3m1(1T) structure. Fig. 2(b) and Fig. 2(c) are the Rietveld refinement results for the PdSe_{1.9}Te_{0.1} and PdSe_{0.6}Te_{1.4} samples. The refined lattice parameters for $PdSe_{1,9}Te_{0,1}$ are a = 5.7687(2)Å b =5.8973(6)Å c=7.7160(8)Å, which are slightly bigger than those of $PdSe_2(a = 5.7410\text{\AA}, b = 5.8660\text{\AA}, c = 7.6910\text{\AA})$ as expected. The refined lattice parameters for the $PdSe_{0.6}Te_{1.4}$ sample are a = b = 3.9460(9)Å and c =5.0371(5)Å, slightly less than that of CdI₂-type PdTe₂ as Se atoms are smaller than Te atoms.

A small crystal from the PdSe_{1.2}Te_{0.8} sample is selected for the X-ray single crystal diffraction. The determined crystallographic parameters, refinement details, atomic coordinates, occupancies and equivalent anisotropic displacement parameters are included in Table 1 (for more crystal structure details, see suplemental materials [44]). The refined ratio of Se:Te =1.23(2):0.77(2), is rather close to the nominal composition and consistent with the chemical analysis results from SEM analysis. The crystal structure of the refined $PdSe_{1,23(2)}Te_{0,77(2)}$ is shown in Fig. 3 (a). The refined $PdSe_{1.23(2)}Te_{0.77(2)}$ crystallizes in a polymorphic structure in a monoclinic cell with space group C2/c (# 15), the chalcogen-chalcogen interlayer interactions now appear as covalent bonds, and a three-dimensional framework is visualized in Fig. 3(a). The chalcogenPdchalcogen stacking is maintained similar stacking order with a fundamental building motif $Pd(Se,Te)_6$ octahedra, just like in the 1T PdTe₂ phase. However, the Pd(Se,Te)₆ octahedra are much more distorted, and severely elongated along one direction within the plane, compared to the $PdTe_6$ octahedra in $PdTe_2$ as is shown in the Fig. 3(b).

As a result, the chalcogen-chalcogen distance between the adjacent stacking layers decreases down to 2.6338(5)



FIG. 2. (a) Powder X-ray diffraction patterns of $PdSe_{2-x}Te_x$ ($0 \le x \le 2$). (b) Rietveld refinement for $PdSe_{1.9}Te_{0.1}$ and (c) $PdSe_{0.6}Te_{1.4}$. The black circulars, red lines, pink bars, blue lines denote the observed diffraction intensities, calculated diffraction intensities, calculated locations of diffraction peaks and the difference between the calculated and observed diffraction intensities, respectively.

in the PdSe_{1.23(2)}Te_{0.77(2)} from the value of 3.463 in the 1T-PdTe₂ system, suggesting an effective covalent bonding interaction between the two adjacent layers. Another distinct difference of this Verbeekite polymorphic phase from PdSe₂ and PdTe₂ is on the Pd metal configurations. As shown in the Fig. 3(c), the Pd metals in the Verbeekite polymorphic phase PdSe_{1.23(2)}Te_{0.77(2)} have a rhombus packing with an obtuse angle measuring ~ 117° , which is completely different from the rectangular Pd packing (*i.e.*, 90°) in the PdSe₂ and trigonal antiprismatic Pd packing (*i.e.*, 120°) in the PdTe₂ structure. The Pd-Pd distance is ~ 3.99 Å, which is smaller than 4.10 Å in the PdSe₂ phase and 4.04 Å in the PdTe₂ phase.

It is worthwhile to mention that this new polymorphic phase $PdSe_{1.2}Te_{0.8}$ is isostructural with the rare mineral Verbeekite $PdSe_2$ phase. The mineral Verbeekite was discovered naturally in 2002, and synthesized in the lab via multianvil high-pressure high-temperature methods at 11.5 GPa and 1300 C in 2017. On the other hand, a different structural phase transition from orthorhombic to a cubic pyrite-type structure is observed in a direct diamond anvil cell studies of $PdSe_2$ crystals under high pressure above 6GPa without applying the high temperature. $PdSe_{1.2}Te_{0.8}$ has bigger unit cell parameters (a = 11.2450Å, b =4.1877Å, and c = 6.8110 Å) than the high pressure Verbeekite $PdSe_2$ phase (a = 8.914Å, b =4.1542

