



This is the accepted manuscript made available via CHORUS. The article has been published as:

Giant negative thermal expansion at the nanoscale in the multifunctional material Gd {5}(Si,Ge) {4}

João H. Belo, Ana L. Pires, Isabel T. Gomes, V. Andrade, João B. Sousa, Ravi L. Hadimani, David C. Jiles, Yang Ren, Xiaoyi Zhang, João P. Araújo, and André M. Pereira Phys. Rev. B **100**, 134303 — Published 7 October 2019

DOI: 10.1103/PhysRevB.100.134303

Giant Negative Thermal Expansion at Nanoscale in Gd₅(Si,Ge)₄ Multifunctional Material

João H. Belo $^{\psi}$, Ana L. Pires $^{\psi}$, Isabel T. Gomes $^{\psi}$, V. Andrade $^{\psi}$, João B. Sousa $^{\psi}$, Ravi L. Hadimani $^{\#}$, David, C. Jiles $^{\square}$, Yang Ren $^{\square}$, Xiaoyi Zhang $^{\square}$, João P. Araújo $^{\psi}$ * and André M. Pereira $^{\psi}$ *

^ψ IFIMUP and IN-Institute of Nanoscience and Nanotechnology, Departamento de Física e Astronomia da Faculdade de Ciências da Universidade do Porto, Rua do Campo Alegre, 687, 4169-007 Porto, Portugal.

Department of Mechanical and Nuclear Engineering, Virginia Commonwealth University, Richmond, Virginia 23284, United States of America.

□ Department of Electrical and Computer Engineering, Iowa State University, Ames, IA 50011, USA. And Ames Laboratory, US Department of Energy, Iowa State University, Ames, IA 50011, United States of America.

□ X-ray Science Division, Argonne National Laboratory, Argonne, Illinois 60439, United States of America.

*E-mail: jearaujo@fc.up.pt and ampereira@fc.up.pt

PHYSH KEYWORDS: Condensed Matter & Materials Physics, Magnetocaloric effect,

Magnetic phase transitions, Thermal properties, Thermal expansion, Nanocrystals, Electrical
conductivity.

ABSTRACT

In this work the thermal expansion of $Gd_5Si_{1,3}Ge_{2,7}$ magnetic nanogranules was studied. A cross-over from positive (macro and micro scale in the literature) towards a negative thermal expansion (at nanoscale) is observed. The NTE behavior was found in two temperature windows: 90-160K ($\beta^{LT}\sim$ -32.2 ppm K⁻¹) and within the room temperature 255-340K, where a giant NTE was observed (-69 ppm K⁻¹). The key atomic mechanism driving the NTE is identified as an atomic triplet chain flexibility. The results suggest that the NTE behavior emerges as a size reduction effect caused by the intrinsic nanoparticles surface pressure (estimated at 11kBar), leading to the PTE-NTE cross-over at the nanoscale.

I. INTRODUCTION

A majority of materials are known to expand when heated up and contract when cooled down, the so-called - positive thermal expansion (PTE). However, in the last 30 years, the number of materials exhibiting the opposite behavior, negative thermal expansion (NTE), has increased significantly. The paramount example is ZrW_2O_8 [1–3], where the observed NTE is associated with low energy vibration modes which have a negative Gruneisen parameter. NTE has also been observed in magnetic materials with strong magnetovolume coupling below their magnetic ordering temperature [4]. In parallel, there has been an increasing effort to study nanomaterials thermal expansion, as remarkably illustrated by the emergence of NTE in Au [5] and CuO [4] nanoparticles, in opposition with their bulk counterparts PTE, arising from the effects of nanoconfinement and band structure modifications.

The $R_5Si_xGe_{4-x}$ family (R stands for Rare-earth) has been intensively studied since 1997, when the giant magnetocaloric effect (GMCE) was found in the $Gd_5Si_2Ge_2$ compound [6]. Besides GMCES, that growing research effort lead to the discovery of giant magnetoresistance, spontaneous generation voltage and colossal magnetostriction [7,8]. Such variety of effects arise from the coupling between the magnetic and structural phases, ultimately leading to magnetostructural transitions at a transition temperature (T_{MS}). These transitions can be driven by multi-stimuli such as temperature, magnetic field, pressure or combinations of these [7,8]. At the macroscale, these materials exhibit positive thermal expansion (PTE), either in the ferromagnetic ($T < T_{MS}$) and the paramagnetic states ($T > T^{MS}$) [9–11]. Unfortunately, there is scarce information available on the size-reduction effects on these materials [12–15] and even scarcer concerning the temperature dependence of their atomic structure at nanoscale. The film here presented was grown from a $Gd_5Si_{1,3}Ge_{2,7}$ target ablated with a high energy femtosecond

