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Magnetism and Faraday Rotation in Oxygen-Deficient Polycrystal-1 line and Single-Crystal Iron-Substituted Strontium Titanate 2

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19 ABSTRACT

20 21 22 23 24 25 26 Both polycrystalline and single-crystal films of iron-substituted strontium titanate, Sr(Ti_{0.65}Fe_{0.35})O₃₋₆, prepared by pulsed laser deposition (PLD), show room temperature magnetism and Faraday rotation (FR), with the polycrystalline films exhibiting higher saturation magnetization and FR. The magnetic properties vary with the oxygen pressure at which the films are grown, showing a maximum at pressures of $\sim 4 \mu$ Torr at which the unit cell volume is largest. The results are discussed in terms of the oxygen stoichiometry and corresponding Fe valence states, the structure and strain state, and the presence of small volume fractions of metallic Fe in single-crystal films grown at the optimum deposition pressure. Integration of magneto-optical polycrystalline films on an optical waveguide device demonstrates a nonreciprocal phase shift.

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I. INTRODUCTION

28 Complex oxides exhibit a wide range of useful properties such as high temperature superconductivity, ferroelectricity, colossal 29 30 magnetoresistance, magnetism, magneto-electricity, magneto-optical activity, and multiferroicity that have inspired spintronic, magnetoelectronic, and optical applications [1-6]. Transition-metal-substituted perovskites or perovskite-derived structures are par-31 ticularly interesting because their properties can be tuned by cation substitution over extensive composition ranges, and because 32 33 they can be epitaxially grown on a buffered Si substrate [7-10]. Beyond cation composition, the properties are also very sensitive to the growth conditions and substrate used, which affect the microstructure, film strain [7], oxygen deficiency, the cation valence 34 states, and the ordering of cations within the A and B sites.

35 Iron-substituted strontium titanate or $SrTi_{1-x}Fe_xO_{3-\delta}$ (STF) (here δ represents the oxygen deficiency with respect to a stoichiometric 36 37 perovskite $SrTi_{1-x}Fe_xO_3$, i.e., one in which the Fe is present as Fe^{4+}) has been extensively characterized as a mixed electronic-ionic conductor [11-15] and for its magnetic, magneto-optical, ferroelectric, and catalytic properties [16-19]. Characterization of STF has 38 included determining its oxygen content via thermogravimetric analysis and coulometric titration [20,21], and measurements of 39 ionic/electronic conductivity and impedance [12,13,15,22-24]. The magnetic properties are believed to be related to oxygen vacan-40 cies [16,20,25], although the origin of the magnetism is not fully understood. We have previously found that epitaxial STF and 41 $Sr(Ti_{1-x}Co_x)O_{3-\delta}$ (STCo) films on $SrTiO_3$ (STO), LaAlO₃, and CeO₂/yttria-stabilized zirconia (YSZ)-buffered (001) Si substrates 42 grown in vacuum with x = 0.1 to 0.5, i.e. non-dilute systems, exhibited room-temperature magnetism and anisotropy with an out-of-43 plane magnetic easy axis that was attributed to magnetoelastic effects [16,17,26-28]. However, films deposited in oxygen had no 44 significant room-temperature magnetism. The magnetic moment of vacuum-deposited epitaxial STF films persisted to a tempera45 ture of $>1000^{\circ}$ C and the magnetization decreased slowly as a function of temperature [16,29,30]. Depending on the thickness and 46 deposition conditions, STF could be grown as a single crystal film on STO, or as a double epitaxial film consisting of an (001)-47 oriented STF film with (011)-oriented crystals growing within it, both orientations being epitaxial with the substrate [31]. The sin-48 gle crystal and double epitaxial microstructures differed in their strain state, uniformity of the Fe distribution, and magnetic proper-49 ties [19,31].

50 These experimental results led to the conclusion that strain and oxygen stoichiometry were key structural factors in determining the 51 52 53 54 magnetic properties of the STF and STCo films (factors that are also important in many other oxide systems [29,32],) but there are few reports on the effect of oxygen pressure during growth [33,34], and polycrystalline films have not been investigated. We report on the correlations among the microstructure, valence, and distribution of the Fe ions as well as the room temperature magnetic properties of STF films as a function of base pressure P, characterized by high-resolution x-ray diffraction (XRD), transmission 55 electron microscopy (TEM), Faraday rotation (FR), magnetometry, and Mössbauer spectroscopy. The STF films on Si were poly-56 crystalline but exhibited a magnetic moment and FR up to 65% higher than the single-crystal film deposited on an STO substrate. 57 From the viewpoint of applications, the formation of polycrystalline magnetic perovskite films on silica or other non-epitaxial sub-58 strates allows the useful properties of STF to be easily introduced into complementary metal-oxide semiconductor (CMOS) devices 59 and into optical devices such as isolators and magnetophotonic crystals [4,35-39], spatial light modulators [40-42], or magneto-60 optical holographic memories [43-46]. First-principles calculations demonstrate the importance of oxygen defects and exchange 61 interactions in determining the magnetic properties.

II. FILM GROWTH AND CHARACTERIZATION

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A. Film preparation

65 In this work, STF films were fabricated on silicon covered by a native oxide and on STO substrates under various P using pulsed 66 laser deposition (PLD). The SrTi_{0.62}Fe_{0.38}O_{3- δ} target was prepared using a conventional mixed oxide sintering method [47]. During 67 film deposition, the KrF excimer laser (wavelength $\lambda = 248$ nm) was pulsed at a frequency of 10 Hz with a pulse duration of 25 ns. 68 The spot size on the target was 0.5 mm \times 2.0 mm, and the energy density at the target surface was \sim 1.3 J cm⁻² during the pulse. The 69 deposition rate was 1.8-2.7 nm min⁻¹. STF films of thickness 70 ± 10 nm were deposited by PLD onto 0.38-mm-thick double-side-70 polished (001) Si substrates and onto 0.5-mm-thick double-side-polished (001) STO substrates. Films were grown simultaneously 71 72 73 74 on both substrate types in each deposition run. The substrates were placed 80 mm from the target, and the substrate holder was heated to 650°C and was rotated for uniform growth. This temperature was chosen to minimize the formation of secondary phases such as iron oxides. Samples were prepared at a range of base pressures below 10 µTorr by pumping the chamber until a specific base pressure P was reached. The substrate was then heated, which raised the total pressure of the chamber to about 30 μ Torr due to 75 chamber outgassing. The chamber pressure was then pumped back down to P before deposition started. Therefore the pressure P 76 during deposition consists of residual gas which includes oxygen and water vapor. Pre-deposition (5×10^3 laser pulses) was started 77 with the shutter placed between the PLD target and the substrate, leading to an increase in the total chamber pressure by $\sim 2 \mu$ Torr. 78 After opening the shutter for film deposition (2×10^4 pulses), the chamber pressure gradually decreased and returned to near P at the 79 end of the deposition. The substrate was cooled to room temperature at a rate of -5° C min⁻¹ while pumping the chamber.