TABLE I. Crystallographic Data, Atomic Coordinates and Equivalent Isotropic Displacement Parameters of $PdSe_{1.2}Te_{0.8}$

Temperature	295 K				
Wavelength	0.71073				
Cell parameters	$a = 11.2450(9) \text{\AA} b = 4.1877(4) \text{\AA}$ $c = 6.8110(5) \text{\AA}, \ \beta = 124.326^{\circ}$				
	$V = 264.88(4) \text{Å}^3, Z = 4$				
Space group	C2/c (No. 15)				
Absorption coefficient	31.834 mm^{-1}				
F (000)	511.4				
θ range for data collection	4.389 - 32.921				
Reflections collected	2302				
Independent reflections	$487 [[R_{int}] = 0.0139]$				
Data/restraints/parameters	487/0/18				
Goodness-of-fit on ${\rm F}^2$	1.104				
Final R indices [I >2 $\sigma(I)$]	$R_1 = 0.0225, wR_2 = 0.0623$				
R indices (all data)	$R_1 = 0.0228, wR_2 = 0.0624$				
Largest diff. peak and hole	3.854 and -1.249 $\mathrm{e}^{\text{-3}}$				
Atom Wyck, x	v z Occ. $U_{\alpha\alpha}^{\alpha}(A^2)$				

1100111	or you.		3	2	000.	eq (11)
Pd1	4c	3/4	1/4	1/2	1	0.0120(2)
Se1	8f	0.60958(3)	0.31783(9)	0.05186(5)	0.615(8)	0.0133(2)
Te1	8f	0.60958(3)	0.31783(9)	0.05186(5)	0.385(8)	0.0133(2)

Å, and c = 6.710 Å) as expected, since atomic size of Te elements is bigger than that of Se atoms. However, it is rather puzzling why larger cell of $PdSe_{1.2}Te_{0.8}$ could crystallize in this high pressure $PdSe_2$ analog Verbeekite-type structure. This unique Se-Se dimer packing structure in Verbeekite-type $PdSe_{1.2}Te_{0.8}$ phase could potentially offer another promising platform to investigate interesting structural and electrical behaviors near the critical point of two- and three-dimensionality for TMD materials.

Fig. 4 shows the formation energy for $PdSe_{2-x}Te_x$, in the *Pbca* phase, the C2/c phase and the *P*-3*m*1 phase, calculated using DFT. The theoretical calculations show that the *Pbca* phase is the most stable phase for x=0 (PdSe₂), for x=0.3 up to x=0.8, the C2/c phase appears the most stable, and for x >0.8, the *P*-3*m*1 phase has the lowest energy. These results are in excellent agreement with the experimentally grown material, characterized by the X-ray diffraction results shown in Fig. 2. Overall, the agreement is remarkable although a slight discrepancy between theory and experiment is observed in the x = 0.3 to x = 0.6 region where experimentally both the *Pbca* and the C2/c co-exist while our theoretical calculations clearly favor the C2/c phases. We note that the phonon



FIG. 3. (a) Ball and stick models for $PdSe_{1.2}Te_{0.8}$ with space group C2/c along different directions. (b) Comparison of octahedra in $PdSe_{1.2}Te_{0.8}$ and $PdTe_2$ structures. (c) Metal configurations in Pd planes of $PdSe_2$, $PdSe_{1.2}Te_{0.8}$ and $PdTe_2$.

contribution is a significant component to the Gibbs free energy and our theoretical results reveal that the entropy associated with the phonons is an important component in determining which phase is most stable.