pulsed laser, resulting in an ensemble of nanogranules whose diameters follow a Lorentzian distribution with 80nm average value, as detailed in previous work [14]. In this work and the ones following it, the main focus was towards the understanding of nanostructuring effects on the magnetocaloric effect, critical for magnetic refrigeration applications [13-15]. At room temperature, the atomic structure of the nanogranules is composed of a mixture of two concurrent Orthorhombic phases: O(I) and O(II), with 35% and 65% fractions, respectively. Below ~ 150K only the O(I) phase exists, whereas at the magnetostructural transition temperature ($T_{MS} \sim 190K$), it converts incompletely (65%) into O(II), while 35% remain in the O(I) phase with ferromagnetic order up to $T_C \sim 250$ K. The O(I) has a smaller unit cell volume (1.2% smaller) than O(II) and their unit cells can be decomposed in two rigid slabs that lie in the ac plane and are stacked by interslab Ge3-Ge3 bonding (or its absence in O(II)) along the b axis, seen in Figure below and reported in several works [16, 32, 34]. as

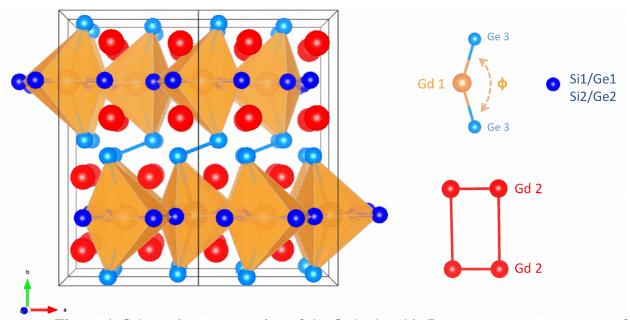


Figure 1. Schematic representation of the Orthorhombic Pnma space group structure of Gd₅Si_{1.3}Ge_{2.7}. The octahedra constituent Ge3-Gd1-Ge3 triplet system (Ge3 represented as small

light blue spheres and Gd1 as larger orange spheres), the slabs constituent Gd2 atoms (as red spheres) and the Si1,2/Ge1,2 (as dark blue spheres) are highlighted.

Each rigid slab consists of two pseudo body-centered cubes with Gd atoms at their vertices and center, having two Si/Ge as next neighbors. Alternatively, these structures can also be viewed as a stacking of corner-linked octahedrons with Gd1 at the center, Si/Ge1 and Si/Ge2 in the four vertices in the horizontal ac plane and the Ge3 in the two vertices along the b direction, (Figure 1). The Ge3 form an important triplet chain with the Gd1: Ge3-Gd1-Ge3, whose angle is ϕ , (Figure 1 top-right corner).

Herein, these nanomaterials thermal expansion behavior will be presented in detail. In particular, the finding that nanostructuring implied a cross-over from the typical PTE observed at the macroscale towards NTE will be detailed together with the main atomic mechanism driving such drastic behavior change. Furthermore, the combination of these unpredicted features has potential to be applied in multifunctional devices on the field of thermomechanical systems/sensors.

II. EXPERIMENTAL DETAILS

A multigrain sample of Gd₅Si_{1.3}Ge_{2.7} prepared from high purity materials by Tri-arc method was used as a target material in a pulsed laser deposition (PLD) system. A femtosecond laser with the pulse energy of 3.5mJ and a repetition rate of 1000 Hz was used for the deposition. The granular film was deposited on a 1 µm SiO2 layer on the top of a (001) silicon substrate at 200° C and 1.2×10^{-6} Torr. The rate of deposition was ~ 0.65 nm/sec. The use of a femtosecond laser with higher repetition rate caused Coulomb ablation due to multi-photon absorption or Coulomb explosion resulting in an ensemble of nanogranules obeying to a Lorentzian distribution with 80nm average. High-resolution Transmission Electron Microscopy (HR-TEM200-SE), JEOL, model 2200FS, STEM mode integrated was used to observe isolated nanogranules, extracted from the ensemble by ultra-sonic stimulation immersed on ethanol. Synchrotron X-ray diffraction measurements of the thin film sample were carried out at the beamline 11-ID-D, Advanced Photon Source, Argonne National Laboratory. The wavelength of the x-rays was 0.765335 Å, and an Oxford cryo-stream system was used for the temperature dependent measurements. Two-dimensional diffraction patterns were collected using a Pilatus area detector, and converted to the standard one-dimensional diffraction curves using the Fit2D program. A very high-resolution monochromator for the incident X-ray beams with an energy resolution of $\Delta E/E = 10^{-4}$ was used. The beam size was 10 μ m x 50 μ m and the Pilatus 2M detector pixel size is 172 µm x 172 µm. NIST standard CeO2 sample was used for calibration, yielding a XRD angular resolution of $\Delta Q/Q = 1.0\%$. Synchrotron X-ray diffraction measurements of the bulk sample were carried out at ESRF, Grenoble, beamline ID31 at 0.117 Å wavelength. Similarly to the thin film diffraction setup, the two-dimensional diffraction patterns of the bulk sample were collected using a Pilatus area detector, and converted to the conventional one-dimensional