B. X-ray characterization

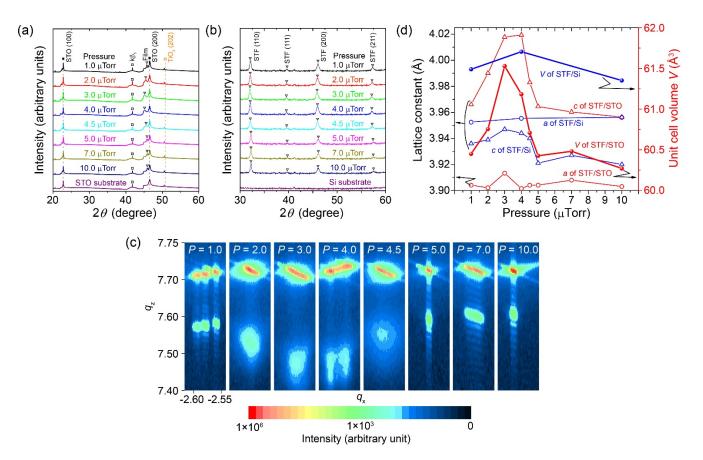
81 In-plane and out-of-plane lattice constants of single crystalline films on STO and polycrystalline films on Si were characterized by 82 XRD (PANalytical, X'Pert) and two high-resolution XRD systems (HRXRD, Bruker-Axs D8 DISCOVER and Rigaku Smartlab). 83 XRD patterns are shown in Figs. 1(a) and 1(b) for STF/STO and STF/Si, taken using a Cu-K α radiation source at a wavelength of 84 0.15418 nm. Figure 1(a) was measured by XRD (PANalytical, X'Pert), in which $K\beta_1$ peaks from the Cu source were observed. 85 Figure 1(b) was measured by HRXRD (Rigaku Smartlab). The STF/Si samples were tilted by 1° to eliminate the strong substrate 86 peak. The out-of-plane lattice parameters of STF on STO and on Si, $c_{\text{STF/STO}}$ and $c_{\text{STF/Si}}$, varied with the deposition pressure, and 87 exhibited a maximum at $P = 3 - 4 \mu$ Torr, with the variation stronger for the STF/STO. This behavior was also confirmed with the 88 (002) peak obtained from XRD (PANalytical X'Pert). The TiO₂ (202) peak in Fig. 1(a) originated from the substrate.

89 Reciprocal space mapping (RSM) was used to measure the in-plane lattice parameter a_{STF/STO} for the STF/STO. Grazing incidence 90 diffraction (GID) was used to measure $a_{\text{STF/Si}}$ for the STF/Si samples grown at P = 1, 4, and 10 μ Torr with the incident angle of the 91 x-rays ω fixed at 0.4°. Figure 1(c) shows the RSM data for the STF/STO samples for the asymmetric (310) peak. The intense upper 92 peak arises from diffraction from the STO substrate, and the bottom peak corresponds to the STF films. Multiple peaks are attribut-<u>9</u>3 ed to twinning of the substrates. The film and substrate peaks align in q_x indicating coherent cube-on-cube epitaxial growth of the 94 STF films on the STO substrate, i.e. $a_{\text{STF/STO}} = a_{\text{STO}}$, whereas q_z gives the out-of-plane lattice parameter c, which differs for the film 95 and substrate. The RSM data gave a standard deviation of $< 10^{-3}$ Å for $a_{\text{STE/STO}}$. The difference in values for c from different diffrac-96 tometers, given by (HRXRD value – XRD value) / HRXRD value, was less than 0.1%, and the value of $c_{\text{STF/STO}}$ from RSM matched 97 that measured by XRD. All the STF films on STO had out-of-plane lattice parameters greater than that of STO ($a_{\text{STO}} = 3.905$ Å) and 98 were tetragonally distorted, with $(c/a)_{\text{STF/STO}} = 1.013$ to 1.030 indicating an in-plane compressive strain in the STF, similar to that 99 found in prior work [16]. Figure 1(d) shows the variation of $a_{\text{STF/STO}}$, $c_{\text{STF/STO}}$ and unit cell volume $V_{\text{STF/STO}}$ with base pressure P.

100 For the samples of STF/Si, the in-plane and out-of-plane lattice parameters, $a_{\text{STF/Si}}$ and $c_{\text{STF/Si}}$, and the unit cell volume $V_{\text{STF/Si}}$ are shown in Fig. 1(d). The data gave a standard deviation of $\sim 6 \times 10^{-4}$ Å for *a*, and 2×10^{-3} Å for *c*. The value of $a_{\text{STF/Si}}$ was

101 102 3.955 ± 0.003 Å and showed little variation with P. Of the three STF/Si samples whose unit cell volume was measured, the one 103 grown at $P = 4.0 \,\mu\text{Torr}$ showed the largest unit cell volume, the same trend as for films on the STO substrate. The value of $c_{\text{STF/Si}}$ 104 was similar to or slightly smaller than $a_{\text{STF/Si}}$, i.e. $(c/a)_{\text{STF/Si}} = 0.990$ to 0.998, depending on P. Thermal mismatch on cooling would 105 lead to a tensile strain in the STF/Si film (c/a < 1), since Si has a lower thermal expansion coefficient than STF. This differs from 106 the films on STO where lattice mismatch governs the strain state and produces an in-plane compressive strain in the STF, with c/a107 > 1.