B. Electronic properties

Fig. 5 shows the changes of electrical transport properties in the $PdSe_{2-x}Te_x$ system upon Te doping. As one can see from Fig. 5(a), a clear crossover in electrical transport accompanied by the semiconductor-tometal transition is observed with increasing Te content. Fig. 5(b) shows the plot of ln $(\rho/\rho_{300\rm K})$ vs (1/T) for a PdSe_{1.9}Te_{0.1} sample, where thermally activated conduction behavior is observed. Two clear linear relations of $\ln (\rho / \rho_{300 \text{K}})$ vs. (1/T) are found in the range of 140-300 K and 30-69 K, indicating an energy gap of 92 meV and 13 meV, respectively. Fig. 5(c) shows the temperature dependent electrical resistivity ρ (T) for the Verbeekite polymorphic phase of PdSe_{1.3}Te_{0.7}. The resistivity value is weakly dependent on the temperature and shows an overall semiconducting behavior from room temperature down to 50 K. An anomalous kink arises at the temperature ~ 50 K and a clear slope change of resistivity is observed afterwards. The resistivity increases more gently and has a tendency of saturation below 50 K, which may be related with the weak localizations of electrons caused by the disorders in this system.

The semiconducting behavior of the Verbeekite phase is also verified with theoretical DFT calculations. Fig. 6 shows the calculated band structure and density-of-states (DOS) of PdSe_{1.25}Te_{0.75}. The DFT results show a small (15 meV) indirect bandgap for the C2/c phase in agreement with the experimental transport measurements. The conduction band minimum appears at the S point while the valence band maximum appears along the Γ -Y line. We note that we chose a Hubbard U of 4 eV for all our calculations. To verify if this is a physically realistic choice, we determined the Hubbard U using linear response theory and found values of 3.1 eV, 2.9 eV and 2.7



FIG. 4. (Formation energy (Gibbs free energy) of *Pbca*, C2/c and *P-3m1* phases of PdSe_{2-x}Te_x at 300K, calculated using DFT.

eV for P3m1, C2c, and the *Pbca* phases of PdSe₂. This indicates that our Hubbard U may be slightly overestimated although the linear-response is known to underestimate the Hubbard U. Finally, the difference in Hubbard U between the three phases indicates that follow-up theoretical work should consider the impact of different Uvalues for different phases and compositions.

The inset of Fig. 5(c) highlights the plot of $\ln \rho$ vs (1/T) below 50 K. The curves become almost Tindependent at low temperature between 2 and 30K, indicating the existence of a hopping process in the materials. Several general hopping models could contribute to the electrical transport in PdSe_{2-x}Te_x. One is the nearest-neighbor hopping model (NNH) where ρ is proportional to $\exp(E_A/k_BT)$. Here E_A is the activation en-



FIG. 5. (a) Temperature dependence of normalized resistivity $(\rho / \rho_{300\text{K}})$ in log scale for PdSe_{2-x}Te_x solid solution with 0.1 $\leq x \leq 0.9$. (b) Ln $(\rho / \rho_{300\text{K}})$ vs (1/T) for PdSe_{1.9}Te_{0.1}. The blue dashed lines represent the fitting using the standard activation model. (c) Electrical resistivity ρ vs. temperature T for PdSe_{1.3}Te_{0.7}. The inset shows Ln (ρ) vs. (1/T) of PdSe_{1.3}Te_{0.7} in the temperature range 2 - 50 K. The solid red line is fitted to Eq.1. (d) Hall resistivity of PdSe_{1.3}Te_{0.7} at 10K, 30K, 60 K, 100 K,150 K and 200K. The inset shows the evolution of the Hall coefficient R_H with temperature. (e) Temperature-dependent normalized resistance (R/R_{300K}) of PdSe_{2-x}Te_x solid solution for $0.9 \leq x \leq 1.5$. (f) Enlarged view of R/R_{300K} in the low temperature region 1.8 - 4.5 K.

ergy and $k_{\rm B}$ is the Boltzmann constant[45]. Another model describes variable range hopping (VRH) where $\rho \propto \exp[T_0/T]^{1/(1+d)}$ and often occurs between states with larger spatial distance but closer energy[46]. Here T_0 is a characteristic temperature and d is the dimensionality of the solid-state materials. VRH has been found in many disordered systems [47–49].