diffraction curves. The electrical resistivity was measured with the standard four-point potentiometric method using a direct current of 2 mA, stable to 1:106. The four electrical contacts were established by Gold sputtering four points evenly spaced along a straight line and Silver paint to bond them together with Copper wires on top of a 3 x 6 mm piece cut from the larger deposited thin film. The voltage was measured with a Keithley 182 nanovoltmeter with a resolution of 10 nV during the measurements. The thin film was glued with a thin layer of GE varnish to a massive Copper block support in order to minimize any possible temperature gradient, and its electrical resistance (R) was measured in the [90 - 300]K temperature range with a 0.5 K min⁻¹ rate.

III. RESULTS

High-resolution synchrotron measurements were performed in a wide range of temperatures [90, 340]K. Two temperature regions, below (a)) and above (b)) T_{MS}, were selected since the magnetostructural transition temperature window was already thoroughly discussed in references [13–15]. Figures 2 a) and b) display the most intense X-ray reflections, associated with the (1 3 2), (2 3 1), (2 0 2) and (0 4 2)/(2 1 2) Miller indices of O(I) and O(II) phases, suggesting the polycrystalline nature of the nanogranules. In accordance with the obtained diffractograms, the high-resolution transmission electron microscopy (HRTEM) images corroborate the polycrystalline nature of the nanogranules, even for the smaller nanoparticles.

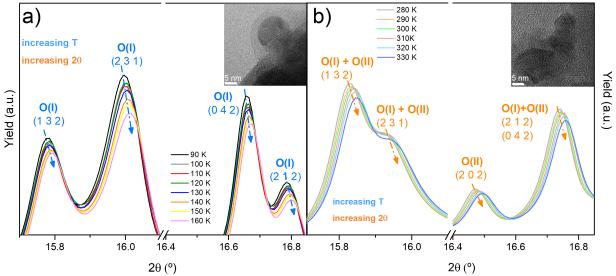


Figure 2. Synchrotron X-ray diffraction data for Gd₅Si_{1.3}Ge_{2.7} nanogranules extracted for fixed temperatures with a 10K step in the 90-160K (LT) (a)) and 280-330K (HT) (b)) temperature intervals, respectively. In the insets, high-resolution transmission microscopy images of isolated nanogranules are displayed, highlighting their crystalline nature.

The crystallite size was estimated in 95nm by Williamson-Hall analysis (Figure S14 in [22], whereas in Figure S15 the angular resolution is presented [22]), in accordance with the average nanoparticle size obtained by SEM surface image analysis.

Inspecting the diffraction patterns in more detail, it is observed an up-shift of all diffracted peaks upon heating, - a clear fingerprint of NTE - in both temperature regions: low temperature NTE (LT-NTE) and high temperature NTE (HT-NTE). Each Synchrotron X-ray diffractogram was fitted by Rietveld refinement with FullProf software in order to monitor the atomic structure temperature dependence [17]. The relative unit cell volume, V, and lattice parameters, a, b and c are plotted in Figure 3 a) and b) as a function of temperature.

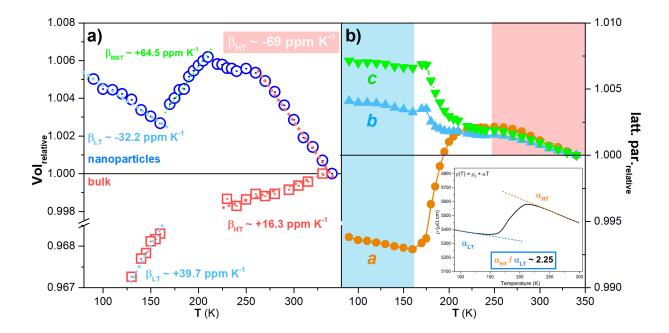


Figure 3. a) Temperature dependence of the $Gd_5Si_{1.3}Ge_{2.7}$ nanogranules (blue open circles) and bulk relative unit cell volume (red open diamonds). The volume values were normalized to V(T=340K, T=330K, for nano and bulk, respectively) and the linear fits in the 90-150K, 175-

210K and 255-340K temperature ranges are presented. b) $Gd_5Si_{1,3}Ge_{2,7}$ nanogranules relative a, b and c lattice parameters values as a function of temperature. The represented nanogranules volume values in the T > 250K region correspond to the averaged values of the two phases O(I) and O(II) volumes weighted to their corresponding phase fractions. The resistivity of the ensemble of nanogranules as a function of temperature in the 90-300K temperature range is displayed in Figure b) inset, where the dashed blue and orange lines are the linear fits of the $\varrho(T)$ curve at the LT-NTE and HT-NTE regions, respectively.