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FIG. 1. Structural analysis of the STF films with XRD. (a) XRD patterns of STF/STO. The triangles show the film peaks. Black squares are peaks from bulk STO. A circle shows a TiO₂ peak present in the substrate scan. (b) XRD patterns of STF/Si. Positions of STF peaks are shown with triangles. There are no Si peaks in this angular range. (c) Reciprocal space mapping of single crystalline STF on STO substrate, plotted in q_z - q_x for the asymmetric (310) peak. (d) Lattice constants and unit cell volumes of STF films versus deposition pressure for both STF/STO and STF/Si.

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116 The larger unit cell volume of the films compared to that estimated for bulk STF without oxygen vacancies [SrTi_{0.65}Fe_{0.35}O₃, V =117 ~60.13 Å³ interpolated linearly from the unit cell volumes of STO (a = 3.9051 Å, $V = \sim 59.55$ Å³) [48] and SrFeO₃ (a = 3.9410 Å, V 118 $= -61.21 \text{ Å}^3$ (49) is a result of the greater oxygen vacancy content in the films compared to the bulk, which drives the Fe to lower 119 valence states, expanding the lattice because of the larger ionic radii of the Fe. Other work on STO films [50] has showed changes 120 in cation stoichiometry and hence lattice parameter at different laser focus conditions, but here we kept the deposition conditions 121 nominally the same except for the base pressure, so we attribute changes in lattice volume to the oxygen vacancies and correspond-122 ing changes in Fe ionic radius. MacChesney et al. and Taguchi et al. showed that oxygen vacancies in SrFeO_{3- δ} and SrCoO_{3- δ} re-123 spectively increased the unit cell volume [51,52], and a sufficiently large oxygen deficiency ($\delta > 0.12$) promoted a cubic-to-124 125 tetragonal structural change in SrFeO_{3- δ} [42]. Perry *et al.* showed that the largest lattice parameter is obtained for STF annealed under the most reducing conditions, based on chemical expansion measurements [34]. Chakraverty et al. demonstrated that, for 126 SrFeO_{2.5} films on STO grown by PLD, the laser spot size, which affects the growth process, kinetics and stoichiometry, controlled 127 the lattice parameter [53].

C. Film morphology.

129 Cross-sectional TEM images of STF/Si and STF/STO grown at P = 1, 4 and 10 μ Torr are given in Fig. 2. All three STF/Si films

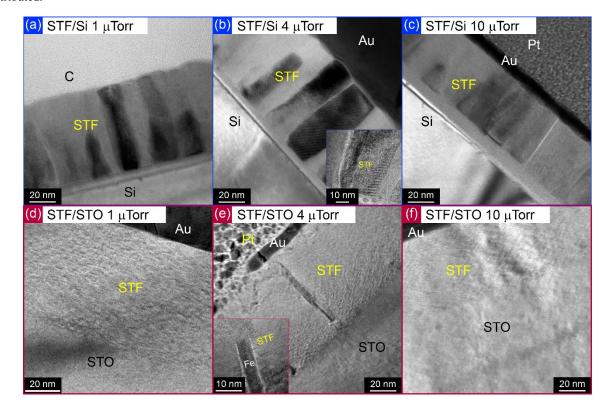
had a polycrystalline columnar microstructure, and energy-dispersive x-ray spectroscopy (EDX) mapping of the 4 μ Torr sample, Fig. 3, indicates a spatial variation in the Fe content, with higher Fe content corresponding to a few of the columnar features. In 132 agreement with the Mössbauer spectroscopy (described below), there was no clear evidence from high resolution TEM that these 133 134 columns were metallic Fe, and we presume they represent STF with a higher Fe content than neighboring grains. A similar inhomogeneity was found at $P = 1 \mu$ Torr, but at 10 μ Torr the Fe was uniformly distributed. The compound SrTi_{1-x}Fe_xO_{3- $\delta}$} is stable 135 throughout the composition range [13,14], therefore we assume that the inhomogeneity is different in origin from that seen in spi-136 nodal decomposition such as that of Cr-rich (Zn, Cr)Te [54], In-rich (Ga, In)N [55], or (Mn, Ga)As [56] systems. In comparison, 137 the STF grown on STO was single-crystal but showed contrast that indicated small scale structural inhomogeneity, possibly repre-138 senting layered vacancy ordering as seen in brownmillerite-structured SrFeO_{2.5} [57] and in (La,Sr)CoO₃ [58,59]. At $P = 4 \mu$ Torr, a 139 few straight, vertical nanopillars were observed perpendicular to the substrate within the perovskite matrix. Lattice imaging re-140 vealed that the pillars were metallic α -Fe. The pillars were sparsely distributed, with spacings of ~200 nm in the TEM image of a 141 lamella ~20–50 nm thick. This suggests a density of 100–250 pillars per μ m².

142 The pillar width was 3–5 nm, giving an estimate of the volume fraction of metallic Fe of 0.1–0.4%. This volume fraction is small,

but it could contribute a significant amount of the net magnetization because metallic Fe has a high saturation magnetization of

144 1700 emu cm⁻³. However, the majority of the Fe remains in the perovskite, e.g. the formation of even 0.5 vol.% metallic Fe would 145 deplete Fe from ~8% of the STF volume. Films grown on STO at $P = 1 \mu$ Torr and 10 μ Torr had no pillars, and the Fe was uniform-

146 ly distributed.



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148 FIG. 2. Transmission electron microscopy of STF samples. Cross-sectional TEM images of STF/Si deposited at $P = (a) 1 \mu$ Torr, (b)

149 4 μ Torr, and (c) 10 μ Torr. Cross-sectional TEM images of STF/STO deposited at $P = (d) 1 \mu$ Torr, (e) 4 μ Torr, showing a Fe rod,

 $150 \qquad \text{and (f) 10} \ \mu\text{Torr. Specimens were covered with conductive materials, either Au or C.}$

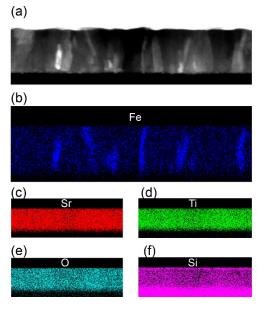


FIG. 3. Composition analysis of STF/Si. (a) STEM images and STEM-EDX element mappings of STF deposited on Si at P = 4 μ Torr. STEM-EDX Element map of (b) Fe, (c) Sr, (d) Ti, (e) O, (f) Si. A Sr peak overlaps the Si peak leading to a false signal for the region of the film in (f).

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D. Fe valence state.