In our system, one can clearly observe that the slope decreases continuously with decreasing temperature, which suggests the existence of multiple conducting channels. Therefore, we utilized one NNH model together with one VRH model to fit our low temperature resistivity data below 50 K using the following equation:

$$R(T)^{-1} = R_0^{-1} + R_1^{-1} \exp\left[-E_1/2k_BT\right] + R_2^{-1} \exp\left[-(T_2/T)^{\frac{1}{3}}\right]$$

Where E_1 is the activation energy, T_2 is the characteristic temperature, R_0 , R_1 and R_2 are the related resistance coefficients. Here the temperature independent R_0^{-1} represent the scattering of electrons by disorders in low temperature and is extracted to be 0.0647. The extracted activation energy E_1 is about 2 meV. The low activation energy E_1 reveals inhomogeneous localizationdistributions in this system and the nearly free carriers are weakly localized by the disorder. When the temperature decreases below 30 K, the VRH dominates and the characteristic temperature T_2 is extracted to be ~ $4.8 \times$ 10^4 K. Here the characteristic temperature has a positive relation with the optimal hopping distance of the localized electrons and the yielding value is consistent with many other TMD materials [49]. The coefficient R_1^{-1} and R_2^{-1} are extracted to be 0.004 and 19, respectively. We have also tried to fit out data with 1D, 2D and 3D VRH models and all of them fail to converge. When combining one NNH with a VRH model of 1D and 3D no convergence on the fit is observed either. The fact that only the 2D VRH combined with the NNH mode fits the data well, suggests that the structure of this new Verbeekite polymorphic phase is closer to quasi-2D rather than 3D.



FIG. 6. Band structure and density-of-states states with the most important orbital contributions indicated of $PdSe_{1.25}Te_{0.75}$ determined from first principles.

Fig. 5(d) shows the temperature dependence of the Hall resistivity ρ_{xy} for PdSe_{1.3}Te_{0.7}, exhibiting the new C2/c phase, at 10, 30, 60, 100, 150 and 200 K. Here ρ_{xy} shows linearly dependent behaviors with the magnetic field. The slopes are all negative, indicating that the charge carriers are dominantly electrons near the Fermi surface. The Hall coefficient $R_H = \rho_{xy}/\mu_0 H$ estimated by the linearly fitting of ρ_{xy} , is shown in the inset of Fig. 4(d). $R_{\rm H}$ scales almost monotonically with temperature and there are no significant changes of $R_{\rm H}$ around 50 K. The charge-carrier density from single band approximation is estimated to be 5.12×10^{19} cm⁻³ at 10 K, which is about 2 orders smaller than the 5.5×10^{21} cm⁻³ extracted for PdTe₂ in Ref.[50].

Finally, Fig. 5(e) and 5(f) show the temperature dependence of the in-plane resistivity ratio ρ/ρ_{300K} of PdSe_{2-x}Te_x samples with x in the range $0.9 \leq x \leq 1.5$. All of these samples show metallic behavior with residual resistivity ratios (*RRR*) around 2, which are smaller than 75 of the PdTe₂ single crystal[50], reflecting the substantial disorder induced by Te substitution. Remarkably, in the composition range $1.0 \leq x \leq 1.2$, superconductivity suddenly occurs with an onset T_c close to 2.7 K. For compositions beyond x >1.2, no superconductivity is detected above 1.8 K (the limit of our instrument), but the samples could possibly still be superconducting below 1.8 K, as 1T-PdTe₂ is superconducting at 1.65 K.

C. Phase diagram

Fig. 7 summarizes the rich phase diagram for the $PdSe_{2-x}Te_x$ system where both crystal symmetry and lattice parameters are listed. With increasing Te content



FIG. 7. Phase diagram of $PdSe_{2-x}Te_x$ versus Te content x. Sc denotes superconducting phase. Open squares, circulars and triangles show the lattice parameters a, b and c respectively. Solid purple balls show the onset superconducting transition temperature determined from transport measurements.

x, the structure of $PdSe_{2-x}Te_x$ retains an orthorhombic phase up to x = 0.3. The new Verbeekite-type polymorphic phase starts to emerge when x \geq 0.3 and remains stable up to x = 0.8. With further doping beyond x = 0.9, the monoclinic phase becomes unstable and the samples transform to a *P*-3*m*1 (1T) phase. Within a narrow region of *P*-3*m*1(1T) phase when $1.0 \leq x \leq 1.2$, the samples show superconductivity with onset T_c close to 2.7 K, as reported by us previously[37].