The nanogranules volume V contracts in the whole temperature range, except in the temperature interval where the magnetostructural transition occurs. As mentioned above, the nanogranules NTE behavior splits into two major temperature ranges: 90-160 K (LT-NTE) and the 255-340 K (HT-NTE)) as highlighted in blue and red, respectively in Figure 3 b). In the LT-NTE regime, all three lattice parameters contract linearly as T increases, with $\alpha_a^{LT} = \Delta a^{LT}/\Delta T \sim -$ 13.0 ppm K⁻¹ > $\alpha_b^{LT} = \Delta b^{LT}/\Delta T > \alpha c^{LT} = \Delta c^{LT}/\Delta T$, summing up to a large and negative volume thermal expansion, $\beta_{LT} = \Delta V^{LT}/\Delta T \sim -32.2$ ppm K⁻¹. The inherent contraction is interrupted by the previously reported magneto-structural transition (for details see [14]), across which the unit cell greatly expands with $\beta^{MST} = \Delta V^{MST}/\Delta T \sim +64.5$ ppm K⁻¹, up to T= 210K. Then the contraction resumes, exhibiting a transient regime ($\beta^{TR} \sim 23.8$ ppm K⁻¹) up to 240K. The HT-NTE regime begins at 255K, enhancing the volume contraction up to a giant $\beta^{HT} \sim -69$ ppm K⁻¹ NTE value along a wide temperature interval (~ 90K) around room temperature, 255-340K - the highest temperature measured. The a-contraction increases again, surpassing its LT-NTE value, as occurs for the b and c values: $\alpha_a^{HT} \sim 23.2 \text{ ppm K}^{-1} > \alpha_c^{HT} > \alpha_b^{HT}$. The contraction force is clearly enhanced in this regime. One notes that the observed β^{HT} is much larger than the one found in

Graphene (-7 ppm K⁻¹) [18,19] and is more than the double of that observed in ZrW2O8 (-28.2 ppm K-1) [20]. Although it is smaller than the one exhibited by CuO nanoparticles (-110 ppm K-1) [4] and bulk Ca₂RuO_{3,74} (-115 ppm K-1) samples [21], it is retained even above the magnetic ordering temperature (250K). These results are unexpected and represent a significant contrast with the bulk counterpart, as can be seen in Figure 3 a) (red diamond), where the bulk PTE is observed throughout the [130, 330]K temperature range: $\beta^{LT} \sim +39.7$ and $\beta^{HT} \sim +16.3$ ppm K⁻¹. Therefore, it is suggested that the NTE here observed emerges as a consequence of the size reduction of the Gd₅Si_{1.3}Ge_{2.7} material. The represented nanogranules volume values in the HT-NTE region correspond to the averaged values of the two phases O(I) and O(II) volumes weighted to their corresponding phase fractions. Both the O(I) and O(II) phases present NTE with similar V(T) dependence as can be seen in figure S6 in Supplementary Information [22]. Besides synchrotron diffraction experiments, electrical resistivity (o(T)) measurements were performed in the [90, 300]K interval and are presented in Figure 3 b) inset. Noticeably, the $\varrho(T)$ curve follows the same overall trend as the thermal expansion, presenting a negative temperature dependence along the LT-NTE and HT-NTE regions, which contrasts with the positive overall temperature dependence observed in electrical resistance measurements in Gd₅(Si,Ge)₄ bulk materials [23,24]. The Q(T) presents a smoother linear negative temperature dependence $(\rho(T)=\alpha T+B)$ with $\alpha < 0$ in the LT-NTE region followed by a sharper positive temperature dependence across the magnetostructural transition and again presenting a stronger negative linear temperature dependence in the HT-NTE region. The correlation between the electrical resistivity and the thermal expansion curves is further reinforced by the comparison between their linear slopes ratio of the HT-NTE and the LT-NTE regions of the V(T) and o(T) curves: $\alpha^{\rm HT}$ / $\alpha^{\rm LT}$ ~ 2.25 versus and $\beta^{\rm HT}$ / $\beta^{\rm LT}$ ~ 2.16. Both ratio values are similar, differing only by 4%. In fact, such correspondence results from the linear dependence of the phonon contribution to the total resistivity of a metal/alloy on the thermal expansion coefficient [25].