157 Measurement of the atomic ratio of Ti and Fe is shown in Fig. 4(a) from energy-dispersive x-ray spectroscopy (EDS, FEI Philips XL 30), confirming that there was no systematic change in the ratio of Fe:Ti (= 0.35:0.65) with *P*.

159 The Fe valence state was characterized by conversion electron Mössbauer spectrometry (CEMS) and by x-ray photoelectron spec-160 troscopy (XPS). CEMS showed a difference between two films grown at $P = 2 \mu$ Torr on different substrates, as shown in Figs. 4(c), 161 and 4(d). For the STF/Si, a weak paramagnetic signal of Fe³⁺ was detected (~0.7% max emission). The signal may be fitted to a 162 singlet (Gaussian) or a doublet (2 pseudo-Voigts) as shown in Fig. 4(c), yielding an isomer shift relative to α -Fe of 0.25 ± 0.04 mm 163 s⁻¹ and a quadrupole splitting of 0.89 ± 0.08 mm s⁻¹. This signal may arise from Fe³⁺ associated with oxygen vacancies [60]. No 164 other signals were resolved.

The STF/STO showed 0.5–0.6 % emission after optimization. Unlike the sample on Si, the STF/STO showed a metallic iron magnetic phase, identified not just because of the presence of a sextet but also because of the isomer shift of 0.00 ± 0.02 mm s⁻¹, equal to that of α -Fe (Fig. 4(d)). The peak-intensity ratio of the sextet state indicates that the iron is not uniformly magnetized perpendicular to the substrate [61-63]. There was also a paramagnetic ferric doublet with an isomer shift of 0.35 ± 0.04 mm s⁻¹ that accounted for 23% of the peak area. A measurement using Auger electrons sensitive to the top 20–30 nm of the film showed a lower peak area fraction (15%) for the paramagnetic iron suggesting a depth dependence of the iron valence state. These results indicate that metallic Fe was present in the STF/STO but not in the STF/Si grown at $P = 2 \mu$ Torr.

Figure 4(b) shows XPS results in which the Fe peaks were similar over the range of deposition pressures and the two substrate types. The peak positions were calibrated so that the Cu peak was present at 284.8 eV. There were no significant differences in the Sr, Ti, and O XPS peaks. The satellite peaks around the Fe $2p_{3/2}$ and $2p_{1/2}$ doublet indicate the presence of Fe³⁺ but not Fe²⁺, similar to epitaxial BiFeO₃ [64] and iron oxides [65]. The lack of a metallic Fe peak in the 4 µTorr sample on STO differs from the TEM results and from the Mössbauer analysis of the $P = 2 \mu$ Torr sample. Since XPS is surface-sensitive, we assume that the top parts of the Fe nanorods at the surface would be subject to oxidation, or their very small area coverage may have precluded the observation of a metallic Fe peak.

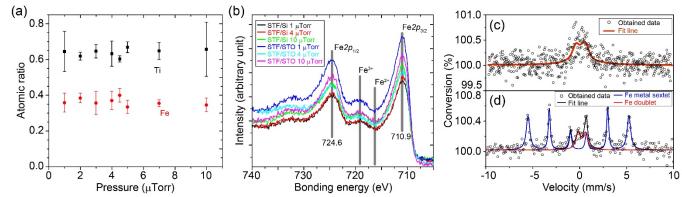


FIG. 4. Composition analysis of the STF films. (a) Atomic ratio of Fe to Ti in STF/STO measured by EDS. Error bars show the standard deviation. (b) XPS peaks from Fe 2p in STF films on STO and Si substrates. Mössbauer spectroscopy of (c) STF/Si, and (d) STF/STO, both deposited at $P = 2 \mu$ Torr, measured over the depth of the films. The STF film on the Si substrate does not show the metal sextet phase, but instead, a weak paramagnetic signal.

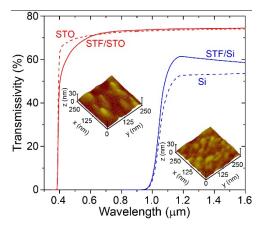
E. Optical and magnetic characterization

188 Transmissivity of the films on the two substrates from visible to near-IR wavelengths is shown in Fig. 5. For energies below the bandgap, the transmissivity of STF/Si was higher than that of bare silicon because of the anti-reflection effect that arises from introducing a lower refractive index film on a higher refractive index substrate. The transmissivity of STF/STO and bare STO were similar except for energies just below the bandgap where the STF/STO became absorbent.

Atomic force microscope (AFM) images of the STF/Si and STF/STO grown at $P = 4 \mu$ Torr are shown as insets in Fig. 5. The films had a root-mean-square roughness of 1.8 and 2.2 nm, respectively, leading to reflective film surfaces. The STF/Si grown at P = 1, 4,

and 10 μ Torr had similar surface topography, with in-plane feature sizes of 20–40 nm. In contrast, the STF/STO grown at P = 1

195 μ Torr and 10 μ Torr was much smoother than the film grown at 4 μ Torr.



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FIG. 5. Optical properties of STF films. Transmission spectra of STF/Si, STF/STO, and their substrates; the insets show the AFM images of the surface topography.

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200 Figure 6 shows the room temperature magnetic hysteresis loops from vibrating sample magnetometry (VSM) for films on both sub-201 strates as a function of P. Linear substrate signals were subtracted. Uncoated STO substrates subjected to thermal cycling in vacu-202 um in the PLD chamber similar to the thermal cycling during film growth showed a diamagnetic response with little change on 203 annealing. For both substrates, the films grown at $P = 4.0 \mu$ Torr showed the largest saturation magnetization, while films grown at 204 or above $P = 5 \mu$ Torr did not exhibit a hysteresis loop. The magnetic easy axis was along the out-of-plane direction, with switching 205 fields of 3-4 kOe and a high remanence. The hysteresis loop shape, switching field, and the anisotropy field are generally consistent 206 with those seen in previous studies of STF/STO [16,27,28]. The STF/STO grown at $P = 4 \mu$ Torr had $M_s = 17 \text{ emu cm}^{-3}$, whereas-207 remarkably—the polycrystalline STF/Si had $M_s = 28$ emu cm⁻³, which is higher by a factor of nearly two and corresponds to 0.48 208 $\mu_{\rm B}$ /Fe. In the STF/STO, the metallic Fe would contribute ~2–6 emu cm⁻³ based on the estimated volume fraction.