IV. CONCLUSION

In conclusion, we have carried out a systematic study of isoelectronic substitution of Te for Se in PdSe_{2-x}Te_x solid solutions. A structural evolution with increasing Te fraction from the *Pbca* PdSe₂ phase to the *P*-3*m*1 PdTe₂ phase was observed accompanied with clearly distinct electrical transport behavior. A new Verbeekitetype polymorphic C2/c structure was discovered within a narrow range of Te composition (x = 0.3 to x = 0.8), and the C2/c structure has a distinct packing structure which is different from all known TMDs to date. Theoretical first principles calculations agree with the experimental findings and revealed that the phonon-contribution to the free energy is an important factor in making the *Pbca* and C2/c favorable phases. The C2/c phase showed 7

electron-dominated charge carriers and displayed an unusual electrical conductivity behavior which could be well explained through the combined nearest-range-hopping and variable-range-hopping model. In the P-3m1 (1T) phase region, an enhanced superconductivity emerges in a narrow range with onset T_c about 2.7 K. The rich phase diagram and very rare polymorphic structure discovered in this system could provide a new material platform to further investigate transport and electronic properties of different types of polymorphs in the TMD structure that of significant interest.

ACKNOWLEDGMENTS

This work at University of Texas at Dallas is supported by US Air Force Office of Scientific Research Grant No. FA9550-19-1-0037 and National Science Foundation (DMR-1921581). We also acknowledge the support from Office of Research at University of Texas at Dallas through Seed Program for Interdisciplinary Research (SPIRe) and the Core Facility Voucher Program. The project or effort depicted is sponsored by the Department of Defense, Defense Threat Reduction Agency. The content of the information does not necessarily reflect the position or the policy of the federal government, and no official endorsement should be inferred.

- G. Fiori, F. Bonaccorso, G. Iannaccone, T. Palacios, D. Neumaier, A. Seabaugh, S. K. Banerjee, and L. Colombo, Nat. Nanotechnol. 9, 768 (2014).
- [2] K. S. Novoselov, A. Mishchenko, A. Carvalho, and A. H. Castro Neto, Science 353, aac9439 (2016).
- [3] Q. Yun, L. Li, Z. Hu, Q. Lu, B. Chen, and H. Zhang, Adv. Mater. 32, 1903826 (2020).
- [4] Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman, and M. S. Strano, Nat. Nanotechnol. 7, 699 (2012).
- [5] W. Choi, N. Choudhary, G. H. Han, J. Park, D. Akinwande, and Y. H. Lee, Mater. Today 20, 116 (2017).
- [6] Y. Zhang, Y. Iwasa, J. Miyazaki, W. Shi, N. Inoue, J. Ye, M. Yoshida, R. Suzuki, and Y. Saito, Sci. Rep. 5, 12534 (2015).
- [7] A. Ciarrocchi, A. Avsar, D. Ovchinnikov, and A. Kis, Nat. Commun. 9, 919 (2018).
- [8] J. Li, X. Yang, Y. Liu, B. Huang, R. Wu, Z. Zhang, B. Zhao, H. Ma, W. Dang, Z. Wei, K. Wang, Z. Lin, X. Yan, M. Sun, B. Li, X. Pan, J. Luo, G. Zhang, Y. Liu, Y. Huang, X. Duan, and X. Duan, Nature **579**, 368 (2020).
- [9] S. Wu, V. Fatemi, Q. D. Gibson, K. Watanabe, T. Taniguchi, R. J. Cava, and P. Jarillo-Herrero, Science 359, 76 (2018).
- [10] M. Chhowalla, H. S. Shin, G. Eda, L. J. Li, K. P. Loh, and H. Zhang, Nat. Chem. 5, 263 (2013).
- [11] C. Jin, E. C. Regan, A. Yan, M. Iqbal Bakti Utama, D. Wang, S. Zhao, Y. Qin, S. Yang, Z. Zheng, S. Shi, K. Watanabe, T. Taniguchi, S. Tongay, A. Zettl, and F. Wang, Nature 567, 76 (2019).