IV. ANALYSIS AND DISCUSSION

In the literature four different mechanisms are identified as main causes for NTE behavior: (i) abnormal electronic band temperature dependence, (ii) magnetovolume coupling, (iii) atomic radius contraction and (iv) the tension effect, as reviewed in reference [26]. (i) The abnormal electronic bands temperature dependence is enhanced at the nanoscale, however, this effect should only develop for ultra-small nanoparticles namely through the quantum-like induced separation of its discrete energy levels, as was observed in 4nm Au nanoparticles [5]. However, this effect should only develop for nanoparticles where the electron mean free path is higher than the nanoparticles size, hence leading to perturbations of the energy level state. Considering the nanogranules here presented have a broad distribution and a mean diameter of 80 nm, the electronic confinement is not expected in the majority of the nanogranules, neither significant perturbation of its discrete electronic levels which should be detected at electrical resistivity level. Nevertheless, the 80nm size does not rule out phonon confinement, as observed in Si-Ge \sim 100nm nanowires [27]. (ii) In magnetovolume based NTE bulk materials, such as La(FeSi)₁₃ [28] or MnCoGe-based materials [29], the magneto-volume coupling is strong enough to induce a large structural change, yet conventional PTE is recovered at temperatures below and above the magnetostructural transition. Another interesting magnetovolume based NTE was found while reducing the dimensions of CuO (5nm) [4]. Although in the present work the NTE also emerges with size reduction, here the magnetostructural transition results on a PTE whereas the NTE behavior occurs on the outside regions of the magnestructural transition. Moreover, the fact that it is retained even at temperatures above the magnetic ordering

temperature discards the magneto-volume coupling as NTE primary cause. (iii) The atomic radius contraction phenomenon occurs in systems where there is a charge transfer between two neighboring atoms, as reported for Sm_{2.75}C₆₀ [30] or LaCu₃Fe₄O₁₂ [31] bulk materials. Nonetheless, similarly to the magnetovolume coupled NTE materials, this phenomenon occurs only below a critical temperature.

The unexpected NTE behavior here reported demands further studies in order to be completely understood, however some important underlying mechanisms are identified which will contribute significantly for this goal. It should be noticed that when considering the unit cell parameters behavior as a function of temperature, the a contracts more, followed by the b and then the c parameters. Such anisotropic behavior is in accordance to the one observed along the magnetostructural transition in this nanostructure [14] and in general by the bulk Gd₅(Si,Ge)₄ materials, corroborating the image of rigid slabs alternated with flexible interslab regions. Inspecting the unit cell in greater detail and recalling the alternative octahedra, it becomes clear the structural importance of the triplet chain Ge3-Gd1- Ge3 on both the octahedra stability and its connection with the next-near octahedra along the b axis. A closer look into the Ge3-Gd1-Ge3 angle temperature dependence, $\phi(T)$, is given in Figure 4 a). As can be seen, $\phi(T)$ mimics the volume behavior and a linear correlation between the two is observed for LT-NTE and HT-NTE, as represented in Figure 4 a) inset. This striking resemblance helps to unveil the potential atomic mechanism behind the NTE behavior. As temperature increases, the Ge3-Gd1-Ge3 chain flexes, decreasing its angle ϕ and leading to a sliding movement of adjacent slabs (mostly along a-axis) decreasing the two adjacent slabs distance (along b-axis), which results in the observed anisotropic contraction and consequent volume reduction -the interslab Gd1-Gd1 and the Ge3-Ge3 interatomic distances behavior corroborate this phenomenon as can be seen in Figures S11

and S10 in [22]. This complex angle behavior contrasts with the almost invariant intraslab distances, Gd2-Gd2 (Figure S9 in [22]), both in the LT-NTE and HT-NTE temperature regions and the smoother variations of the in-slab triplet chains angles Ge,Si2-Gd1-Ge,Si2 (Figure S12 in [22]), which strengthens the role played by Ge3-Gd1-Ge3 triplet chain in the NTE behavior. This is schematically shown in Figure 4 b). In fact, the importance of triplet chain Ge3-Gd1-Ge3 in Gd₅(Si,Ge)₄ bulk materials was previously shown by Mudryk and co-workers when they have shown its critical role on the magneto-volume coupling of these materials [32]. However, in contrast with the behavior presented by the triplet angle in the Gd₅Si_{1.3}Ge_{2.7} nanogranules, the bulk counterpart $\phi(T)$ varies smoothly with temperature (Figure S13 in [22]). In fact, such $\phi(T)$ smoother behavior had already been observed in may Ge-rich bulk materials, where an angle change, $\Delta \phi$, smaller than 3° angle variation was observed across the structural transition and within a 100K temperature window [33,34]. This discrepancy of behaviors further underlines the role of the triplet angle on the observed NTE behavior as a consequence of size-reduction. Following this reasoning, the potential for Ge3-Gd1-Ge3 transverse vibrations, which correspond to a bond bending static image, is energetically more favorable than for longitudinal vibrations [3]. In fact, the triplet chain Ge3-Gd1-Ge3 is analogous to the M-O-M triplet found in the Metal-Organic Frameworks (MOF, where M = Ti, Zr, P, V,...) in the sense that both involve the simultaneous presence of lighter and mobile (Ge/O) bonded to heavier and more inert atoms (Gd/M) - in particular, their atomic mass ratio is identical to the NTE material ZrP₂O₇ $[Z(Gd)/Z(Ge) \sim Z(P)/Z(O)].$