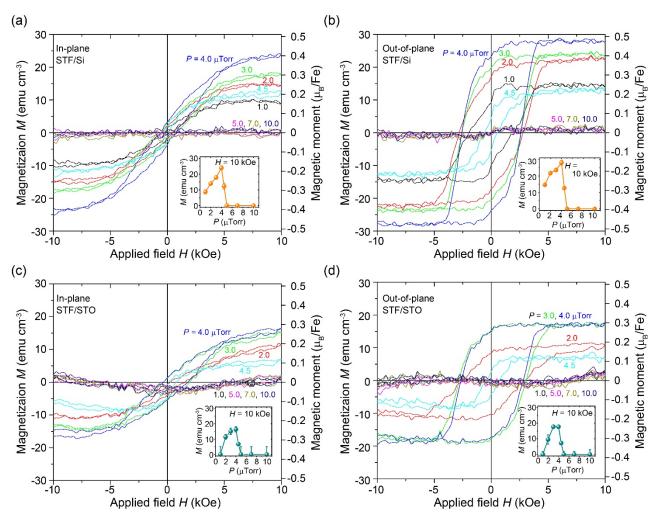


FIG. 6. Magnetic properties of STF films measured by VSM. (a),(b) Magnetic properties of STF/Si specimens (c),(d) Magnetic properties of STF/STO specimens, (a),(c) in-plane and (b),(d) out-of-plane. Insets show the dependence of the magnetization on pressure at H = 10 kOe.

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The magneto-optical response of STF (FR loop versus applied magnetic field) was measured at a wavelength of 1550 nm by the Faraday cell modulation method [65] and is shown in Fig. 7. The field and light were perpendicular to the film in the Faraday measurement, and the Faraday loop showed a comparable shape and coercivity (~3 kOe) to the out-of-plane magnetic hysteresis loops shown in Fig. 6. Low-spin (S = 1, 2 μ_B per ion) Fe⁴⁺ can provide splitting of the two circular modes of a linearly polarized optical wave required for FR [29]. The saturation FR angle of STF/Si was approximately two times larger than that of STF/STO, reaching -1900 °cm⁻¹, i.e. the FR scaled with the saturation magnetization. This FR is ~60% of that reported for single crystalline cerium-substituted yttrium iron garnet (CeYIG, Ce₁Y₂Fe₅O₁₂) [66,67] at 1550 nm. CeYIG is a candidate material for magnetooptical devices in the near-IR. (CeYIG films [4,5,68-70] differ from STF in having an in-plane easy axis.)

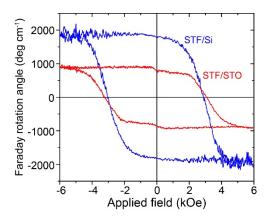


FIG. 7. Magneto-optical properties of STF films. FR loops of STF/Si and STF/STO samples at a wavelength of 1550 nm.

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F. Discussion

These results illustrate a sensitivity of the STF structure and the optical and magnetic properties to both the pressure during deposition and the substrate type. Both STO and SrFeO₃ structures can accommodate significant amounts of oxygen deficiency, maintaining a perovskite-like structure with oxygen vacancies that may be disordered or that can be arranged in layers, e.g. brownmillerite SrFeO_{2.5} when δ =0.5. The unit cell volume increases with δ and there are dramatic changes in conductivity and other properties with δ [11,15,20,22-24,72]. Vacancies can even be introduced into an STO substrate at low pressure and high temperature [73], which affects the surface of the substrate prior to film growth.

The oxygen content of the films and therefore δ is difficult to quantify because the sample volume is too small for thermogravimetric analysis, although other methods such as photoluminescence, direct imaging, or measurements of cation valence may be applicable [74,75]. However, Rothschild *et al.* [14] showed that Fe³⁺ was the dominant valence state in STF over a range of pressures and temperatures including those corresponding to the growth conditions, i.e. the composition can be written $SrTi_{1-x}Fe_xO_{3-(x/2)-\gamma}$ with γ representing the deviation from a state containing all Fe³⁺. The presence of Fe²⁺ became significant at extremely low oxygen pressures, e.g., below 10⁻⁹ Torr at 600°C, whereas Fe⁴⁺ was present at higher pressures, e.g., above 10 mTorr. Rodriguez *et al.* showed that a mixture of Fe³⁺ and Fe⁴⁺ was present in STF after synthesis at 1300°C in oxygen [76], and Ferreira suggested that some Fe⁴⁺ was present at room temperature [20]. It is not clear how well the PLD growth conditions correspond to these equilibrium conditions, but we expect that the oxygen content will be kinetically trapped after cooling the samples from the growth temperature, implying that the oxygen content and the average Fe valence state in the films will be lower for films grown at a lower base pressure.

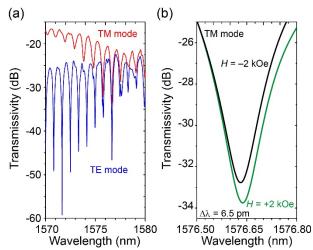
The formation of sparsely distributed metallic Fe nanorods was inferred in the $P = 2 \mu$ Torr and 4μ Torr STF/STO samples. This was not seen in our previous work on STF [16], but there have been several other observations of metallic transition metal nanorods forming in an oxide matrix, notably Fe in a (non-perovskite) LaSrFeO₄ matrix formed from the deposition of LaSr_{0.5}Fe_{0.5}O₃, and Co, Ni or Fe in ceria or STO by codeposition of the matrix phase with a transition metal oxide [77-79]. In the $P = 4 \mu$ Torr STF/STO, it is possible that the low base pressure combined with the large tetragonal distortion due to epitaxy promotes the precipitation of a fraction of the Fe as a strain relief mechanism that was not seen in the non-epitaxial films on Si. The precipitation of the small volume fraction of metallic Fe would lead to an excess of Sr in the perovskite matrix that could be accommodated through additional SrO layers as in Ruddlesden-Popper phases without disrupting the overall perovskite structure. In contrast, in the films on Si it appears that all the Fe was present within the perovskite lattice. The Fe distribution did, however, become less homogeneous at low base pressure. It is not clear why the STF unit cell volumes decreased for $P < 4 \mu$ Torr. This may indicate a different strain relaxation mechanism becoming important, or a structural change due to the ordering of vacancies or the formation of Ti³⁺.