- [12] S. Ulstrup, R. J. Koch, S. Singh, K. M. McCreary, B. T. Jonker, J. T. Robinson, C. Jozwiak, E. Rotenberg, A. Bostwick, J. Katoch, and J. A. Miwa, Sci. Adv. 6, eaay6104 (2020).
- [13] L. Wang, E. M. Shih, A. Ghiotto, L. Xian, D. A. Rhodes, C. Tan, M. Claassen, D. M. Kennes, Y. Bai, B. Kim, K. Watanabe, T. Taniguchi, X. Zhu, J. Hone, A. Rubio, A. N. Pasupathy, and C. R. Dean, Nat. Mater. **19**, 861 (2020).
- [14] Q. Li, C. He, Y. Wang, E. Liu, M. Wang, Y. Wang, J. Zeng, Z. Ma, T. Cao, C. Yi, N. Wang, K. Watanabe, T. Taniguchi, L. Shao, Y. Shi, X. Chen, S. J. Liang, Q. H. Wang, and F. Miao, Nano Lett. 18, 7962 (2018).
- [15] H. X. Zhong, S. Gao, J. J. Shi, and L. Yang, Phys. Rev. B - Condens. Matter Mater. Phys. 92, 115438 (2015).
- [16] B. E. Brown, Acta Crystallogr. 20, 264 (1966).
- [17] Y. Deng, Y. Lai, X. Zhao, X. Wang, C. Zhu, K. Huang, C. Zhu, J. Zhou, Q. Zeng, R. Duan, Q. Fu, L. Kang, Y. Liu, S. J. Pennycook, X. R. Wang, and Z. Liu, J. Am. Chem. Soc. **142**, 2948 (2020).
- [18] E. Bjerkelund and A. Kjkshus, Acta.Chem.Scand., 21, 513 (1965).
- [19] B. Yan and C. Felser, Annu. Rev. Condens. Matter Phys. 8, 337 (2017).
- [20] T. A. Empante, Y. Zhou, V. Klee, A. E. Nguyen, I. H. Lu, M. D. Valentin, S. A. Naghibi Alvillar, E. Preciado, A. J. Berges, C. S. Merida, M. Gomez, S. Bobek, M. Isarraraz, E. J. Reed, and L. Bartels, ACS Nano 11, 900 (2017).

- [21] R. Sankar, G. Narsinga Rao, I. P. Muthuselvam, C. Butler, N. Kumar, G. Senthil Murugan, C. Shekhar, T. R. Chang, C. Y. Wen, C. W. Chen, W. L. Lee, M. T. Lin, H. T. Jeng, C. Felser, and F. C. Chou, Chem. Mater. 29, 699 (2017).
- [22] J. Jiang, Z. K. Liu, Y. Sun, H. F. Yang, C. R. Rajamathi, Y. P. Qi, L. X. Yang, C. Chen, H. Peng, C. C. Hwang, S. Z. Sun, S. K. Mo, I. Vobornik, J. Fujii, S. S. P. Parkin, C. Felser, B. H. Yan, and Y. L. Chen, Nat. Commun. 8, 13973 (2017).
- [23] H. Luo, W. Xie, J. Tao, H. Inouec, A. Gyenis, J. W. Krizan, A. Yazdani, Y. Zhu, and R. J. Cava, Proc. Natl. Acad. Sci. U. S. A. **112**, E1174 (2015).
- [24] F. Gronvold and E. Rost, Acta Chem. Scand. 10, 1620 (1956).
- [25] F. Hulliger, J. Phys. Chem. Solids, 26, 639 (1965)
- [26] R. Kempt, A. Kuc, and T. Heines, Angew. Chemie Int. Ed. 59, 9242 (2020).
- [27] A. D. Oyedele, S. Yang, L. Liang, A. A. Puretzky, K. Wang, J. Zhang, P. Yu, P. R. Pudasaini, A. W. Ghosh, Z. Liu, C. M. Rouleau, B. G. Sumpter, M. F. Chisholm, W. Zhou, P. D. Rack, D. B. Geohegan, and K. Xiao, J. Am. Chem. Soc. **139**, 14090 (2017).
- [28] Y. Yu, F. Yang, X. F. Lu, Y. J. Yan, Y. H. Cho, L. Ma, X. Niu, S. Kim, Y. W. Son, D. Feng, S. Li, S. W. Cheong, X. H. Chen, and Y. Zhang, Nat. Nanotechnol. **10**, 270 (2015).
- [29] Y. Ma, L. Kou, X. Li, Y. Dai, and T. Heine, NPG Asia Mater. 8, e264 (2016).
- [30] Q. Liang, Q.Wang, Q. Zhang, J. Wei, S.X. Lim, R. Zhu, J. Hu, W. Wei, C. Lee, C. Sow, and W. Zhang, Advanced Materials, **31**, 1807609, (2019)
- [31] W. L. Chow, P. Yu, F. Liu, J. Hong, X. Wang, Q. Zeng, C. H. Hsu, C. Zhu, J. Zhou, X. Wang, J. Xia, J. Yan, Y. Chen, D. Wu, T. Yu, Z. Shen, H. Lin, C. Jin, B. K. Tay, and Z. Liu, Adv. Mater. **29**, 1602969 (2017).
- [32] G. Zhang, M. Amani, A. Chaturvedi, C. Tan, J. Bullock, X. Song, H. Kim, D. H. Lien, M. C. Scott, H. Zhang, and A. Javey, Appl. Phys. Lett. **114**, 253102 (2019).