Generally, the intrinsic strain on nano and bulk materials is very different, and the stress at a nanoparticle surface is known to be inversely proportional to its diameter, d [35]. Such stress is the equivalent to a hydrostatic applied pressure in the 1-50 kBar range for nanogranules with

diameter in the 1-100 nm range [35]. Gd₅Si_xGe_{1-x} materials are greatly sensitive to pressure, namely by shifting their Curie temperature, inducing magnetostructural transitions and enhancing its MCE [36,37]. In fact, a 13K increase in nanogranule T_{MS} is observed in comparison with the bulk counterpart, which corresponds to an hydrostatic pressure in the 8-11 kBar range [35]. This large size-induced stress is the major difference between the bulk and the nanoscale, and hence can be considered the driving force of the reported NTE. Such stress is supported by the Williamson-Hall analysis of the XRD peaks full width at half maximum, which unveiled a large strain (1.7 x 10⁻³) present in these nanogranules (Figure S14 in [22]). More generally, the fact that the volume contraction occurs more steeply in the HT-NTE than in the LT-NTE region, can have a two-fold explanation: 1) in the LT-NTE region, the magneto-volume effect is acting in opposition to the overall NTE (towards PTE, as occurs in the bulk counterpart), therefore counterbalancing the nanogranules overall tendency to contract; 2) pressure studies on Gd5(Si,Ge)4 bulk materials unveiled that their corresponding compressibility is larger for structures with higher unit cell volume, in particular $\kappa_{O(II)} = 1.85 \text{ Mbar}^{-1} > \kappa_{O(I)} = 1.35 \text{ Mbar}^{-1}$ ¹ [36,37]. In particular, it is interesting to remark that these pressure studies by Magen and coworkers, disclosed an anisotropic unit cell compressibility which is in accordance with what was found here: $\kappa a > \kappa b > \kappa c$.

The pressure-induced lattice softening was already predicted and observed in bulk NTE materials, including simple metals [38–40]. In these materials, pressure was able to dramatically change the Young's modulus, softening the atomic lattice and changing its phonon spectra. Hence, considering this scenario, it is plausible to infer that the Gd₅Si_{1.3}Ge_{2.7} nanogranules surface stress is inducing a lattice softening, a phenomenon also known as the reverse Hall–Petch effect [41], recently observed in Mn-Co-Ge-In granules [29], changing the

phonon spectra and promoting the Ge3-Gd1-Ge3 transverse vibrations and the consequent NTE behavior.

As suggested by Barrera and co-authors review [3], nanogranules higher surface/volume ratio generally enhances the role of the tension effect and therefore leads to changes in the thermal expansion behavior, as was observed here.

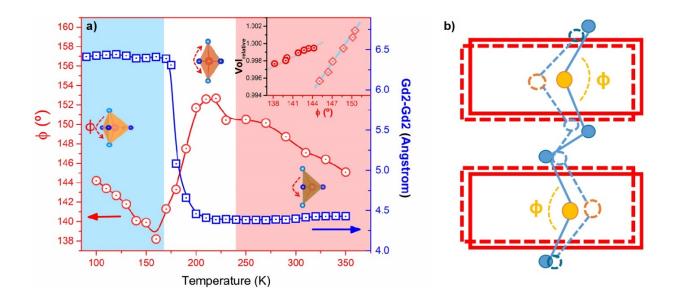


Figure 4. a) Temperature dependence of the φ angle between the key Ge3-Gd1-Ge3 triplet atomic chain in the left y-axis and of the Gd2-Gd2 intraslab interatomic distance in the right y-axis. In the inset the relative unit cell volume is plotted against the φ angle. b) Schematic illustration of the key distances along the interslab region leading to the overall negative thermal expansion behavior. With increasing temperature, the slab and atoms move to new positions, here represented by dashed rectangles and circles, respectively.