257 In previous work, we have suggested that oxygen-deficient STF exhibits mixed valence Fe ions, leading to both ferromagnetic and 258 259 antiferromagnetic interactions between nearest-neighbors. A range of other Fe-based perovskites including Sr₂FeMoO₆ and Ba_2FeReO_6 also display room temperature magnetism that originates on the Fe spins [80-82]. Moreover, the magnetic moment of 260 Fe ions is dramatically affected by adjacent oxygen vacancies that break the symmetry of the oxygen coordination octahedron, as 261 predicted theoretically in STCo [83]. Film strain leads to magnetoelastic anisotropy, which can orient the overall direction of mag-262 netization in the film, accounting for the out-of-plane easy axis for STF/STO [16]. The resulting net magnetic moment of the STF 263 and the hysteresis behavior is therefore highly dependent on the concentration and arrangement of the oxygen and Fe and the strain 264 state, as well as the presence of Fe nanorods, and is expected to vary with all the film deposition conditions including P. In STCo, 265 changes in *P* were also found to affect the magnetic moment [22].

The most striking finding in the present work is the higher magnetization and FR in films made on Si compared to those on STO. The films differ in their microstructure and strain state, but may also have subtle differences in oxygen content, vacancy clustering, or Fe distribution despite being deposited simultaneously; this could result from different surface kinetics or surface chemistry during film nucleation, or differing substrate temperatures due to the differences in emissivity and thermal conductivity between the Si and STO substrates. The magnetic effects in STF/Si appear not to originate from metallic Fe, and are therefore intrinsic to the substituted perovskite. Metallic behavior has been calculated for STO surfaces [84], suggesting that the grain boundaries in the STF may play a role in its electronic properties. Unlike the STF/STO, which has a significant tetragonal distortion (c/a > 1) leading to magnetoelastic anisotropy [27,85], the polycrystalline STF/Si has little strain, suggesting that the origin of its magnetism is microstructural (influenced by grain boundaries, oxygen content, vacancy clustering, or the inhomogeneous Fe distribution observed in Figs. 2(a-c) and 3(b)) rather than magnetoelastic.

III. INTEGRATED MAGNETOOPTICAL ISOLATOR

278 The polycrystalline STF/Si is suitable for magnetooptical devices because it is expected to grow on a variety of substrates and it has 279 a large remanent Faraday rotation angle. As an illustration of the integration of polycrystalline STF into a magneto-optical device, a 280 ring resonator was prepared from a silicon-on-insulator wafer consisting of a racetrack placed adjacent to a straight waveguide, 281 282 283 similar to the devices shown in Refs. [4,5] When light passing through the waveguide is resonant with the racetrack (i.e., when the racetrack circumference is an integral number of wavelengths), the light couples into the racetrack, and there is a drop in transmittance through the waveguide, as shown in Fig. 8. If the racetrack is clad on top with a magneto-optical material magnetized perpen-284 dicular to the waveguide, the transverse magnetic (TM) mode, but not the transverse electric (TE) mode, experiences a nonrecipro-285 cal phase shift (NRPS). This leads to a different resonant wavelength for clockwise and counterclockwise propagating TM light, 286 which forms the basis of an isolator.



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FIG. 8. Integration of STF on a waveguide device. (a) Resonant peaks for TE and TM modes at wavelengths between 1570 nm and 1580 nm. (b) Example of a peak shift for the TM mode that showed a NRPS of 6.5 nm when the field *H* direction was reversed.

The waveguides had a thickness of 250 nm and a width of 450 nm. The racetrack was 290 μ m long and 90 μ m wide. A 200- μ mlong, 20- μ m-wide window was opened in the silica cladding of one of the sides of the racetrack and STF (at *P* = 4 μ Torr) was deposited over the sample, so that the STF contacted the waveguide directly within the window. The transmittance of the waveguide/resonator was measured at a wavelength of around 1550 nm for a magnetic field of ±2 kOe applied in-plane, perpendicular to the side of the racetrack. The resonant wavelength of both the TM and TE modes varied between successive measurements by up to 10 pm because of thermal drift or vibration, but the TM mode showed an NRPS of ~4 pm compared to the TE mode when the sign of the applied field was changed.

299 Similar devices made with CeYIG cladding [5] (FR \sim -1300 °cm⁻¹) with in-plane magnetization exhibited a NRPS of \sim 20 pm. The 300 lower NRPS in the STF devices is a result of incomplete in-plane saturation, which requires a field > 5 kOe. Based on the hysteresis 301 loops, a 2 kOe in-plane field would lead to an in-plane magnetization of approximately 1/3 of the saturation magnetization. The FR 302 scales with the magnetization, so we expect the STF to produce an NRPS of 1/3 of what would be produced at in-plane saturation. 303 The NRPS of the STF devices is in reasonable agreement with the expected NRPS, considering the geometry of the resonator and 304 the magnitude of the FR. The absorption of the STF at a wavelength of $\lambda = 1550$ nm was 31 ± 2 dB cm⁻¹, derived as shown in [69]. 305 Hence, the figure of merit defined as the ratio of saturation FR/absorption was 58 °dB^{-1} , which is good compared with other thin 306 film magneto-optical materials [17,33].

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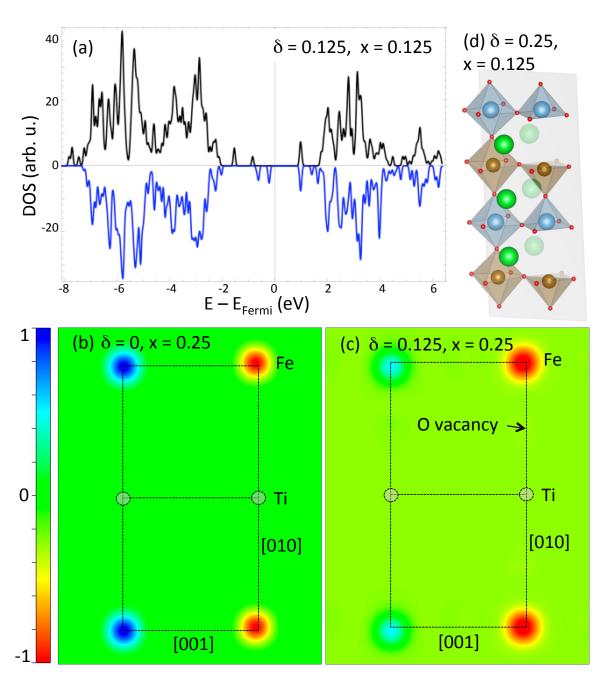
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IV. ELECTRONIC STRUCTURE CALCULATIONS

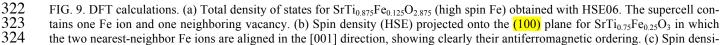
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We now present a brief discussion of the theoretical predictions for the electronic structure and the change of the lattice parameters as a function of the chemical pressure created by oxygen vacancies of STF based on hybrid density functional theory (DFT) [83], as described in the methods section. All possible vacancy and Fe configurations for STF with x = 0.125 or 0.25 were evaluated.