- [33] M. A. Elghazali, P. G. Naumov, H. Mirhosseini, V. S, L. Mchler, W. Schnelle, C. Felser, and S. A. Medvedev, Phys. Rev. B 96, 060509(R) (2017).
- [34] W. Zheng, R. Schnemann, N. Aryal, Q. Zhou, D. Rhodes, Y. C. Chiu, K. W. Chen, E. Kampert, T. Frster, T. J. Martin, G. T. McCandless, J. Y. Chan, E. Manousakis, and L. Balicas, Phys. Rev. B 97, 235154 (2018).
- [35] F. Fei, X. Bo, R. Wang, B. Wu, J. Jiang, D. Fu, M. Gao, H. Zheng, Y. Chen, X. Wang, H. Bu, F. Song, X. Wan, B. Wang, and G. Wang, Phys. Rev. B 96, 041201(R) (2017).
- [36] H. Leng, C. Paulsen, Y. K. Huang, and A. De Visser, Phys. Rev. B 96,220506(R) (2017).
- [37] W. Liu, S. Li, H. Wu, N. Dhale, P. Koirala, and B. Lv, Phys. Rev. Mater. 5, 014802 (2021).
- [38] A. C. Roberts, W. H. Paar, M. A. Cooper, D. Topa, A. J. Criddle, and J. J. Verbeekite, Mineral. Mag. 66, 173 (2002).
- [39] E. Selb, M. Tribus, and G. Heymann, Inorg. Chem. 56, 5885 (2017).
- [40] B. H. Toby and R. B. Von Dreele, J. Appl. Crystallogr. 46, 544 (2013).
- [41] G. Kresse and J. Furthmller, Phys. Rev. B 54, 11169 (1996).
- [42] M. Dion, H. Rydberg, E. Schrder, D. C. Langreth, and B. I. Lundqvist, Phys. Rev. Lett. **92**, 246401 (2004).
- [43] J. Klime, D. R. Bowler, and A. Michaelides, J. Phys. Condens. Matter 22, 022201 (2010).
- [44] See Supplemental Material at for crystal structure details of the Verbbeekite $PdSe_{1.2}Te_{0.8}$.
- [45] V. F. Gantmakher, *Electrons and Disorder in Solids* (Oxford University Press, 2005).
- [46] N. F. Mott and E. A. Davis, *Electronic Processes in Non-Crystalline Materials* (Oxford university press, 2012).
- [47] Z. G. Yu and X. Song, Phys. Rev. Lett. 86, 6018 (2001).
- [48] M. Y. Han, J. C. Brant, and P. Kim, Phys. Rev. Lett. 104, 056801 (2010).
- [49] H. Qiu, T. Xu, Z. Wang, W. Ren, H. Nan, Z. Ni, Q. Chen, S. Yuan, F. Miao, F. Song, G. Long, Y. Shi, L. Sun, J. Wang, and X. Wang, Nat. Commun. 4, 3 (2013).
- [50] Amit and Y. Singh, Phys. Rev. B 97, 054515 (2018).