V. CONCLUSIONS

In summary, a giant negative thermal expansion has been observed through Synchrotron measurements of Gd₅Si_{1,3}Ge_{2,7} nanogranules in two temperature regions including room temperature 255-340K, expanding the set of giant effects observed in these materials (magnetocaloric, magnetoresistive and magnetostrictive). Size-induced NTE behavior had already been observed in several magnetic nanoparticles, however here, unprecedentedly, the NTE behavior is retained even at temperatures above the magnetic ordering temperature. The observed NTE emerges as a consequence of the scale reduction and its main atomic mechanism is thought to be the bond flexing of the key Ge3-Gd1-Ge3 triplet chain which is enhanced at the nanoscale. The size reduction leads to a noticeable increase of the material magnetostructural transition temperature which is explained by the enhanced stress at the nanoparticle surface, estimated at 8-11kbar. This stress leads to lattice softening and a consequent enhancement of low-frequency vibrations resulting in an overall contraction behavior that competes with the vmagnetovolume expansion force below the magnetic ordering temperature. The Gd₅Si₁3Ge_{2.7} ~80nm range nanogranules can be particularly suited to compensate PTE materials when mixed in a composite [42,43], which has a wide range of technological applications such as Bragg gratings optical fiber coatings [44] or in printed circuit boards [45].

VI. ACKNOWLEDGEMENTS

The authors acknowledge FCT and COMPETE 2020 FEDER for financial support through the projects: PTDC/CTM-NAN/115125/2009, EXPL/EMS-ENE/2315/2013, FEDER/POCTIn0155/94, NECL - NORTE-01-0145-FEDER-022096 and UID/NAN/50024/2019. J.H. Belo thanks FCT for the Grant SFRH/BD/88440/2012, the project PTDC/FIS-MAC/31302/2017 and his contract DL57/2016 reference SFRH-BPD-87430/2012. I.T. Gomes acknowledge QREN for the project NORTE-070124-FEDER-000070 for the financial support. V. M. A thanks CNPq for the grant n°156077/2018-3. Work at Ames Laboratory was supported by the U.S. Department of Energy, Office of Basic Energy Sciences, Division of Materials Science and Engineering. Ames Laboratory is operated for DOE by Iowa State University under Contract no.DE-AC02-07CH11358.

REFERENCES

- [1] T. A. Mary, J. S. O. Evans, T. Vogt, and A. W. Sleight, Science **272**, 90-92 (1996).
- [2] J. P. Attfield, Nature **480**, 465 (2011).
- [3] G. D. Barrera, J. A. O. Bruno, T. H. K. Barron, and N. L. Allan, J. Phys. Conden. Matt. 17, R217 (2005).
- [4] X. G. Zheng, H. Kubozono, H. Yamada, K. Kato, Y. Ishiwata, and C. N. . Xu, Nat. Nanotechnol. 3, 724 (2008).
- [5] W. H. Li, S. Wu, C. Yang, S. Lai, K. Lee, H. L. Huang, and H. D. Yang, Phys. Rev. Lett. 89, 135504 (2002).
- [6] V. K. Pecharsky and K. A. Gschneidner, Phys. Rev. Lett. **78**, 4494 (1997).
- [7] V. K. Pecharsky and K. A. Gschneidner Jr, Adv. Mater. **13**, 683 (2001).
- [8] V. K. Pecharsky, A. P. Holm, K. A. Gschneidner, and R. Rink, Phys. Rev. Lett. 91, 197204 (2003).
- [9] M. Nazih, A. de Visser, L. Zhang, O. Tegus, and E. Bruck, Solid State Commun. **126**, 255 (2003).
- [10] M. Han, PhD Thesis, Iowa State Univ. (2004).
- [11] Y. Mudryk, D. Paudyal, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B 77, 24408 (2008).

- [12] A. L. Pires, J. H. Belo, J. Turcaud, G. N. P. Oliveira, J. P. Araújo, A. Berenov, L. F. Cohen, A. M. L. Lopes, and A. M. Pereira, Mater. Des. 85, 32 (2015).
- [13] A. L. Pires, J. H. Belo, I. T. Gomes, R. L. Hadimani, D. C. Jiles, L. Fernandes, P. B. Tavares, J. P. Araújo, A. M. L. Lopes, and A. M. Pereira, Mater. Lett. 159, 301-304 (2015).
- [14] R. L. Hadimani, J. H. B. Silva, A. M. Pereira, D. L. Schlagel, T. A. Lograsso, Y. Ren, D.C. Jiles, and J. P. Araújo, Appl. Phys. Lett. 5, 32402 (2015).
- [15] A. L. Pires, J. H. Belo, I. T. Gomes, A. M. L. Lopes, J. P. Araújo, A. M. Pereira, R. L. Hadimani, D. C. Jiles, R. L. Hadimani, D. L. Schlagel, T. A. Lograsso, D. C. Jiles, and T. A. Lograsso, Thin Solid Films 621, 247-252 (2017).
- [16] G. J. Miller, Chem. Soc. Rev. **35**, 799-813 (2006).
- [17] J. Rodriguez-Carvajal, Physica B **192**, 55 (1993).
- [18] W. Bao, F. Miao, Z. Chen, H. Zhang, W. Jang, C. Dames, and C. N. Lau, Nat. Nanotechnol. 4, 562 (2009).
- [19] D. Yoon, Y.-W. Son, and H. Cheong, Nano Lett. **11**, 3227 (2011).
- [20] J. S. O. Evans, T. A. Mary, T. Vogt, M. A. Subramanian, and A. W. Sleight, Chem. Mater.8, 2809 (1996).
- [21] K. Takenaka, Y. Okamoto, T. Shinoda, N. Katayama, and Y. Sakai, Nat. Commun. 8, 14102 (2017).