From Fig. 9(a), we find that STF with $\delta = 0.125$ and x = 0.125 (one Fe atom and one vacancy in the 2×2×2 perovskite supercell) is a semiconductor with a band gap of ~1.1 eV. The eg orbitals are partially occupied, which gives rise to the ferrimagnetism. From molecular orbital theory, we find that the high-low spin-gap energy is ~70 meV, which is actually larger than the ~40 meV required to take a spin to the eg from the t_{2g} level [30]. A detailed analysis of the spin states in this system will be discussed in a subsequent publication. We confirmed the oxidation states were Fe²⁺ and Ti⁴⁺ rather than both ions being 3+, and the Ti⁴⁺ magnetic moment is negligible.







 $\begin{array}{l} 325\\ 326\\ 326\\ 326\\ 327\\ 328\\ 328\\ 328\\ 329\\ 330 \end{array}$ ty (HSE) projected onto the (100) plane for SrTi_{0.75}Fe_{0.25}O_{2.875} with an oxygen vacancy adjacent to one of the two Fe ions per unit cell. In both cases (b) and (c), the second-nearest Fe neighbors are also depicted and the color scale indicates the spin density normalized to the largest magnetic moment. The ferrimagnetic order in the O-deficient system is clear in (c). (d) Image showing the model in (c) in which the atomic positions are visible. The plane shown in (c) bisects the model passing through the Fe and Ti sites. Blue: Fe; brown: Ti; green: Sr; red: O. 330

When x = 0.25, the two Fe ions can be arranged along [001], [110], or [111]. In each case, the Fe-Fe distances are ~0.9x the Ti-Ti distance in STO, and the final unit cell is distorted from cubic symmetry. In the $\delta = 0$ case the Fe ions have an average moment of $3.7 \mu_B$, with an antiferromagnetic (AF) ground state in the Fe-[001] arrangement. For the [110] and [111] arrangements where the Fe ions are further separated, the lowest energy ferromagnetic (FM) and antiferromagnetic (AF) states are almost degenerate within a small energy range < 40 meV. The preponderance of [110] and [111] configurations in a crystal with random Fe site distribution suggests an overall low moment due to dominant AF interactions.

An oxygen vacancy ($\delta = 0.125$) leads to a significant increase in the volume of the unit cell compared to $\delta = 0$, in agreement with experiment. The volume increase is not due to the Fe-Fe interaction but to the chemical pressure created by the vacancy. When calculating the ground state for $\delta = 0.125$, the spin and valence of the two Fe ions were initialized in all possible combinations (i.e. both Fe³⁺, one Fe⁴⁺ and one Fe²⁺, high and low spin, and FM and AF) and the lowest energy state after relaxation was defined as the ground state. The low spin states tended to shift to high spin with ~ 4.1–4.2 μ_B , with the two Fe spins coupled AF in most but not all cases. However, even the AF-initialized cells had a net moment, i.e., they were ferrimagnetic, because the spin at each Fe location was not identical owing to their different locations with respect to the oxygen vacancy.

In the Fe-[001] case, the lowest energy configuration is one where the vacancy is between the two Fe ions, which eliminates the superexchange between the Fe ions that coordinate the vacancy. The magnetic moment was ~0.2 μ_B /Fe. In the Fe-[110] case, the two possible vacancy positions lead to similar energies, and the magnetization in the ground state is 0.3–0.4 μ_B /Fe. In the more symmetric case of Fe-[111], the FM and AF configurations are almost degenerate (i.e., < 0.1 meV) because of the large separation between the Fe atoms. The final magnetic moment of the AF-initialized arrangement is once again 0.1–0.2 μ_B /Fe.

352 To better understand the effects of the vacancies on the magnetic behavior, we calculate the HSE spin density for a system with x =353 0.25 and $\delta = 0$ or $\delta = 0.125$. Figure 9(b) shows the spin density projected onto the (100) plane for SrTi_{0.75}Fe_{0.25}O₃ in which the two 354 Fe ions are aligned in the [001] direction. The AF ordering of the two Fe ions is seen from the opposite sign but equal value of the 355 356 spin density. In Fig. 9(c), a vacancy is introduced, i.e. the composition is SrTi_{0.75}Fe_{0.25}O_{2.875} with the vacancy placed adjacent to one Fe ion but not between the two Fe ions. The magnetic moment of the Fe next to the vacancy is enhanced, as evident by the slightly 357 larger radius of the red spin-density projections, while the other Fe which is completely O-coordinated has a smaller magnetic mo-358 ment (blue) coupled antiferrimagnetically to the larger moment of the first Fe. While the two Fe magnetic moments per unit cell in 359 the first case are equal in magnitude, in the O-coordinated Fe case there is a difference of 0.1-0.3 $\mu_{\rm B}$ /Fe between the moments 360 (i.e. the system is ferrimagnetic), in accordance with our previous discussion. These results have been qualitatively con-361 firmed using LMTO (linear muffin-tin orbital). In the case of Fig. 9(c) the magnetic moment of the incompletely coordinated 362 Fe was higher, and this is the case for most of the possible combinations of Fe valence and spin state. However, a few con-363 figurations with a slightly higher energy showed an increase of the moment of both Fe ions or even an increase just of the 364 magnetic moment of the completely coordinated Fe ion.