- [22] J. H. Belo, A. L. Pires, I. T. Gomes, V. Andrade, J. B. Sousa, R. L. Hadimani, D. C. Jiles, X. Zhang, Y. Ren, J. P. Araújo, and A. M. Pereira, (Supplementary Information).
- [23] L. Morellon, J. Stankiewicz, B. Garcı´a-Landa, P. A. Algarabel, and M. R. Ibarra, Appl. Phys. Lett. **73**, 3462 (1998).
- [24] Y. Mudryk, V. K. Pecharsky, and K. A. J. Gschneidner, *Handbook on the Physics and Chemistry of Rare Earths Includind Actinides Vol. 44*, Chapter 262 (North Holland, 2014).
- [25] D. K. Palchaev, Z. K. Murlieva, S. H. Gadzhimagomedov, M. E. Iskhakov, M. K. Rabadanov, and I. M. Abdulagatov, Int. J. Thermophys. **36**, 3186 (2015).
- [26] K. Takenaka, Sci. Technol. Adv. Mater. 13, 013001 (2012).
- [27] A. Malhotra and M. Maldovan, Sci. Rep. 6, 25818 (2016).
- [28] R. Huang, Y. Liu, W. Fan, J. Tan, F. Xiao, L. Qian, and L. Li, J. Am. Chem. Soc. 135, 11469 (2013).
- [29] F. Shen, H. Kuang, F. Hu, H. Wu, Q. Huang, F. Liang, K. Qiao, J. Li, F. Shen, H. Kuang, F. Hu, and H. Wu, APL Mater. 5, 106102, (2017).
- [30] J. Arvanitidis, K. Papagelis, S. Margadonna, K. Prassides, and A. N. Fitch, Nature **425**, 599 (2003).
- [31] Y. W. Long, N. Hayashi, T. Saito, M. Azuma, S. Muranaka, and Y. Shimakawa, Nature 458, 60 (2009).
- [32] Y. Mudryk, D. Paudyal, V. K. Pecharsky, K. A. Gschneidner, S. Misra, and G. J. Miller,

- Phys. Rev. Lett. 105, 66401 (2010).
- [33] L. Morellon, J. Blasco, P. A. Algarabel, and M. R. Ibarra, Phys. Rev. B 62, 1022 (2000).
- [34] W. Choe, G. J. Miller, J. Meyers, S. Chumbley, and A. O. Pecharsky, Chem. Mater. **15**, 1413 (2003).
- [35] D. Vollath, *Nanomaterials, An Introduction to Synthesis, Properties and Application*, 2nd ed. (WILEY-V C H VERLAG GMBH, 2013).
- [36] C. Magen, L. Morellon, P. A. Algarabel, M. R. Ibarra, Z. Arnold, J. Kamarad, T. A. Lograsso, D. L. Schlagel, V. K. Pecharsky, A. O. Tsokol, and K. A. Gschneidner Jr., Phys. Rev. B 72, 24416 (2005).
- [37] C. Magen, Z. Arnold, L. Morellon, Y. Skorokhod, P. A. Algarabel, M. R. Ibarra, and J. Kamarad, Phys. Rev. Lett. **91**, 2072021 (2003).
- [38] S. Li and Y. Chen, Phys. Rev. B **96**, 134104 (2017).
- [39] C. Pantea, A. Migliori, P. B. Littlewood, Y. Zhao, H. Ledbetter, J. C. Lashley, T. Kimura, J. Van Duijn, and G. R. Kowach, Phys. Rev. B 73, 1 (2006).
- [40] D. T. Ho, S.-Y. Kwon, H. S. Park, and S. Y. Kim, Nano Lett. **17**, 5113 (2017).
- [41] J. Schiotz, F. D. Di Tolla, and K. W. Jacobsen, Nature **391**, 561 (1998).
- [42] L. C. Kozy, N. Tahir, and W. Tremel, J. Mater. Chem. 7, 2760 (2009).
- [43] J. Sheng, L. D. Wang, D. Li, W. P. Cao, Y. Feng, M. Wang, Z. Y. Yang, Y. Zhao, and W. D. Fei, Mater. Des. 132, 442 (2017).

- [44] K. Kintaka, J. Nishii, and P. G. Kazansky, Opt. Lett. 27, 1394 (2002).
- [45] J. S. O. Evans, J. Chem. Soc. Dalt. Trans. 19, 3317 (1999).

GRAPHICAL TABLE OF CONTENTS