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The STF film is expected to comprise randomly distributed Fe ions and vacancies, and we would therefore expect a variety of nearest neighbor configurations and resulting interactions to be present, including those addressed in the modeling. The crystal would then exhibit a relatively low net magnetic moment, dominated by AF coupling, analogous to a ferrimagnet but without long range order. The presence of vacancies is predicted to increase the unit cell volume, raise the magnetic moment at the Fe sites and to lead to a net moment even when AF interactions are dominant. These predictions are broadly in line with the experimental results.

V. CONCLUSION

374 In conclusion, STF (SrTi_{0.65}Fe_{0.35}O_{3- δ}) films grown on Si or STO substrates by PLD showed a strong dependence of microstructure 375 and magnetic properties on the base pressure during deposition. At an optimum base pressure for deposition ($P = 4 \mu Torr$) the films 376 showed room temperature magnetism of up to 0.53 μ_B /Fe with out-of-plane magnetic anisotropy, and a coercivity of 3 kOe as well 377 as a FR of -1900 ° cm⁻¹ at a wavelength of 1550 nm. STF grown on STO at the optimum base pressure of $P = 4 \mu$ Torr showed a 378 low (< 0.5%) volume fraction of metallic Fe nanorods not present at other pressures that contribute to the magnetic signal, but the 379 polycrystalline STF films grown on Si showed 65% greater magnetization and FR compared to the epitaxial films on STO. There 380 was no evidence of metallic Fe in the STF/Si, indicating that the magnetic and magneto-optical properties are intrinsic to the substi-381 tuted perovskite. Polycrystalline STF was grown onto a silicon ring resonator, and its FR produced a non-reciprocal phase shift.

382 First-principles calculations showed that oxygen vacancies increase both the unit cell volume and the magnetic moment of the Fe 383 ions. The dominant interactions are antiferromagnetic, leading to a material with a net magnetization below about 0.5 $\mu_{\rm B}/{\rm Fe}$, which 384 is reasonable compared with the experimental data. These results illustrate the important role of oxygen vacancies in determining 385 the magnetic properties and structure of a Fe-substituted perovskite, and the integration of the perovskite in magneto-optical devic-386 es 387 388 APPENDIX A: TARGET PREPARATION 389 SrCO₃ (99.99%), TiO₂ (99.99%) and Fe₂O₃ (hematite, 99.945%) powders were stoichiometrically mixed, ball milled for 24 hours, 390 and calcined at 1200°C for 5 hours, showing a single perovskite phase. The powder was pressed into a 1-inch-diameter pellet and 391 sintered at 1300°C for 5 hours. 392 393 APPENDIX B: SAMPLE PREPARATION FOR TEM 394 The samples were prepared using a focused ion beam (FEI-600), and then imaged on a JEOL 2010F field emission TEM at 200 kV 395 acceleration voltage. The element distribution in the sample was mapped by STEM EDX. 396 397 APPENDIX C: X-RAY DIFFRACTION 398 For STF/STO, the PANAlytical X'pert was used. The wavelength was Cu-K α ($\lambda = 1.540598$ Å), the source current was 40 mA, the 399 voltage was 45 kV, and the scan step in 2θ was 0.017°. There was no offset of ω . For STF/STO reciprocal space maps, the Bruker-400 Axs D8 DISCOVER was used with Cu-K α radiation. The x-ray beam was monochromated with the Ge(220)×4 and the Gobel mir-401 ror slit was 0.8 mm. The first and second monochromator slits were 0.2 mm and 1 mm. For STF/Si, the Rigaku Smartlab was used 402 because of its capability for in-plane GID. The wavelength was Cu-Ka with a tube current of 200 mA and a voltage of 45 kV. The 403 step of the 2θ goniometer was 0.024°, and the incident angle of the x-rays ω was fixed at 0.4°. 404 405 APPENDIX D: MÖSSBAUER SPECTROSCOPY 406 ⁵⁷Fe Mössbauer spectroscopy with a ⁵⁷Co/Rh source (of approximately 20 mCi of activity) was performed at 300 K in order to char-407 acterize the iron in the samples using the conversion electron mode. The films were first metallized with a thin layer of aluminum 408 (~30-40 nm) to provide a good Ohmic contact and subsequently inserted into the cathode structure of a He-CH₄ filled (5% at 1.7 409 bar) gas-flow proportional counter, for measurements lasting approximately one week in each case. Energy filtering enabled Möss-410 bauer spectra were acquired for the film (deep electrons and x-rays) and primarily from the surface ($\sim 20-30$ nm penetration depth, 411 due to Auger electrons) for the sample on STO. No Mössbauer signal could be resolved for the lowest escape energies. The sample 412 on Si had a weaker signal and no energy filtering was performed so the data corresponded to the full thickness. The spectra were 413 fitted using the standard least-squares method, using a pure Fe metal foil for velocity calibration. 414 415 APPENDIX E: THEORETICAL METHOD 416 We performed spin-polarized DFT calculations with a Heyd-Scuseria-Ernzerhof (HSE) functional [86,87] as implemented in the 417 Vienna Ab initio Simulation Package (VASP) and within a PAW-PBE framework [88]. Compared to semi-local DFT methods, 418 HSE improves the accuracy of standard local magnetic moments as well as features of interest such as valence spin states, band-419 gaps, and lattice parameters, as shown previously for SrCoO₃ and SrTiCoO₃ [83]. A 2×2×2 perovskite supercell with 39 or 40 at-420 oms for $\delta = 0$ or 0.125, respectively, was used. Monkhorst k-point meshes of 2×2×2 and 4×4×4 were used for relaxation and static 421 calculations, respectively. The energy cutoff used was 500 eV, and forces were converged to within 10^{-5} eV Å⁻¹. 422 423 We explored all possible FM and AFM initializations in both high and low spin configurations for different combinations of valence states for Fe, i.e., 4 and $0\mu_{\rm B}$ (high and low spin Fe²⁺), 5 and $1\mu_{\rm B}$ (high and low spin Fe³⁺), and 4 and $2\mu_{\rm B}$ (high and low spin 424 Fe⁴⁺). We also explored perturbations of $\pm 20\% \mu_{\rm B}$ for every initialization. In total, a few hundred calculations were performed, and 425 in all cases, the magnetic moments were allowed to fully relax to their final value. Some cases with 4 x 4 x 4 supercells were also 426 examined with x = 0.25 and δ = 0.125, of which a few reached convergence within a reasonable time. These configurations showed 427 antiferromagnetic coupling between the Fe moments and agreed with the 2x2x2 supercell results. 428

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